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Epitaxial NiWO₄ films on Ni(110): experimental and theoretical study of surface stability

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ABSTRACT

Despite the application potential of nickel tungstate (NiWO_4) in heterogeneous catalysis, humidity and gas sensing, etc, its surfaces have essentially remained unexplored. In this work, NiWO_4 nanoparticles and films with the wolframite structure have been grown via a solid-state reaction of $(\text{WO}_3)_3$ clusters and a $\text{NiO}(100)$ film on a $\text{Ni}(110)$ crystal surface and characterized by a variety of experimental techniques, including x-ray photoelectron spectroscopy (XPS), scanning tunneling microscopy (STM) and x-ray diffraction (XRD), combined with ab-initio density functional theory (DFT) calculations. NiWO_4 grows initially as three-dimensional (3D) crystalline nanoparticles displaying mainly two crystalline facets vicinal to the (100) surface, which merge with increasing the $(\text{WO}_3)_3$ coverage into a quasi-continuous epitaxial film. The DFT results provide an account of the energetics of NiWO_4 low index surfaces and highlight the role of faceting in the stabilization of extended polar (100) terraces. These combined experimental and theoretical results show that interaction with a metal substrate and vertical confinement may stabilize oxide nano-objects with high energy facets, able to enhance their reactivity.

INTRODUCTION

The first row transition metal tungstates MWO_4 ($M = Mn, Fe, Co, Ni$) belong to a technologically important family of materials, which crystallize with the monoclinic wolframite structure, space group $P2/c$. Among them, nickel tungstate ($NiWO_4$) has a great application potential in the heterogeneous catalysis,¹⁻⁷ humidity and gas sensors,⁸⁻¹⁰ microwave devices,¹¹ and photoanodes.^{12,13} Also, $NiWO_4$ has been demonstrated recently to exhibit electrochromic properties,¹⁴⁻¹⁶ which are superior to those of its binary NiO and WO_3 constituents. With the advance of nanotechnologies there is a growing interest in preparing $NiWO_4$ structures at the nanoscale, where surfaces and interfaces are playing a key role in determining novel physical and chemical properties of these materials. Whereas the bulk crystal structure of $NiWO_4$ has been well established by x-ray¹⁷ and neutron¹⁸ diffraction, the $NiWO_4$ surfaces are essentially unexplored. This is due to the fact that nickel tungstate compounds have been mostly synthesized in a polycrystalline form, while only few studies report on the growth of well-shaped $NiWO_4$ single crystals.¹⁹ Fundamental studies of wolframite surfaces are scarce, in general, and even more so when polar orientations are considered. Growing epitaxial oxide thin films on single crystal metal supports allows one to prepare well-defined surfaces, which can be investigated with atomic precision using modern surface analytic tools.²⁰

Recently, we have fabricated two-dimensional (2-D) Cu^{21} – and Fe^{22} – tungstate nanolayers on $Cu(110)$ and $Pt(111)$ surfaces, respectively, via an ultrahigh vacuum (UHV) surface epitaxial growth route using an on-surface solid-state chemical reaction in two dimensions. This preparation approach differs drastically from the conventional methods for bulk $NiWO_4$ production, where typically chemical reaction of molten powders, mechanical grinding in a vibrating mill, or high-temperature decomposition of complex precursors are used.

Here, instead, the reacting components are brought together in the form of structurally well-defined oxide phases, a $NiO(100)$ film and $(WO_3)_3$ clusters, on a single crystal $Ni(110)$ support; their reaction to form an epitaxial stoichiometric $NiWO_4(100)$ film has been initiated by oxidative annealing at elevated temperature and has been investigated *in situ* by a combination of x-ray photoelectron spectroscopy (XPS) with synchrotron radiation and scanning tunneling microscopy (STM) with atomic resolution, and *ex situ* by x-ray diffraction (XRD). In contrast to any previous work on $NiWO_4$ films, this study is the first one showing the growth of well-ordered epitaxial $NiWO_4$ films, whose surface structure was characterized at the atomic level. The stability of various low index surfaces of $NiWO_4$ has been explored

via DFT+U calculations, with a specific focus on the observed polar (100) orientation. We conclude that the observed morphology of the NiWO₄ film is likely driven by the interaction with the Ni substrate and by the vertical confinement in the ultrathin oxide film. Two structural phases are highlighted which show that the NiWO₄(100) surface is faceted into two vicinal surfaces, (16̄11) and (8̄11), with low energy (211) ledges. Models of edge and terrace configurations are proposed.

METHODS

The STM experiments were performed in an UHV system (base pressure $\sim 1 \times 10^{-10}$ mbar) containing a VT-STM/AFM (Omicron), a LEED optics and a cylindrical mirror analyzer (CMA) for Auger electron spectroscopy (AES), as well as the usual facilities for sample manipulation, surface cleaning and thin film evaporation.²³ STM images were recorded at room temperature in constant current mode with electrochemically etched W tips; the bias values cited are referred with respect to the sample.

High-resolution x-ray photoelectron spectroscopy (XPS) experiments were conducted at the Materials Science beamline, at the Elettra synchrotron light source in Trieste, Italy. The UHV chamber was equipped with a spherical electron energy analyzer (mean radius of 150 mm), rear view LEED optics and facilities for sample cleaning and thin film deposition. The photoemission spectra were taken at photon energies of 120 eV, 130 eV, 640 eV and 950 eV for the valence band and the W 4f, O 1s and Ni 2p core levels, respectively, with a total energy resolution of ~ 200 meV. The binding energy scale was calibrated with respect to the Fermi energy of the metal substrate, and all spectra were normalized to the secondary electron background at a few eV lower binding energy than the respective core level peak. A Shirley background was subtracted from the core level spectra prior to the peak decomposition analysis. The latter has been performed with the help of free-ware least-squares minimization software, FitXPS.²⁴ The core-level spectra have been decomposed into several components, whose individual shape consists of a Donjach-Šunjić profile convoluted with a Gaussian distribution.²⁵

X-ray diffraction measurements were performed at atmospheric conditions with a Philips X'Pert diffractometer equipped with an ATC3 texture cradle using chromium radiation from a sealed tube. At the secondary side a flat graphite single crystal was used to select monochromatic K_α radiation ($\lambda = 2.291$ Å), but the monochromator was transmissive also for

the wavelengths $\lambda/2$, $\lambda/3$, The experimental data are presented in terms of scattering vector q instead of scattering angles 2θ obtained by the following transformation: $q = (4\pi/\lambda) \sin\theta$. Specular scans were performed to detect the orientation of the single crystalline substrate, of the NiO and NiWO₄ crystallites relative to the substrate surface. The results are evaluated by comparing the experimentally observed peak positions with peak positions calculated by using the single crystal structure of the materials. An fcc structure of nickel with a lattice constant of 3.5228 Å,²⁶ an NaCl type structure of NiO with a lattice constant of 4.168 Å²⁷ and a monoclinic structure of NiWO₄ with $a = 4.5992$ Å, $b = 5.6606$ Å, $c = 4.9068$ Å and $\beta = 90.03^\circ$ were used.^{17,18} The peak positions were computed by the software POWDERCELL 2.3.²⁸ The in-plane alignments of the different types of crystallites relative to each other were determined by pole figure measurements.²⁹ The experimentally observed pole directions (net plane normals) were compared with stereograms using the software STEREOPOLE.³⁰

The Ni(110) surface was cleaned by cycles of Ar ion sputtering (1.5 keV) and annealing to 1000 K. In order to remove residual carbon contaminations the Ni crystal was exposed to oxygen (pressure of up to 1×10^{-7} mbar) at 570 K, followed by a brief flash to 1000 K in UHV. Surface order and cleanliness were monitored by LEED, STM and AES and XPS. The NiWO₄(100) films have been prepared by a three-step procedure. First, the Ni(110) surface has been oxidized in 5×10^{-6} mbar O₂ at 773 K for 10 min to produce a NiO(100) layer³¹ with a thickness of ~ 10 Å, as established by XPS (see below). In a second step, (WO₃)₃ clusters from a molecular beam²¹ have been deposited in UHV onto the NiO film at 300 K. Finally, the mixed oxide film has been subjected to oxidative annealing in 5×10^{-6} mbar O₂ at 973 K for 15 min. The last step was crucial to initiate the solid-state reaction between the binary NiO and (WO₃)₃ phases, forming the ternary NiWO₄ phase. The latter was found to exhibit a moderate structural order in STM, but annealing in UHV to 823 K resulted in better ordered films. The (WO₃)₃ clusters beam was generated by thermal sublimation of WO₃ powder at 1300 K in a thermal or e-beam evaporator, with the evaporation flux monitored by a quartz microbalance. The NiWO₄ coverage, cited in the text, is set equal to the (WO₃)₃ coverage measured in the quartz microbalance, with 1 monolayer equivalent (MLE) (WO₃)₃ corresponding to the density of the surface atoms of Ni(110), i.e. 1.14×10^{15} atoms cm⁻².

DFT calculations were performed with the Vienna Ab-initio Simulation Package (VASP)³² using the Projector Augmented Wave (PAW) method³³ to represent the electron-core

interaction. Standard Ni and W, and soft oxygen (energy cutoff of 270 eV) pseudopotentials provided by VASP were used. The typical imprecision with respect to standard oxygen pseudopotential (energy cutoff of 400 eV) is of 0.01 Å (bulk NiWO₄ lattice parameters) and 0.01 J/m² [(1x1)-NiWO₄(100) surface energies]. The Perdew-Wang 91 (PW91)³⁴ gradient-corrected exchange-correlation functional was employed, within the DFT+U approach as proposed by Dudarev,³⁵ with $U - J = 6$ eV for both Ni and W, in order to improve the description of the bulk band gap E_g . The calculated bulk lattice parameters $a = 4.65$ Å, $b = 5.72$ Å and $c = 5.02$ Å are in good agreement with the experimental ones.¹⁷ Considering the very small distortion of the monoclinic cell with respect to orthorhombicity, in all the following, we have assumed that $\beta = 90^\circ$.

NiWO₄ has an antiferromagnetic insulating ground state, associated to a doubling of the unit cell along the a axis. The present DFT+U settings allow to well reproduce this ground state and produce a density of states in agreement with hybrid DFT simulation of bulk NiWO₄.³⁶ The calculated band gap width E_g ($E_g = 3$ eV) is significantly larger than the GGA value of 0.9 eV and close to the experimental $E_g = 3.6$ eV.³⁷ Reproducing a correct band gap is required to correctly describe the ionic-covalent character of the cation-oxygen bonds, i.e. the electron distribution in the system,³⁸ as well as a correct ordering of polar versus non-polar surface energies.³⁹

A sketch of the wolframite structure is shown in Figure 1, using VESTA.⁴⁰ It is built from distorted NiO₆ and WO₆ octahedra, sharing edges, and forming zig-zag chains along the c axis. Structurally inequivalent oxygen atoms exist in the structure, half of them being bound to 1 W and 2 Ni, and the other half to 2 W and 1 Ni. This results in inequivalent Ni-O and W-O distances. The Ni-O distances are not very different from each other (2.09 Å, 2.08 Å and 2.08 Å, two of each type), while the WO₆ octahedra are much more distorted (2.12 Å, 1.95 Å and 1.84 Å).

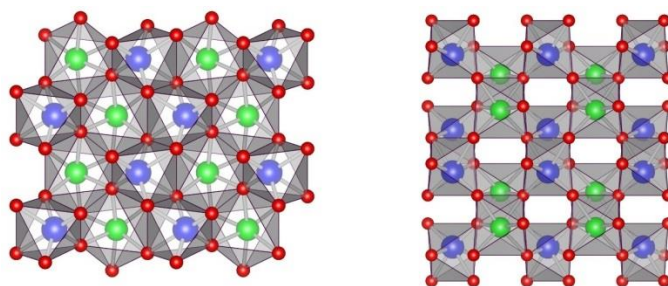


Figure 1. The NiWO₄ bulk structure (W-blue, Ni-green, O-red) seen perpendicular to the a (left) and c (right) axes

EXPERIMENTAL RESULTS

A. X-ray photoelectron spectroscopy characterization

The chemical composition and electronic structure changes occurring during the reaction of 5 MLE (WO₃)₃ clusters with the NiO(100) film, supported on the Ni(110) surface, have been followed by recording XPS spectra in the Ni 2p, O 1s and W 4f core level and valence band regions, the results are presented in Fig. 2. The Ni 2p_{3/2} spectrum of the clean Ni(110) surface (bottom spectrum in Fig. 2a) contains a main emission line with a binding energy (BE) of 852.6 eV (marked Ni⁰) and a satellite peak (marked S) shifted by ~ +6 eV, which is due to a two hole c⁻¹3d⁹4s² (c⁻¹ is a core hole) final state effect.⁴¹ The NiO(100) film is spectroscopically well distinguished from the Ni(110) substrate and displays a characteristic Ni 2p_{3/2} line shape (Fig. 2a, middle spectrum), which can be fitted with four core-level components with BEs of 854.4 eV (marked **A**), 856.1 eV (**B**), 861.0 eV (**C**) and 862.9 eV (**D**), typical for Ni²⁺ species in bulk NiO.⁴² The lowest BE main line **A** has been assigned to a c⁻¹3d⁹L⁻¹ final state, where c⁻¹ and L⁻¹ refer to a hole in the 2p core level and the ligand band, respectively, whereas the component **B** has been ascribed to a non-local screening process from neighbouring NiO₆ units⁴³ and/or surface effects.⁴⁴ The emission structures at the higher binding energy side, **C** and **D**, correspond to a c⁻¹3d¹⁰L⁻² and c⁻¹3d⁸ final states, respectively.⁴² Some remnant intensity of the underlying Ni substrate can be also seen at the low BE side of the Ni 2p_{3/2} spectrum of the NiO film, which proves useful in determining the thickness of the latter. From the attenuation of the substrate Ni⁰ signal by the NiO film and using the electron inelastic mean free length of 4.9 Å for NiO at a kinetic energy of 100 eV,⁴⁵ we estimate a thickness of 10.5 Å, which correspond to ~ 5 NiO layers. The O 1s spectrum of the NiO film (Fig. 1b, bottom curve) contains a single peak with a maximum at a binding energy of 529.9 eV, in agreement with similar values reported in the literature.^{46,47} The corresponding valence band spectrum (Fig. 2d, middle curve) is also typical for NiO(100) surfaces of bulk crystals⁴⁸ or thin films,⁴⁹ containing a doublet emission structure at 2.5 eV and 4.1 eV due to a multiplet of hybridized 3d⁸L⁻¹ final states, a broad band between 5 eV and 7 eV, related to O 2p-derived states, and a satellite peak around 10 eV, which has been attributed to emission from unscreened 3d⁷ final states.

The deposition of 5 MLE (WO₃)₃ onto the NiO(100) film on Ni(110) at room temperature results in the formation of a layer of intact (WO₃)₃ clusters, as confirmed by the W 4f (Fig. 2c, bottom curve) and O 1s (Fig. 1b, middle curve) spectra. The W 4f spectrum of the (WO₃)₃ clusters has been fitted with a single W 4f_{7/2} – W 4f_{5/2} doublet peak, with BEs of 35.9 eV and

38.1 eV, respectively, which are compatible with the oxidation state 6+ of the W atoms in the intact $(\text{WO}_3)_3$ clusters.²¹ The $(\text{WO}_3)_3$ layer exhibits no long-range order (as confirmed by LEED), therefore the W 4f spectrum displays a broad line shape with a full width at half maximum (FWHM) of the W 4f_{7/2} peak of 1.2 eV. The same holds for the $(\text{WO}_3)_3$ related component in the corresponding O 1s spectrum, which is characterized by broad (FWHM = 2.2 eV) line shape with a maximum at 530.8 eV, and is energetically well separated and distinguished from the attenuated O 1s peak of the underlying NiO film at 529.9 eV.

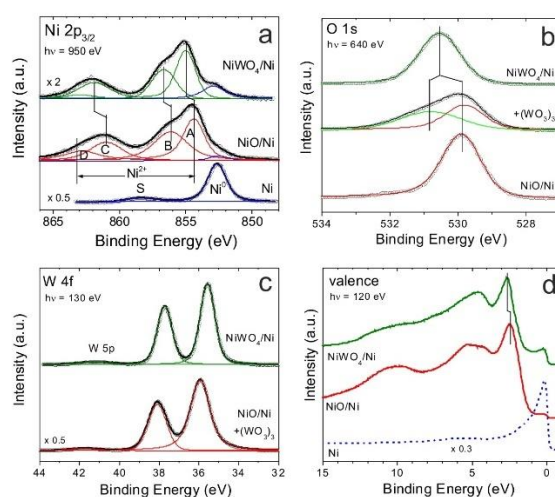


Figure 2. Ni 2p (a), O 1s (b), W 4f (c) core level and valence band (d) spectra of the different surfaces investigated: the Ni(110) surface, the NiO(100) film on Ni(110), the $(\text{WO}_3)_3$ clusters on NiO, and the NiWO₄ film on Ni(110). The core level spectra have been decomposed into different components, described in the text.

After the thermal treatment of the mixed $(\text{WO}_3)_3$ -NiO layer at 973 K in 5×10^{-6} mbar O₂ distinct changes are observed in all analyzed XPS spectra (top curves in Fig. 2), reflecting the generation of a new oxide phase, which has been unambiguously identified as NiWO₄ in the XRD measurements, discussed below. The W 4f spectrum contains a single W 4f_{7/2} -W 4f_{5/2} doublet peak, with the W 4f_{7/2} peak located at 35.6 eV, in good agreement with binding energy values reported for NiWO₄ compounds.^{16,50,51} The corresponding O 1s spectrum contains a single component with a BE of 530.3 eV, which is also compatible with NiWO₄ literature data.¹⁶ The absence of NiO – and $(\text{WO}_3)_3$ – related core-level components in the O 1s spectrum suggests that the original binary oxide phases have completely reacted together to form the ternary NiWO₄ phase. Note, that in the latter both W 4f_{7/2} and O 1s peaks display markedly narrowed line shapes with respect to corresponding spectra of the $(\text{WO}_3)_3$ clusters, with FWHM of 0.75 eV and 1.25 eV, respectively, which is fully in line with the crystalline

nature of the NiWO₄ phase, as reported below. The Ni²⁺ fine structure is generally maintained in the Ni 2p_{3/2} spectrum of the NiWO₄ film (Fig. 2a, top spectrum), but the peaks **A** and **B** are now shifted by +0.6 eV, the peak **C** by ~ +1 eV. These shifts are well documented in the literature^{16,51} and reflect the different local chemical environment of the Ni²⁺ species in the NiWO₄ phase, as compared to NiO. The substrate Ni⁰ signal in the Ni 2p spectrum of the NiWO₄ covered Ni(110) surface can again assist to estimate the average thickness of the NiWO₄ layer. Using the electron inelastic mean free length of 4.6 Å for NiWO₄ at kinetic energy of 100 eV,⁴⁵ the NiWO₄ thickness is determined to be ~ 9.8 Å. Note that the intensity decrease of the W 4f and Ni 2p peaks in the NiWO₄ spectra, with respect to the corresponding spectra of the (WO₃)₃ and NiO layers, respectively, is mostly due to the three-dimensional character of the NiWO₄ film growth, as established by STM (see e.g. Fig. 5), but some sublimation of (WO₃)₃ and/or reduction of the NiO layer at the high reaction temperature (923 K) cannot be excluded. Finally, the valence band spectrum (top curve in Fig. 2d) reflects the changes in the electronic band structure accompanying the formation of the NiWO₄ layer. Here, mainly hybridized Ni 3d and O 2p emissions are present. The 3d⁸L⁻¹ multiplet structure at the valence band maximum is shifted in NiWO₄ by +0.3 eV and partially overlaps with the O 2p band, which can be ascribed to the modified ligand field in the NiWO₄ layer with respect to NiO. The broad satellite structure around 10 eV is strongly attenuated, indicating a smaller weight of the unscreened 3d⁷ emission channel in NiWO₄. On annealing in UHV to 823 K, which was employed to improve the structural order of the NiWO₄ layer, no significant changes in all XPS spectra were observed, besides some decrease of the W 4f and Ni²⁺ intensities, accompanied by an increase of the substrate Ni⁰ signal, which we attribute to morphological changes in the oxide overlayer.

B. X-ray diffraction analysis

We have performed *ex-situ* XRD measurements of the 5 MLE NiWO₄ film on Ni(110), which has been prepared in UHV and characterized *in-situ* with STM (see Fig. 5b). The sample has then been taken out and left for one night in air, and afterwards inserted back in the UHV chamber. No differences in the film morphology and structure were found in STM and LEED, proving that the NiWO₄ film is robust and stable in air.

A specular x-ray diffraction pattern of the sample is depicted in Fig. 3a. Three different types of crystal structures are observed. The Ni(110) single crystal substrate generates a peak at $q_z =$

2.52 Å⁻¹, which originates from the (110) plane and corresponds to the 220 reflection from radiation with the wavelength $\lambda/2$. The NiWO₄ crystallites contribute with the 100 and 200 diffraction peaks and small traces of the 011 peak; from the NiO phase the 200 peak is observed. These results reveal that the NiWO₄(100) plane (and/or $\bar{1}00$ plane), as well as the NiO(100) plane, are parallel to the Ni(110) surface. We note parenthetically that the NiO phase has been formed during the exposure of the sample in air as a result of oxidation of the Ni(110) substrate. This is due to the fact that the NiWO₄ film, although thick enough to generate x-ray diffraction reflections, is not completely closed (see e.g. STM image in Fig 5b). In a next step, the x-ray diffraction pole figures were investigated. From each crystalline phase a set of two individual pole figures was measured. The 111 pole figure of the nickel substrate is shown in Fig. 3b. The experimentally observed pole densities are explained by the peaks 111 and $\bar{1}\bar{1}\bar{1}$. In a subsequent step the in-plane direction $[\bar{1}10]$ of the Ni(110) surface has been determined. A single pole figure of the NiWO₄ crystallites is displayed in Fig. 3c. Two different alignments of the NiWO₄ crystallites (marked by triangles up and triangles down, respectively) are required to explain the observed pole densities. In both cases either the [011] or $[0\bar{1}\bar{1}]$ direction of NiWO₄ phase is parallel to the $[\bar{1}10]$ direction of the Ni(110) substrate, while either the (100) plane is parallel to the substrate surface (triangles down) or the $\bar{1}00$ plane (triangles up). In summary, two different epitaxial alignments of the NiWO₄ crystallites relative to the Ni(110) surface are observed by the x-ray diffraction: either NiWO₄(100) or NiWO₄ $\bar{1}00$ \parallel Ni(110) and either NiWO₄[011] or NiWO₄ $[0\bar{1}\bar{1}]$ \parallel Ni $[\bar{1}10]$.

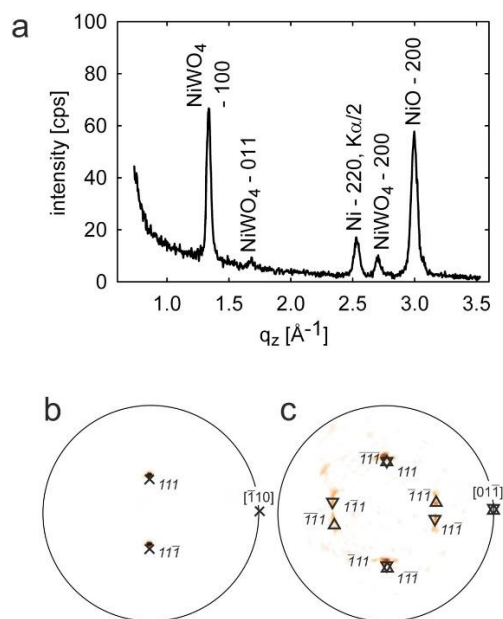


Figure 3. (a) Specular x-ray diffraction scan of the 5 MLE NiWO₄ film on Ni(110); (b) and (c) are x-ray diffraction pole figures of the Ni substrate and NiWO₄ film, respectively.

The most important findings of the XRD analysis are summarized in Fig. 4. The crystal structure of NiWO₄ is that of the mineral *Wolframite*, which is characterized by alternating layers of Ni and W atoms parallel to the (100) plane, separated by layers of quasi-hexagonally close packed oxygen atoms (Fig. 4a). The NiWO₄(100) surface plane, which according to the XRD data is parallel to the Ni(110) substrate, displays a rectangular unit cell with unit cell vectors $b = 5.6606 \text{ \AA}$ and $c = 4.9068 \text{ \AA}$, as exemplified in Fig. 4b. The diagonals of the unit cell are running along the $\langle 011 \rangle$ NiWO₄ directions, and these are parallel to the $\langle 110 \rangle$ Ni(110) substrate directions, as established in the pole figure analysis. This epitaxial alignment is most likely a result of the favorable matching of the NiWO₄(100) and Ni(110) lattices in the Ni[$\bar{1}10$] direction (Fig. 4c): the diagonals of the NiWO₄(100) unit cell (7.49 \AA) fit almost perfectly (mismatch of only 0.25%) to 3 times the Ni atom spacing in the Ni[$\bar{1}10$] direction (7.47 \AA).

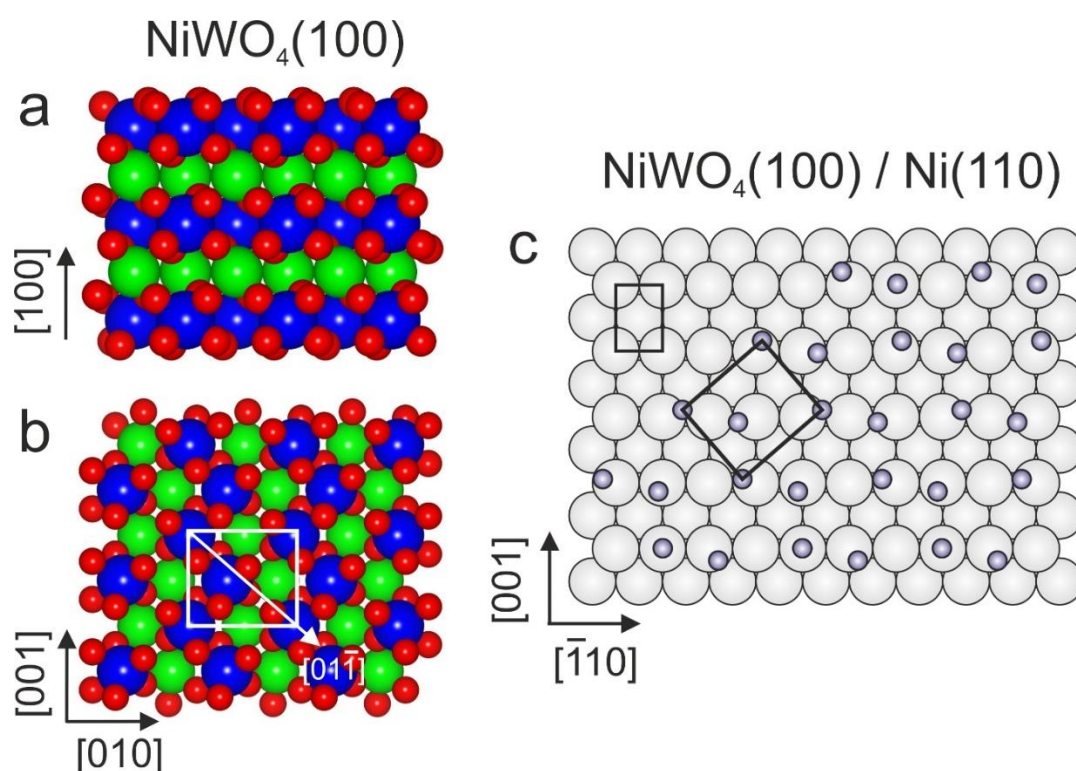


Figure 4. Side (a) and top (b) view of the NiWO₄ structure along the [100] direction. The NiWO₄(100) unit cell and the main crystallographic directions are indicated (W-blue, Ni-green, O-red); (c) lattice model visualizing the epitaxial alignment of the NiWO₄(100) and Ni(110) lattices, with the corresponding unit cells indicated. The large grey spheres correspond to the substrate Ni atoms, the small blue spheres to the NiWO₄ film Ni (or W) atoms.

C. Surface morphology and structure

The large-scale STM images in Fig. 5(a,b) reveal that the NiWO₄ film grows in a three-dimensional (3D) manner on the Ni(110) substrate. At a (WO₃)₃ coverage of 2 MLE well-faceted crystallites form (Fig. 5a), whose height varies between 2 and 5 nm. The most often occurring facets are the top one (indicated A) and the side facet B, which are both atomically flat and exhibit characteristic stripe patterns, with the stripes running along the Ni[$\bar{1}10$] substrate direction, as discussed in more detail below. The top facet is separated from the side facets by straight edges, which include angles compatible with the angles between high-symmetry directions parallel to NiWO₄(100) planes, as sketched in Fig. 5c. In particular, the NiWO₄[011] direction is parallel to the Ni[$\bar{1}10$] substrate direction, which is in line with the results of the XRD analysis. The edge parallel to the NiWO₄[011] direction is also the one, where the top facet A and the side facet B meet each other.

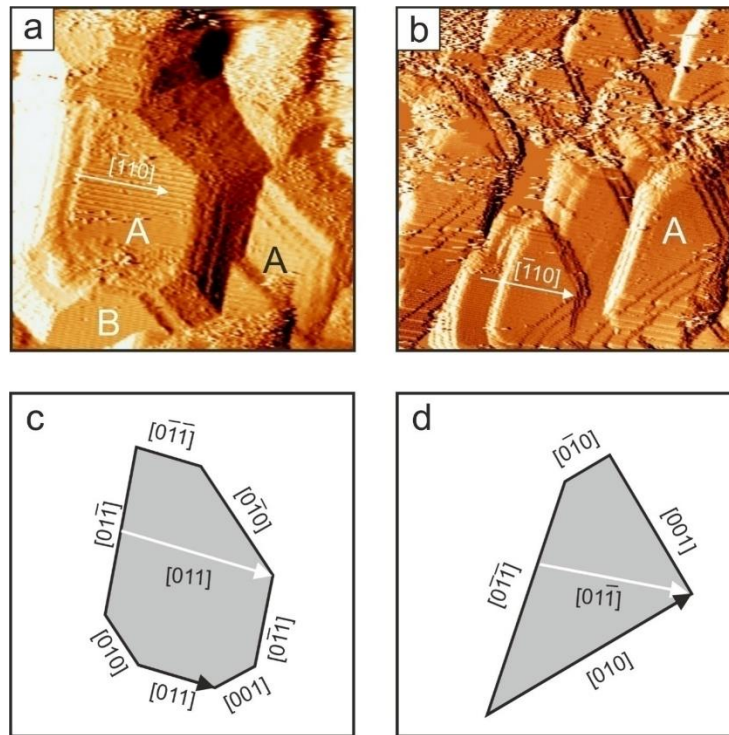


Figure 5. Large-scale STM images ($1000 \times 1000 \text{ \AA}^2$, $U = +1.0 \text{ V}$, $I = 10 \text{ pA}$) of 2 MLE (a) and 5 MLE (b) NiWO₄ films on Ni(110). The dominant facets are indicated A (top) and B (side); (c) and (d) are schematic drawings, highlighting the edges (black line) between the top (gray shaded) and the side facets in the STM images in (a) and (b), respectively. The black arrows indicate the rotation sense of the edge directions, while the white arrows indicate which NiWO₄ $\langle 011 \rangle$ direction is parallel to the Ni substrate [$\bar{1}10$] direction.

With increasing $(\text{WO}_3)_3$ coverage the density of the NiWO_4 crystallites increases and eventually at a coverage above 5 MLE they merge forming more planar, but discontinuous NiWO_4 film with a RMS roughness of 0.5 nm (Fig. 5b). This film morphology is possibly the reason for the oxidation of the underlying $\text{Ni}(110)$ substrate to NiO upon exposing the NiWO_4 film to air, as observed in the XRD. The surface of the NiWO_4 film is terminated exclusively by the A-type facets, which are separated by straight step edges including similar characteristic angles as in Fig. 5a. The analysis of these angles (see the sketch in Fig. 5d) revealed that the stripes of the A-facet (which are parallel to the $\text{Ni}[\underline{1}10]$ direction) are now aligned along the other diagonal of the $\text{NiWO}_4(100)$ unit cell, which is in the $\text{NiWO}_4[01\underline{1}]$ direction, in good agreement with the XRD analysis.

At intermediate oxide coverages (between 2 and 5 MLE) the NiWO_4 film displays an intriguing growth pattern reminiscent of the “hill-and-valley” reconstruction of vicinal surfaces, which is driven by the lowering of the surface energy. Figure 6a shows an STM image of a 3.3 MLE NiWO_4 film surface (Fig. 6b is a 3D view of the image in Fig. 6a), containing a staircase of alternating A- and B- facets, separated by atomically sharp and straight edges running in the $\text{Ni}[\underline{1}10]$ direction. The A-facets are distinguished in STM by their single stripe pattern, whereas the B-facets exhibit a double stripe structure (see Fig. 7 for more detail). A line profile taken along the A-B-A-B staircase is plotted in Fig. 6c and allows us to determine the angle between the A- and B-facets, which measures $6^\circ \pm 1^\circ$.

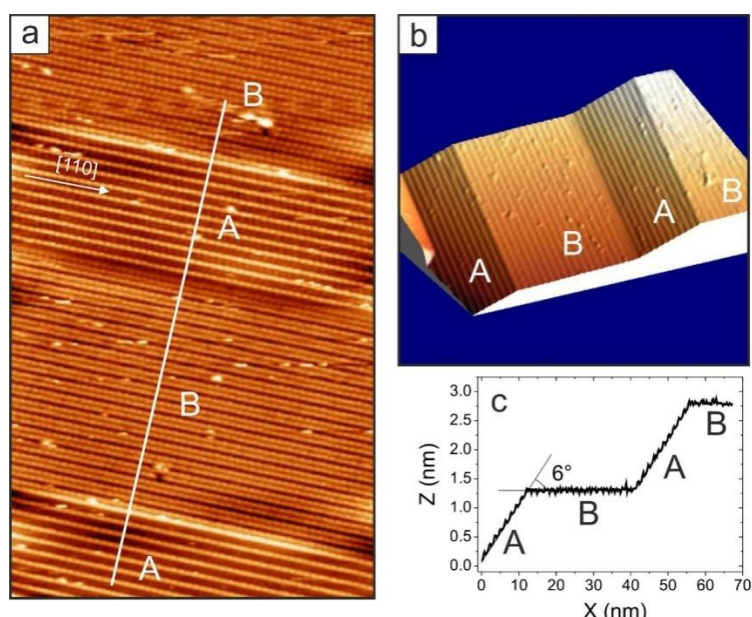


Figure 6. (a) STM image of a 3.3 MLE NiWO_4 film on $\text{Ni}(110)$ ($450 \times 700 \text{ \AA}^2$, $U = +0.5 \text{ V}$, $I = 50 \text{ pA}$). The A and B facets are indicated; (b) 3D view of the STM image in (a) to emphasize the staircase film morphology; (c) line profile across the A – and B – facets, including an angle of 6° .

The high-resolution STM images of the A- and B- facets, presented in Fig. 7a and 7b, respectively, assist identifying their geometrical structure and orientation. The A – facet displays single stripes oriented along the Ni[$\bar{1}10$] direction with a periodicity of $15.8 \text{ \AA} \pm 0.2 \text{ \AA}$ across the stripes (see line profile 2-2' in Fig. 7a and 7c). The stripes consist of a periodic array of bright protrusions, elongated along the stripe direction, with a separation of $7.6 \text{ \AA} \pm 0.2 \text{ \AA}$ (line profile 1-1' in Fig. 7a and 7c). These two periodicities define an oblique unit cell for the A – facet (indicated with a black solid line in Fig. 6a) with an obtuse angle of $100^\circ \pm 2^\circ$. Between two neighboring stripes a row of elongated maxima with a darker contrast can be also recognized in both the STM image (Fig. 7a) and the line profile 2-2' (Fig. 7c). The B – facet shows a more complex appearance in the STM images (Fig. 6b): it consists of double zig-zag stripes running also parallel to the Ni[$\bar{1}10$] direction, with the distance between the maxima along the stripe direction being the same as in the A – facet, i.e. 7.6 \AA , but the periodicity across the stripes is twice larger, i.e. 31.5 \AA . The corresponding unit cell is drawn with the solid line on the STM image in Fig. 7b, with the two unit cell vectors including the same angle of 100° , as in the facet A. We note in passing that the STM images of the A- and B-facets show no significant sample bias and polarity dependence in the range (-2.0 to +2.0 V), with the best contrast achieved for the A-facet at positive sample bias, and for the B-facet at negative sample bias.

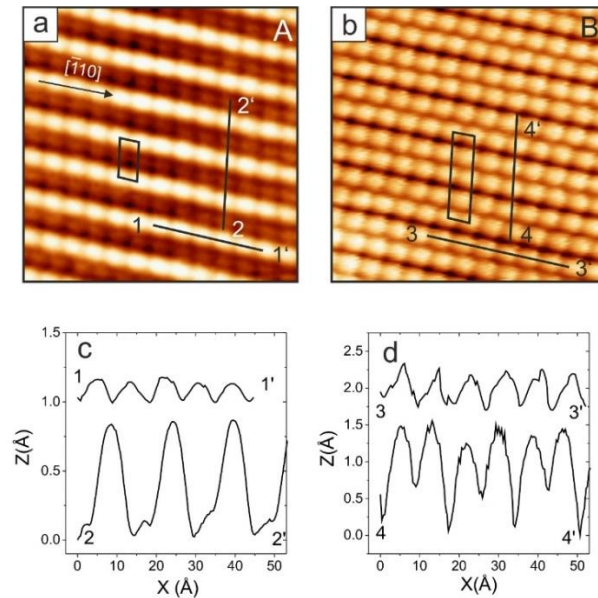


Figure 7. High-resolution STM images of the A (a) and B (b) facets: (a) $100 \times 100 \text{ \AA}^2$, $U = +1.0 \text{ V}$, $I = 100 \text{ pA}$, (b) $100 \times 100 \text{ \AA}^2$, $U = -1.0 \text{ V}$, $I = 150 \text{ pA}$. The corresponding unit cells are indicated with black lines; (c) and (d) display line profiles taken along the directions indicated in (a) and (b) respectively.

COMPUTATIONAL RESULTS

To the best of our knowledge, no NiWO₄ surface has been theoretically studied in the past. We have thus made a preliminary assessment of the stability of several NiWO₄ low index surfaces, prior to focusing on the polar (100) orientation.

A. NiWO₄ low energy surfaces

We have considered a series of low Miller index surfaces, restricting ourselves to configurations with (1x1) surface unit cells. In Table 1 we report results on the most stable of them, in the order of increasing surface energy. The corresponding atomic surface structures are given in Figure S1 (Supplementary material).

(1x1) Surface	Es (J/m ²)	Polar character	2D cell area (Å ²)	N _{Ni-O} ^{broken}	N _{W-O} ^{broken}
(010)	0.62	NP	46.66	4 (0.086)	0
(210)	0.77	NP	73.94	6 (0.081)	0
(211)	0.77	NP	45.54	2 (0.044)	1 (0.022)
(011)	0.82	NP	70.75	4 (0.056)	2 (0.028)
(100)Ni	1.01	P	28.68	3 (0.105)	0
(201)	1.06	NP	78.23	4 (0.051)	2 (0.025)
(100)W	1.10	P	28.68	2 (0.070)	2 (0.070)
(101)	1.18	P	60.43	3 (0.050)	2 (0.033)
(001)	1.26	NP	53.19	2 (0.038)	2 (0.038)

Table 1. Most stable NiWO₄ surfaces with low Miller indexes: their polar character (NP = non-polar, P = compensated polar), their surface energy (in J/m²), the area of their 2D unit cell (in Å²), the number of broken Ni-O and W-O bonds per unit cell and per surface area (in parenthesis, in Å²).

As a general trend, the orientations of lowest energy are non-polar and involve the smallest number of broken bonds per surface area. This is particularly clear in case of W-O bonds, and highlights their larger strength compared to the Ni-O bonds. As a consequence, the most stable surfaces are Ni-terminated. If W atoms are present in the outermost layers, they are bound to an oxo- ligand (e.g. the (210) surface) with a very short W-O bond (~1.74 Å). We note that a simple picture with a constant ratio between the Ni-O and W-O bond strengths cannot account for the complex hierarchy of the surface energies in Table 1, which is also

influenced by fine details of surface bond relaxation and long range electrostatic effects. As a peculiarity of the wolframite structure, the surfaces of lowest Miller indexes have not necessarily the lowest density of broken bonds, and, as a corollary, not the lowest surface energy (e.g. the (100) and the (001) surfaces), at variance with simpler crystallographic structures.

The calculated relatively low stability of the (100) surfaces (to be discussed below) suggests that the large facets vicinal to the (100) orientation observed experimentally are either due to a kinetic hindrance or, more likely, to a limited thickness of the oxide film and to its substantial interaction with the metallic Ni substrate.

B. NiWO₄(100) surface

The NiWO₄(100) surface is polar, according to Tasker's classification,⁵² because the repeat unit perpendicular to the surface consists of successive charged layers 2Ni/4O/2W/4O, and thus bears a non-zero dipole moment. As a consequence, electrostatic stabilization requires compensating charges in the surface layers. Considering the absence of experimental evidence of changes of cation oxidation states, we will focus here on stabilization by non-stoichiometric surface reconstructions, which have proven to efficiently heal polarity.^{53,54}

For the Ni and W terminations, the required compensating surface charge is equal to half the layer charge, so that, e.g., one half of surface cations have to be removed. Conversely, at the two nonequivalent oxygen terminations, either one or three out of four oxygen atoms per unit cell have to be removed, depending if the underlying layer is composed of W or of Ni. We have analyzed these alternative surface terminations with stoichiometric bulk-cut slabs. They are symmetric when terminated by Ni or W (e.g., W/4O/2Ni/4O/.../2W/4O/2Ni/4O/W) and asymmetric for the oxygen terminations (3O/2W/4O/2Ni/4O/.../2W/4O/2Ni/O). As a consequence, only the average surface energy of the oxygen terminations is obtained.

The most stable surface is Ni-terminated ($E_s = 1.01 \text{ J/m}^2$), with only three broken Ni-O bonds per surface unit cell (Fig. 8, left panel). The surface Ni atoms undergo a noticeable inward relaxation which strongly shortens the three remaining Ni-O bonds (1.85 Å compared to 2.12 Å in the bulk) and leads to an energy lowering of nearly 80% (from 1.82 J/m² to 1.01 J/m²). Much stronger atomic rearrangements take place at the W termination, where an oxygen atom bound to two subsurface Ni moves outwards and binds on-top the outermost W (Fig. 8, right panel). With respect to the unrelaxed termination, two Ni-O bonds per unit cell are broken and one strong W=O bond is formed with a length (1.74 Å) typical of a terminal oxo-ligand.⁵⁵

Upon such rearrangement, the surface energy is divided by roughly a factor of 5 (from 5.64 J/m² to 1.10 J/m²), which brings it close to, but slightly higher than, that of the Ni-termination. At this point it is worth noting that the comparison between the Ni- and W=O-terminated unrelaxed surface energies shows that the ratio between W-O and Ni-O bond strengths is roughly equal to 3, consistently with the previous consideration of the relative stability of the different surface orientations (Table 1). As far as the oxygen terminations are concerned, due to the asymmetry of the two sides of the slab, only the average surface energy can be obtained without relying on grand potential analysis, which would be beyond the scope of the present work. Since their average surface energies are much higher than those of the Ni- or W-terminated surfaces (Table S1, Supplementary Material), the oxygen terminations will not be further discussed.

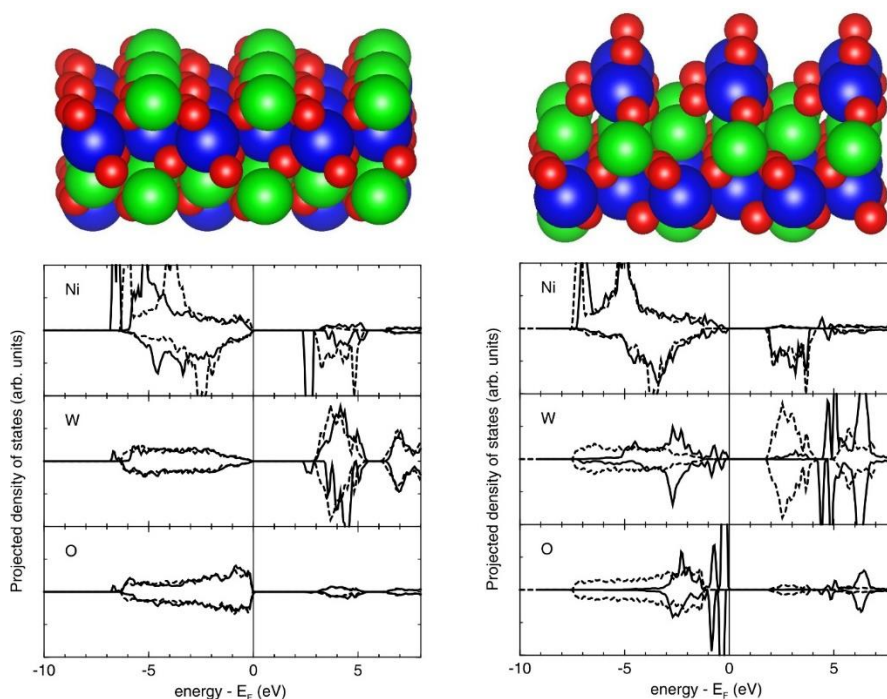


Figure 8. Top: side views of the Ni – terminated (left) and W=O – terminated (right) NiWO₄(100) (1x1) surface. W, Ni and O atoms are represented in blue, green and red spheres, respectively. Only five top layers are shown in the upper panels and three in the lower panels. Bottom: their LDOS projected on surface atoms (solid lines) and central atoms (dashed lines).

Figure 8 (bottom part) displays the local density of states (LDOS) projected on surface atoms at the Ni (left panel) and W=O (right panel) terminations together with the projection on the corresponding atoms in the slab center. On the Ni termination, the main feature is a strong Ni surface state at the bottom of the conduction band (CB), leading to a gap reduction of 0.6 eV.

Due to its low coordination (three oxygen neighbors instead of six) and despite the contraction of the three remaining Ni-O bonds, the surface Ni experiences a reduced electrostatic (Madelung) potential which lowers its unoccupied d states into the bulk band gap. The top of the valence band (VB) has a strong oxygen character, with nevertheless a noticeable mixing with Ni states. These features of the electronic structure and the geometric position of the outermost Ni atoms above the surface oxygen layer produce a very well defined contrast in STM images, simulated under the Tersoff-Hamann approximation, especially at positive bias (Figure S2, Supplementary material).

On the W=O termination, a high density of surface states exists at the top of the VB, inducing a gap reduction of about 1.3 eV (Fig. 8, right panel). These states are strongly localized on the singly-coordinated oxygen atom bound to the surface W. The LDOS displays a double peak, associated to the p_z oxygen orbital for the lowest energy peak and p_x/p_y for the higher one. As expected from its geometric position above the surface and from the contribution of its orbitals to the top of the VB, the on-top oxygen atom gives strong simulated STM contrast, especially at negative bias (Figure S2, Supplementary material).

The surface configurations considered so far correspond to a chessboard arrangement of half a layer of surface cations and consequently produce a dense chessboard-like pattern in simulated STM images. Several alternative regular arrangements of surface cations can be conceived, respecting the polarity compensation criterion. In the following we focus on the row-wise arrangement of the surface cations. Following the experimental indications, we use a surface unit cell spanned by the diagonals of the $\text{NiWO}_4(100)$ unit cell: $\vec{A} = \vec{b} + \vec{c}$ and $\vec{B} = 2 * (\vec{c} - \vec{b})$ ($A = 7.61 \text{ \AA}$, $B = 15.21 \text{ \AA}$), with the cationic rows oriented along the short axis \vec{A} and every second row missing along the \vec{B} axis (Figure S3, Supplementary material).

The Ni-terminated row-wise configuration (Fig. 9a) is only by 0.09 J/m^2 less stable than the chessboard one. It is characterized by a dimerization of the Ni atoms along the rows, with successive Ni-Ni distances equal to 3.24 and 4.53 \AA . Interestingly, the simulated STM image (Fig. 9c) of this configuration displays a single elongated spot per dimer, located close to midway between the Ni-Ni atoms. Such STM signature of Ni dimers is reminiscent of the slightly elongated spots in the experimental images. At the W=O termination (Fig. 9b) the row-wise and chessboard termination are nearly iso-energetic. At this termination, the dimerization does not alter the STM signature (Fig. 9d), with a separate spot for each on-top oxygen atom.

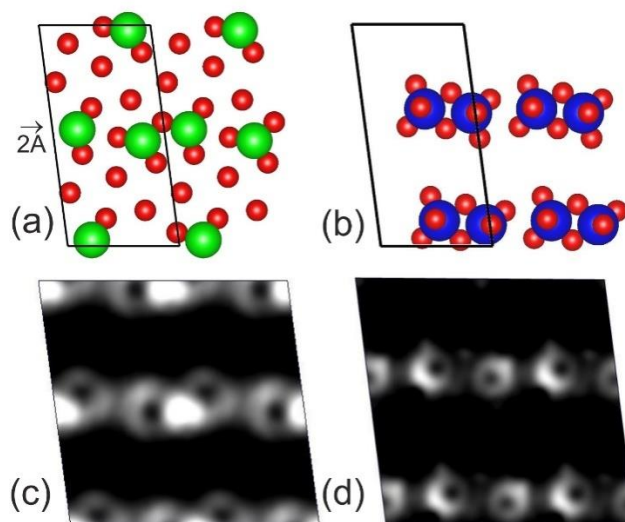


Figure 9. Top view of the outermost atoms of the Ni – terminated (a) and W=O – terminated (b) reconstructed NiWO₄ (100) surfaces. The simulated STM images of the models in (a) and (b), obtained at bias -1V and 3V (with respect to the VB maximum), respectively are shown in (c) and (d). The unit cell is doubled along the \vec{A} axis.

DISCUSSION

Our XPS and XRD results identify unambiguously the chemical composition and the crystalline structure of the nickel tungstate film on the Ni(110) surface as stoichiometric NiWO₄ with the wolframite structure. According to the XRD scans and pole figures (Fig. 3) the NiWO₄ film grows with its (100) and/or ($\bar{1}00$) crystal planes parallel to the Ni(110) substrate. Moreover, the NiWO₄ film is rotationally aligned with the substrate to achieve best lattice matching, which is realized when the diagonals of the NiWO₄(100) unit cell (running in the NiWO₄<011> directions) are parallel to the Ni[$\bar{1}10$] direction. The STM images reveal a more complex picture of the morphology of the NiWO₄ film, which grows initially as 3D crystallites exposing several facets, with the most dominant ones being the top facet A and a side facet B. Upon increasing the coverage the NiWO₄ crystallites coalesce and form a rough film which is terminated exclusively by the A – facets. Both the A – and B – facets display oblique unit cells, whose size and symmetry differ significantly from the rectangular unit cell of the NiWO₄(100) surface. Considering the small angle that they form with respect to the (100) orientation, they may be considered as vicinal to the (100) surface.

To elucidate the effect of misorientation on the stability of the (100) surface, we have considered a series of vicinal terminations, with (100) terraces separated by steps running along $\text{NiWO}_4 \langle 011 \rangle$ direction. Since such steps require $(n\bar{1}1)$ step ledges, we have chosen the $(2\bar{1}1)$ ones due to their low surface energy (Table 1). Construction of such vicinal surfaces, with different proportions of polar (100) terraces and non-polar $(2\bar{1}1)$ ledges, rises a general question of polarity compensation condition at faceted surfaces. Indeed, the standard evaluation of compensating charge density from the dipole density of the repeat unit cell is possible, but becomes cumbersome in such complex case. A method in which the polarity of ledges and terraces is separately compensated raises the question of how to treat edge and counter-edge atoms. The safer method, when possible, consists in representing the vicinal surfaces by stoichiometric symmetric slabs. Since, in such case, a stoichiometric bulk repeat unit without dipole moment can always be found⁵³ and since both the bulk region and each surface region are stoichiometric, the two surfaces are necessarily charge-neutral and thus polar compensated.

Table 2 summarizes results on the series of surfaces with an increasing proportion of $(2\bar{1}1)$ ledges. They all display surface energies smaller than the planar (100) one. The energy reduction is driven by the contribution of the $(2\bar{1}1)$ ledges, whose energy is lower than that of (100) terraces. As a consequence, surface energies strongly decrease as the proportion of $(2\bar{1}1)$ facets increases, thus pointing towards a thermodynamic preference for a strong faceting of the (100) surface. These results suggest that the limited faceting found in the experiment is likely due to the thinness of the NiWO_4 film and to its non-negligible interaction with the Ni substrate.

Surface	E_s (J/m^2)	A (\AA)	B (\AA)	θ ($^\circ$)	ψ ($^\circ$)
(100)	1.01	7.61	5.02	90.0	0
$(8\bar{1}1)$	0.96	7.61	15.91	97.2	16
$(4\bar{1}1)$	0.91	7.61	8.92	96.4	30
$(3\bar{1}1)$	0.80	7.61	15.57	110.0	40
$(2\bar{1}1)$	0.77	7.61	6.84	118.9	43

Table 2. Structural and energetic characteristics of a series of surfaces vicinal to the (100) one with an increasing proportion of (211) ledges: surface energy E_s ; lengths (A and B) and angle θ between their 2D lattice vectors; their misorientation angle ψ with respect to (100) surface.

We note that all structural characteristics of the (811) surface (the lengths of its lattice parameters A and B, the angle θ between them, and its misorientation angle ψ with respect to the (100) surface) match very closely the experimental structure of the A facet. The B facet corresponds most likely to a (1611) vicinal orientation: $A = 7.61 \text{ \AA}$, $B = 30.78 \text{ \AA}$, $\theta = 97.5^\circ$, $\psi = 8.7^\circ$. This surface has not been explicitly included in the present series of ab initio calculations, but from Table 2 its energy can be estimated in the range 0.96– 1.0 J/m².

Figure 10 shows two models of the vicinal (811) surface in which one recognizes a small (211) ledge facet (Figure S1 in Supplementary material). Model I (Fig. 10a) has a Ni-type step edge and two additional Ni atoms on the (100) terrace in a row-wise configuration per unit cell. The corresponding surface energy is equal to 1.09 J/m². More stabilization can be obtained by removing one NiWO₄ formula unit from each step edge (Model II - Fig. 10b), $E_s = 0.96 \text{ J/m}^2$. In this case a W=O-type edge is obtained, with an oxygen atom in an on-top position and a short W=O bond characteristic of an oxo-ligand. It is worthwhile noting that the row-wise arrangement of Ni atoms on the terraces is slightly more stable than the chessboard one, at variance with the result on the planar (100) surface.

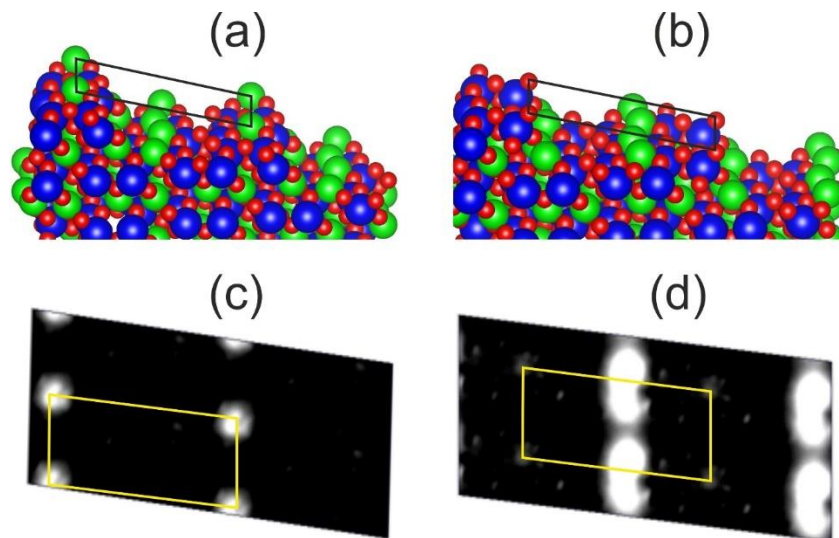


Figure 10. Two proposed models (side views perpendicular to the \vec{A} vector of the structure) of the NiWO₄ (811) vicinal surface. Ni atoms on the terraces are arranged in a row-wise configuration and the step edge is Ni-terminated (a) or W=O terminated (b). The simulated STM images of (a) and (b) obtained at bias 3V (with respect to the VB maximum) are shown in Fig. 10 (c) and (d), respectively.

The NiWO₄ (8 $\bar{1}$ 1) unit cell is indicated on the structure models (black line) and the STM images (yellow line).

For these two vicinal surfaces, we have simulated STM images (Figs. 10c and 10d). On the Ni-edge termination, the bright spots correspond to Ni atoms on the step edges, which are both at the highest position and display a high density of states in the energy window upon consideration. Ni atoms located on the terraces are almost invisible, due to their lower position. On the W=O edge termination, the bright elongated spots in the STM simulated image (Fig. 10d) correspond to the Ni on the terraces, and the very weak central spots to states localized on the W=O moiety and thus reproduces very well the contrast of the experimental STM image of the (8 $\bar{1}$ 1) facets (Fig. 7a).

CONCLUSIONS

The growth and surface structure of ultrathin NiWO₄ films on a Ni(110) surface was studied for the first time experimentally using XPS, XRD and STM and analyzed theoretically by DFT calculations. It is shown that the NiWO₄ film grows epitaxially with the (100) planes of the wolframite structure parallel to the Ni(110) surface and is rotationally aligned to the substrate, such that the NiWO₄[01 $\bar{1}$] direction is parallel to the Ni[$\bar{1}$ 10] direction. The oxide film shows a complex surface morphology displaying mainly two crystalline facets corresponding to (16 $\bar{1}$ 1) and (8 $\bar{1}$ 1) planes, which are vicinal to the NiWO₄(100) surface. Several mechanisms for the stabilization of the polar NiWO₄(100) surface have been investigated in the DFT calculations, including chessboard and row-wise reconstructions of the (100) plane, as well as more complex faceting into high-index stepped surfaces. It is found that the latter mechanism leads to a more substantial lowering of the surface energy thanks to the presence of low energy (2 $\bar{1}$ 1) ledges. Inspired by the experimental STM results, a theoretical model has been proposed for the (8 $\bar{1}$ 1) vicinal surface, which consists of (100) terraces with outermost Ni dimers, (2 $\bar{1}$ 1) ledges, and W=O groups on the step edges. It is concluded that the observed stabilization of the vicinal facets is the result of a competition between the reduction of surface energy, the interface lattice matching with the Ni(110) substrate, and the confinement effects in the ultrathin NiWO₄ film. In the context of various applications for which surface and its reactivity are important, our study shows the possibility

of stabilizing nanoparticles with high-energy facets (thus with lower-coordinated and more reactive surface sites).

ASSOCIATED CONTENT

Supporting Information

Structure of different NiWO₄ low energy surfaces (Figure S1); Structural and electronic characteristics of various NiWO₄(100) (1x1) surface terminations (Table S1); Simulated STM images of the Ni- and W-terminated NiWO₄(100)-(1x1) surfaces (Figure S2); Top view of the bulk-truncated (100) surface with the indication of the (1x1) unit cell and the rotated cell with $\vec{A}=\vec{b} + \vec{c}$ and $\vec{B} = 2 * (\vec{c} - \vec{b})$ axis (Figure S3).

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Notes

The authors declare no competing financial interest.

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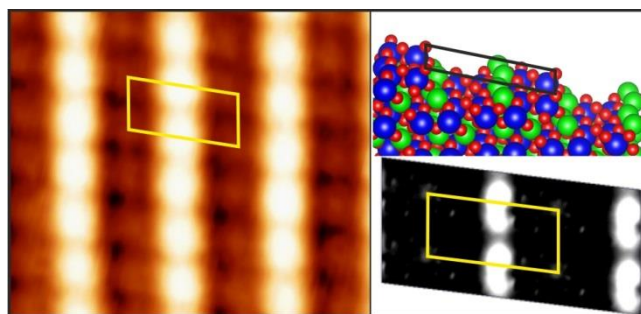
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