

## A New Family of Titanium–Oxo Clusters: Complementarity of Solid State NMR and XRD

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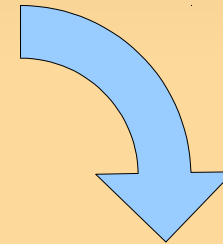
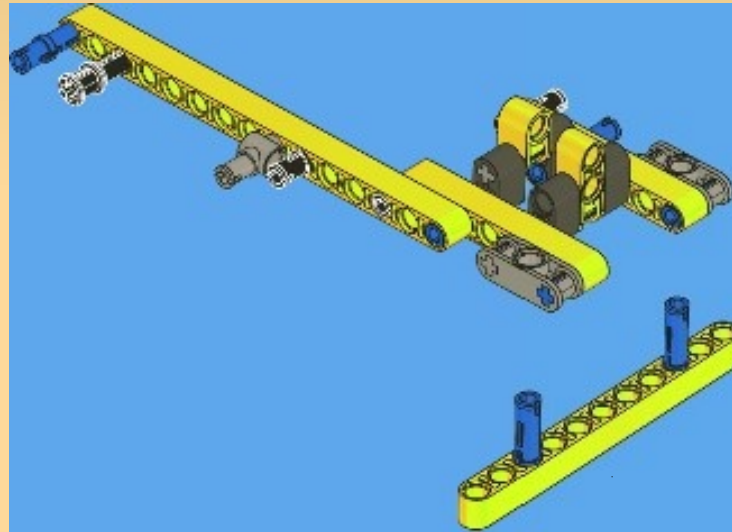
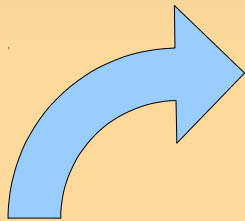
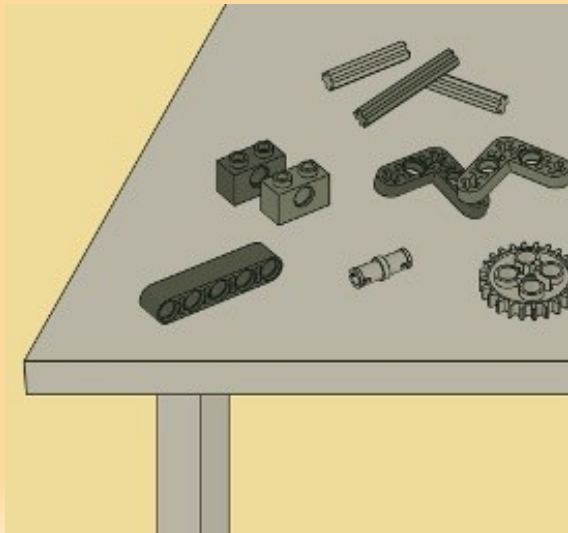
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# Outline

- Synthesis
- XRD characterization
- First NMR characterization
- Solvent exchange
- Functionalization
- NMR assignments

# Synthesis

- Hybrid materials
- Bottom-up approach (NBU/SBU)

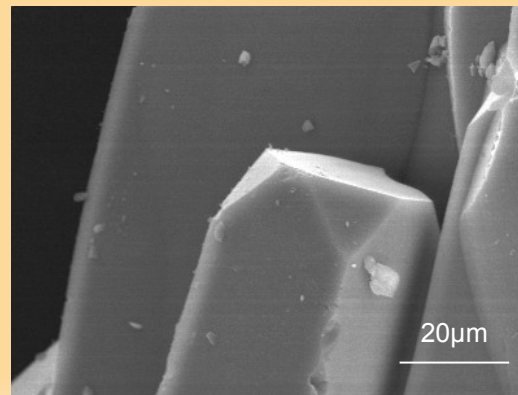
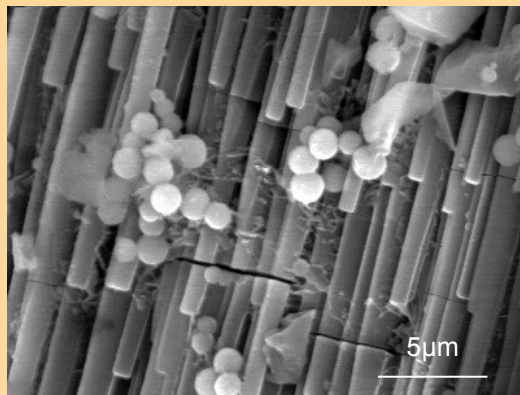


# Synthesis

- Rich and versatile Ti chemistry
- Sol-gel route, soft conditions
- In situ water generation :
  - $\text{Ti(OR)}_4 + \text{R'COOH} \rightarrow \text{Ti(OR)}_3(\text{OOCR'}) + \text{ROH}$
  - $\text{ROH} + \text{R'COOH} \rightarrow \text{R'COOR} + \text{H}_2\text{O}$
- Hydrolysis :  $\text{H}_2\text{O} + \text{Ti(OR)}_4 \rightarrow \text{Ti(OR)}_3(\text{OH}) + \text{ROH}$
- Condensation :
  - $\text{Ti-OH} + \text{RO-Ti} \rightarrow \text{Ti-O-Ti} + \text{ROH}$
  - $\text{Ti-OH} + \text{HO-Ti} \rightarrow \text{Ti-O-Ti} + \text{H}_2\text{O}$
- Low stability of alkoxo ligands  $\text{Ti(OR)}_4$

# Synthesis

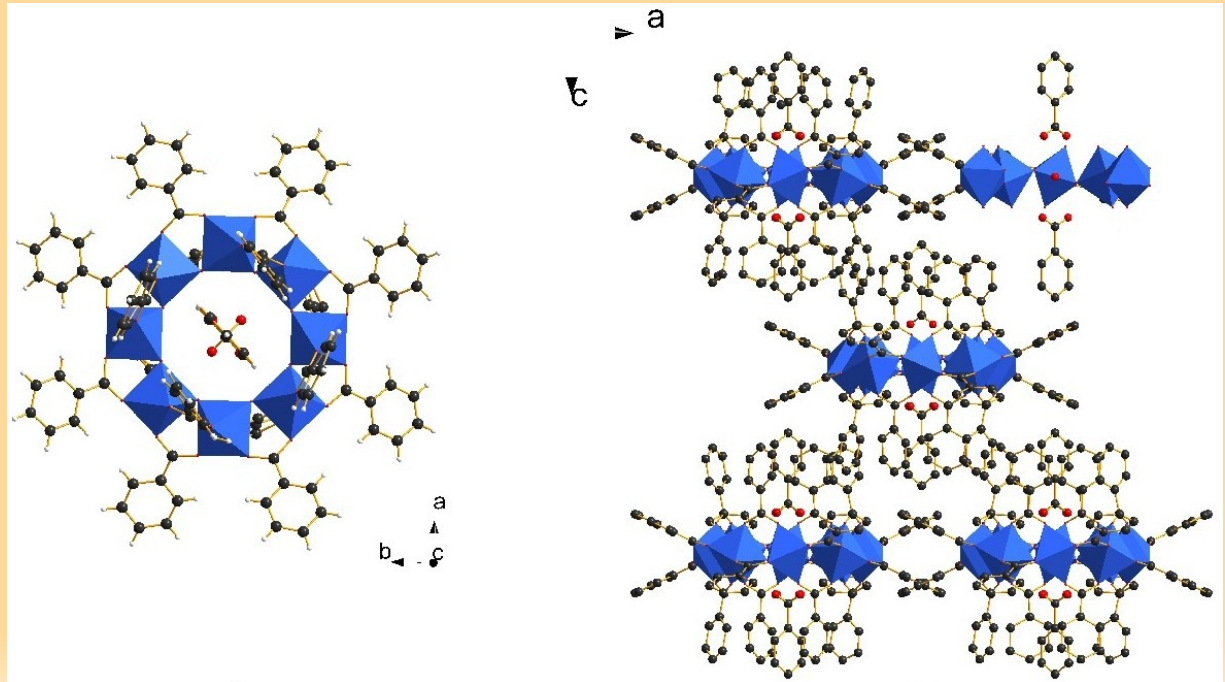
- $\text{Ti}(\text{O}^i\text{Pr}) + \text{PhCOOH}$  (excess)
- 5 days at  $105^\circ\text{C}$  (solvothermal)  $\longrightarrow$
- $\text{Ti}_8\text{O}_8(\text{OOCPh})_{16} \cdot (\text{PhCOOH})_2 \cdot \text{H}_2\text{O}$
- Crystalline pure oxo-carboxo cluster
- Needles
- Stable



# Characterizations

- XRD:
  - 8-member ring vertex shared  $\text{TiO}_6$  octahedra
  - Bridging bidentate benzoates
  - Body centered tetragonal with central ring inversed
  - No pi-stacking

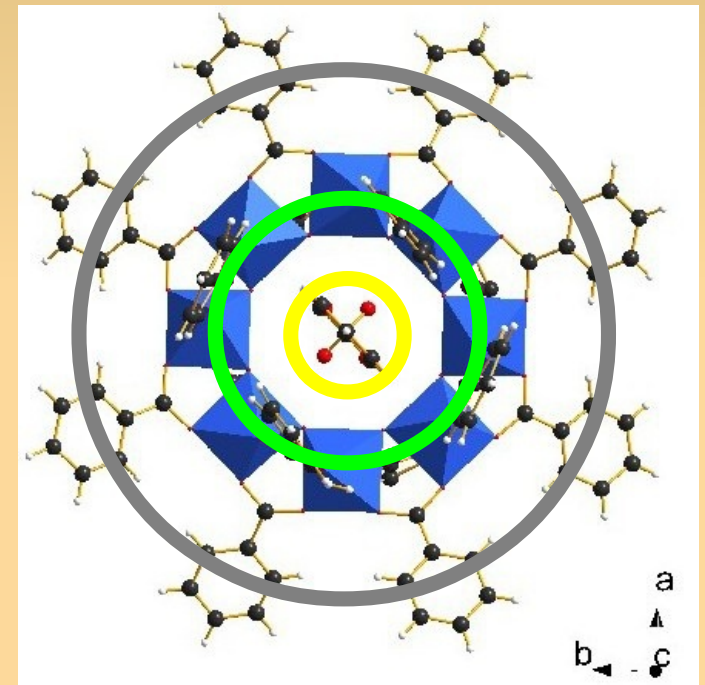
Internal diameter : 9 Å  
External diameter : 20 Å



# Characterizations

- XRD:
  - 3 types of carboxy groups :
    - Equatorial (4+4)
    - Axial (4+4)
    - Trapped (1+1)
  - Different thermal agitations

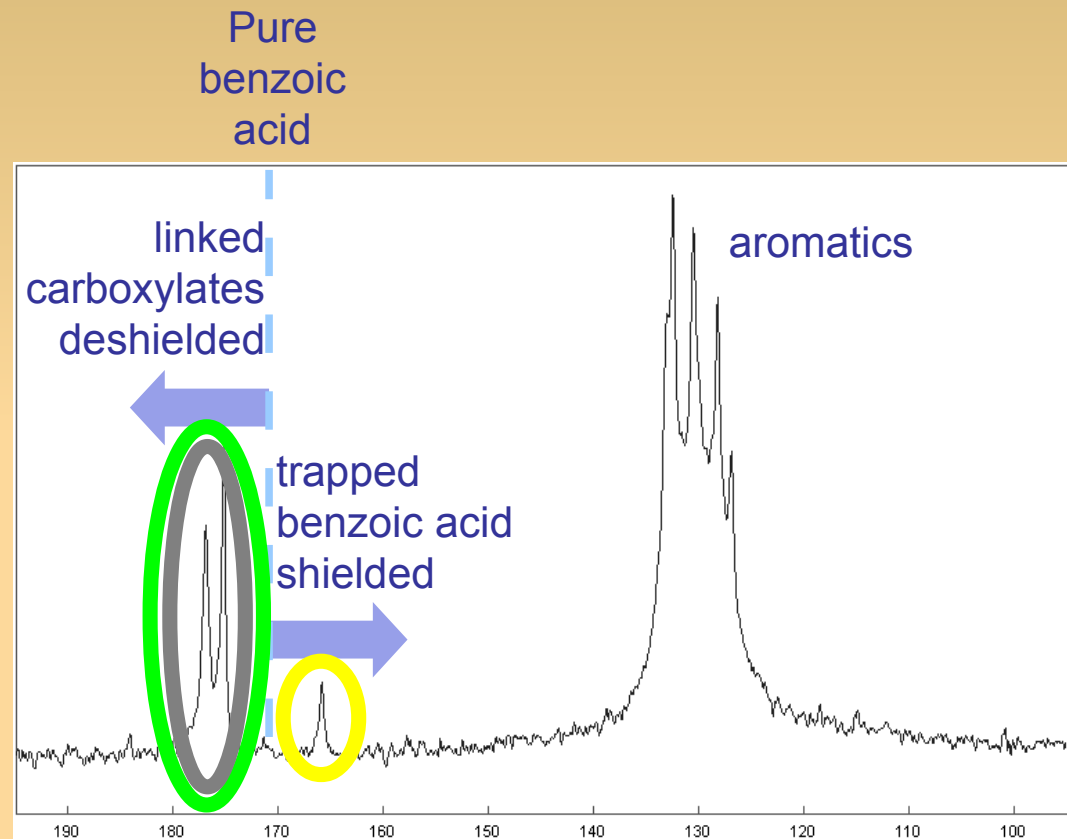
Aromatic ring	Ued (*10 <sup>3</sup> Å <sup>2</sup> )	Occupancy
C <sub>eq</sub>	35	1
C <sub>ax</sub>	47	0.5 (2 sites)



- Difficulties to solve the structure : single crystal, 150 K, disorder

# Characterizations

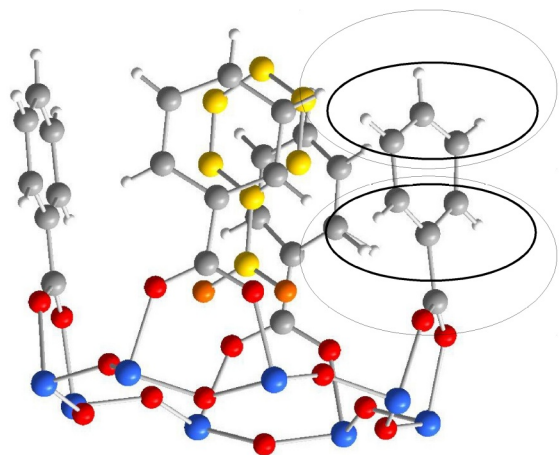
- $^{13}\text{C}$  solid state NMR:
  - Quick, Tamb
  - 3 carboxy :  $\text{C}_{\text{eq}}$ ,  $\text{C}_{\text{ax}}$ ,  $\text{C}_{\text{free}}$
  - Ratio 4/4/1
  - Different  $\delta$  compared to pure benzoic acid
  - Deshielded ones:
    - linked to the Ti
    - 2 line widths
    - $\text{C}_{177} > \text{C}_{175}$



7 T, MAS 14 kHz, rotor 4 mm, hpdec  $90^\circ$  30 s



# Trapping inside the pores

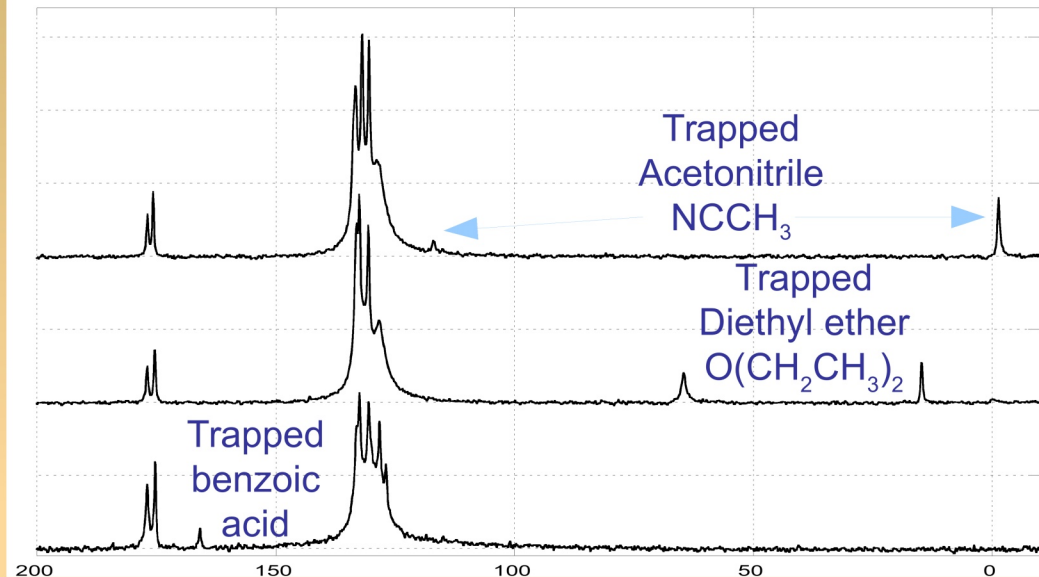


Induced magnetic field



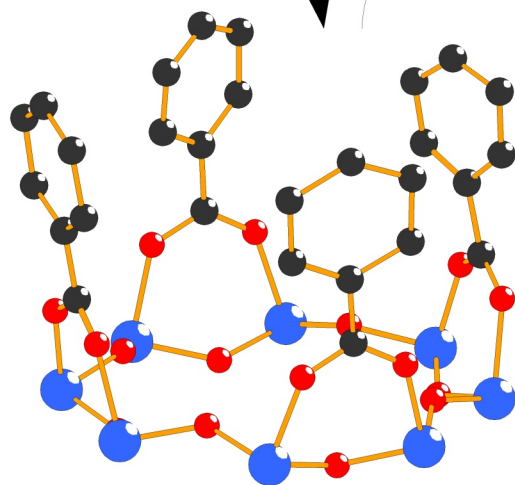
Shielded region  
(-6 ppm for benzoic acid)

Only upper axial groups are represented

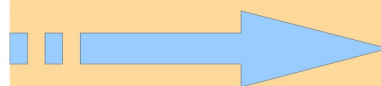


Acetonitrile,  
diethyl ether...

Benzoic acid

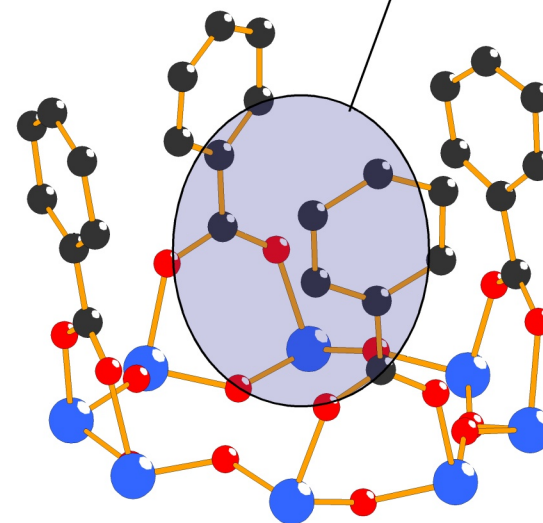


Storage  
& release



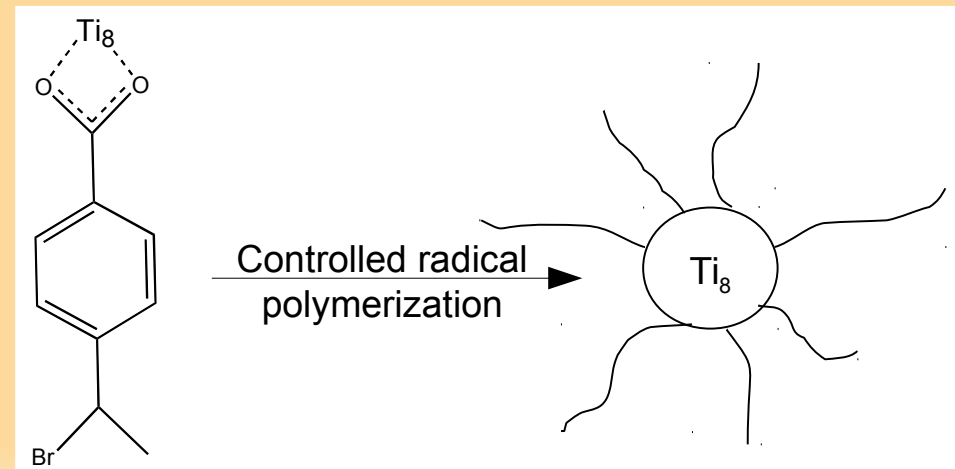
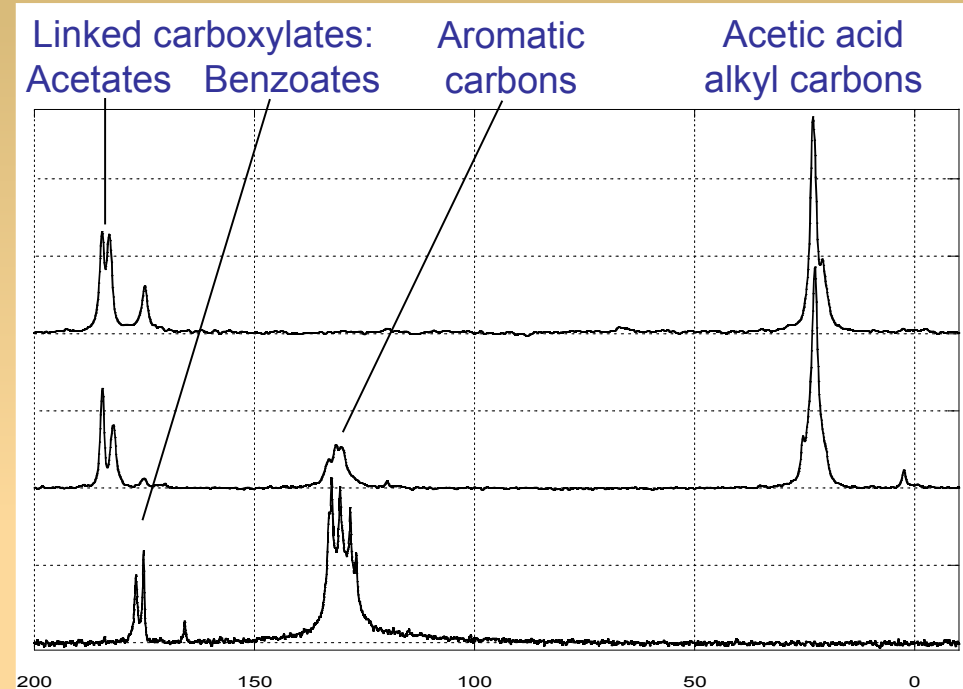
Easy characterization of the trapped molecules by NMR

Target molecule



# Ligand exchange

- To have new functionalities
- Structure is kept with acetate ligand
- Selectively functionalize axial or equatorial ?
- Need to assign carboxylate peaks

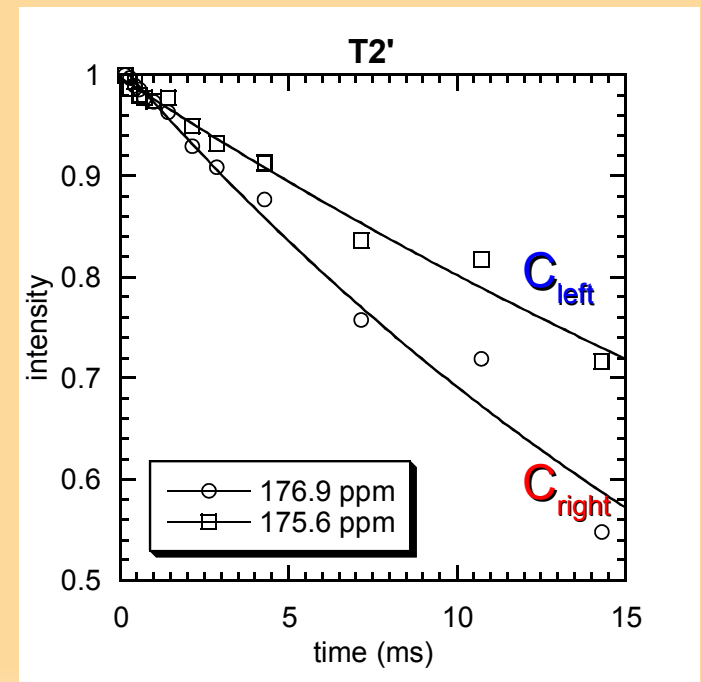
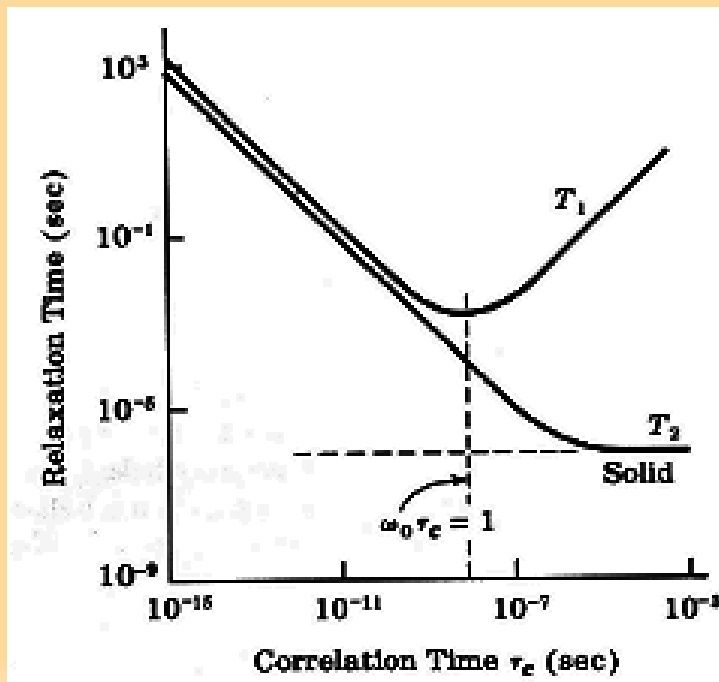
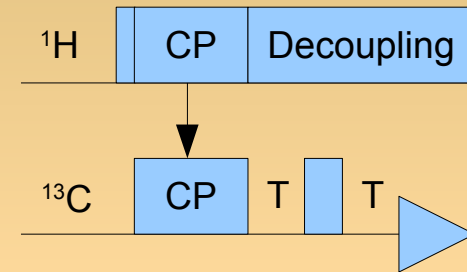


# NMR assignments

## $^{13}\text{C}$ $T_2'$ measurement

$$T_2^* < T_2' < T_2$$

Carboxy	$T_2^*$ (ms)	$T_2'$ (ms)	Assignment
$C_{\text{left}}$	5.3	26	$C_{\text{eq}}$
$C_{\text{right}}$	8.4	46	$C_{\text{ax}}$

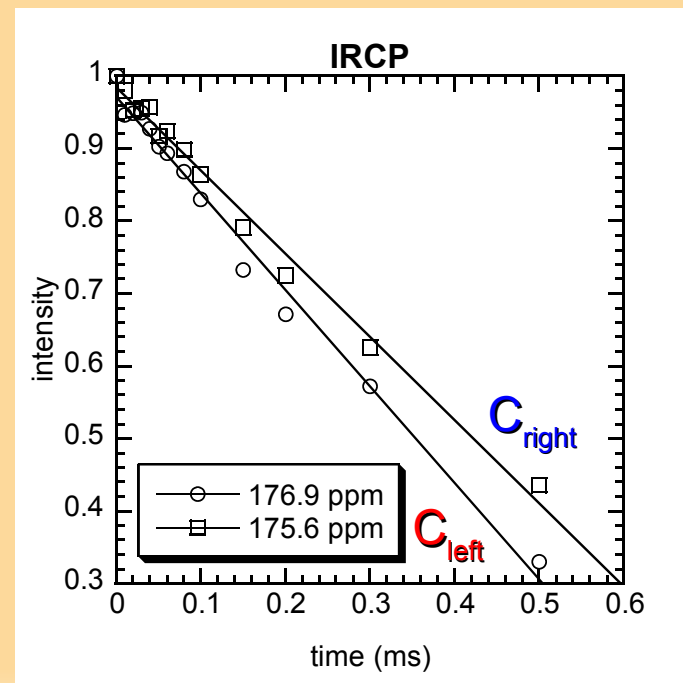
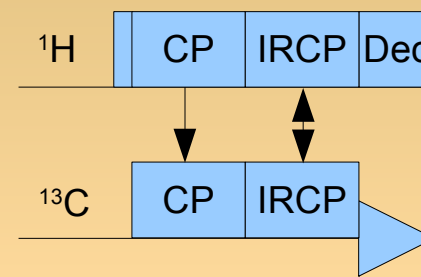


7 T, MAS 14 kHz, rotor 4 mm

# NMR assignments

## $^{13}\text{C}$ IRCP measurement

- $T_{\text{IRCP}}$  : time to inverse the magnetization
- $D_{\text{CH}_n} \sim \langle n(\gamma_1 \cdot \gamma_2) / r^3 \rangle_t$
- $T_{\text{CH}_2} < T_{\text{CH}} < T_{\text{CH}_3} < T_{\text{Cq}}$
- No complete inversion

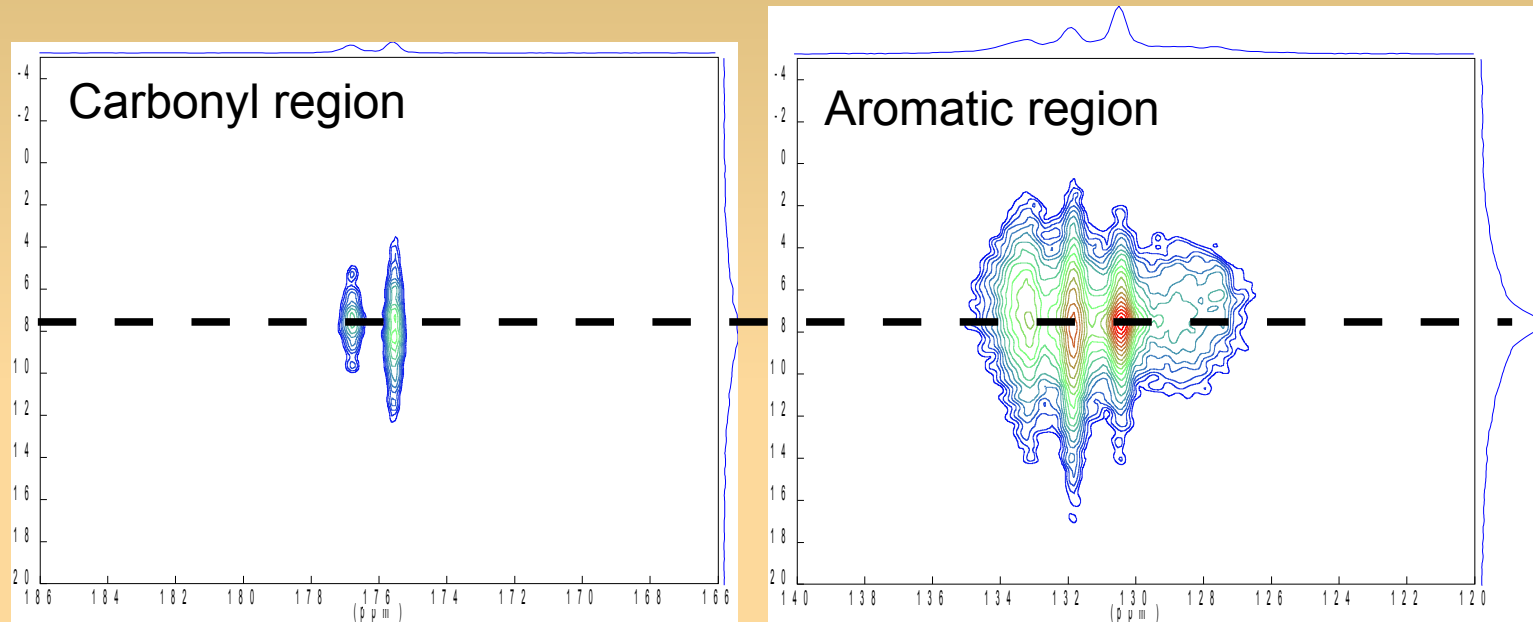


7 T, MAS 14 kHz, rotor 4 mm

Carboxy	$T_2^*$ (ms)	$T_2'$ (ms)	$T_{\text{IRCP}}$ (ms)	Assignment
$\text{C}_{\text{left}}$	5.3	26	0.75	$\text{C}_{\text{eq}}$
$\text{C}_{\text{right}}$	8.4	46	0.93	$\text{C}_{\text{ax}}$

# NMR assignments

- 2D HETCOR: 7 T, MAS 14 kHz, rotor 4 mm,  $t_{cp}=2$  ms, 512 scans, 122 slices




- $^1\text{H}$  Broadening sources :
  - $^1\text{H}$  :  $D_{\text{HH}} > \Delta\delta > T_2$ , ~~CSA~~ (MAS), ~~D~~ $D_{\text{eH}}$  (nat. ab.)

# NMR assignments

- 2D HETCOR:

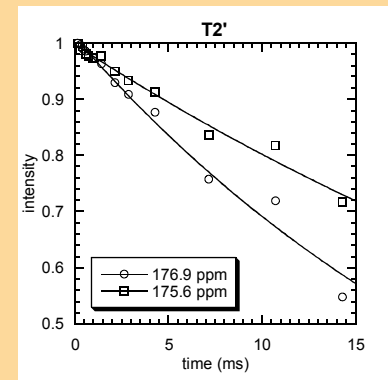
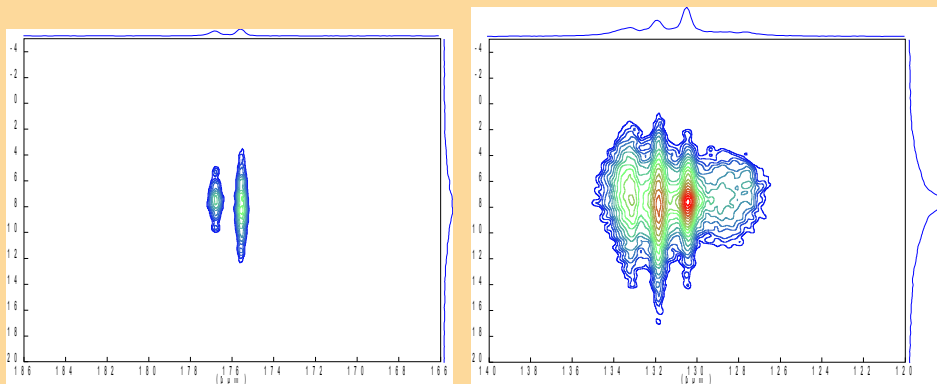
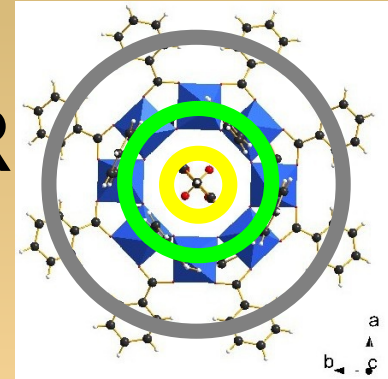
Carboxy	$T_2^*$ (ms)	$T_2'$ (ms)	$T_{IRCP}$ (ms)	$\Delta\delta(^1H)$ (ppm)	Attribution
$C_{left}$	5.3	26	0.75	5	$C_{ax}$
$C_{right}$	8.4	46	0.93	8	$C_{eq}$



- How can we explain the contradiction ?
- Different local field fluctuations between axial and equatorial carboxylates ?  $\overline{B_{loc}^2}$

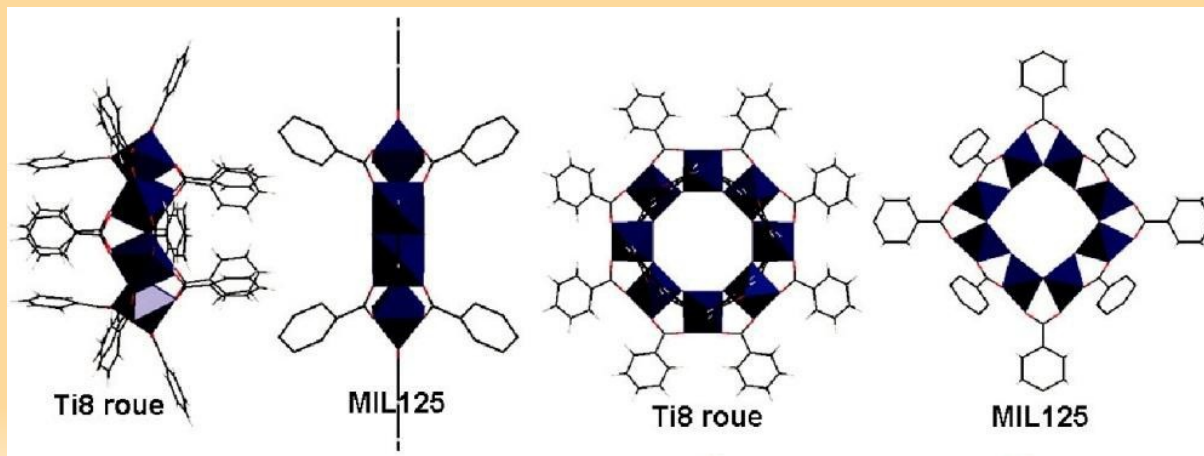
# Conclusion

- High complementarity of XRD and NMR
  - XRD → structure at low T, disorder
  - NMR →  $T_{amb}$ , exchanges
- Axial and equatorial benzoates with different dynamics
- Different local field fluctuations ?



# Further work

- $T_{1\rho}(^1\text{H})$  and  $T_{1\rho}(^{13}\text{C})$
- $T_2(^1\text{H})$  via  $^{13}\text{C}$ 
  - Check that H width is  $D_{\text{HH}}$  and not distribution
  - Spectral edition : only axial or equatorial benzoates
- MOF (collaboration with UVSQ) – PhD
- Ti8 + terephthalic acid  $\rightarrow$  MIL125





# Acknowledgements

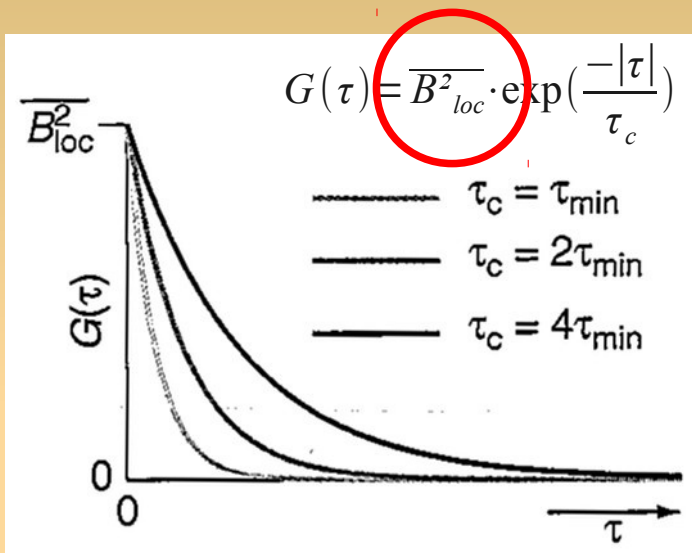


All the lab for their help and friendship

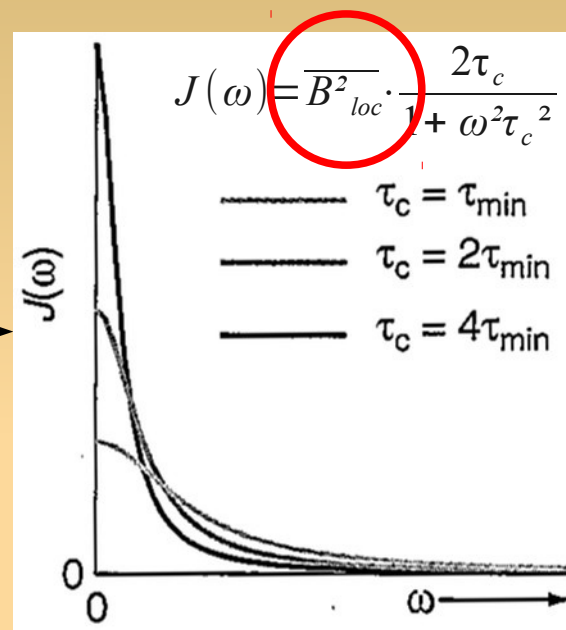
*Thank you for your attention*

# Relaxation

- Directed by local field fluctuation



FT →



$1/T_2 : J(0) = \overline{B_{loc}^2} \cdot 2\tau_c$     
 $1/T_1^{fast} : J(\omega_0) = \overline{B_{loc}^2} \cdot 2\tau_c$     
 $1/T_1^{slow} : J(\omega_0) = \overline{B_{loc}^2} \cdot \frac{2}{\omega_0^2\tau_c}$

Similar with  $T_{1\rho}$  and  $\omega_1$

175 ppm :  $B_{loc}^2 \downarrow, D_{HH} \uparrow, H \text{ width} \uparrow$   
 $B_{loc}^2 \downarrow, T_2' \uparrow$   
 $B_{loc}^2 \downarrow, T_{1\rho} \downarrow, T_{IRCP} \uparrow$

OK } Equatorial position  
 OK } (lowest thermal  
 OK } agitation by XRD)