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# Bronze Age Iron: Meteoritic or not? A Chemical Strategy.

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## Abstract

Bronze Age iron artifacts could be derived from either meteoritic (extraterrestrial) or smelted (terrestrial) iron. This unresolved question is the subject of a controversy: are some, all or none made of smelted iron? In the present paper we propose a geochemical approach, which permits us to differentiate terrestrial from extraterrestrial irons. Instead of evaluating the Ni abundance alone (or the Ni to Fe ratio) we consider the relationship between Fe, Co and Ni abundances and their ratios. The study of meteoritic irons, Bronze Age iron artifacts and ancient terrestrial irons permit us to validate this chemical approach. The major interest is that non-invasive p-XRF analyses provide reliable Fe:Co:Ni abundances, without the need to remove a sample; they can be performed in situ, in the museums where the artifacts are preserved. The few iron objects from the Bronze Age *sensu stricto* that could be analyzed are definitely made of meteoritic iron, suggesting that speculations about precocious smelting during the Bronze Age should be revised. In a Fe:Co:Ni array the trend exhibited by meteoritic irons departs unambiguously from modern irons and iron ores. The trend of Ni/Fe vs Ni/Co in different analysis points of a single object corroded to variable extents provides a robust criterion for identifying the presence of meteoritic iron. It opens the possibility of tracking when and where the first smelting operations happened, the threshold of a new era. It emphasizes the importance of analytical methods for properly studying the evolution of the use of metals and metal working technologies in our past cultures.

**Keywords:** Iron, Bronze Age, Iron Age, meteorite, iron ore

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## 1. Introduction

Paradoxically, a number of iron artifacts from the Bronze Age have been found in a variety of Old World culture areas (e.g. [Li Chung 1979](#); [Waldbaum 1980, 1999](#) and references therein; [Yalçin 1999](#); [Jean 2001](#)) with the recurrent question about the origin of the iron: extraterrestrial (meteoritic) or terrestrial (smelted)? "Bronze Age Iron" means iron that appears within Old World culture areas, prior to the advent of iron smelting on some scale in those areas.

The two possibilities are supported by a number of valuable arguments which are summarized as follows: Nickel (Ni) is the signature element for meteoritic iron. The Ni content is sometimes too low to be meteoritic which could be explained either by the use of Ni rich iron ores (Ni poor relative to meteorites) or by weathering with preferential loss of Ni. The confusion is increased by some conflicting results on the same artifacts by different methods at different times, with the difficulty that metallographic analyses which could solve this contention are impossible on such rare and fragile objects ([On Line](#)

[Supplementary Material: A1 Meteoritic vs. Terrestrial Iron: a controversy](#)).

In this work we examine a new geochemical approach involving the analysis of three elements (Fe:Co:Ni) instead of two (Fe:Ni) with the aim of differentiating between the above mentioned possibilities. This is enabled by the recent development of high performance portable XRF analyzers ([see On Line Supplementary Material : A2 Analytical methods](#)). Our argument proceeds as follows:

- Consider a data set of meteoritic irons, including oxidized specimens.
- Analyze irons of diverse ages: Bronze Age, Bronze to Iron Age transition and Iron Age, and take benefit of recent high quality analyses ([see On Line Supplementary Material : A3. Samples](#)).
- Iron ore compositions are also considered. Lateritic alteration products derived from peridotitic rocks are common from Croatia to Greece, Turkey, Iran, Cyprus ... These may be valuable iron ores and contain significant amounts of Co and Ni unlike sedimentary iron ores which are more common in western Europe

(see [On Line Supplementary Material : A3.6. Iron ores](#)). Since Brun in [Schaeffer \(1939\)](#) suggested that the 13<sup>th</sup> century BCE Ugarit (Syria) iron axe could be derived from iron sulfide ore (pyrrhotite  $Fe_{1-x}S$ ;  $x = 0$  to 0.2), such material will be considered in this study using data from the literature ([Bamba, 1981](#)).

## 2. Results

**2.1 Iron meteorites:** The analytical results for polished surfaces of iron meteorites, outer oxidized surfaces of iron meteorites ([OLSM table A1](#)) and literature compositions for different classes of iron meteorites are presented in [figure 1](#), in addition to a compilation of iron meteorite compositions from the Meteoritical Bulletin database (<http://www.lpi.usra.edu/meteor/>): 176 meteorites classified from 1986 to 2016 ([OLSM, table A2](#)). The variations within this population are best illustrated in a Ni/Fe vs Ni/Co diagram ([Fig. 1](#)).

Fresh meteorites exhibit a Ni/Fe range from 0.058 to 0.40 (average 0.102) whereas oxidized/weathered surfaces of iron meteorites exhibit lower values down to 0.009. We checked that the variations observed are not due to the variability within one single

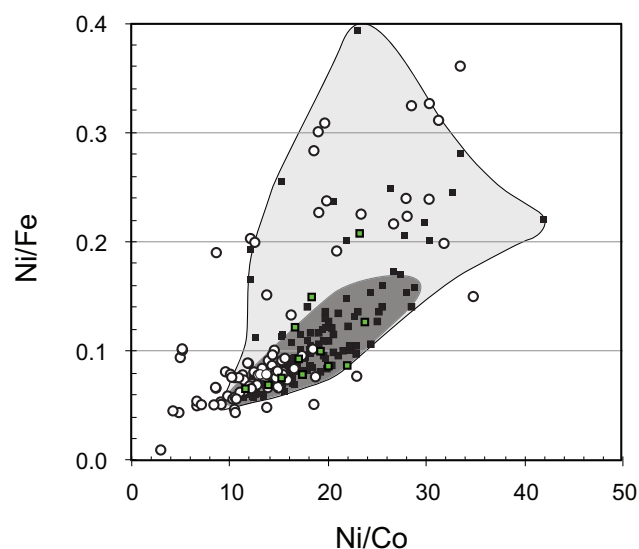


Fig. 1. Ni/Fe vs Ni/Co in iron meteorites. Black squares and gray area are from literature data for fresh iron meteorites. Green squares: average compositions of the major iron meteorite groups ([Mittlefehldt et al. 1998](#)). White circles: p-XRF analyses of both polished and oxidized outer surface (this work). P-XRF measurements on fresh surfaces are similar to literature data. On the average, oxidized compositions extend to slightly lower Ni/Co and Ni/Fe ratios. 80% of the data for fresh meteorites are enclosed in the high-density field (dark gray). (<http://www.lpi.usra.edu/meteor/>)

meteorite as illustrated by the results for Morasko meteorite, which we could analyze in 31 points, hence assessing the internal variability of its Co and Ni content, and the different compositions of fresh

and weathered surface. The results obtained for fresh metal surfaces is within the range of those reported in the literature (NAA analysis of large samples, [Pilski et al. 2013](#)) with restricted standard deviations smaller than the variability range observed for the whole data set ([On line supplementary Material; A3.2 Iron meteorites](#))

Surfaces oxidized during the atmospheric flight exhibit similar ratios. The case is different for finds, which have been weathered. This indicates that during weathering, a surficial layer is impoverished in Ni relative to Fe whence the Ni/Fe ratio cannot be used as a reliable indicator of the meteorite type for weathered samples. A comparable variation is observed for the Ni/Co ratio. Notice that the chemical properties relative to weathering are in the order  $Fe > Co > Ni$ , whence the positive correlation between Ni/Fe and Ni/Co as will be better illustrated in the following subsection for the case of the Ugarit axe, analyzed in different spot analyses ([Fig. 2](#)). It follows that the Ni/Fe ratio alone cannot be used as an indicator of the source of iron.

### 2.2 Bronze Age archaeological artifacts:

Analytical results are listed in the [on line supplementary material, table A4](#) with references to where a description can be found. Additional information can be found in OLSM section A4.1.

#### Ugarit axe (Syria 1400 BCE).

We performed ten spot analyses at different places on both sides of the blade (see [Jambon et al. 2017](#)). The Ni concentrations of 1.7 up to 7.6% Ni (calculated on an oxygen free basis) document nicely the effect of weathering. The high Ni contents are undoubtedly the signature of meteoritic iron whereas the lowest values correspond to pervasively oxidized spots. These differences probably result from rust flakes detachment from the surface. The variations of Ni/Co and Ni/Fe correlate fairly well with Ni content, which can be viewed as an index of weathering, as displayed in [figure 2](#). The Ni/Fe and Ni/Co ratios plot on the trend defined previously for iron meteorites, [figure 3](#), which is interpreted as corresponding to different degrees of weathering. Our results for the Ugarit axe show both higher and lower Fe/Ni ratios compared to the analysis reported in [Schaeffer \(1939\)](#). The sampling made by Schaeffer being undocumented, we assume that it was a surface chip, an average of more and less oxidized material, but no obvious mark of sampling is presently visible on the axe blade.

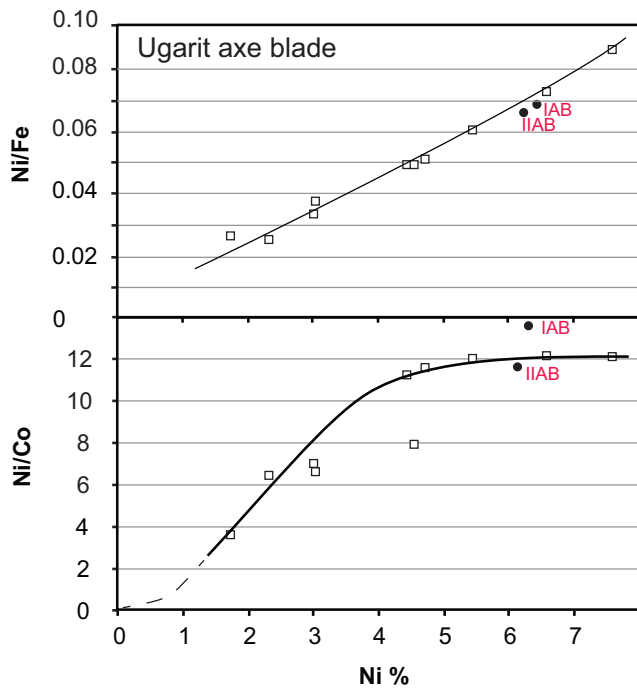


Fig 2. Plot of Ni/Fe and Ni/Co against Ni abundance for Ugarit Axe. The steady variation of the Ni/Fe ratio against Ni indicates that Ni is preferentially leached during weathering whereas, Fe oxidized to  $Fe^{3+}$  is not. The Ni/Co ratio remains constant for mild weathering and then decreases when part of Co is oxidized to  $Co^{3+}$ . Average composition of IAB and IIAB meteorites are plotted for reference.

#### Other Bronze Age artifacts.

The results for the Gerzeh beads (Egypt; 3200 BCE), Umm el Marra pendant (Syria; 2300 BCE), Tut's dagger, bracelet and headrest (Egypt; 1350 BCE), Shang Dynasty axes (China; 1400 BCE) and Alaca Höyük dagger, (Turkey; 2500 BCE) are commented in detail in the [On line supplementary Material; A4.1](#). The only specimen for which one single analysis is available is Tut's dagger the meteoritic origin of which is beyond any doubt (Comelli et al. 2016; Ströbele et al. 2016). For all other specimens we have 2 to 9 analytical points. Both Co and Ni are highly variable (0.8 to 8.5 % Ni) but in the Ni/Fe vs. Ni/Co plot, figure 3, the data points fall nicely on the same trend as Ugarit axe data. For some of the artifacts the highest Ni concentrations are in the range of meteoritic values (e.g. Gerzeh beads, Alaca Höyük dagger, Tut's jewelry) while some data points fall below 3.5 % Ni. Taken at face value and if the conservative threshold of 5 % Ni were considered (e.g. Yalçın, 1999), one would infer the presence of both terrestrial and meteoritic iron in one and the same sample!

In the present interpretation however, replicate analyses including Co analyzes, indicate that all artifacts tested are made of meteoritic iron. For the other artifacts (e.g. Umm el Marra pendant, Shang axe) the low Ni concentrations cannot

therefore be considered as a proof of being terrestrial. The data points falling on the same correlation as meteoritic artifacts strongly suggests that they are meteoritic as well. In other words, the

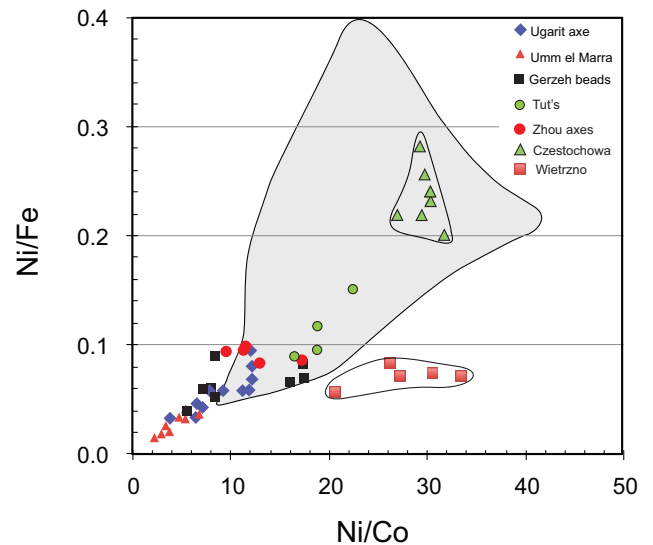


Fig 3. Same as figure 1 for Bronze Age and Iron Age iron artifacts. The gray area is taken as a reference from figure 1. Most compositions are clearly displaced to lower Ni/Co and Ni/Fe values. Well documented specimens like Umm el Marra pendant and Ugarit axe exhibit a clear positive correlation corresponding to variable extents of weathering, undistinguishable from the bulk trend. Tut's dagger and Gerzeh beads, which have been demonstrated to be of meteoritic origin fall on the same correlation. This work, except Tut's, Alaça Höyük and Zhou axes data taken from the literature. Notice the apparent out-of-place field for Wietrzno axe.

Ni/Fe ratio alone is not appropriate to conclude whether an artifact is made of meteoritic iron or not, or may be the threshold should be dramatically decreased. This indicates that the weathered surface exhibits variable Ni/Fe and Ni/Co. The trend for artifacts is similar to that observed for weathered meteorites with lower Ni/Fe and Ni/Co on the average. This is not surprising since the meteorites analyzed were only marginally weathered. To summarize, the range in both Ni/Fe and Ni/Co for meteorites and iron artifacts overlap, with a significant variability in Ni/Fe and possibly low Ni contents (down to less than 1%). According to the present observations, none of the Ni bearing iron artifacts, with low Ni content, was proved to be made of terrestrial iron.

Our preliminary conclusions may therefore read :

1) All results, the present p-XRF analyses and literature results as well, illustrate than one single object may exhibit various Ni contents depending on the analytical spot. This is not due to primordial metal heterogeneity but rather to a variable extent of weathering (corrosion). Most of us

will agree that contents exceeding 5% Ni should be considered as meteoritic; the lower Ni contents from the same objects, especially when analyzed with the same technique, must be considered meteoritic as well.

2) When the Co results are considered, even though the Ni data alone could appear inconclusive, the Fe:Co:Ni correlation falls on one and the same trend with iron meteorites. Weathering is an important factor of variation in the Ni content. One way of accounting for this effect is to consider the Fe:Ni:Co correlations.

3) No nickel-poor iron from the Bronze age can be proved not to be meteoritic.

4) No nickel-rich iron object of the Bronze Age consists of smelted iron. We therefore may ask whether one single iron object from this time is not made of meteoritic iron, since the low level or even absence of Ni can no more be considered as a proof for the smelting origin of iron.

5) If our interpretation that the three iron objects from Tut's treasure are made of three different iron meteorites, is correct, this suggests an active search about iron meteorites.

### 2.3. Transition artifacts

Results are listed in OLSM table A5. Complementary information is given in the OLSM A5.1.

**Le Louvre specimens.** For both the halberd and the adze (Luristan, ca. 1300- 650 BCE), the Ni fall below the detection limit while Co is not (detection limit is 0.02 %) which indicates a Co/Ni ratio > 10 and a Ni/Fe ratio < 0.005. Both artifacts are without ambiguity made of terrestrial iron. Casting a bronze socket indicates that the ability at smithing was not well mastered for these objects, which may be considered as benchmarks in bloomery iron smithing.

**Zhou dynasty axes.** The objects from early Zhou (about 1000 BCE) were first investigated by Foshag (1950) (in [Gettens et al., 1971](#)) who detected no nickel and concluded that the iron was not meteoritic. Gettens et al. however made insightful analyses, including microprobe analyses and metallographic examinations. They showed unambiguously that both artifacts are made from meteoritic iron. What interests us more are the chemical analyses of both metal and oxidation products, which fall within the field of iron meteorites ([Fig. 3 and table A4](#)).

**Neuchatel artifacts.** Three needles, two nails and one hook were analyzed. Despite their chemical composition, the needle typology is identical to and typical of Bronze Age needles of that area. In other words they were probably produced in the same region, between 950 and 850 BCE according to dendrochronological dating ([Rychner 1979; Rychner-Faraggi 1993](#)) in the Late Bronze Age of western Switzerland. Their composition is typical

of smelted iron which indicates that late Bronze Age in Europe is not equivalent (technologically speaking) to Near Eastern final Bronze Age. We know that iron was smelted further east at the same time (possibly as early as 1200 BCE in SW Asia) and iron ingots may have been imported as precious metal to make jewelry.

### 2.4 Iron age artifacts.

The results are listed in the [On line supplementary material; table A5.2 with complementary informations in section A4.3.](#)

#### 2.4.1. Present study

**The Marsal ingots** (NE France) are quite fresh. From the archaeological context they are dated at about 700±100 BCE. Their Ni and Co contents are exceedingly low, mostly below the detection limit of our equipment. When Co and Ni are detectable the Co/Ni ratio is observed to exceed unity in strong contrast with meteoritic iron (<0.2) and clearly outside of the trend defined by meteoritic material. In addition the amount of Cu is sometimes significant (up to 0.5% in one of six ingots) in contrast to the composition of meteoritic metal. The different chemical compositions among the various ingots analyzed suggest several provenances that cannot be specified for the moment. For such low Ni values it is more convenient to show the data in an Fe/Co vs. Ni plot with a log scale. The field for Marsal ingots falls unambiguously apart from the meteoritic compositions ([Fig. 4](#)).

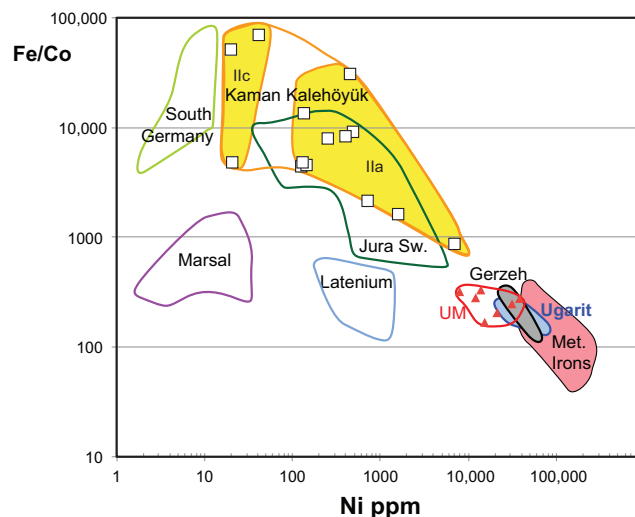


Fig. 4. Plot of Fe/Co against Ni for Bronze Age and Iron Age artifacts. The log scale permits to illustrate the variations for low Ni irons (e.g. smelted irons from sedimentary ores) and high Ni irons (e.g. meteoritic irons). At 1% Ni (10,000 ppm) the Fe/Co ratio permits to distinguish samples with otherwise similar Ni abundance. UM= Umm el Marra pendant (Syria) ([Schwartz et al.](#)). Latenium= Neuchatel ([Rychner, 1987](#)). Jura Sw(itzerland) ([Eschenlohr et al. 2007](#)). White squares are data for Kaman KaleHöyük levels IIA and IIc (Turkey) from ([Akanuma, 2006](#)).



## Irons from Poland

**Czestochowa-Rakowa bracelets** are also dated from Hallstatt C (800-600 BCE). Our Ni results are in agreement with those of [Kotowiecki \(2004\)](#) 18.5 and 12.5 % Ni for CrZ1 and CrZ2 respectively, to be compared to our values of 15.5-18.4 and 18.1-21.2 % Ni, an unusually high Ni content. [Piaskowski, \(1982\)](#) gives comparable results for Crz1 of 18.25% Ni, 0.58% Co and 12.4 % Ni for CrZ2. These compositions fall nicely in the field of iron meteorites. The suggestion of previous workers (e.g. [Photos, 1989](#)) that such high Ni metal could be produced from terrestrial ores is not supported in the present case by the chemical composition.

**Wietrzno Axe** (Hallstatt unspecified) is a special case, which illustrates the potential of the method. The data points in figure 3 plot below the field of iron meteorites despite a high Ni content (5.5 to 7.6 % Ni) typical of iron meteorites indicating that for its Ni/Co ratio, the Ni content should be higher ([Fig. 3](#)). One simple explanation is that its metal is a mixture of terrestrial iron (low Co, low Ni) with meteoritic iron similar to that used for the bracelets. Doing so the Ni/Co ratio remains unchanged (the contribution from terrestrial iron is negligible) whereas the Ni/Fe ratio is significantly decreased. This explanation is substantiated by the observation that the blade is made of five layers (two high in Ni and three devoid of Ni; [Piaskowski, 1982](#)), which cannot be resolved with the p-XRF analyzer. This unexpected result suggests that the similarity between meteoritic iron and smelted iron was recognized and that the use of meteoritic iron was still a viable practice. Because of the rarity of iron meteorites with such a high Ni content exceeding 15 %, it is likely that the same meteorite was used for the bracelets and the axe.

### 2.4.2. Literature data

We selected examples where archaeological artifacts were analyzed for Co and Ni on sufficiently well preserved samples (some metal is preserved): Fragments from Kaman Kalehöyük (Turkey) ([Akanuma, 2006](#)), Western European irons of Manching (South Germany) ([Schwab et al. 2006](#)) and two from Renningen and one from Grösseltal (South Germany) ([Brauns et al. 2013](#)). More information can be found in the [On line supplementary material; section A4.2](#).

The results are plotted in [figure 4](#). All data fields are presented in [figure 4](#) and, as expected fall quite far from the meteoritic field.

Finally the more recent objects from Devélier-Courtetelle (Jura, Switzerland), a set of 67 objects, analyzed by [Eschenlohr et al., \(2007\)](#) are low in Ni (40-4000 ppm; average 1200 ppm) with a Co/Ni ratio ranging 0.02 to 3 (average 0.43).

## 2.5. Ni bearing iron ores

We showed that smelted and meteoritic irons can be distinguished from their Fe:Co:Ni composition, still according to previous suggestions we must investigate whether lateritic iron ores which reportedly contain some nickel, could produce iron distinguishable from meteoritic iron. As noticed by [Pryce and Natapintu \(2009\)](#) laterites cover a wide domain of compositions, but the ones we are concerned with are those developed on peridotites. These are rocks with dominant (>60%) olivine of formula  $(\text{Mg,Fe})_2\text{SiO}_4$ . These contain significant amount of iron oxide (about 10% on the average) and little aluminum oxide (less than 3%).

**Lateritic ores:** The Fe, Co and Ni contents were measured across alteration profiles, from the fresh mother rocks to the iron oxide cap. The iron content increases from bottom to top, while the Ni first increases then decreases at the top where the iron ore is of the best quality (low silica content). The Ni/Fe and Ni/Co ratios vary along the profile and the observed correlation passes through the starting composition of fresh peridotite, which is also the terrestrial mantle composition ([Fig. 5](#)). In order to avoid confusion, we selected the data for potential iron ores, that is rocks with less than 20% silica and more than 50 % Fe. The results taken from the literature are presented in [figure 5 and 6](#). In [figure 5](#), the correlation lies below the field of iron meteorites and it appears that for low Ni contents there might be some ambiguity between weathered meteoritic and lateritic compositions. Some of the scatter is due to the small size of analyzed samples (on the order of hundred mg) and we expect that the charge of a

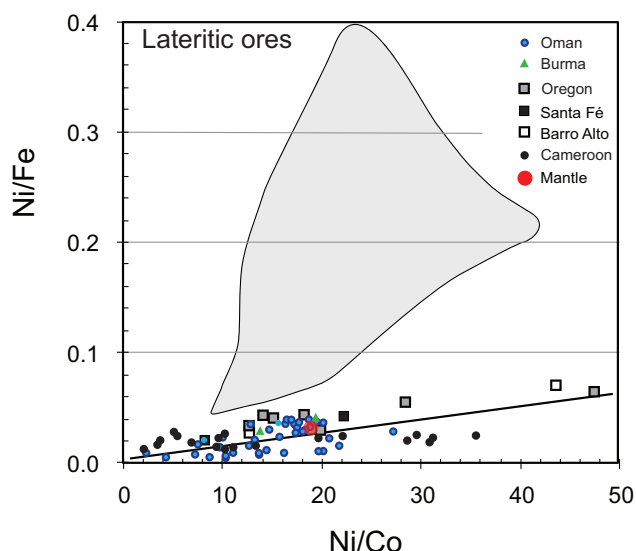


Fig. 5. Same as fig. 1 for diverse lateritic iron ores. The trend for terrestrial iron is clearly different from that of extraterrestrial material. Barro Alto and Santa Fé (Brazil) are from [Trescases and Oliveira, \(1981\)](#), Cameroon from [Yongue-Fouateu et al., \(2006\)](#), Burma from [Schellmann, \(1989\)](#), Oregon from [Hotz, \(1964\)](#) and Oman from [Al Kirbash, \(2015\)](#).

Terrestrial mantle composition falls on the same trend.

smelter (several kg) would exhibit less dispersion due to the averaging effect. The same is observed in figure 6. Some overlap is observed in the field representing the Umm el Marra iron pendant, which was shown to be the most weathered artifact.

More interestingly, the data for Kaman Kalehöyük, stratum II a (Fig 4-6), overlap with the lateritic field but extend to significantly lower Ni contents. It is well known that such lateritic ores are commonplace in Anatolia (see e.g. Pigott, 1989).

These data however are for lateritic ores not metal. There is an additional possibility, which we did not consider yet: some fractionation might occur during the reduction process, thus changing the Ni/Fe and/or Ni/Co ratios. One obvious question then is whether some fractionation occurs between lateritic ores and the metal (Photos, 1987). This point will be discussed below.

**Pyrrhotite:** Some pyrrhotites (FeS) may contain significant concentrations of Ni. This is illustrated by the composition of pyrrhotite of the Oshirabetsu Mine, Hokkaido (Bamba, 1981 ; *On line supplementary material; table A3*) ranging from 0.02 to 1.1 % Ni. Their Ni/Fe (<0.02) and Ni/Co (0.5-39) plot along a trend below the terrestrial mantle correlation defined by laterites. They are significantly different from all meteoritic values and cannot explain the composition of Bronze Age irons.

### 3. Discussion

The starting point of this study was the controversy about the ultimate source of iron: meteoritic or smelted. According to previous results, the abundance of Ni in metal or the Ni/Fe ratio was considered a strong indicator of the origin of iron. We

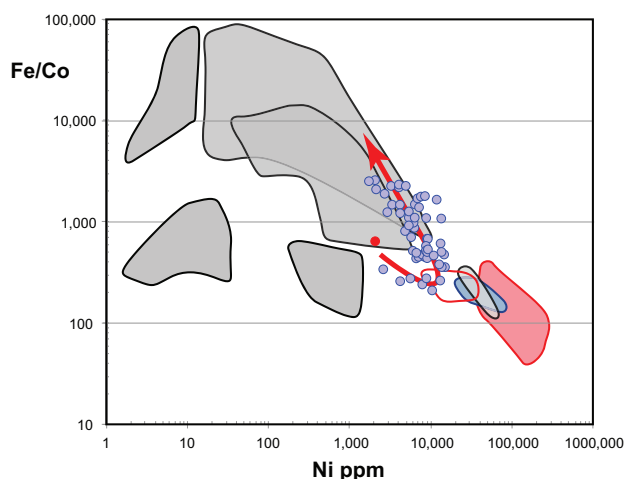


Fig. 6. Same as figure 4 for lateritic samples (dots; same set of data points as in fig. 5). Notice the weathering path (red arrow) from the mantle value (red dot) with Ni increasing and then decreasing; the opposite being observed for Fe/Co. The fields for iron meteorites and iron artifacts from figure 4 are shown for comparison.

confirm that when the Ni is low, especially for weathered specimens, it is not sufficient as a criterion, whence the search for a more robust tracer of origin. According to the above results, the Fe/Co/Ni composition can provide the required information, which can be obtained using p-XRF. Analyzing several spots is highly recommended: on one single artifact, variable Ni is measured due to the extent of weathering, but the trend in a Ni/Fe vs Ni/Co appears to be a robust information as it permits to distinguish terrestrial from meteoritic iron.

**The effect of weathering.** The above results show that weathering of meteoritic iron affects the Ni content and the Ni/Fe ratio (Fig. 2-3). However, the Ni/Co and Ni/Fe variations still correlate. The low Ni/Fe ratio (or the Ni abundance) of terrestrial irons could be similar to that of some weathered meteoritic irons but at the same time their Ni/Co ratio does not fall on the correlation exhibited by meteoritic irons. The effect of weathering was ignored in most previous investigations but is now well established.

**The effect of smelting.** Photos (1989) found metal prills with comparatively high Ni/Fe ratios in some iron slags from Petres (N. Greece). It is noteworthy that no iron artifacts with high Ni content were ever found in the same context. The slags in question were high in iron oxide (wüstite) and contained small amounts of metal prills. She concluded that this could explain the abundance of Ni in all irons from the Bronze Age without the need of the extraterrestrial iron hypothesis. I cannot share this view and claim that the high Ni content of iron prills in slag ensues anytime, when the ore reduction fails. Ni is more easily reduced than iron, therefore if the smelting conditions were slightly too oxidizing, it is quite possible that nearly all Ni was reduced while only a small fraction of iron was. It can be shown that the fraction of Ni in the metal is a measure of the oxygen partial pressure in the furnace. The observed heterogeneity of Ni in the iron is a good indicator of out-of-control oxidizing conditions. Then a very small quantity of metal is obtained with a high Ni content (tens of %) if the starting ore contained some Ni. This is what the experiments of Photos (1989) yielded and possibly the case in Petres discarded slag as well. Extracting metal prills from a large quantity of slag would be a very hard task for little reward and most probably therefore the slag was discarded. The recovery of metal by a second smelting operation seemed questionable from an economical point of view. For this second reduction stage to be efficient, crushing the slag to permit exchange with carbon in the furnace would be required. This operation would have been too demanding when compared with using fresh ore. The experiments of Photos confirm our view. Some of the metal prills she analyzed after her smelting experiments contain up to 67 % nickel.

When starting from an ore with 1% NiO (0.7 % Ni) and 72% Fe<sub>2</sub>O<sub>3</sub> (50 % Fe), this suggests that only 0.3 % Fe was reduced (for a metal with 20 % Ni, 2.8 % of Fe), a very poor yield! In addition the composition of chrome-spinel in the slag, indicates that 17% of its Fe is actually Fe<sup>3+</sup> indicating that oxygen fugacity was close to the magnetite-wüstite buffer, that is far above the reducing conditions necessary to obtain iron metal.

Another important question is: are such Ni rich ores really common? I would say, in principle not. As already mentioned above, the best iron ores of the lateritic type contain little nickel ( see also the discussion by [Pryce and Natapintu \(2009\)](#) on the quality of laterite in order to be a qualified iron ore) . Those high in nickel (actually nickel ores as reported in [figures 5-6](#)) contain also quite large amounts of silica, which makes them poor iron ores; their color is orange yellowish and they look quite different from the true iron ores having a dark rusty color. This is because during the final stages of lateritization, silica and nickel (Ni<sup>2+</sup>) are leached away, whereas Fe<sup>3+</sup> remains immobile as iron oxides. It is important to notice that no smelted iron object of the Iron Age has been reported with significant amounts of Ni. In particular, the irons of Kaman Kalehöyük fall in the range 0.7 to 0.01 % Ni ([Akanuma et al. 2006](#)), whereas numerous lateritic ores are present in Anatolia. According to their composition ([Fig.4-6](#)) we think that Kaman Kalehöyük artifacts are good candidates for iron derived from lateritic iron ores. Such irons have Fe/Co higher than meteoritic irons and also higher than the lowest Fe/Co ratio of nickel rich ores, but the overlap is significant.

#### 4. Conclusions

We conclude that it appears now important to measure correctly Bronze Age irons for their Fe:Co:Ni abundances, in order to determine whether or not, any specimen from that time is made from terrestrial iron. Replicate analyses are necessary since weathering leads to variably depleted Ni contents and a Ni content below 1% alone is no proof of origin; for Ni in excess of 1% the Ni:Co:Fe correlation will be conclusive. Scraps of rust are strongly biased samples, depleted in Ni and should be avoided, whereas oxidized artifacts are acceptable.

The present results complementing high quality analyzes from the literature suggest that (most or) all irons from the Bronze Age are derived from meteoritic iron, until some transition period, which occurred supposedly close to about 1200 BC. The next step will be to determine where and when terrestrial iron smelting appeared for the first time.

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## On Line Supplementary material

### A1. Meteoritic vs. Terrestrial Iron: a controversy.

From the beginning of the twentieth century, starting with Dörpfeld's report in 1902 of nickel-rich iron in a mace head at Troy, and followed shortly by the find of iron beads in two tombs from the early Bronze Age at Gerzeh, (Egypt) in 1911, it was suggested that the iron was extraterrestrial as the metal contained a significant amount of nickel (Ni), the signature of meteoritic iron (e.g. Götze, 1902; Petrie et al. 1912; Desch, 1929; Bjorkman, 1973). This disturbing idea was challenged very early, on the basis of two strong arguments: a number of irons contained too low amounts of Ni compared to meteoritic metal, and some iron ores were shown to contain significant amounts of Ni. Conversely, a number of iron artifacts from the Bronze Age contain some Ni whereas none from the Iron Age are reported as Ni-bearing (in excess of 1%) except for particular instances where it was shown that the ultimate iron source was meteoritic (e.g. tools from Greenland, Buchwald, 1975, 2005). In a number of instances, duplicate analyses proved to be contradictory. For instance the Ni content analyzed in Gerzeh beads was found negligible by Gowland and Bannister (1927) and El Gayar (1995), in contradiction with the results of Desch (1929), Wainwright (1932), Johnson et al. (2013) and Rehren et al. (2013). In order to be as little invasive as possible, surface chips or scraps were sometimes retrieved for analytical purpose, with the consequence that significantly weathered material was analyzed, which as we now know could have lost most of its Ni (Li Chung 1979; Craddock, 1995). Photos (1989) reported on archaeological slags containing Ni-rich metal prills, from Petres (N-E Greece) providing support to the idea of Ni bearing terrestrial iron but unfortunately no iron object with such a composition was ever found. The experimental production of Ni rich metal seemed to support this argument. As a consequence of these contradictory observations, some confusion followed: a number of authors decided that, to be considered extraterrestrial, the metal should contain more than a minimal amount of Ni (e.g. 5 %) otherwise it would be considered as terrestrial (see e.g. Waldbaum 1980-1999; Photos, 1989; Yalcin 1999; Merkel and Barrett, 2000), which we consider as a totally irrational conclusion: the analytical results of Gowland and Bannister (1927) and El Gayar (1995) already mentioned, show that meteoritic iron, when corroded, can be totally deprived from its original Ni (see also Craddock 1995). Tutankhamen's dagger as analyzed by Helmi and Barakat (1995), with 2.8% Nickel was considered terrestrial while the more recent analyses of Comelli et al. (2016) and Ströbele et al. (2016) with 10.8 and 12.9 % Ni respectively show that it is undoubtedly meteoritic. We conclude that analytical difficulties are a serious possibility and interpretation of the analytical results must be reconsidered. We

must admit however that most of the time, conclusive arguments are missing partly because a metallurgical study, which could prove an extraterrestrial origin, is missing, since a fresh sample cannot be taken for analysis.

Another difficulty is in dating archaeological artifacts. Some irons, reputedly from the Bronze age, are actually more recent. A good example is provided by the irons of Timna (Gale et al. 1990), which were dated from the 14th-12th century BCE (Rothenberg, 1988). More recent dating indicates that they are Iron Age (Ben Yosef et al. 2012).

### An how to overcome it...

Recent analytical improvements permit non-invasive textural studies (see e.g. Johnson et al. 2013; Rehren et al. 2013) and provide an unambiguous answer but the analytical procedures are arduous and require the transfer of the artifacts to an analytical facility, which is time and money consuming and not always safe for fragile artifacts. The recent development of portable XRF equipment (p-XRF) permits performing non-destructive, on-site chemical analyses, which are of particular interest whenever the chemical analyses permit the discrimination between the two types of iron (see e.g. Nakai et al., 2008). More generally, the trace element abundances in meteoritic iron are expected to differ significantly from iron derived from iron ore and this offers a powerful tool, provided that the artifacts are not pervasively weathered. The p-XRF analysis is quite efficient but the analysis of a weathered surface limits the interpretation as far as trace elements are concerned. In this paper we develop a strategy illustrated on a few examples, which permits the recognition of extraterrestrial iron conclusively, whenever the Ni content exceeds 1 wt. %. We analyzed meteoritic samples, Bronze Age iron artifacts some of which have been proved to be meteoritic, Iron Age artifacts and considered literature data for additional artifacts and for Ni bearing ores. Critical comparison of the different data sets permits us to clearly establish a distinction between the two possible iron sources.

### A2. Analytical methods

#### A2.1. P-XRF

For all artifacts we used a p-XRF analyzer (Niton X3t 900) using a source operated at 50 kV. A set of filters operated sequentially permits us to explore four windows in the spectrum from 1.25 keV (Mg, K line) to 13.6 keV (U, K line). In the present case only two were used in order to analyze light elements (from Al to Ca) and transition elements from Ti to Cu (mostly). Analyses were performed in ambient air for two minutes which for transition elements is acceptable as long as the analyzer is in close contact with the surface to be analyzed, the surface is smooth and elements exhibit X-ray lines close to one another,

which is the case for Fe, Co and Ni. Notice that under these conditions, elements lighter than Al could not be analyzed (e.g. oxygen), even though we obtain a different result for a metal and its oxide. The counting duration of 2 min. was selected as the best compromise in order to decrease the limit of detection for trace elements and permit a number of replicate analyses within a reasonable time. The analyzer was calibrated using synthetic references (Fe: Ni alloys containing 100: 90: 83: 70 wt.% Fe) and polished iron meteorites of known compositions. A special investigation on Morasko (IAB iron meteorite) permitted us to evaluate the reproducibility in analyzing natural fresh samples and accuracy by comparison with NAA (neutron activation analysis) (Pilski et al. 2013). We were particularly concerned by Co analysis. The resolution of the EDS (Energy Dispersive Spectrometer) is 0.180 keV at 5.9 keV (Mn K $\alpha$  line). Co K $\alpha$  line (6.92 keV) and Fe K $\beta$  line (7.07 keV) on the one hand and Co K $\beta$  line (7.65 keV) and Ni K $\alpha$  line (7.47 keV) on the other interfere with one another, which is a significant difficulty when the abundance of Co is low compared to both Fe and Ni, as is the case in meteorites with Fe:Co:Ni mass fractions of 90:0.5:8 on average. Obtaining the correct value for Co requires subtraction of Ni and Fe contributions from the whole spectrum, with a routine integrated in the analyzer software. The results compared to reference samples of known composition indicate that the precision on Co under these conditions is on the order of 300 ppm, i.e. 6% relative. The subtraction of Fe and Ni peaks from the whole spectrum gives an idea of the effectiveness of the procedure (Fig. A1). The reference Fe-Ni alloys (0, 9.6, 17.6 and 30.1 % Ni) are Co free (below detection level) as checked with EMPA analysis. Their spectra permit to obtain the Co-free, Fe and Ni peaks used to compute the results presented in figure A1.

### A2.2. Electron microprobe analyses.

In order to check for the composition of meteorite specimens, polished sections of iron meteorites: Cape of Good Hope, Coahuila, Copiapo, Juromenha were provided by MNHN Paris and analyzed with an Electron microprobe, CAMECA SX-100 equipped with four WDS spectrometers, at UPMC in Paris. The analytical conditions were: 15 kV accelerating voltage, 10 nA beam current for Fe and Ni and 300 nA for Co, focused beam, counting time 10 s at peak position and 10 s for background. The reference Fe-Ni alloys (0, 9.6, 17.6 and 30.1 % Ni) were analyzed for calibration of Fe and Ni and also to check the p-XRF Co concentrations. Under these conditions, the detection limits are 0.1 % (Fe), 0.14 % (Ni) and 30 ppm (Co). The  $2\sigma$  error on Co is 20 ppm.

### A3. Samples:

In this paper we investigate the different chemical characteristics of meteoritic and smelted irons. This is not as trivial as it seems because of the effects of

weathering. Archaeological artifacts are quite often severely corroded and their chemical composition

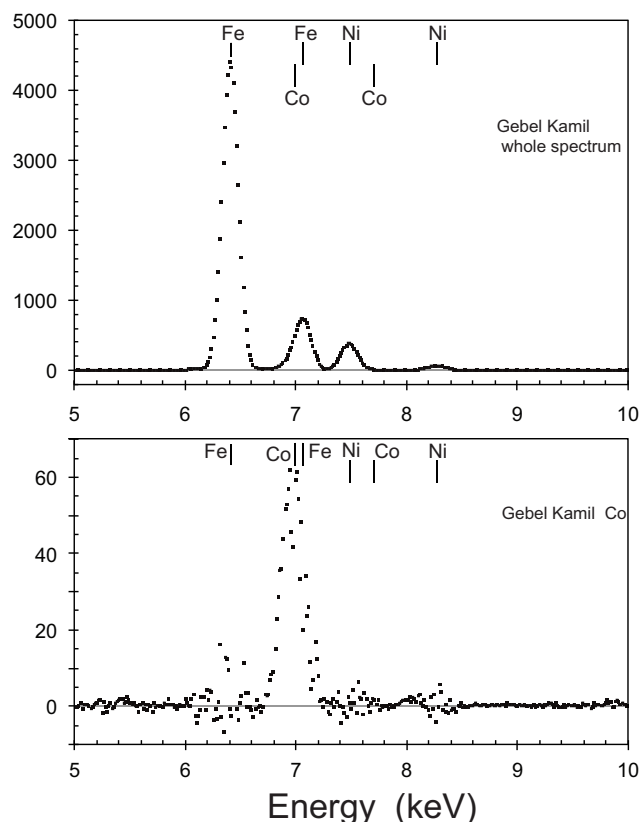


Figure A1. p-XRF spectra for Gebel Kamil iron meteorite. Top: whole spectrum. Bottom: same spectrum after subtraction of iron and nickel peaks (notice the different scales). The Co peak (K $\alpha$  line) is clearly observed, affected by a significant background noise due to the presence of abundant Fe and Ni. The K $\beta$  line of Co (7.65 keV) is negligible due to absorption by Fe (K edge at 7.11 keV).

weathering. Archaeological artifacts are quite often severely corroded and their chemical composition may be affected especially at their surface. It is of utmost importance to assess the difference between weathered meteoritic iron, depleted in Ni, and smelted iron with low Ni content on samples of well-defined origin, terrestrial and meteoritic, from the Bronze and Iron Age.

### A3.1. Synthetic reference samples (Table A1)

As already mentioned, such samples were necessary to check that the analytical procedures are correct, especially the analysis of Co. A set of synthetic Fe:Ni alloys in the range 100:0 to 70:30 free of Co and any other contaminants were analyzed several times to check for the accuracy of the p-XRF analyzer. It was important to consider reference samples without Co because of possible interference between Co, Ni and Fe peaks. We analyzed the same polished samples with our p-XRF and also using EMPA.

### **A3.2. Iron meteorites.** (Table A1)

In order to validate the comparison between artifacts and extraterrestrial irons we analyzed 15 different iron meteorites of various types with p-XRF. Agoudal, Campo del Cielo, Canyon Diablo, Cape of Good Hope, Chinga, Copiapo, Gebel Kamil, Juromenha, Morasko, Mundrabilla, Northwest Africa (NWA) 854 (Ziz), NWA 859 (Taza), NWA 5549, Sikhote Alin, and UAI an Algerian unnamed iron ataxite. Samples were from Museum National d'Histoire Naturelle (MNHN) Paris, Moroccan Import (Asnières), Poznan University for Morasko or the private collection of the author. The overall Ni contents of polished samples range from 5 to 24 % in agreement with literature values. For most samples, we analyzed both a fresh polished surface and the outer oxidized, more or less weathered surface. At the university of Poznan we had access to about thirty different pieces of Morasko on which we performed 31 analyses of the metal fraction. This permitted us to assess the internal variability of about 10 % for Co and Ni, and the different compositions of fresh and weathered surfaces. The Morasko fall date is about 5000 years B.P and the finds of the different pieces are between 1914 and the present, which gives an idea of the time during which the buried samples were subject to terrestrial weathering. The pieces analyzed however had had their surface layers, mixtures of clay and iron oxide cleaned off. In addition the results obtained on freshly polished surfaces (26 measurements) permitted us to calculate an average composition in agreement with that reported in the literature within 10 % (Pilski et al. 2013).

### **A3.3. Bronze Age archaeological artifacts.**

For the purpose of this study we considered a set of high quality analyses from the literature, which we complemented with our own p-XRF analyses.

We selected a set of accessible objects from various Bronze Age levels. 1) Gerzeh (Egypt) beads from the Petrie Museum (UCL, London) (Wainwright 1912) studied recently using various methods, dated at 3200 BCE and shown to be meteoritic: see Rehren et al. (2013); ref numbers UC10738, UC 10739, UC10740. Actually our p-XRF measurements of the UCL beads (Jambon, 2010), motivated the more in-depth studies mentioned above. 2) Umm el Marra (Syria) pendant from a tomb dated at 2300 BCE (Schwartz et al. 2003; Jambon 2017), in Aleppo Archaeological Museum and suggested to be meteoritic (ref. number M00299). 3) Ugarit (Ras Shamra, Syria; 1400 BCE) axe (Schaeffer, 1939; Jambon et al. 2017) in Aleppo Archaeological Museum (ref. number RS9-250).

Additional artifacts will be considered in the discussion: one gold-handled dagger from Alaca

Höyük (Çorum, Turkey; Ankara Museum ref. number Al. K. 14) ca. 2400-2300 B.C., (Nakai et al. 2008; Kosay, 1951; Wertime, 1973), Tutankhamen's dagger 14<sup>th</sup> century B.C. (Cairo Museum. Carter ref. 256K JE 61585), analyzed recently (Comelli et al. 2016; Ströbele, 2016), one axe from Shang dynasty, (Kao Ch'eng, Hubei, China) ca. 1400 B.C., analyzed by Li Chung (1979).

### **A3.4. Bronze to Iron Age transition**

Two objects curated at Le Louvre Museum (Paris, France), one halberd axe (AO22953) and one adze (AO20882/AO20159) both with an iron blade and a bronze socket and reported from Luristan, were analyzed. According to their typology these two objects are considered to be quite primitive and Overlaet (2008) indicates « the characteristic Luristan style objects all belong to the Iron Age (ca. 1300-650 BCE) ». Those analyzed here with a bronze socket appear rather primitive, technically speaking.

Seven objects were analyzed at Latenium Museum in Neuchâtel (Switzerland): three pins (AUV-2242; HR-93 and HR-110), one fish hook (AUV-2290), one nail (AUV-2736), one nail head (AUV-2740) and one adze (N-1). The objects are from lake sediments near Neuchâtel, Auvernier site (AUV) and Hauterive site (H), and dated by dendrochronology, the pins having a well-characterized Late Bronze Age typology. According to their stratigraphic age (Rychner, 1987) they are from the Bronze Age, meaning European Bronze Age. They are contemporaneous to Iron Age objects from the Near East and for this reason we decided to consider their age in this work, to be transitional. In addition the adze (N) is from an unspecified location in Norway.

Iron axes from early Zhou dynasty about 1000 B.C. (Hunan, China) (possibly Shang-Zhou transition period), reportedly meteoritic, were also considered (Gettens et al. 1971). According to the technical evolution in China these axes are dated from before the knowledge of iron smelting in China, but as already mentioned the iron could have been imported from the west where iron smelting was already mastered. As for the Latenium samples, we considered these axes to be transitional, just to avoid confusion.

### **A3.5. Archaeological irons from the Iron Age.**

We know that the beginning of the Iron Age becomes younger when going westwards from the Middle East, but also eastward (e.g. China and India) and southward (Egypt and Africa) from what appears to be a crucial area. Therefore iron objects from the Bronze Age in western European countries may be imported irons from the Iron Age in more eastern countries. We considered in this section iron artifacts from Iron Age layers. As for artifacts from the Bronze Age, we considered high quality analyses from the



literature as well as a few analyses of our own, using our p-XRF analyzer.

Six ingots from Marsal (Moselle, France), well preserved bi-pyramidal iron bars (Bertaux, 1990; L. Olivier pers. com. 2016), were analyzed at the Musée d'Archéologie Nationale in St Germain-en-Laye (France). According to the archaeological context they are dated at ca. 700±100 BCE. A summary of the results is reported in table A3.

One oven bottom from Morocco of unspecified age (provided by A. Hmani). Bloomery iron recovered from Gresham shipwreck was kindly provided by M. Martinon-Torres (UCL) (wreck after 1574; excavation 2003) Auer and Firth (2007). In addition we analyzed three artifacts from Hallstatt time, as polished sections at the University of Poznan (Poland): Two bracelets (Zimny J., 1965) from Czestochowa-Rakowa (Poland) already studied by Kotowiecki (2004) and Piaskowki (1982) and shown to have a high Ni content and one axe from Wietrzno (Podkarpackie Province, Poland; Hallstatt, unspecified) already mentioned by Piaskowki (1960) to be a composite made of 5 slices (2 and 3 of two different kinds).

In addition we collected data from the literature. Artifacts from the Iron Age excavated at Kaman Kalehöyük (Kaman, Turkey) (Akanuma, 2006) provide good examples of smelted iron in regions where Ni rich iron ores are available. Eschenlohr et al. (2007) analyzed a great number of irons from Develier –Courtételle (Jura, Switzerland), Brauns et al. (2013) one ingot from Renningen and one bloomery iron from Grösseltal (Baden-Württemberg, Germany) dated from late Hallstatt to early La Tène and Schwab et al. (2006) irons from Manching (Bavaria, Germany) dated from La Tène.

### A3.6. Iron ores.

There exist a number of different types of iron ores. Sedimentary ores contain negligible amounts of Ni and Co (see e.g. Schwab et al. 2006) and are not relevant for the present study. The ultimate source of Ni in the Earth is the terrestrial mantle, which is composed of peridotite, a rock group with dominant olivine with about 0.3% Ni. Peridotite crop out in orogenic massifs forming the substratum of former oceanic crust now included in orogenic belts. Ni and Co are contained mostly in olivine and their concentrations in peridotites are marginally variable. The average composition of the mantle has been estimated by a number of workers considering a significant database (see e.g. McDonough and Sun, 1995). Weathered peridotites are transformed to laterites which ultimately become highly concentrated in iron and are potential ores. Lateritic iron ores are considered here for two reasons: (1) they may contain variable but significant amounts of Ni and Co. (2) Such ores are quite common in the Near East for instance in Greece, Anatolia, Cyprus and Iran. The

compositions of such ores are available in the literature. We selected data obtained at various levels in alteration profiles (ore beds) with a significant amount of iron oxide in order to document varied Fe/Ni ratios corresponding to different levels in the alteration profile from Oman (Al Kirbash, 2016), Barro alto and Santa Fé, Brazil (Trescases and Oliveira, 1981), Oregon, USA (Hotz, 1967), Burma (Schellmann, 1989) and Cameroon (Yongue-Fouateu et al. 2006).

### A.3.7. Pyrrhotite. (Table A3)

Some pyrrhotites (FeS) may contain significant amounts of Ni. These are high temperature rocks. They are high temperature sulfides precipitated from magmas formed at depth in the mantle. We considered a set of pyrrhotite analyses from Oshirabetsu mine, Hokkaido (Bamba, 1981).

## A4. Results

### A4.1 . Bronze Age Artifacts (Table A4)

**Gerzeh beads.** Several spots were analyzed on each of the three different beads from the Petrie Museum (UCL, London). The Ni contents (corrected on an oxygen free basis) vary from 3.6 to 8.5 with an average value of 5.7 % Ni. This can be compared to the 0.9 to 4.8 % Ni measured by EDS (Energy dispersive spectrometry) in an SEM (scanning electron microscope) over rastered areas 0.25 x 0.20 mm<sup>2</sup> by Johnson et al. (2013) on the bead preserved in Manchester Museum and 4.7 to 6.8 % by PGAA on Petrie Museum beads (Rehren et al. 2013).

The analyses of Rehren et al. (2013) on the same beads and those of Johnson et al. (2013) are completely consistent with the p-XRF data once the effects of weathering are accounted for. The penetration depth of the analyses (PGAA>p-XRF>EDS-SEM) is another important parameter since outer layers are more affected by weathering.

**Umm el Marra pendant.** This specimen is highly weathered with an apparent density of 2.4 g/cm<sup>3</sup> (instead of 8 for fresh meteoritic iron). It is the sample exhibiting the highest state of weathering considered in this study.

The magnetic susceptibility was measured with a portable susceptibility probe SM30, according to the procedures given by Gattacceca et al. (2004). One finds a value of  $\log \chi = 4.9$  ( $10^{-9}$  m<sup>3</sup>/kg). This value is far higher than that of any terrestrial rock, which excludes a terrestrial origin. For a meteorite this would correspond to an ordinary chondrite of type L. According to the state of weathering, and the total amount of Fe (47.8 %), it is more reasonable to interpret this value as representing some residual metal. This estimate can be done from the experiments of Rochette et al. (2009) and one obtains about 15 wt. % metal confirming that the sample was

highly, but not completely weathered. The Ni content is quite low (0.8 to 3.2 % Ni on an oxygen free basis).

**Literature data.** We restricted our compilation to sufficiently well preserved artifacts (those containing some preserved metal) analyzed for Fe, Co and Ni. The number of such cases is very limited but we hope that this may change drastically on a short time scale. Among those are prestigious artifacts: Tutankhamen's dagger (Comelli et al. 2016; Ströbele et al. 2016,), Chinese axes (Li Chung, 1979) and the Alaça Höyük dagger (Nakai et al. 2008).

**Tut's dagger** is obviously the best preserved iron as discussed recently by Comelli et al. (2016) with a fresh meteoritic composition, quite high in Ni. We notice that the composition given by Comelli is quite high in Ni (10.8 %) and within error and/or variability similar to that of Ströbele et al. (2016) (12.9%), therefore unambiguously meteoritic, compared to the previous results of Helmi and Barakat (1995) giving only 2.8 % Ni (Co was not documented). This again indicates that the idea of a threshold value is meaningless. Interestingly Horus eye (on the bracelet) and the miniature headrest exhibit significantly less nickel. According to their exceptional state of preservation we can say that, because of their significantly different compositions, that the dagger is not made from the same meteorite as the other two objects. The dagger is homogeneous (Standard Deviation of 1.1%) in relation with its high Ni content (small size of the exsolution domains). Both the Horus eye (S.D. =4.8 %) and the headrest (S.D. =7.2 %) are far less homogeneous. This suggests that the Widmanstätten pattern is still present and coarser. The low Ni phase (kamacite) cannot be much different from 5% Ni, whereas the Ni rich phase (taenite) is less homogeneous with on the average about 20% Ni. The more Ni rich metal (on average) corresponds to a larger fraction of taenite and a larger variability in the composition, at the scale of analysis (4 mm diameter). The slightly different compositions and variabilities suggest that the headrest and the Horus eye were also made from two different meteorites. In others words Tutankamun's treasure was the first meteorite collection ever! A set of 16 additional iron miniature artifacts are presently investigated and may add to the list (F. Stroebel, pers. com)..

**The Chinese axe from Shang dynasty** (14<sup>th</sup> century BC) at Kao Ch'eng (Hopei, China) studied by Li Chung (1979) is not as well preserved but careful analysis showed the presence of residual metal. Unlike the Zhou axes (see below) this one is

undoubtedly from the Bronze Age. The typology is typically one of the Bronze Age with a bronze socket cast around an iron blade. According to Li Chung (1979) metallographic investigations on the blade structure indicate a meteoritic structure. The Ge content exceeds 100 ppm, a meteoritic signature. The Ni and Co contents average (0.8-4% and 0.2-0.4 % respectively on an oxygen free basis). Despite a significant uncertainty the Ni/Co and Ni/Fe plot along the meteoritic correlation with a lower than meteoritic Ni content.

**The Alaça Höyük dagger** is the worst preserved among those three. Still, Nakai et al. (2008) report variable Ni contents from 2.4 to 5.99 %, with the higher value in the meteoritic range. The calculation of the Co concentration is not documented (see A2.1) and according to the severe interference between Fe and Co this point is critical. Their Ni/Co ratios range 15-26 suggesting a factor of two in their Co estimates. The high Ni concentration suggests a typical meteoritic composition but because of the possible analytical bias the Co data were not considered.

#### **A4.2. Iron Age artifacts (Table A5)**

##### **Literature data**

**Fragments from Kaman Kalehöyük (Turkey).** (Akanuma, 2006). The only objects, which were not pervasively oxidized (O<10%) are dated from the Iron Age (stratum II a and c). The chemical composition exhibits very low contents of both Ni (<0.08%) with low Ni/Fe (<0.001) and Co with a high Co/Ni ratio (> 0.8). This composition is that of typical smelted iron, as expected from Iron Age artifacts. Some of the compositions are comparatively high in Ni (up to 0.7%), which may correspond to a Ni rich iron ore.

**Western European irons.** The irons of Manching, Germany (Schwab et al. 2006) are low in Ni (between 100 and 1600 ppm), the Co/Ni ratio are on the order of 0.5-1.0, typical values for smelted irons. The iron objects analyzed by Brauns et al. (2013) (2 from Renningen and one from Grösseltal, both Germany) are low in Ni (300-1000 ppm) with Co/Ni ranging 0.2 to 0.5.

Finally the more recent objects from Devélier-Courtetelle (Jura, Switzerland), a set of 67 objects, analyzed by Eschenlohr et al., (2007) are low in Ni (40-4000 ppm; average 1200 ppm) with a Co/Ni ratio ranging 0.02 to 3 (average 0.43)

For these three sets the corresponding fields are well separated from the meteoritic composition field.

Specimen	surface type	Fe	Co	Ni	Specimen	surface type	Fe	Co	Ni
Agoudal	cleaned	91.30	0.69	8.01	Morasko	polished	92.09	0.49	7.42
Agoudal	oxidized	93.10	0.73	6.17	Morasko	polished	92.16	0.50	7.34
Agoudal	oxidized	94.97	0.89	4.14	Morasko	polished	91.31	0.53	8.15
Agoudal	oxidized	92.17	0.65	7.18	Morasko	average	91.79	0.53	7.69
Campo del Cielo	rough	93.45	0.42	6.13	Mundrabilla	polished	94.65	0.71	4.64
Campo del Cielo	rough	93.19	0.53	6.28	Mundrabilla	rough	94.10	0.79	5.11
Campo del Cielo	rough	92.09	0.50	7.41	Nantan	rough	92.50	0.46	7.04
Campo del Cielo	rough	92.43	0.44	7.14	NWA 854 ZIZ	oxidized	94.30	0.50	5.20
Canyon Diablo	oxidized	95.06	0.44	4.50	NWA 854 ZIZ	oxidized	94.33	0.49	5.18
Canyon Diablo	oxidized	92.05	0.51	7.44	NWA 859 Taza	oxidized	52.25	0.72	47.03
Canyon Diablo	oxidized	94.37	0.57	5.06	NWA 859 Taza	oxidized	80.94	0.79	18.27
Canyon Diablo	oxidized	92.70	0.46	6.84	NWA 859 Taza	oxidized	81.22	0.65	18.13
Canyon Diablo	oxidized	90.25	0.63	9.12	NWA 859 Taza	oxidized	89.11	1.82	9.07
Cape of Good Hope	polished	83.16	0.78	16.07	NWA 859 Taza	oxidized	89.28	1.82	8.90
Chinga	oxidized	87.67	0.72	11.61	NWA 859 Taza	oxidized	89.80	1.77	8.43
Chinga	oxidized	80.05	0.69	19.26	NWA 5549 IAB-MG	oxidized	92.88	0.45	6.67
Copiapo	polished	93.90	0.51	5.59	NWA 5549 IAB-MG	oxidized	91.99	0.76	7.25
Gebel Kamil	polished	75.93	1.21	22.86	NWA 5549 IAB-MG	oxidized	92.40	0.69	6.91
Gebel Kamil	rough	75.50	1.19	23.31	NWA 5549 IAB-MG	oxidized	92.40	0.66	6.94
Gebel Kamil	rough	77.00	1.18	21.82	NWA 5549 IAB-MG	oxidized	92.12	0.72	7.16
Gebel Kamil	rough	80.09	0.96	18.95	NWA 5549 IAB-MG	oxidized	93.56	0.52	5.92
Gebel Kamil	rough	80.73	0.97	18.30	NWA unclassified	polished	91.35	0.52	8.13
Gebel Kamil	rough	82.56	1.83	15.61	Sikhote Alin	rough	93.90	0.57	5.53
Juromenha	polished	94.31	0.51	5.18	Sikhote Alin	rough	94.60	0.54	4.86
Juromenha	polished	90.73	0.51	8.76	Sikhote Alin	rough	92.56	0.60	6.84
Morasko	weathered	86.37	0.78	12.86	Sikhote Alin	rough	92.42	0.52	7.06
Morasko	weathered	91.20	0.69	8.12	Sikhote Alin	rough	93.82	0.47	5.71
Morasko	weathered	90.70	0.61	8.68	UAI	polished	74.83	0.86	24.31
Morasko	weathered	98.75	0.32	0.92	UAI	oxidized	86.11	0.96	12.93
Morasko	weathered	95.46	0.40	4.14	UAI	oxidized	72.89	0.80	26.31
Morasko	weathered	92.19	0.57	7.24	UAI	oxidized	74.75	0.81	24.44
Morasko	weathered	92.77	0.43	6.80	UAI	oxidized	75.67	0.76	23.57
Morasko	weathered	94.91	0.26	4.83					
Morasko	weathered	93.42	0.50	6.08	<b>Standard</b>				
Morasko	weathered	91.69	0.59	7.72	0 Ni	polished	99.74	BDL	BDL
Morasko	polished	92.59	0.38	7.03	10 Ni	polished	90.27	BDL	9.73
Morasko	polished	92.37	0.47	7.15	10 Ni	polished	90.29	BDL	9.71
Morasko	polished	92.45	0.64	6.91	17 Ni	polished	82.42	BDL	17.58
Morasko	polished	91.86	0.48	7.66	17 Ni	polished	82.40	BDL	17.60
Morasko	polished	86.84	0.64	12.52	30 Ni	polished	69.89	BDL	30.11
Morasko	polished	90.49	0.68	8.83	30 Ni	polished	69.88	BDL	30.12
Morasko	polished	92.11	0.51	7.39	<b>Irons average</b>				
Morasko	polished	92.42	0.55	7.03	IAB		93.14	0.46	6.40
Morasko	polished	92.54	0.48	6.98	IC		92.44	0.46	7.10
Morasko	polished	92.09	0.59	7.32	IIAB		93.32	0.53	6.15
Morasko	polished	90.31	0.50	9.18	IIC		88.55	0.65	10.80
Morasko	polished	91.62	0.55	7.83	IID		88.33	0.47	11.20
Morasko	polished	92.33	0.51	7.16	IIE		90.40	0.47	9.13
Morasko	polished	92.50	0.46	7.04	IIF		86.40	0.70	12.90
Morasko	polished	93.40	0.45	6.15	IIIAB		91.01	0.50	8.49
Morasko	polished	92.22	0.54	7.23	IIIE		91.01	0.50	8.49
Morasko	polished	91.78	0.47	7.75	IIIF		91.69	0.36	7.95
Morasko	polished	92.22	0.55	7.23	IVA		91.60	0.40	8.00
					IVB		82.16	0.74	17.10

Table A1 : p-XRF analyses of reference iron specimens of the present study. All totals normalized to 100%. Average irons (bottom) are for NAA analyses from the literature (Willis et al. 1980 in Mittlefehldt et al. 1998). BDL = below detection limit.

Specimen	Fe	Co	Ni	Ni/Fe	Ni/Co	Specimen	Fe	Co	Ni	Ni/Fe	Ni/Co
Açfer 234	90.72	0.45	8.73	0.10	19.5	Maria da Fé	92.07	0.38	7.45	0.08	19.6
Agoudal	93.99	0.41	5.50	0.06	13.4	Miles	91.50	0.44	7.96	0.09	18.0
Apache Junction	90.98	0.53	8.39	0.09	15.9	Mohawk	92.05	0.44	7.41	0.08	16.7
Assamakka	90.70	0.39	8.81	0.10	22.8	Mont Dieu	91.92	0.43	7.55	0.08	17.4
Ban Rong Du	91.43	0.57	7.90	0.09	13.8	Morasko	92.85	0.46	6.59	0.07	14.4
Baygorria	92.97	0.43	6.50	0.07	15.1	Mundrabilla	91.68	0.72	7.50	0.08	10.4
Bernic Lake	92.97	0.43	6.50	0.07	15.1	Nantan	89.11	0.79	10.00	0.11	12.7
Blanca Estela	92.84	0.46	6.60	0.07	14.5	NEA 002	89.04	0.66	10.20	0.11	15.5
Buffalo Gap	91.42	0.48	8.00	0.09	16.7	Nothing	89.11	0.66	10.13	0.11	15.3
Bur-Abor	91.40	0.51	7.99	0.09	15.6	Nova 011	87.54	0.47	11.89	0.14	25.2
Burns	88.96	0.57	10.37	0.12	18.2	NWA 10224	89.60	0.60	9.70	0.11	16.2
Campinorte	92.22	0.58	7.10	0.08	12.2	NWA 10417	91.32	0.48	8.10	0.09	16.9
Campo del Cielo	92.73	0.43	6.74	0.07	15.7	NWA 1430	91.51	0.50	7.89	0.09	15.7
Canyon Diablo	92.33	0.44	7.13	0.08	16.3	NWA 176	90.83	0.41	8.66	0.10	21.0
Canyon Diablo	92.61	0.46	6.83	0.07	14.8	NWA 2151	88.93	0.55	10.42	0.12	19.1
Cape Good Hope	92.42	0.47	7.01	0.08	15.0	NWA 2311	89.63	0.49	9.78	0.11	20.1
Catalina 003	82.53	0.76	16.61	0.20	21.9	NWA 2428	87.57	0.56	11.77	0.13	21.1
Catalina 107	91.40	0.50	8.00	0.09	15.9	NWA 2677	86.52	0.59	12.80	0.15	21.9
Cedar Station	88.99	0.60	10.31	0.12	17.3	NWA 2678	94.00	0.45	5.45	0.06	12.2
Cerro del Inca	77.55	0.65	21.70	0.28	33.4	NWA 2679	88.16	0.56	11.18	0.13	20.0
Cheder	92.22	0.44	7.24	0.08	16.5	NWA 2680	85.70	0.54	13.66	0.16	25.5
Chinga	82.69	0.55	16.66	0.20	30.3	NWA 2743	92.67	0.48	6.75	0.07	14.2
Chisenga	93.10	0.60	6.20	0.07	10.3	NWA 3200	88.68	0.55	10.67	0.12	19.3
Chuckwalla	92.85	0.46	6.59	0.07	14.3	NWA 3201	94.18	0.51	5.21	0.06	10.2
Cotopaxi	89.60	0.50	9.80	0.11	19.6	NWA 3202	93.84	0.45	5.61	0.06	12.4
Cruz Alta	93.68	0.45	5.77	0.06	12.8	NWA 3204	87.84	0.59	11.48	0.13	19.6
DaG 406	92.75	0.44	6.71	0.07	15.3	NWA 3205	87.38	0.81	11.71	0.13	14.5
Dayet el aam	93.74	0.46	5.70	0.06	12.4	NWA 3206	79.78	0.60	19.52	0.24	32.7
Digor	91.70	0.49	7.71	0.08	15.7	NWA 3208	91.83	0.50	7.58	0.08	15.3
Djebel In-Azzene	89.10	0.50	10.30	0.12	20.6	NWA 3322	92.09	0.51	7.30	0.08	14.3
Dolores	91.92	0.50	7.48	0.08	15.0	NWA 4217	92.59	0.43	6.88	0.07	15.9
Domeyko	89.87	0.55	9.48	0.11	17.1	NWA 4233	87.12	0.48	12.30	0.14	25.5
Dronino	89.54	0.55	9.81	0.11	17.7	NWA 468	87.33	0.72	11.85	0.14	16.5
Dutch Flat	93.63	0.49	5.78	0.06	11.8	NWA 4700	91.40	0.50	8.00	0.09	16.0
El Medano 300	92.80	0.50	6.60	0.07	13.1	NWA 4701	82.36	0.61	16.93	0.21	27.9
Faina	90.62	0.61	8.67	0.10	14.2	NWA 4702	94.03	0.44	5.43	0.06	12.4
Fish Canyon	93.01	0.40	6.50	0.07	16.5	NWA 4703	90.90	0.48	8.52	0.09	17.8
Foum Zguid	93.59	0.50	5.81	0.06	11.6	NWA 4704	90.51	0.49	8.90	0.10	18.2
Gapyeong	91.14	0.49	8.27	0.09	16.8	NWA 4705	78.57	1.32	20.01	0.25	15.2
Gebel Kamil	79.35	0.75	19.80	0.25	26.4	NWA 4706	88.57	0.55	10.78	0.12	19.6
Guanaco	94.96	0.51	4.43	0.05	8.7	NWA 4707	91.20	0.52	8.18	0.09	15.8
Hagersville	92.53	0.48	6.89	0.07	14.3	NWA 4708	91.50	0.51	7.89	0.09	15.6
Hidden Valley	91.21	0.50	8.19	0.09	16.4	NWA 4709	92.53	0.47	6.91	0.07	14.9
High Island creek	91.53	0.51	7.86	0.09	15.4	NWA 4710	88.57	0.55	10.78	0.12	19.6
Horh Uul	90.65	0.48	8.77	0.10	18.3	NWA 4711	88.64	0.53	10.73	0.12	20.4
Javorje	91.59	0.48	7.83	0.09	16.3	NWA 4713	84.92	0.53	14.45	0.17	27.3
Juromenha	79.45	0.75	19.80	0.25	26.4	NWA 4861	87.40	0.60	11.90	0.14	19.8
Juromenha	90.78	0.52	8.60	0.09	16.5	NWA 5289	90.48	0.40	9.02	0.10	22.3
Kavarpura	90.01	0.39	9.50	0.11	24.4	NWA 5549	92.57	0.46	6.88	0.07	15.1
La Yesera 005	92.06	0.39	7.45	0.08	19.2	NWA 5804	87.85	0.51	11.54	0.13	22.7
Las Cruces	90.22	0.53	9.15	0.10	17.2	NWA 6163	91.04	0.43	8.43	0.09	19.7
Left Hand Creek	92.47	0.47	6.96	0.08	14.9	NWA 6164	92.39	0.50	7.01	0.08	14.0
Licking	90.20	0.41	9.29	0.10	22.7	NWA 6165	88.61	0.49	10.80	0.12	22.0
Lixian	94.03	0.47	5.40	0.06	11.5	NWA 6166	81.50	0.60	17.80	0.22	29.8

Table A2 (part 1). Iron compositions according to the Met Bull data base : 1986-2016

Specimen	Fe	Co	Ni	Ni/Fe	Ni/Co	Specimen	Fe	Co	Ni	Ni/Fe	Ni/Co
Lone Island Lake	91.86	0.43	7.62	0.08	17.9	NWA 6167	79.97	0.93	19.00	0.24	20.5
Longtian	92.09	0.50	7.32	0.08	14.8	NWA 6203	93.18	0.44	6.28	0.07	14.2
NWA 6279	92.94	0.49	6.47	0.07	13.3	Ocotillo	92.42	0.47	7.01	0.08	14.9
NWA 6583	81.47	0.43	18.00	0.22	41.9	Oglat Sidi Ali	84.66	1.17	14.07	0.17	12.0
NWA 6716	91.71	0.42	7.77	0.08	18.4	Pontes e lacerda	91.31	0.49	8.10	0.09	16.5
NWA 6903	90.98	0.52	8.40	0.09	16.2	Porto alegre	90.38	0.50	9.02	0.10	18.2
NWA 6931	92.47	0.45	6.98	0.08	15.5	Prospector Pool	90.53	0.44	8.94	0.10	20.6
NWA 6932	86.95	0.68	12.27	0.14	17.9	Qijiaoqing	88.76	0.54	10.60	0.12	19.8
NWA 7335	88.08	0.84	10.98	0.12	13.1	Rainy River	92.21	0.46	7.23	0.08	15.6
NWA 7795	85.83	0.47	13.60	0.16	28.8	Sahara 03505	90.47	0.41	9.02	0.10	21.8
NWA 8154	94.16	0.47	5.27	0.06	11.1	Santo Antonio d.	93.68	0.46	5.76	0.06	12.5
NWA 8155	86.13	0.48	13.30	0.15	28.0	Sapopema	91.55	0.39	7.96	0.09	20.4
NWA 8156	87.17	0.43	12.30	0.14	28.5	Sarvestan	91.79	0.39	7.72	0.08	19.8
NWA 8302	93.75	0.45	5.70	0.06	12.6	Shawnee	92.04	0.46	7.40	0.08	16.1
NWA 8337	88.27	0.45	11.18	0.13	25.0	Sikhote Alin	92.67	0.46	6.88	0.07	15.1
NWA 8346	90.99	0.48	8.43	0.09	17.6	Sikhote Alin	93.73	0.47	5.70	0.06	12.1
NWA 8347	91.47	0.44	7.99	0.09	18.2	Sikhote Alin	93.58	0.38	5.94	0.06	15.6
NWA 8348	84.70	0.55	14.65	0.17	26.7	Tamarack	93.54	0.50	5.86	0.06	11.7
NWA 8370	90.93	0.53	8.44	0.09	16.0	Tarahumara	91.52	0.47	7.91	0.09	16.8
NWA 8442	91.42	0.50	7.98	0.09	15.9	Tartak	92.03	0.50	7.37	0.08	14.7
NWA 8443	89.81	0.50	9.59	0.11	19.0	Tequisquiapan	93.05	0.39	6.46	0.07	16.6
NWA 8444	92.33	0.50	7.07	0.08	14.2	Tongan	92.87	0.43	6.60	0.07	15.2
NWA 8480	92.89	0.61	6.40	0.07	10.5	Tres Castillos	90.16	0.51	9.23	0.10	18.1
NWA 8539	85.98	0.47	13.45	0.16	28.6	Turgut	90.44	0.42	9.04	0.10	21.5
NWA 854	92.51	0.47	6.92	0.07	14.8	Uruaçu	93.01	0.46	6.43	0.07	14.0
NWA 8568	94.54	0.26	5.10	0.05	19.6	Verkhnyi Saltov	91.45	0.50	7.95	0.09	15.9
NWA 859	82.69	1.31	15.90	0.19	12.1	Villa regina	91.47	0.50	7.93	0.09	15.8
NWA 860	91.02	0.52	8.36	0.09	16.0	Vitoria da Con.	90.09	0.41	9.40	0.10	22.9
NWA 8719	91.62	0.50	7.78	0.08	15.4	Whitecourt	91.30	0.50	8.11	0.09	16.4
NWA 959	89.99	0.43	9.48	0.11	22.2	Willow Grove	70.79	1.21	27.90	0.39	23.1
NWA 968	86.11	0.55	13.25	0.15	24.3	Yarovoye	89.93	0.52	9.45	0.11	18.2
NWA8441	87.43	0.52	11.95	0.14	23.1	Yuanyang	92.42	0.46	7.02	0.08	15.2
Ocate	92.44	0.47	6.99	0.08	15.0	Zapaliname	92.74	0.46	6.70	0.07	14.7

Table A2 (part 2). Iron compositions according to the Met Bull data base : 1986-2016

sample #	S	Fe	Co	Ni	Total	Ni/Co	Ni/Fe
29	38.51	60.25	0.05	0.95	99.76	19.0	0.016
36	38.52	60.83	0.02	0.47	99.84	23.5	0.008
45	39.19	58.78	0.10	1.09	99.16	10.9	0.019
89	38.23	59.88	0.06	1.08	99.25	18.0	0.018
55	38.70	60.19	0.02	0.79	99.70	39.5	0.013
12	38.60	61.63	0.06	0.51	100.80	8.5	0.008
44	38.61	61.57	0.04	0.02	100.24	0.5	0.000
51	37.88	61.74	0.10	0.52	100.24	5.2	0.008

Table A3. Chemical composition of pyrrhotites from the Oshirabetsu Mine, Hokkaido (Japan). (Bamba, 1981)



Specimen		Fe	Co	Ni	Locality	Age BCE	analysis #	ref
RS9-250	axe A	90.16	0.54	6.55	Ugarit	1350	217	1,2
RS9-250	axe A	89.43	0.43	2.99	Ugarit	1350	218	1,2
RS9-250	axe A	91.38	0.36	2.32	Ugarit	1350	219	1,2
RS9-250	axe A	80.33	0.46	3.01	Ugarit	1350	220	1,2
RS9-250	axe A	65.64	0.47	1.72	Ugarit	1350	221	1,2
RS9-250	axe B	89.57	0.40	4.43	Ugarit	1350	224	1,2
RS9-250	axe B	89.58	0.46	5.44	Ugarit	1350	225	1,2
RS9-250	axe B	92.46	0.41	4.71	Ugarit	1350	226	1,2
RS9-250	axe B	91.97	0.57	4.54	Ugarit	1350	227	1,2
RS9-250	axe B	87.67	0.63	7.59	Ugarit	1350	228	1,2
Petrie 10738	Bead	92.47	0.58	3.45	Gerzeh	3200	281	3,4,5
Petrie 10738	Bead	86.24	0.99	8.10	Gerzeh	3200	282	3,4,5
Petrie 10739	Bead	81.37	0.73	5.24	Gerzeh	3200	283	3,4,5
Petrie 10740	Bead	89.18	0.54	4.53	Gerzeh	3200	284	3,4,5
M00299	pendant	95.18	0.53	2.62	Umm el Marra	2500	209	6,7
M00299	pendant	93.95	0.59	2.63	Umm el Marra	2500	210	6,7
M00299	pendant	97.22	0.40	0.84	Umm el Marra	2500	211	6,7
M00299	pendant	91.20	0.42	1.14	Umm el Marra	2500	212	6,7
M00299	pendant	95.77	0.49	2.52	Umm el Marra	2500	213	6,7
M00299	pendant	92.96	0.63	3.21	Umm el Marra	2500	214	6,7
M00299	pendant	92.29	0.37	1.31	Umm el Marra	2500	215	6,7
M00299	pendant	95.69	0.58	1.87	Umm el Marra	2500	216	6,7
M00299	pendant	96.23	0.46	2.99	Umm el Marra	2500	233	6,7
Tut's dagger	blade	88.57	0.58	10.85	Thebes	1350	p-XRF	8
Tut's dagger	blade	86.70	0.58	12.91	Thebes	1350	p-XRF	12
Tut's bracelet	Horus eye	90.03	0.49	8.03	Thebes	1350	p-XRF	12
Tut's	Headrest	90.51	0.47	8.76	Thebes	1350	p-XRF	12
Al.K.14	Dagger	97.41	-	2.43	Alaca Höyük	2400	p-XRF #1	9
Al.K.14	Dagger	95.68	-	4.07	Alaca Höyük	2400	p-XRF #2	9
Al.K.14	Dagger	93.78	-	5.99	Alaca Höyük	2400	p-XRF #3	9
Al.K.14	Dagger	95.31	-	4.50	Alaca Höyük	2400	p-XRF #4	9
Shang axe	blade	91.10	0.40	3.40	Kao Ch'eng	14th cent.	wet chemical	10
Shang axe	blade	96.84	0.21	1.95	Kao Ch'eng	14th cent.	XRF	10
Zhou axe 34.10	Broad axe	95.85	0.15	3.00	Hunan	10th cent.	wet chemical	11
Zhou axe 34.10	Broad axe	88.31	0.21	11.47	Hunan	10th cent.	wet chemical	11
Zhou axe 34.10	Bd axe oxide	87.62	0.39	11.99	Hunan	10th cent.	wet chemical	11
Zhou axe 34.10	Bd axe oxide	88.44	0.26	11.30	Hunan	10th cent.	wet chemical	11
Zhou axe 34.10	Bd axe oxide	88.27	0.28	11.44	Hunan	10th cent.	wet chemical	11
Zhou axe 34.10	Bd axe oxide	88.31	0.21	11.47	Hunan	10th cent.	wet chemical	11
Zhou axe 34.10	Bd axe Metal	92.20	0.60	6.90	Hunan	10th cent.	EMPA	11
Zhou axe 34.10	Bd axe Metal	92.60	0.70	6.70	Hunan	10th cent.	EMPA	11
Zhou axe 34.10	Bd axe Metal	91.60	0.60	6.80	Hunan	10th cent.	EMPA	11
Zhou axe 34.10	Bd axe Metal	65.30	0.30	23.00	Hunan	10th cent.	EMPA	11
Zhou axe 34.10	Bd axe Metal	70.00	0.30	22.60	Hunan	10th cent.	EMPA	11
Zhou axe 34.10	Bd axe Metal	62.40	0.30	27.30	Hunan	10th cent.	EMPA	11
Zhou axe 34.10	Bd axe Metal	65.40	0.20	29.30	Hunan	10th cent.	EMPA	11
Zhou axe 34.11	Dagger axe	98.45	0.05	0.50	Hunan	10th cent.	wet chemical	11
Zhou axe 34.11	Dagger axe	98.85	0.07	1.08	Hunan	10th cent.	wet chemical	11
Zhou axe 34.11	Dagger axe	86.00	0.40	5.20	Hunan	10th cent.	EMPA	11
Zhou axe 34.11	Dagger axe	81.50	0.30	5.20	Hunan	10th cent.	EMPA	11

Table A4 : Present p-XRF analyses (top) and literature analyses of Bronze Age artifacts.

1 Schaeffer 1939  
2 Jambon et al. 2017  
3 Jambon 2010

4 Rehren et al. 2013  
5 Johnson et al. 2013  
6 Schwartz et al. 2003

7 Jambon 2017  
8 Comelli et al. 2016  
9 Nakai et al. 2008

10 Li Chung 1979  
11 Gettens et al. 1971  
12 Ströbele et al. 2016

Specimen	type	Fe	Co	Ni	Locality	Age BCE	analysis #	ref
p-XRF	detection limit	0.24	0.05	0.02	-			
Zhou axe 34.10	Broad axe	95.85	0.15	3.00	Honan	10 th century	wet chem.	11
Zhou axe 34.10	Broad axe	88.31	0.21	11.47	Honan	10 th century	wet chem.	11
Zhou axe 34.10	Bd axe oxide	87.62	0.39	11.99	Honan	10 th century	wet chem.	11
Zhou axe 34.10	Bd axe oxide	88.44	0.26	11.30	Honan	10 th century	wet chem.	11
Zhou axe 34.10	Bd axe oxide	88.27	0.28	11.44	Honan	10 th century	wet chem.	11
Zhou axe 34.10	Bd axe oxide	88.31	0.21	11.47	Honan	10 th century	wet chem.	11
Zhou axe 34.10	Bd axe metal	92.20	0.60	6.90	Honan	10 th century	EMPA	11
Zhou axe 34.10	Bd axe metal	92.60	0.70	6.70	Honan	10 th century	EMPA	11
Zhou axe 34.10	Bd axe metal	91.60	0.60	6.80	Honan	10 th century	EMPA	11
Zhou axe 34.10	Bd axe metal	65.30	0.30	23.00	Honan	10 th century	EMPA	11
Zhou axe 34.10	Bd axe metal	70.00	0.30	22.60	Honan	10 th century	EMPA	11
Zhou axe 34.10	Bd axe metal	62.40	0.30	27.30	Honan	10 th century	EMPA	11
Zhou axe 34.10	Bd axe metal	65.40	0.20	29.30	Honan	10 th century	EMPA	11
Zhou axe 34.11	Dagger axe	98.45	0.05	0.50	Honan	10 th century	wet chem.	11
Zhou axe 34.11	Dagger axe	98.85	0.07	1.08	Honan	10 th century	wet chem.	11
Zhou axe 34.11	Dagger axe	86.00	0.40	5.20	Honan	10 th century	EMPA	11
Zhou axe 34.11	Dagger axe	81.50	0.30	5.20	Honan	10 th century	EMPA	11
AUV-2242 Latanium	pin	93.17	bdl	bdl	Neuchatel	950-850	#283	12
AUV-2242 Latanium	pin	96.15	bdl	bdl	Neuchatel	950-850	#285	12
AUV-2290 Latanium	hook	87.17	bdl	bdl	Neuchatel	950-850	#286	12
AUV-2736 Latanium	nail	96.64	bdl	bdl	Neuchatel	950-850	#287	12
AUV-2740 Latanium	nail head	97.06	0.17	0.09	Neuchatel	950-850	#288	12
AUV-2740 Latanium	nail head	96.98	bdl	bdl	Neuchatel	950-850	#289	12
AUV-2740 Latanium	nail head	95.33	bdl	bdl	Neuchatel	950-850	#290	12
AUV-2740 Latanium	nail head	96.00	bdl	bdl	Neuchatel	950-850	#291	12
HR-93 Latanium	pin	91.66	0.35	bdl	Neuchatel	950-850	#292	12
HR-93 Latanium	pin	89.14	0.38	bdl	Neuchatel	950-850	#293	12
HR-93 Latanium	pin	90.67	0.19	bdl	Neuchatel	950-850	#295	12
HR-93 Latanium	pin	91.84	0.31	bdl	Neuchatel	950-850	#296	12
HR-110 Latanium	long pin	82.11	0.68	bdl	Neuchatel	950-850	#297	12
HR-110 Latanium	long pin	95.80	0.23	bdl	Neuchatel	950-850	#298	12
HR-110 Latanium	long pin	93.96	0.51	bdl	Neuchatel	950-850	#299	12
HR-110 Latanium	long pin	92.61	0.49	bdl	Neuchatel	950-850	#300	12
AO22953 le_Louvre	Halberd sideA	89.00	0.35	bdl	Luristan	IA transition	#587	12
AO22953 le_Louvre	Halberd sideA	91.99	0.31	bdl	Luristan	IA transition	#588	12
AO22953 le_Louvre	Halberd sideA	90.72	0.30	bdl	Luristan	IA transition	#589	12
AO22953 le_Louvre	Halberd sideB	88.83	0.36	bdl	Luristan	IA transition	#591	12
AO20882 le_Louvre	Adze side A	79.53	bdl	bdl	Luristan	IA transition	#592	12
AO20882 le_Louvre	Adze side A	84.89	0.26	bdl	Luristan	IA transition	#593	12
AO20882 le_Louvre	Adze side B	85.77	0.32	bdl	Luristan	IA transition	#594	12
N-1 Latanium	adze	96.68	0.21	bdl	Sweden	?	#301	12
N-1 Latanium	adze	96.98	0.48	bdl	Sweden	?	#302	12
N-1 Latanium	adze	96.28	0.56	bdl	Sweden	?	#303	12
Gresham ship UCL	bloomery iron	98.81	0.10	0.00	London	ca. 1580 AD	#131	13
Gresham ship UCL	bloomery iron	98.85	0.20	0.02	London	ca. 1580 AD	#132	13
Gresham ship UCL	bloomery iron	98.88	0.19	0.01	London	ca. 1580 AD	#133	13
M1 St_Germain-en-laye	Ingot 01	86.12	0.45	0.27	Marsal	ca 600 BCE	#178	14
M1 St_Germain-en-laye	Ingot 01	94.73	0.29	0.21	Marsal	ca 600 BCE	#179	14
M1 St_Germain-en-laye	Ingot 01	93.24	0.30	0.16	Marsal	ca 600 BCE	#180	14
M1 St_Germain-en-laye	Ingot 01	96.00	0.28	0.16	Marsal	ca 600 BCE	#181	14
M1 St_Germain-en-laye	Ingot 01	92.52	0.33	0.20	Marsal	ca 600 BCE	Mean	14

Table A5 (part 1): p-XRF analyses of transition and Iron Age specimens discussed during this study.  
12 This study (p-XRF) 14 Bertaux (1990) and Olivier (pers. com. 2016) 13 Auer and Firth (2007)

Specimen	type	Fe	Co	Ni	Locality	Age BCE	analysis #	ref
M2 St_Germain-en-laye	Ingot 02	97.11	0.15	bdl	Marsal	ca 600 BCE	#182	14
M2 St_Germain-en-laye	Ingot 02	97.72	0.09	bdl	Marsal	ca 600 BCE	#183	14
M2 St_Germain-en-laye	Ingot 02	92.90	0.23	bdl	Marsal	ca 600 BCE	#184	14
M2 St_Germain-en-laye	Ingot 02	96.88	0.20	bdl	Marsal	ca 600 BCE	#185	14
M2 St_Germain-en-laye	Ingot 02	96.15	0.17	bdl	Marsal	ca 600 BCE	Mean	14
M3 St_Germain-en-laye	Ingot 03	97.98	0.15	bdl	Marsal	ca 600 BCE	#186	14
M3 St_Germain-en-laye	Ingot 03	97.67	0.14	0.16	Marsal	ca 600 BCE	#187	14
M3 St_Germain-en-laye	Ingot 03	97.98	0.08	0.00	Marsal	ca 600 BCE	#188	14
M3 St_Germain-en-laye	Ingot 03	98.40	0.07	0.20	Marsal	ca 600 BCE	#189	14
M3 St_Germain-en-laye	Ingot 03	98.01	0.11	0.12	Marsal	ca 600 BCE	Mean	14
M4 St_Germain-en-laye	Ingot 04	97.37	0.00	bdl	Marsal	ca 600 BCE	#192	14
M4 St_Germain-en-laye	Ingot 04	95.94	0.10	bdl	Marsal	ca 600 BCE	#193	14
M4 St_Germain-en-laye	Ingot 04	97.50	0.08	bdl	Marsal	ca 600 BCE	#194	14
M4 St_Germain-en-laye	Ingot 04	95.50	0.09	bdl	Marsal	ca 600 BCE	#195	14
M4 St_Germain-en-laye	Ingot 04	96.58	0.07	bdl	Marsal	ca 600 BCE	Mean	14
M5 St_Germain-en-laye	Ingot 05	98.64	bdl	bdl	Marsal	ca 600 BCE	#170	14
M5 St_Germain-en-laye	Ingot 05	97.38	bdl	bdl	Marsal	ca 600 BCE	#171	14
M5 St_Germain-en-laye	Ingot 05	98.18	bdl	bdl	Marsal	ca 600 BCE	#172	14
M5 St_Germain-en-laye	Ingot 05	98.83	0.26	bdl	Marsal	ca 600 BCE	#173	14
M5 St_Germain-en-laye	Ingot 05	98.26	0.26	bdl	Marsal	ca 600 BCE	Mean	14
M6 St_Germain-en-laye	Ingot 06	92.41	0.19	0.14	Marsal	ca 600 BCE	#174	14
M6 St_Germain-en-laye	Ingot 06	97.71	0.13	0.28	Marsal	ca 600 BCE	#175	14
M6 St_Germain-en-laye	Ingot 06	97.60	0.11	0.17	Marsal	ca 600 BCE	#176	14
M6 St_Germain-en-laye	Ingot 06	96.51	0.33	0.37	Marsal	ca 600 BCE	#177	14
M6 St_Germain-en-laye	Ingot 06	96.06	0.19	0.24	Marsal	ca 600 BCE	Mean	14
Bloomery iron	polished	97.85	bdl	0.09	Morocco	?	#136	13
Bloomery iron	oxidized	83.28	0.05	0.11	Morocco	?	#137	13
Bloomery iron	oxidized	90.40	0.17	bdl	Morocco	?	#134	13
Bloomery iron	oxidized	94.32	0.15	bdl	Morocco	?	#135	13
Bloomery iron	polished	99.82	0.18	bdl	Morocco	?	#162	13
Cz.I-9:60	bracelet	78.20	0.49	15.50	Czestochowa	Hallstatt D	#417	15
Cz.I-9:60	bracelet	80.70	0.66	17.50	Czestochowa	Hallstatt D	#418	15
Cz.I-9:60	bracelet	77.20	0.61	18.40	Czestochowa	Hallstatt D	#419	15
Cz.I-9:60	bracelet	78.70	0.59	17.13	Czestochowa	Hallstatt D	Mean	15
Cz.I-294:61	bracelet	78.90	0.60	18.10	Czestochowa	Hallstatt D	#12	15
Cz.I-294:61	bracelet	75.90	0.73	21.20	Czestochowa	Hallstatt D	#13	15
Cz.I-294:61	bracelet	77.40	0.67	19.65	Czestochowa	Hallstatt D	Mean	15
polished section	axe	92.11	0.29	7.60	Wietrzno	Hallstatt C	#29	15
polished section	axe	92.91	0.22	6.87	Wietrzno	Hallstatt C	#34	15
polished section	axe	93.21	0.20	6.59	Wietrzno	Hallstatt C	#35	15
polished section	axe	94.50	0.25	5.25	Wietrzno	Hallstatt C	#14	15
polished section	axe	93.18	0.24	6.58	Wietrzno	Hallstatt C	Mean	15

Table A5 (part 2): p-XRF analyses of Iron Age specimens discussed during this study.  
14 Bertaux (1990) and Olivier (pers. com. 2016)                      15 Piaskowki (1982)