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1 **TITLE**

2 **Assessing the physiological responses of the gastropod *Crepidula fornicata***
3 **to predicted ocean acidification and warming**

4

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19 **Running title:** Responses of *C. fornicata* to OA and warming

20

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22

23

24 **Abstract**

25 Organisms inhabiting coastal waters naturally experience diel and seasonal physico-
26 chemical variations. According to various assumptions, coastal species are either considered
27 to be highly tolerant to environmental changes or, conversely, living at the thresholds of their
28 physiological performance. Therefore, these species are either more resistant or more
29 sensitive, respectively, to ocean acidification and warming. Here, we focused on *Crepidula*
30 *fornicata*, an invasive gastropod that colonized bays and estuaries on northwestern European
31 coasts during the 20th century. Small (< 3 cm in length) and large (> 4.5 cm in length),
32 sexually mature individuals of *C. fornicata* were raised for 6 months in three different $p\text{CO}_2$
33 conditions (390, 750 and 1400 μatm) at four successive temperature levels (10, 13, 16 and
34 19°C). At each temperature level and in each $p\text{CO}_2$ condition, we assessed the physiological
35 rates of respiration, ammonia excretion, filtration and calcification on small and large
36 individuals. Results show that, in general, temperature positively influenced respiration,
37 excretion and filtration rates in both small and large individuals. Conversely, increasing $p\text{CO}_2$
38 negatively affected calcification rates, leading to net dissolution in the most drastic $p\text{CO}_2$
39 condition (1400 μatm) but did not affect the other physiological rates. Overall, our results
40 indicate that *C. fornicata* can tolerate ocean acidification, particularly in the intermediate
41 $p\text{CO}_2$ scenario. Moreover, in this eurythermal species, moderate warming may play a
42 buffering role in the future responses of organisms to ocean acidification.

43

44 **Keywords:** calcification, coastal system, invasive species, metabolism, mollusk, $p\text{CO}_2$,
45 temperature

46

47

48

49 **Introduction**

50 Predictions indicate that coastal ecosystems will be strongly affected by ocean
51 acidification and warming, currently two of the most prominent anthropogenic processes
52 influencing marine life (Harley et al. 2006). Due to the increase in atmospheric CO₂ partial
53 pressure ($p\text{CO}_2$), pH in surface waters is predicted to decline by 0.06 to 0.32 units and sea
54 surface temperatures to increase by 1.0 to 3.7°C by the end of the century, depending on the
55 Intergovernmental Panel on Climate Change (IPCC) representative concentration pathway
56 considered (Stocker et al. 2013). Modifications in seawater carbonate chemistry due to ocean
57 acidification lead to a decrease in carbonate ion concentrations (CO_3^{2-}) (Orr et al. 2005) and a
58 reduction in the calcium carbonate saturation state (Ω), which regulates the thermodynamics
59 of calcium carbonate (CaCO_3) precipitation (Feely et al. 2009). In estuarine and coastal
60 waters, pH is more variable than in the open ocean due to intense biological and
61 biogeochemical processes (Andersson and Mackenzie 2011). In these habitats, ocean
62 acidification and warming will shift the baselines, exacerbate natural variations in pH and
63 temperature, and probably threaten the communities living there (Waldbusser and Salisbury
64 2013).

65 Mollusks constitute a major taxonomic group in estuarine and coastal waters in terms
66 of community structure and ecosystem functioning (Gutiérrez et al. 2003). Because most
67 marine mollusk taxa accumulate significant amounts of CaCO_3 to form protective external
68 shells, they may be sensitive to the changes in pH and carbonate chemistry induced by ocean
69 acidification (for review, see Gazeau et al. 2013), although recent studies have shown that
70 some species could be resistant to elevated $p\text{CO}_2$ (Range et al. 2011; Ries et al. 2009). Along
71 with direct impacts on calcification, high CO₂ concentrations may also have indirect effects
72 on metabolism by disturbing the extracellular acid-base equilibrium, leading to general

73 internal acidosis (Melzner et al. 2009). These potential shifts in acid-base homeostasis have
74 the potential to change organisms' energy balance (Pörtner et al. 2005).

75 In mollusks, the effects of elevated $p\text{CO}_2$ and/or decreased pH alone are highly
76 species-specific (see review in Gazeau et al. 2013), and depend on species sensitivity and any
77 existing compensation mechanisms (Michaelidis et al. 2005). To better estimate future ocean
78 acidification effects on mollusk species, various physiological processes have been studied in
79 bivalves and gastropods such as respiration (Beniash et al. 2010; Bibby et al. 2007), excretion
80 (Fernandez-Reiriz et al. 2011; Liu and He 2012), feeding (Fernandez-Reiriz et al. 2012;
81 Marchant et al. 2010), immune response (Bibby et al. 2008; Matozzo et al. 2012) and protein
82 or enzyme production (Matozzo et al. 2013; Tomanek et al. 2011). However, few studies have
83 simultaneously assessed the responses of more than three physiological processes to ocean
84 acidification and warming. The concomitant increase in seawater temperature and $p\text{CO}_2$ are
85 likely to affect mollusk metabolism because, in addition to changes in gas solubility and the
86 proportion of carbon species (Zeebe 2011), temperature also strongly affects physiological
87 and biochemical reactions (Cossins and Bowler 1987). Because warming can modulate the
88 metabolism responses to ocean acidification (Ivanina et al. 2013; Melatunan et al. 2013),
89 investigations of both pH and temperature effects are valuable for understanding the
90 responses of mollusks in the future ocean.

91 One of the most abundant and widespread shelled mollusks on the French
92 northwestern Atlantic and Channel coasts is the slipper limpet *Crepidula fornicata*, Linnaeus
93 1758 (Blanchard 1997). This gastropod native to the northeastern American coast was
94 introduced in Europe at the end of the 19th century, mainly via oysters imported for farming
95 (Blanchard, 1995). It then colonized European coasts from southern Sweden to southern
96 France, becoming invasive in some places (Blanchard 1997). *C. fornicata* lives in shallow
97 sites, especially in bays and estuaries where it can reach very high densities of more than 1000

98 individuals per m² (Blanchard 1995). This species is known to be highly robust to
99 environmental stress, in particular temperature and salinity (Diederich and Pechenik 2013;
100 Noisette et al. 2015), parameters that have diel and seasonal variations in these coastal
101 habitats. Established *C. fornicata* populations have largely affected biodiversity and
102 ecosystem functioning in terms of sediment modifications (Ehrhold et al. 1998), changes in
103 faunal assemblages (De Montaudouin et al. 1999) and trophic structure (Chauvaud et al.
104 2000). This species also affects benthic biogeochemical cycles by enhancing filtration,
105 metabolic activities, CaCO₃ production, and the recycling of nutrients and dissolved carbon
106 back into the pelagic ecosystem (Martin et al. 2006; Martin et al. 2007; Ragueneau et al.
107 2002)

108 Although *C. fornicata* is likely highly tolerant to environmental fluctuations, the
109 combined effects of decreased pH and increased temperature may push this species away
110 from its physiological optimum. Thus the objective of this work was to quantify the
111 respiration, ammonia excretion, filtration and calcification responses of small and large
112 specimens of *C. fornicata* in different temperature and *p*CO₂ conditions. Investigating the
113 physiology of this key engineer in some coastal ecosystems in a context of climate change is
114 one way to better understand the sensitivity of this species and its potential future ecological
115 impact.

116

117 **Methods**

118

119 *Sampling site and in situ conditions*

120 *C. fornicata* stacks were collected by SCUBA divers on 30 November 2011, in
121 Morlaix Bay (northwestern Brittany, France), at the “Barre des Flots” site (3°53.015'W;
122 48°40.015'N) at approximately 11 m depth. No temporal series of abiotic parameters were

123 available for this exact location. However, variations in the physico-chemical parameters
124 (surface measurements) at a station (called Estacade), located approximately 10 km from the
125 Barre des Flots site, were obtained from the *Service d'Observation des Milieux Littoraux*
126 (SOMLIT) between 2010 and 2013, with a sampling step of 15 days. Between October 2010
127 and March 2013, temperature varied between 8.1°C (January 2011) and 16.5°C (August
128 2011) with mean values (\pm SE) of $10.1 \pm 0.2^\circ\text{C}$ in winter, $12.7 \pm 0.4^\circ\text{C}$ in spring and $15.8 \pm$
129 0.02°C in summer.

130 In Morlaix Bay (2009 to 2011), phytoplankton groups ($> 5\mu\text{m}$), the most important
131 food resource of *C. fornicata* (Decottignies et al. 2007), were mainly dominated by planktonic
132 diatoms in concentrations varying between 10 to 300 cells mL^{-1} (depending on the season)
133 and dinoflagellate species that were found at lower abundances (ca. 25 cells mL^{-1} ; Leroy
134 2011).

135

136 *Biological material*

137 *C. fornicata* forms stacks of several individuals in which each individual adheres to the
138 dorsal surface of the shell of the subjacent partner in the stack. It is a protandrous
139 hermaphrodite, meaning that the small individuals at the top of the stacks are generally males
140 and the large ones at the bottom, females (Coe 1936). After sampling, stacks were brought
141 directly to the *Station Biologique de Roscoff* where they were kept in natural, unfiltered
142 seawater for 6 weeks at a temperature gradually lowered to 10°C , reflecting the seasonal drop
143 in temperature between autumn and winter. Sexually mature individuals (more than 1 cm in
144 length) were selected and separated into two class sizes: small individuals (29.5 ± 0.9 mm
145 length) from the top of the stack and larger ones (45.4 ± 0.6 mm length) from the bottom.
146 They were separated from the stack and individually labeled with tags glued on their shell.
147 Empty subjacent shells, whose soft tissue was removed, served as substratum for the sampled

148 live individuals. Other empty shells whose size was similar to that of the substratum shell of
149 live individuals were also selected for flux corrections (see part “Metabolic rates and O:N
150 ratios” below). All the shells were gently brushed to remove epibionts without altering
151 periostracum layer.

152 Length (in mm), volume (in mL) and tissue dry weight (DW in g) of the live
153 individuals were determined for each incubated specimen at the end of the whole experiment.
154 Length was measured with calipers, volume was estimated as the volume of seawater moved
155 when individual was immersed and DW was determined after drying fresh samples at 60°C
156 for 48 h.

157

158 *Experimental conditions*

159 Single small and large individuals, along with their substratum shell, were randomly
160 distributed into nine 10 L aquaria with 10 individuals of each class size per each aquarium.
161 Empty shells were also distributed into nine other 10 L aquaria (4 shells per aquarium). At the
162 beginning of the experiment, pH was gradually decreased over 2 weeks by 0.02 pH unit per
163 day from 8.1 until the different pH treatments were reached. *C. fornicata* individuals and
164 empty shells were then subsequently held for 24 weeks (12 January to 27 June 2012) in three
165 $p\text{CO}_2$ treatments selected according to the recommendations in Barry et al. (2010): (1) 390
166 μatm ($\text{pH}_T = 8.07$) represented current $p\text{CO}_2$, (2) 750 μatm ($\text{pH}_T = 7.82$) corresponded to the
167 elevated $p\text{CO}_2$ level predicted by the IPCC for the end of the century (Solomon et al. 2007)
168 and (3) 1400 μatm ($\text{pH}_T = 7.56$) represented a $p\text{CO}_2$ five-fold higher than preindustrial $p\text{CO}_2$
169 (280 μatm) also predicted for 2100 (Stocker et al. 2013). $p\text{CO}_2$ was adjusted by bubbling
170 CO_2 -free air (current $p\text{CO}_2$) or pure CO_2 (elevated $p\text{CO}_2$) in three 100 L header tanks supplied
171 with unfiltered seawater pumped directly from the foot of the *Station Biologique de Roscoff*.
172 Each of the three $p\text{CO}_2$ treatments had six replicate 10 L aquaria, three for live organisms and

173 three for empty shells. They continuously received CO₂-treated seawater at a rate of 9 L h⁻¹
174 (i.e. a renewal rate of 90% h⁻¹) from the header tanks. pCO₂ was monitored and controlled by
175 an offline feedback system (IKS Aquastar, Karlsbad, Germany) that regulated the addition of
176 gas in the header tanks. The pH values of the IKS system were adjusted from daily
177 measurements of pH_T in the 18 aquaria using a pH meter (826 pH mobile, Metrohm AG,
178 Herisau, Switzerland) calibrated with Tris HCl and 2-aminopyridine HCl buffers (Dickson et
179 al. 2007).

180 In each pCO₂ treatment, temperature was raised from 10 to 19°C with an incremental
181 step of 3°C. The first three temperature levels (10 to 16°C) simulated the natural change in
182 temperature from winter to summer in Morlaix Bay whereas the last level (19°C)
183 corresponded to a temperature increase of 3°C predicted for the end of the century (Solomon
184 et al. 2007). *C. fornicata* individuals were held for three weeks at each temperature before
185 carrying out the metabolic measurements (see below). This acclimation time was long enough
186 to overcome the immediate stress response (Meistertzheim et al. 2007). Temperature was
187 maintained at (1) 10°C (1st trial period) from 16 January to 12 February 2012; (2) 13°C (2nd
188 trial period) from 27 February to 25 March 2012; (3) 16°C (3rd trial period) from 9 April to 6
189 May 2012, and (4) 19°C (4th trial period) from 21 May to 27 June 2012. Between two
190 temperature levels, temperature was gradually increased by 0.2°C day⁻¹ over two weeks. The
191 18 aquaria were placed in thermostatic baths in which temperature was regulated to within ±
192 0.2°C using submersible 150 to 250 W heaters controlled by the IKS system.

193 Three independent 10 L aquaria named “control” were maintained at 10°C under
194 ambient pH (with no pCO₂ control) until the end of the experiment in order to estimate a
195 potential bias on metabolism induced by the mesocosm experiment over time. Each aquarium
196 contained 10 small and 10 large slipper limpets on their substratum shell and was supplied

197 with the same seawater sourced from the header tanks. They were kept in a thermostatic bath
198 regulated at 10°C by an aquarium chiller (TC5, TECO®, Ravenna, Italy).

199 In addition to the natural phytoplankton found in the unfiltered seawater, all slipper
200 limpets were fed twice a week with a stock solution composed of the diatom *Chaetoceros*
201 *gracilis* ($\sim 15 \times 10^6$ cells mL⁻¹) and the dinoflagellate *Isochrysis affinis galbana* ($\sim 26 \times 10^6$
202 cells mL⁻¹); 400 mL of this microalgal mix was added to each aquarium at each feeding.
203 Seawater flow was stopped for 2 h when organisms were fed and filtering actively. During
204 this feeding time, pH variation did not exceed 0.05 units.

205 Individuals that did not adhere to their substratum shell and that showed no reaction
206 when their foot was stimulated were counted as dead and removed from the tanks. Mortality
207 reached only 8% at the end of the experiment among all $p\text{CO}_2$ conditions.

208

209 *Seawater parameter monitoring*

210 Seawater parameters were monitored throughout the experiment. pH_T and temperature
211 were recorded daily in each of the 21 aquaria (18 + 3 controls) using a pH meter (826 pH
212 mobile, Metrohm AG, Herisau, Switzerland) as described above. Total alkalinity (A_T) was
213 measured at each trial period by 0.01 N HCl potentiometric titration on an automatic titrator
214 (Titroline alpha, Schott SI Analytics, Mainz, Germany). Salinity was also measured at each
215 trial period with a conductimeter (LF 330/ SET, WTW, Weilheim, Germany). Seawater
216 carbonate chemistry, i.e. dissolved inorganic carbon (DIC), $p\text{CO}_2$ and the saturation state of
217 aragonite (Ω_{Ar}) were calculated for each $p\text{CO}_2$ level and temperature with CO₂SYS software
218 (Lewis and Wallace 1998) using constants from Mehrbach et al. (1973) refitted by Dickson &
219 Millero (1987).

220

221 *Metabolic rates and O:N ratios*

222 Metabolic rates were assessed at each temperature level after a four-day starvation
223 period and after the shells were gently cleaned to remove biofilm-forming organisms. Two
224 small and two large individuals were selected per aquarium. They were incubated individually
225 in 185 mL (small) and 316 mL (large) acrylic chambers (Engineering & Design Plastics Ltd,
226 Cambridge, UK) filled with seawater from their respective aquaria. They were put on a plastic
227 grid above a stirring bar (speed 100 rpm.), which ensured water homogeneity. Chambers were
228 placed in their original aquaria for incubation to keep the temperature constant. Incubations
229 were carried out in dark for 2 to 10 h, depending on temperature and limpet size, to maintain
230 oxygen saturation above 80% until the end of the incubation. At each temperature period,
231 empty shell incubations were carried out to correct individual rates for fluxes related to the
232 substratum shell. Blank incubations containing only seawater from the aquarium also helped
233 to correct fluxes for any microbiological activity in seawater.

234 Oxygen concentrations were measured at the beginning and the end of the incubation
235 period with a non-invasive fiber-optics system and reactive oxygen spots attached to the inner
236 wall of the chambers (FIBOX 3, PreSens, Regensburg, Germany). Spots were calibrated at the
237 beginning of each trial period with 0% and 100% oxygen buffers. Seawater was sampled for
238 ammonium (NH_4^+) concentration and A_T measurements with 100 mL syringes at the
239 beginning of the incubation, directly in the aquaria just after the chambers were closed, and at
240 the end of the incubation, in the incubation chamber itself. Samples were filtered through 0.7
241 μm Whatman GF/F filters into 100 mL glass bottles and fixed with reagent solutions for
242 ammonium or poisoned with mercuric chloride (0.02% vol/vol; Dickson et al. 2007) for A_T
243 measurements. Vials were stored in the dark pending analysis. NH_4^+ concentrations were then
244 determined using the Solorzano method (Solorzano 1969) based on spectrophotometry at a
245 wavelength of 630 nm (spectrophotometer UV-1201V, Shimadzu Corp, Kyoto, Japan). A_T (in
246 $\mu\text{Eq L}^{-1}$) values were determined by 0.01 N HCl potentiometric titration on an automatic

247 titrator (Titroline alpha, Schott SI Analytics, Mainz, Germany) and by using the Gran method
248 (non-linear least-squares fit) applied to pH values from 3.5 to 3.0 (Dickson et al. 2007).

249 Respiration (in $\mu\text{mol O}_2 \text{ g}^{-1} \text{ DW h}^{-1}$; equation [1]) and excretion (in $\mu\text{mol NH}_4^+ \text{ g}^{-1}$
250 DW h^{-1} ; equation [2]) were directly calculated from oxygen and ammonium concentrations,
251 respectively. Net calcification (in $\mu\text{mol CaCO}_2 \text{ g}^{-1} \text{ DW h}^{-1}$; equation [3]) was estimated using
252 the alkalinity anomaly technique (Smith and Key 1975) based on a decrease in A_T by 2
253 equivalents for each mole of CaCO_3 precipitated (Wolf-Gladrow et al. 2007). As ammonium
254 production increases alkalinity in a mole-per-mole ratio (Wolf-Gladrow et al. 2007), the
255 alkalinity variation was corrected by the ammonium flux to calculate CaCO_3 fluxes.

256 [1] $R = \frac{\Delta\text{O}_2 \times V}{\Delta t \times \text{DW}}$

257 [2] $E = \frac{\Delta\text{NH}_4^+ \times V}{\Delta t \times \text{DW}}$

258 [3] $G_n = -\frac{(\Delta A_T - \Delta\text{NH}_4^+) \times V}{2 \times \Delta t \times \text{DW}}$

259 where ΔO_2 (in $\mu\text{mol O}_2 \text{ L}^{-1}$) is the difference between initial and final O_2 concentrations; Δ
260 NH_4^+ (in $\mu\text{mol NH}_4^+ \text{ L}^{-1}$) is the difference between initial and final NH_4^+ concentrations; ΔA_T
261 is the difference between initial and final total alkalinity ($\mu\text{mol Eq L}^{-1}$); V (in L) is the volume
262 of the chamber minus *C. fornicata* volume; Δt (in h) is the incubation time and DW (in g) is
263 the soft tissue dry weight of incubated *C. fornicata*.

264 In addition, oxygen consumption of the individuals maintained at 10°C during the
265 experiment were assessed on six small and six large individuals at each trial period, following
266 the technique described above. These “controls” tested if mesocosm conditioning induced
267 metabolic stress over time.

268 The O:N ratio, which corresponds to the amount of oxygen consumed for nitrogen
269 excreted, was calculated from respiration and excretion rates except for the experiments run at
270 10°C for which rates were too low to obtain significant data. Generally, the O:N ratio is

271 considered a common indicator of the proportion of the three metabolic substrates
272 (carbohydrates, lipids and proteins) used in energy metabolism (Mayzaud and Conover 1988).
273 The atomic ratio of oxygen uptake and excreted nitrogen was calculated following the
274 equation [4] based on Thomsen & Melzner (2010):

$$275 \quad [4]: O:N = R / E$$

276 where R is the respiration rate used as a proxy of the quantity of oxygen consumed by the
277 individual and E, the excretion rate representing the concentration of nitrogen excreted.

278

279 *Filtration rates*

280 At each trial period, the filtration rate of three small and three large slipper limpets per
281 $p\text{CO}_2$ condition (i.e. 1 individual per size per aquarium) was determined by calculating
282 clearance rates (Coughlan 1969). To do so, 10 and 20 mL of a microalgae mix (*C. gracilis*, *T.*
283 *affinis galbana*, 1:1) were added to the small and large chambers (same as for metabolic
284 measurements), respectively, using a 10 mL syringe equipped with a thin tube. The mean
285 initial concentration of the mix was $1\,200\,000 \pm 310\,000 \text{ cell mL}^{-1}$. In parallel, control
286 incubations containing only microalgae were carried out to check that phytoplankton cells did
287 not multiply significantly during the incubation. Water from the chambers was sampled with
288 the syringe every 15 min until the water became totally clear (around 2 h). Samples were
289 immediately fixed with 25% glutaraldehyde and frozen at -80°C pending analyses (Marie et
290 al. 1999). The number of microalgal cells in each sample was then determined on 200 μL
291 aliquots using flux cytometry (Cell Lab QuantaTM, SC, Beckman Coulter, USA). Filtration
292 rates (F, in $\text{mL SW g}^{-1} \text{DW min}^{-1}$) were calculated following equation [5]:

$$293 \quad [5] \quad F = V \times \frac{\ln[C_i] - \ln[C_f]}{\Delta t \times DW}$$

294 where $[C_i]$ and $[C_f]$ (in cell mL^{-1}) were respectively the initial and final cell concentrations in
295 the chamber water; V (in L) is the volume of the chamber minus individual *C. fornicata*

296 volume; Δt (in h) is the incubation time and DW (in g) is the tissue dry weight of the
297 individual incubated.

298

299 *Statistical analyses*

300 All statistical analyses were performed using the R software, version 2.15.0 (R Core
301 Team 2013). Normality and homoscedasticity were checked using Kolmogorov-Smirnov's
302 test and Levene's test, respectively, before each statistical test. Spatial pseudoreplication
303 effect was first tested by considering "aquarium" as a random factor (p -value < 0.05). Then,
304 statistical analyses were simplified to two-way ANOVAs with repeated measurements on the
305 same individual through the four trial periods (different temperature levels) separately for
306 small and large individuals. These analyses were performed for the four physiological rates
307 (respiration, excretion, calcification and filtration) and the O:N ratio, assuming $p\text{CO}_2$ and
308 temperature as fixed factors. Student-Newman-Keuls (SNK) post hoc tests were applied to
309 identify differences among treatments with a confidence level of 95% when ANOVA showed
310 significant results. In parallel, any changes in the respiration rate of individuals constantly
311 maintained at 10°C through time were assessed using a non-parametric Friedman test for
312 repeated measurements, separately for small and large slipper limpets. All results are given as
313 mean \pm standard error (SE).

314

315 **Results**

316

317 *Seawater parameters*

318 The mean temperature and carbonate chemistry parameters among the $p\text{CO}_2$ and
319 temperature conditions are presented in Table 1. Temperature was stable at each trial period
320 with a variability lower than 0.5°C. The different $p\text{CO}_2$ levels remained close to the selected

321 values of 390, 750 and 1400 μatm except at 19°C where all $p\text{CO}_2$ increased from the baseline
322 (+ 100-200 μatm). A_T ranged from 2365 ± 2 to 2422 ± 2 $\mu\text{Eq kg}^{-1}$. Ω_{Ar} decreased by less than
323 1 only in the 1400 μatm $p\text{CO}_2$ condition. Salinity varied between 34.2 ± 0.1 and 35.1 ± 0.1
324 among the different $p\text{CO}_2$ and temperature levels with no effect of the temperature increase
325 on salinity.

326

327 *Respiration, excretion and O:N ratio*

328 Respiration and excretion rates changed significantly with temperature, but not with
329 $p\text{CO}_2$, in small and large individuals (Figure 1, Table 2). After pooling results for all $p\text{CO}_2$
330 conditions, mean respiration rates in small *C. fornicata* increased from 3.78 $\mu\text{mol O}_2 \text{ g}^{-1} \text{ DW}$
331 h^{-1} at 10°C to 11.76 $\mu\text{mol O}_2 \text{ g}^{-1} \text{ DW h}^{-1}$ at 19°C. In large individuals, the lowest mean
332 respiration rate was recorded at 10°C (4.82 $\mu\text{mol O}_2 \text{ g}^{-1} \text{ DW h}^{-1}$) whereas rates did not differ
333 from 13 to 19°C with a mean value of 11.50 $\mu\text{mol O}_2 \text{ g}^{-1} \text{ DW h}^{-1}$. Oxygen fluxes measured on
334 empty shells represented only 4% of the whole organism fluxes measured and decreased only
335 slightly with temperature.

336 Mean excretion rates calculated among $p\text{CO}_2$ conditions for small *C. fornicata*
337 individuals gradually increased from 0.15 $\mu\text{mol NH}_3 \text{ g}^{-1} \text{ DW h}^{-1}$ at 10°C to 1.47 $\mu\text{mol NH}_3 \text{ g}^{-1}$
338 DW h^{-1} at 19°C. Excretion rates of large individuals showed a parabolic trend with an
339 increase from 10°C (0.16 $\mu\text{mol NH}_3 \text{ g}^{-1} \text{ DW h}^{-1}$) to 16°C (1.34 $\mu\text{mol NH}_3 \text{ g}^{-1} \text{ DW h}^{-1}$)
340 followed by a decrease at 19°C (0.74 $\mu\text{mol NH}_3 \text{ g}^{-1} \text{ DW h}^{-1}$). The ammonium fluxes of empty
341 shells represented less than 1% of the fluxes estimated for whole organisms and were higher
342 at 10°C than at the other temperature levels (rates practically nil).

343 O:N ratios varied greatly, ranging from 2.86 to 31.68 with a mean value of $12.91 \pm$
344 0.56. They varied with $p\text{CO}_2$ or temperature according to size (Table 2, Figure 2). In small *C.*
345 *fornicata* individuals, O:N ratios were the highest at 750 μatm and similar between 380 and

346 1400 μatm . In large individuals, the O:N ratios varied with temperature and were significantly
347 higher at 16°C.

348

349 *Filtration*

350 Temperature significantly affected filtration rates in both small and large individuals
351 (Figure 1, Table 2). In small *C. fornicata*, mean filtration rates among $p\text{CO}_2$ were similar
352 between 10 and 16°C (25.50 mL g^{-1} DW min^{-1}), but increased at 19°C (54.30 mL g^{-1} DW min^{-1}).
353 $p\text{CO}_2$ alone did not affect the filtration rate but the interaction of $p\text{CO}_2$ and temperature was
354 significant (Table 2, p-value < 0.001). At 19°C, filtration rates increased significantly with the
355 increase in $p\text{CO}_2$. In large individuals, mean filtration rates increased gradually from 10°C
356 (5.43 mL g^{-1} DW min^{-1}) to 19°C (25.78 mL g^{-1} DW min^{-1}) without any effect of $p\text{CO}_2$
357 conditions.

358

359 *Calcification*

360 Calcification rates were significantly affected by $p\text{CO}_2$ increase in both small and
361 large individuals but not by temperature (Figure 1, Table 2). Pooling all temperature levels
362 together, mean calcification rates were similar at $p\text{CO}_2$ of 390 μatm (1.88 and 1.63 μmol
363 CaCO_3 g^{-1} DW h^{-1} in small and large individuals, respectively) and 750 μatm (1.02 and 0.60
364 μmol CaCO_3 g^{-1} DW h^{-1} in small and large, respectively), but significantly lower at 1400
365 μatm $p\text{CO}_2$ (-2.53 and -1.77 μmol CaCO_3 g^{-1} DW h^{-1} in small and large individuals,
366 respectively). In the highest $p\text{CO}_2$ condition (1400 μatm), net calcification rates were
367 negative, corresponding to dissolution. Although the interaction between $p\text{CO}_2$ and
368 temperature was not significant for either small or large limpets, $p\text{CO}_2$ response appeared to
369 vary as a function of temperature, particularly at 1400 μatm . In this drastic $p\text{CO}_2$ condition,
370 organisms globally dissolved at 10, 13 and 16°C and calcified (or dissolved less) at 19°C.

371 Calcification rates decreased with the decrease in the mean aragonite saturation state
372 (Ω_{Ar}) which correlated with pCO_2 increase (Figure 3). When Ω_{Ar} decreased below the
373 threshold of 1, calcification rates were always negative reflecting a dissolution process. At the
374 750 and 1400 μatm pCO_2 conditions, Ω_{Ar} was higher at 19°C than at the other temperature
375 levels because the saturation state increases with temperature.

376

377 *Mesocosm controls*

378 In the aquaria maintained at 10°C throughout the entire experiment, temperature was
379 stable over the first weeks of the experiment and slowly increased from 8 April to the end of
380 the experiment until reaching a mean of 12.4°C between 21 April and 15 June because we had
381 technical problems with the chiller (Table 1). Respiration in small individuals showed high
382 variation over time (Figure 4, white bars) but no time effect was detected (Friedman test, $df =$
383 3, $\chi^2 = 6.6$, $p = 0.086$, $n = 6$). Conversely, respiration rates of large individuals increased
384 throughout the experiment (Figure 4, gray bars) with a significant time effect (Friedman test,
385 $df = 3$, $\chi^2 = 9.4$, $p = 0.024$, $n = 6$).

386

387 **Discussion**

388 An increase in temperature affected three of the four physiological processes assessed
389 on small and large *C. fornicata* individuals. In particular, respiration and ammonia excretion
390 rates clearly increased along the tested temperature gradient. In contrast, increases in pCO_2
391 affected only net calcification of the slipper limpets. Interestingly, the coupled effect of
392 temperature and pCO_2 improved the rate of calcification in the most drastic conditions,
393 particularly in small individuals.

394

395 *Temperature effect*

396 The respiration ($0.6 - 34.6 \mu\text{mol O}_2 \text{ g}^{-1} \text{ DW h}^{-1}$) and excretion rates ($-2 - 4.4 \mu\text{mol NH}_3$
397 $\text{ g}^{-1} \text{ DW h}^{-1}$) measured at $390 \mu\text{atm } p\text{CO}_2$ in small and large *C. fornicata* individuals ranged
398 metabolic rates recorded *in situ* in the Bay of Brest in northwestern France (4 to $45 \mu\text{mol O}_2$
399 $\text{ g}^{-1} \text{ DW h}^{-1}$ and 0.5 to $2.3 \mu\text{mol NH}_3 \text{ g}^{-1} \text{ DW h}^{-1}$; Martin et al. 2006). Both rates increased with
400 temperature in small and large individuals regardless of $p\text{CO}_2$. Although respiration rates
401 gradually increased with temperature in small *C. fornicata* individuals, they only increased
402 from 10°C to 13°C , remaining stable at higher temperatures in large *C. fornicata*. This
403 increase is a common response due to the rate-enhancing effects of temperature on
404 physiological and biochemical reactions in ectotherms (Cossins and Bowler 1987). The
405 intensity of respiratory and excretory processes were also dependent of body size. The
406 respiration and excretion rates of small individuals were higher than those of large individuals
407 because the metabolic rate (per unit biomass) decreases with increasing individual size
408 (Parsons et al. 1984; Von Bertalanffy 1951). Small individuals have higher energy
409 consumption because they grow faster than the large individuals (Von Bertalanffy 1964).

410 The filtration rates measured in small and large *C. fornicata* fall into the range of
411 maximum feeding rates calculated by Newell and Kofoed (1977) in *C. fornicata* between 11
412 and 20°C (18 to $41 \text{ mL g}^{-1} \text{ min}^{-1}$; 15°C acclimated individuals). Rates were higher in small
413 than in large individuals because, again, small organisms feed more actively per unit body
414 mass (Sylvester et al. 2005). Filtration rates increased with temperature as previously
415 described in other studies (Newell and Kofoed 1977). In small individuals, rates were constant
416 between 10 and 16°C and increased only at 19°C while they increased regularly with
417 temperature in the large individuals. In Calyptraeidae, small individuals — i.e. males with low
418 mobility — utilize two feeding strategies: grazing with radula and filtration with gills
419 (Navarro and Chaparro 2002). Therefore, small individuals may have supplemented their diet
420 between 10 and 16°C by grazing. For the increased energy requirements at 19°C , small

421 slipper limpets may also increase their filtration rate to meet these supplementary needs. In
422 large sedentary individuals (usually females), filtration is the only feeding mechanism
423 (Navarro and Chaparro 2002) and filtration rate increases with temperature to help cover the
424 higher energy needs.

425 Surprisingly, temperature did not affect calcification rates although an increase was
426 expected in response to the increase in metabolism and energy requirements (Martin et al.
427 2006). Because mollusk shell production is an energetically costly process (Gazeau et al.
428 2013), the absence of any change in calcification rates may be due to food limitation during
429 the experiment, especially at elevated temperatures (16 and 19°C). At these temperatures,
430 providing additional food only twice a week may not have been sufficient to support maximal
431 individual shell growth under pH stressful conditions. If food had been provided more
432 regularly and/or in higher quantities, *C. fornicata* calcification may not be potentially
433 restricted and individuals may have better mitigated the effect of high $p\text{CO}_2$ (Thomsen et al.
434 2014). Future experiments should include measuring integrated shell growth at each
435 temperature level to determine the food effect more completely.

436 Mesocosm experiments cannot perfectly reproduce *in situ* conditions such as natural
437 diet or tidal cycles. This may lead to an increased stress for the organisms grown in these
438 systems (Bibby et al. 2008). The mesocosm effect on organisms was tested through O_2
439 consumption measurements in individuals kept at a constant temperature throughout the
440 experiment (“controls”). The respiration rates did not change over time in small individuals,
441 whereas the respiration in large individuals increased slightly in correlation with a +2°C
442 temperature increase from the beginning to the end of the experiment, because of technical
443 problems with the chiller. Although food may have constituted a bias, particularly in the one-
444 off calcification response to temperature, the absence of strong changes in respiration rates in
445 “controls”, unexceptional metabolic rates ranging from those measured *in situ* and very low

446 mortality during the experiment (only 8%) all suggest the absence of any acute mesocosm
447 effect on the other physiological traits of *C. fornicata*.

448

449 *pCO₂ effect*

450 In contrast to temperature, *pCO₂* did not affect *C. fornicata* respiration or excretion
451 rates regardless of size. Other studies have underlined a lack of any *pCO₂* effect on bivalve
452 and limpet respiration (Dickinson et al. 2012; Fernandez-Reiriz et al. 2012; Marchant et al.
453 2010), although some mollusk species exposed to high *pCO₂* levels have shown metabolic
454 depression (i.e. decrease in oxygen uptake) to compensate — albeit often drastic — *pCO₂*
455 increases (Michaelidis et al. 2005; Navarro et al. 2013). Responses of ammonia excretion to
456 high *pCO₂* in mollusks are also specific: increase in ammonia excretion can occur under
457 elevated *pCO₂* (Fernandez-Reiriz et al. 2011; Langenbuch and Pörtner 2002; Range et al.
458 2014) while some bivalves show opposing trends (Liu and He 2012; Navarro et al. 2013). The
459 increase in ammonia excretion under increased *pCO₂* conditions can be interpreted as an
460 internal pH regulatory mechanism, sometimes based on protein catabolism (Fernandez-Reiriz
461 et al. 2012; Thomsen and Melzner 2010). In our study, neither change in excretion rates nor in
462 O:N ratios calculated were detected between the 390 and 1400 μatm conditions. This
463 similarity indicates that potential intracellular pH regulation of *C. fornicata* was not induced
464 by enhancing protein metabolism (Fernandez-Reiriz et al. 2012). Thus, the potential for
465 metabolic resistance of *C. fornicata* to elevated *pCO₂* is likely due to another effective
466 acidosis-buffering system, such as the increase in internal HCO_3^- concentrations (Gutowska et
467 al. 2010; Michaelidis et al. 2005) or higher H^+ excretion (Melzner et al. 2009; Pörtner et al.
468 2005).

469 Similarly to the respiration and excretion processes, filtration rates did not change as a
470 function of *pCO₂* in either small or large *C. fornicata* in our study. Filtration responses with

471 respect to $p\text{CO}_2$ depend most of the time on the presence of metabolic depression (Fernandez-
472 Reiriz et al. 2011; Liu and He 2012; Navarro et al. 2013). The absence of variation in
473 filtration rates at the different $p\text{CO}_2$ levels indicates that the quantity of food ingested by *C.*
474 *fornicata* did not vary either. Food is known to interact with other stressors, such as $p\text{CO}_2$,
475 and significantly influence metabolic responses (Melzner et al. 2011; Pansch et al. 2014).
476 Quality or quantity changes in the diet can even worsen the condition of invertebrates (Berge
477 et al. 2006; Vargas et al. 2013). Although our microalgal mix did not perfectly match the
478 natural diet of *C. fornicata* (Barillé et al. 2006; Decottignies et al. 2007), the diatoms and
479 dinoflagellate microalgae provided in the experiment correspond to the main taxa present in
480 Morlaix Bay, assuming a nutritional quality close to the natural diet. However, we cannot
481 assure that the quantity of food was not a limiting factor in our experiment. To be sure that
482 microalgae supplied would not represent a bias, the slipper limpets should be fed *ad libitum*
483 which represented a technical issue on a 6 month experiment.

484 In our study, net calcification was similar between 390 and 750 $\mu\text{atm } p\text{CO}_2$ and
485 strongly decreased at 1400 $\mu\text{atm } p\text{CO}_2$ regardless of size, which is a common response in
486 mollusks (Beniash et al. 2010; Melatunan et al. 2013; Range et al. 2011). This pattern
487 contrasts with that reported in Ries et al. (2009), with a parabolic response in *C. fornicata*
488 calcification with the highest rates observed at 600 $\mu\text{atm } p\text{CO}_2$. The stability of calcification
489 rate at 750 $\mu\text{atm } p\text{CO}_2$ (compared to 390 $\mu\text{atm } p\text{CO}_2$) may be due to the biological control of
490 the calcification process and/or the presence of the periostracum, the organic layer covering
491 the crystalline layers of the shell. This organic layer has been predicted to play a great role in
492 maintaining shell integrity of mollusks in elevated $p\text{CO}_2$ conditions (Ries et al. 2009) and to
493 protect them from dissolution in CaCO_3 -undersaturated waters (Huning et al. 2013).
494 Moreover, mollusks may be able to maintain extrapallial fluid in chemical conditions favoring
495 CaCO_3 precipitation at the calcification site, even if external seawater $p\text{CO}_2$ is high

496 (Hiebenthal et al. 2013). Regulation of enzymes involved in the calcification process, such as
497 chitinase (Cummings et al. 2011) or carbonic anhydrase (Ivanina et al. 2013), may also help
498 maintain calcification in high $p\text{CO}_2$ conditions. In our study, at 1400 μatm , calcification rates
499 dropped, perhaps due to physiological changes in the internal acid-base balance affecting shell
500 deposition (Waldbusser et al. 2011) or to an eroded and/or damaged periostracum (pers. obs.).
501 Degradation of this protective layer may lead to higher vulnerability of the shell to external
502 dissolution processes (Range et al. 2012; Ries et al. 2009), which occurs not only in dead
503 shells but also in live animals (Harper 2000). Furthermore, chemical dissolution increased
504 with an increase in $p\text{CO}_2$ and a correlated decrease in Ω_{Ar} ; the combined effect led to a
505 decrease in net calcification rates observed in both small and large *C. fornicata* individuals at
506 high $p\text{CO}_2$ conditions.

507

508 *Combined effects of temperature and $p\text{CO}_2$*

509 In the range of $p\text{CO}_2$ and temperatures tested, the interaction of these two variables
510 had no negative effect on *C. fornicata* respiration and excretion rates. As a eurythermal
511 species even coping with high temperature in some bays during summer (e.g. Bassin
512 d’Arcachon in southwestern France; De Montaudouin et al. 1999), *C. fornicata* can have an
513 optimal temperature of 19°C or higher (Diederich and Pechenik 2013; Noisette et al. 2015).
514 Thus, 19°C may not constitute a real thermal stress and not transgress the metabolic optimal
515 threshold for this species. Increase in temperature is predicted to enhance sensitivity to high
516 $p\text{CO}_2$ levels beyond the optimal temperature of the species and close to its upper limit of
517 thermal tolerance (Pörtner and Farrell 2008). However, at the cold side of a species optimal
518 temperature, warming can increase resilience to ocean acidification (Gianguzza et al. 2014).
519 Therefore, an increase in temperature may actually improve tolerance to $p\text{CO}_2$ increases in *C.*
520 *fornicata*.

521 Calcification rates of both small and large *C. fornicata* showed a positive trend with
522 temperature in the most drastic $p\text{CO}_2$ conditions (1400 μatm). Temperature-mediated
523 increases in metabolism and feeding rates may potentially offset reductions in calcification
524 rates caused by high $p\text{CO}_2$ conditions (Melzner et al. 2011; Thomsen et al. 2014). In addition
525 to this physiological effect, moderate warming can mediate the effects of ocean acidification
526 by the chemical effect on seawater chemistry (Kroeker et al. 2014). Temperature affects CO_2
527 solubility in seawater as well as the equilibrium coefficients governing carbonate chemistry
528 (Millero 2007). As shown in our study, the saturation state of aragonite was greater in warmer
529 water than in colder water for a given $p\text{CO}_2$, thereby enhancing calcification and reducing the
530 dissolution processes in the high $p\text{CO}_2$ conditions. These results highlight the importance of
531 considering the physiological and geochemical interactions between temperature and
532 carbonate chemistry when interpreting species' vulnerability to ocean acidification. A better
533 understanding of how warming influences species' responses to high $p\text{CO}_2$ levels through
534 both direct (e.g. increases in metabolic rates) and indirect pathways (e.g. changes in carbonate
535 chemistry) is thus necessary.

536

537 *Conclusion*

538 A trade-off between stressors may affect the physiology of organisms in an
539 unexpected way (Kroeker et al. 2014). In our case, *C. fornicata* appeared to be able to tolerate
540 slight increases in $p\text{CO}_2$ but its calcification was affected by drastic conditions with a positive
541 effect of temperature, thereby mitigating any ocean acidification effects. This outcome
542 highlights the need of multistressor studies to understand the future of marine species in a
543 context of climate change in which different physico-chemical factors vary in different ways.
544 Furthermore, our results indicate that some species can be highly tolerant to future $p\text{CO}_2$
545 increases. *C. fornicata* tolerance likely stems from mechanisms that allow it to acclimate or

546 adapt to environmental fluctuations in its habitat (Clark et al. 2013), because species living in
547 environments with large abiotic variations tend to have high phenotypic plasticity, allowing
548 them to survive in stressful conditions (Somero 2010). This capacity to resist decreases in pH
549 may reinforce the ecological role of *C. fornicata* populations in the ecosystems in which they
550 are established, even under projected future conditions anticipated due to climate change.

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Tables

Table 1: Mean seawater temperature and parameters of the carbonate system in each $p\text{CO}_2$ treatment (3 aquaria per treatment) and at each trial period (i.e. temperature level). The pH_T (pH on the total scale) and total alkalinity (A_T) were measured whereas the other parameters were calculated. Mean A_T calculated for each trial period ($n = 3$ for controls 10°C and $19 < n < 30$ for other condition $p\text{CO}_2$ conditions) and $p\text{CO}_2$ condition was used for the calculations. $p\text{CO}_2$, CO_2 partial pressure; DIC, dissolved inorganic carbon and Ω_{Ar} , saturation state of seawater with respect to aragonite.

	n	Temperature ($^\circ\text{C}$)		pH_T		$p\text{CO}_2$ (μatm)		A_T ($\mu\text{Eq kg}^{-1}$)		DIC ($\mu\text{mol C kg}^{-1}$)		Ω_{Ar}	
		Mean	SE	Mean	SE	Mean	SE	Mean	SE	Mean	SE	Mean	SE
1st trial period (10°C)													
390 μatm	23	9.7	0.2	8.14	0.01	322	7	2365	2	2138	4	2.47	0.04
750 μatm	23	9.8	0.2	7.82	0.01	729	19	2368	2	2270	4	1.33	0.03
1400 μatm	23	9.5	0.2	7.55	0.03	1486	75	2376	2	2366	11	0.78	0.08
control 10°C	40	9.2	0.2	8.19	0.02	288	17	2370	3	2115	8	2.73	0.07
2nd trial period (13°C)													
390 μatm	27	12.9	0.2	8.12	0.02	356	25	2418	2	2167	8	2.76	0.07
750 μatm	27	13.0	0.1	7.81	0.01	781	20	2416	2	2304	3	1.48	0.03
1400 μatm	27	12.8	0.1	7.53	0.01	1557	43	2422	2	2405	4	0.82	0.02
control 10°C	41	11.0	0.1	8.18	0.01	297	12	2419	2	2152	5	2.88	0.05
3rd trial period (16°C)													
390 μatm	28	15.9	0.1	8.08	0.01	376	10	2379	5	2126	5	2.80	0.05
750 μatm	28	16.1	0.1	7.82	0.00	748	8	2369	5	2238	2	1.66	0.01
1400 μatm	28	16.0	0.1	7.55	0.01	1492	19	2380	5	2345	2	0.94	0.01
control 10°C	42	11.4	0.1	8.23	0.01	253	6	2376	4	2083	5	3.13	0.05

4th trial period (19°C)

390 μ atm	23	18.4	0.5	8.02	0.01	450	10	2391	2	2152	5	2.70	0.05
750 μ atm	23	18.6	0.5	7.77	0.01	858	19	2395	3	2266	4	1.68	0.04
1400 μ atm	23	18.4	0.5	7.51	0.01	1652	41	2394	2	2359	4	0.96	0.03
control 10°C	23	12.4	0.1	8.20	0.01	280	12	2393	1	2107	8	3.07	0.08

1 **Table 2:** Summary of two-way repeated measurements ANOVAs followed by Student-Newman-Keuls post hoc tests testing the effect of $p\text{CO}_2$,
 2 temperature and their interaction on *Crepidula fornicata* physiology. Numbers in bold indicate significant p-values and values with different
 3 letters are significantly different at $p < 0.05$.

	Two-way repeated measurements ANOVAs						Post hoc SNK tests									
	Factors			Factors			Factors			Factors						
	$p\text{CO}_2$		Temperature		$p\text{CO}_2 \times \text{Temperature}$		$p\text{CO}_2$ (μatm)			Temperature ($^\circ\text{C}$)						
df	F	p	df	F	p	df	F	p	390	750	1400	10	13	16	19	
Small individuals																
Respiration	2	1.685	0.219	3	14.530	< 0.001	6	1.893	0.103				a	b	b	c
Excretion	2	0.386	0.686	3	5.840	0.002	6	1.257	0.296				a	a,b	b	b
Filtration	2	0.271	0.766	3	15.439	< 0.001	6	5.996	< 0.001				a	a	a	b
Net calcification	2	6.705	0.008	3	1.849	0.152	6	2.307	0.050	a	a	b				
O:N ratio	2	4.944	0.022	2	2.214	0.127	4	0.382	0.819	a	b	a				
Large individuals																
Respiration	2	0.377	0.692	3	8.398	< 0.001	6	0.523	0.788				a	b	b	b
Excretion	2	0.563	0.581	3	17.850	< 0.001	6	0.371	0.893				a	b	c	b
Filtration	2	1.593	0.236	3	19.311	< 0.001	0	2.012	0.083				a	b	b	c
Net calcification	2	13.615	< 0.001	3	0.878	0.459	6	0.911	0.496	a	a	b				
O:N ratio	2	0.739	0.494	2	20.714	< 0.001	4	1.728	0.170				-	a	b	a

4

5 **Figures**

6

7 **Figure 1:** Individual respiration, ammonia excretion, filtration and net calcification rates in the three
8 $p\text{CO}_2$ treatments (shaded in grey) at 10, 13, 16 and 19°C for small (< 3 cm in length) and large (> 4.5 cm
9 in length) *C. fornicata* individuals. Different letters above bars or before $p\text{CO}_2$ caption indicate significant
10 differences between temperature or $p\text{CO}_2$ conditions, respectively. Results are expressed as mean \pm
11 standard error, n = 6 individuals.

12

13 **Figure 2:** O:N ratios for the three $p\text{CO}_2$ treatments (shaded in grey) at 13, 16 and 19°C for small and
14 large *C. fornicata* individuals. Different letters above bars or before $p\text{CO}_2$ caption indicate significant
15 differences between temperature or $p\text{CO}_2$ conditions, respectively. Results are expressed as mean \pm
16 standard error, n = 6 individuals.

17

18 **Figure 3:** Mean net calcification rates as function of aragonite saturation state, in the three $p\text{CO}_2$
19 treatments (shaded in grey), at 10 (○), 13 (△), 16 (□) and 19°C (◇) for all *C. fornicata* individuals (n =
20 12 individuals).

21

22 **Figure 4:** Respiration rates in the control treatment (10°C) for the different trial periods (i.e. temperature
23 levels) for single small (white bars) and large (grey bars) *C. fornicata* individuals. Results are expressed
24 as mean \pm standard error, n = 6 individuals.

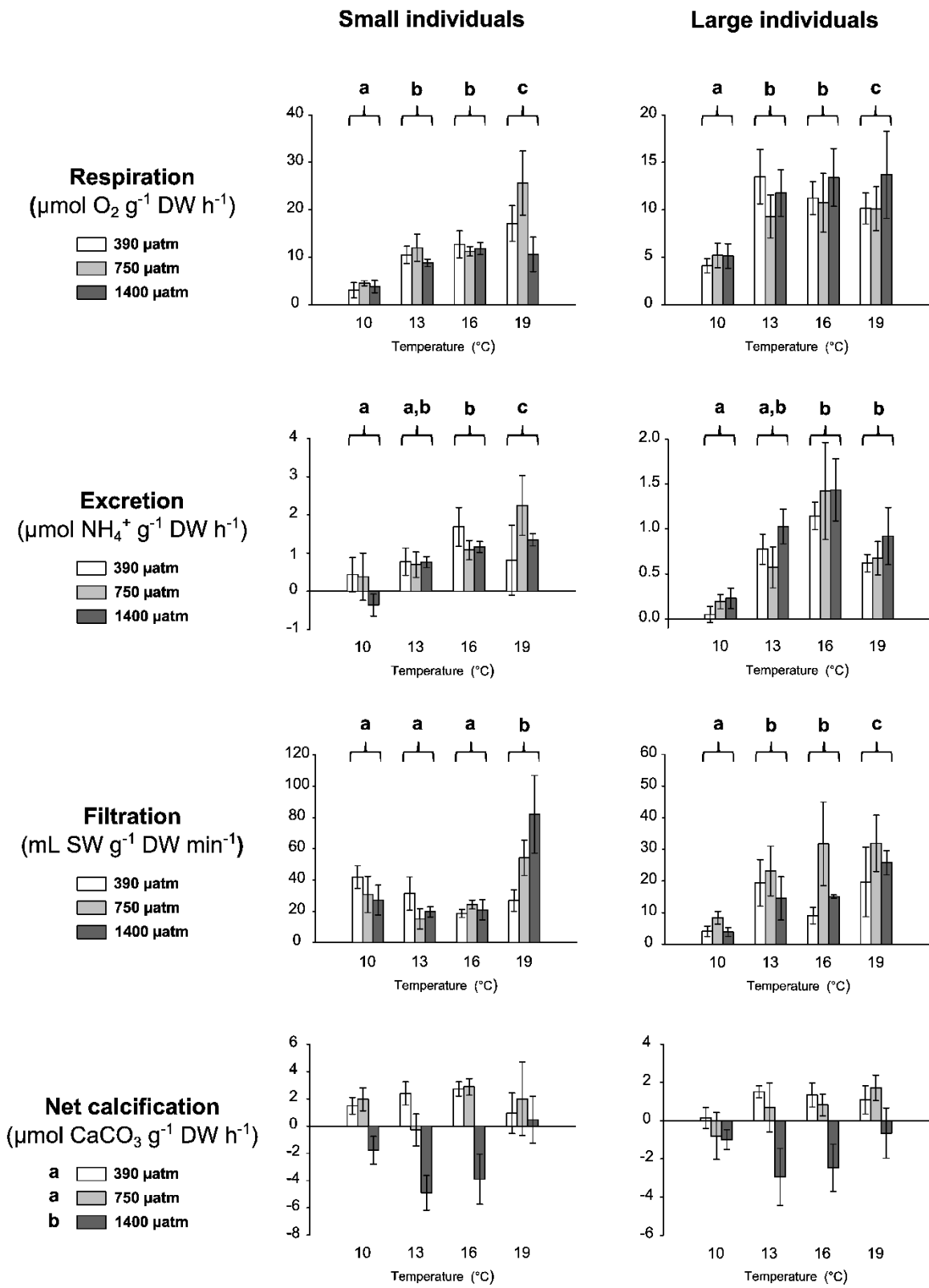
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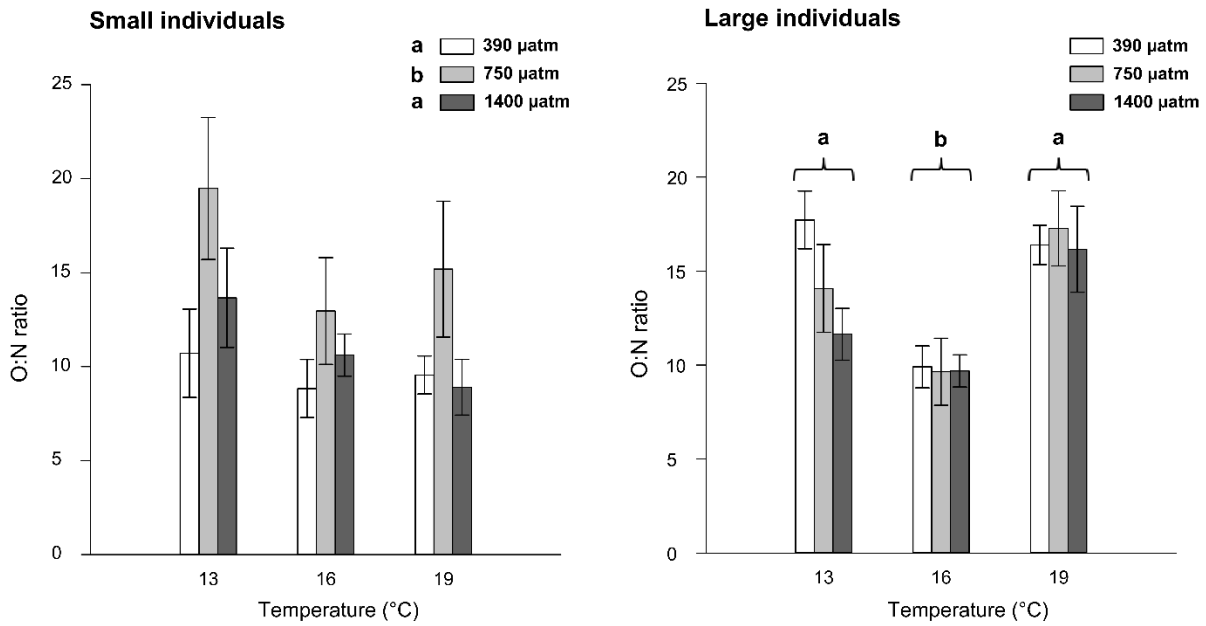
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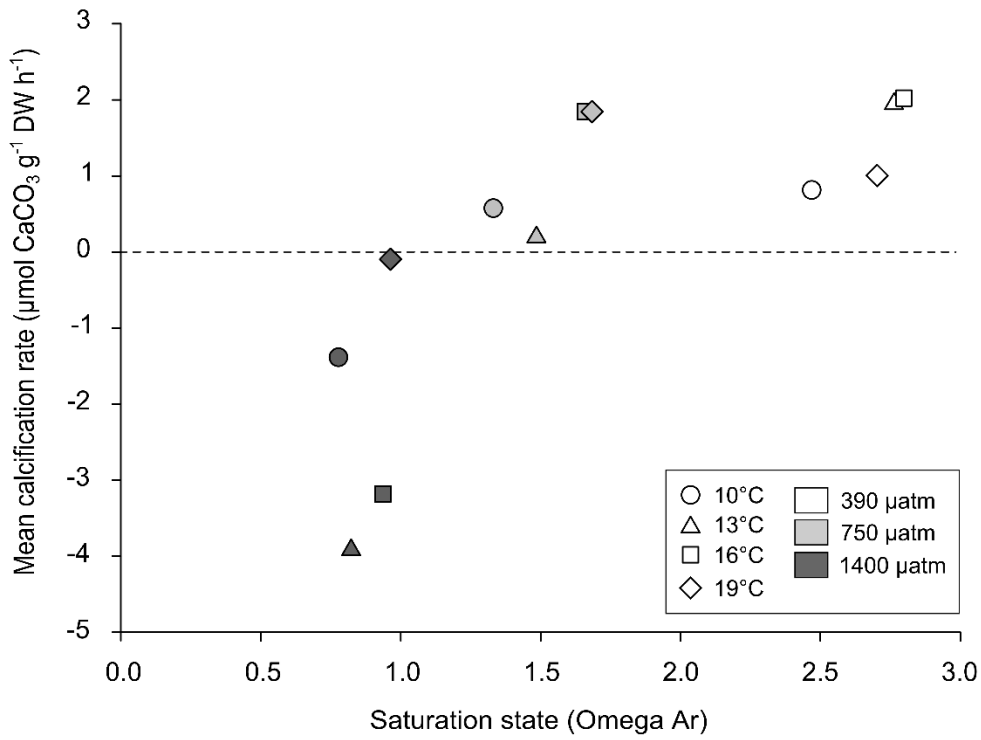
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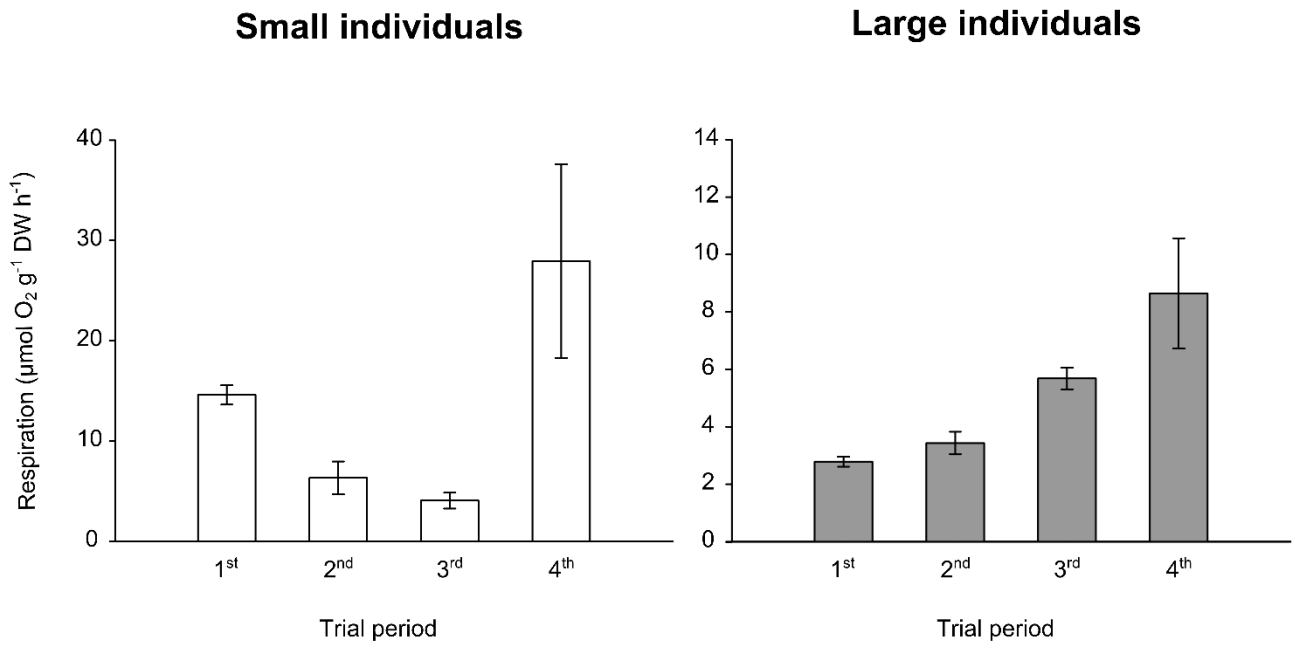
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