

Contributions of oxygen vacancies and titanium interstitials to the band gap states of reduced titania

Jingfeng Li,^{*} Rémi Lazzari,[†] Stéphane Chenot,[‡] and Jacques Jupille[§]
 CNRS, Sorbonne Universités, UPMC Univ Paris 06, UMR 7588,
 Institut des NanoSciences de Paris, 75005 Paris, France

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The spectroscopic fingerprints of the point defects of titanium dioxide remain highly controversial. Seemingly indisputable experiments lead to conflicting conclusions in which oxygen vacancies and titanium interstitials are alternately referred to as the primary origin of the Ti 3d band gap states. We report on experiments performed by electron energy loss spectroscopy whose key is the direct annealing of only the very surface of rutile $\text{TiO}_2(110)$ crystals and the simultaneous measurement of its temperature via the Bose-Einstein loss/gain ratio. By surface preparations involving reactions with oxygen and water vapor, in particular under electron irradiation, vacancies- and interstitials-related band gap states are singled out. Off-specular measurements reveal that both types of defects contribute to a unique charge distribution that peaks in subsurface layers with a common dispersive behavior.

Already used in self-cleaning coatings and dye-sensitized solar cells, titanium dioxide TiO_2 offers promising solutions to water/air purification and water splitting^{42,43}. Its rich physics and chemistry is mostly tied to the reduced TiO_{2-x} form in which dominant point defects, bridging oxygen vacancies ($\text{O}_b\text{-vac}$) and titanium interstitials (Ti_{int}) widely explored on rutile (110)⁴³⁻⁴⁷ (Fig. 1), play a pivotal role. At the origin of the n-type electron conductivity, excess electrons generated by defects populate Ti 3d related band gap states (BGS). They have long been associated to $\text{O}_b\text{-vac}$ (herein labelled V-BGS) because O_2 dissociation both eliminates $\text{O}_b\text{-vac}$ and heals them by a charge transfer toward O_b and O adatoms (O_{ad}) adsorbed on five-coordinated Ti_{5c} ⁴⁸. The model became hotly debated when vacancy-free $\text{TiO}_2(110)$ obtained by reacting off H from hydroxylated TiO_{2-x} was shown to retain most BGS that were consequently related to Ti_{int} (I-BGS)⁴⁵. Their healing by O_2 was explained by a charge donation from Ti_{int} that promotes a non-vacancy-related O_2 dissociation⁴⁵. Consistently, the TiO_x islands formed upon annealing O-covered rutile^{45,49-53} were attributed to reaction of O_{ad} with Ti_{int} which, like the reoxidation of TiO_{2-x} ⁵⁴, relies on the diffusion of Ti_{int} in rutile above 400 K. Finally, extra O_{ad} atoms (relative to dissociation on $\text{O}_b\text{-vac}$) obtained by reacting O_2 with TiO_{2-x} , were associated to charge transfer from Ti_{int} ^{45,55}. Additional supports to the O-vacancy model⁴⁸ feed the controversy. The oxidative chemistry of TiO_{2-x} was suggested to be controlled by donor species ($\text{O}_b\text{-vac}$ and OH) rather than Ti_{int} ⁵⁶ and, based on a relationship of proportionality between $\text{O}_b\text{-vac}$ counting by microscopy and photoemission signal, BGS appeared to mostly stem from $\text{O}_b\text{-vac}$ ⁵⁷.

The crux of the debate is that the signatures of V-BGS and I-BGS look identical. The case is tackled herein by the rarely used high-resolution electron-energy loss spectroscopy (HREELS). The key experiment to single out BGS contributions is the quick annealing (up to 1000 K within a few seconds) of the surface only via a

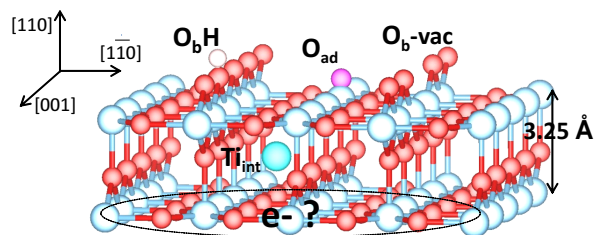


FIG. 1. Ball model of defective $\text{TiO}_2(110)$ (Ti, O, H, Ti_{int} , O_{ad} are blue, red, white, cyan and pink spheres, respectively); bridging oxygen rows involve $\text{O}_b\text{-vac}$ and O_bH ; O_{ad} lie on Ti_{5c} ; Ti_{int} occupy octahedral sites⁵⁸; the localisation of excess electrons (question mark) associated to defects is under debate.

hot filament facing the crystal within the spectrometer, while probing T (temperature) via the Bose-Einstein statistics (Fig. S1 in Supplemental Material⁵⁹) of the loss/gain phonon ratio. -Annealed samples are labelled SA-T- TiO_2 hereafter. Controversial or unexplained data involving annealing treatments, electron bombardment and exposures to O_2 and H_2O , have been revisited to feature I-BGS and V-BGS. Finally, BGS in-depth profiles⁶⁰⁻⁶⁴ were probed by off-specular EELS.

$\text{TiO}_2(110)$ single crystals were mounted in an ultra-high vacuum (UHV) set-up⁶⁵ equipped with an EELS spectrometer (Sect. S1 in Supplemental Material⁵⁹). Sputtering and annealing $\text{TiO}_2(110)$ resulted in reduced dark blue color samples R- TiO_2 of which surfaces could be reoxidized (O- TiO_2) by 30 mins annealing (1100 K)⁵⁰ and cooling in O_2 (5×10^{-6} mbar) at a rate of $\sim 30 \text{ K}\cdot\text{min}^{-1}$ (Sect. S1 in Supplemental Material⁵⁹). EEL spectra were collected between 100 K and 300 K. Unless stated, they were recorded in specular geometry (incident angle $\Theta_I = 60^\circ$, incident plane along $[1\bar{1}0]$ direction) at an energy of $E_I = 38 \text{ eV}$. Typical EELS spectra (Fig. S3 in Supplemental Material⁵⁹) show

broad BGS features at ~ 0.8 eV before the onset of band-to-band excitations at 3.2 eV. Bulk sensitivity at phonon energies provides a reliable intensity normalization (Sect. S3 in Supplemental Material⁵⁹). Exposures are expressed in Langmuir ($1 \text{ L} = 1.33 \times 10^{-6} \text{ mbar}\cdot\text{s}^{-1}$). EELS spectra recorded on R-TiO₂ at 100 K (Fig. S3a in Supplemental Material⁵⁹) and 300 K (Fig. S3b in Supplemental Material⁵⁹) under O₂ show progressive BGS healing, although kinetics are different. Fast and slow decrease in BGS intensity that are observed at 100 K and 300 K, respectively, rely on mechanisms that are under debate^{51,54,66}. Importantly, the similarity of the final BGS intensities (Fig. S3a-inset in Supplemental Material⁵⁹) indicates two equivalent healing processes.

The preparation of SA-TiO₂ surfaces is central in pinpointing categories of point defects. An indisputable proof of the existence of I-BGS is provided by the defect peak growing on SA-420K-TiO₂ (Fig. 2a). The formation of O_b-vac at 420 K being excluded, this peak is assigned to I-BGS formed via an outward Ti_{int} diffusion^{45,67-69} as supported by the square root of time dependence of the BGS intensity (Fig. 2a-inset).

In contrast to the slow kinetics of BGS formation on SA-420K-TiO₂ (Fig. 2a), SA-TiO₂ surfaces prepared by annealing of 10 s above 800 K (Fig. 2b) show strong BGS whose intensity rises with T. Indeed, O_b-vac are expected to form⁴⁴, but it is unclear whether defect states are I-BGS or V-BGS. More is learned on I-BGS by preparing SA-TiO₂ at increasing temperature followed by 20 L of O₂ at 300 K (Fig. 2c, steps 1 to 4; inset: BGS intensities). Firstly, healing I-BGS from SA-420K-TiO₂ by O₂ (step 1) validates the non-vacancy-related O₂ dissociation associated to charge transfer from Ti_{int}^{45,55}. Then, increasing BGS are observed on SA-970K-TiO₂ and SA-1140K-TiO₂ although, surprisingly, the residual BGS observed after O₂ exposure continuously decreases through steps 1, 2, 4 (Fig. 2c). Extra-healing of O_b-vac is excluded since there is no reason why reacting O_b-vac with 20 L of O₂ should leave fewer O_b-vac sites intact while their initial concentration is higher. Therefore, Fig. 2c shows that the commonly observed residual BGS (Fig. S4 in Supplemental Material⁵⁹ and Refs.^{45,57,66}) mostly involve I-BGS. Annealing O-covered surfaces at increasing temperature triggers outward Ti_{int} diffusion^{51,70} from increasingly deeper layers. Assuming that Ti_{int} segregate on step 2 (SA-970K-TiO₂) over deeper layers than that reached by SA-420K-TiO₂ (step 3) explains the marginal BGS change on step 3, in apparent contrast with step 1. The depletion in subsurface Ti_{int} which progressively extends inward (step 1 to 4, Fig. 2c) suggests that annealing/oxidation cycles can lead to a BGS-free surface region.

The existence of V-BGS has not been singled out yet. Often accepted⁵⁷, the attribution to V-BGS of O_b-vac created by electron stimulated desorption (ESD) via the Knotek-Feibelman process^{71,72} is nevertheless strongly

discussed^{73,74}. Electron bombarded surfaces (E-TiO₂) were prepared by illuminating O-TiO₂ by a 75 eV electron beam (current density $\sim 1 \mu\text{A}\cdot\text{cm}^{-2}$)^{71,72}. Totally healed at 100 K by only 1 L O₂ (Fig. 3a), the resulting BGS thus stem from very surface defects but, still, there is no indisputable evidence that they are V-BGS. A clue came from the observation that ESD-induced BGS level off at rather low intensity after 5 min, to remain stable for hours (not shown). Comparison with other groups show that rather weak BGS were obtained with electron fluxes of $1.25 \cdot 10^{12} \text{ e}\cdot\text{cm}^{-2}\cdot\text{s}^{-1}$ ⁷⁵ and $6.25 \cdot 10^{12} \text{ e}\cdot\text{cm}^{-2}\cdot\text{s}^{-1}$ (this work) while $1.25 \cdot 10^{15} \text{ e}\cdot\text{cm}^{-2}\cdot\text{s}^{-1}$ ⁷⁶ and $3 \cdot 10^{15} \text{ e}\cdot\text{cm}^{-2}\cdot\text{s}^{-1}$ ⁷⁷ led to strong BGS, although fluences were similar ($3 \cdot 10^{15} \text{ e}\cdot\text{cm}^{-2}$ ⁷⁵; $2 \cdot 10^{15} \text{ e}\cdot\text{cm}^{-2}$ (this work); $6 - 25 \cdot 10^{15} \text{ e}\cdot\text{cm}^{-2}$ ⁷⁷). Clearly, the electron-induced BGS intensity depends on the flux rather than on the fluence, which suggests a continuous healing of related O_b-vac. We may wonder what is the combined effect of residual water and ESD removal of H adatoms (threshold at 21-22 eV^{71,72}). Although reacting O_b-vac by H₂O does not heal BGS^{45,76-78}, H₂O under 38 eV electrons does heal BGS of SA-970K-TiO₂ (Fig. 3b). Consistently, 22 eV electrons (threshold for H ESD) have no effect (Fig. 3c). In contrast, I-BGS from SA-420K-TiO₂ are not healed at 38 eV when exposed to H₂O (Fig. 3d), although, as in Fig. 2d (step 1), they are healed by O₂. Beyond the demonstration of the existence of V-BGS and I-BGS, the healing of V-BGS via H₂O adsorption provides a mean of distinguishing the two states, as in Fig. 3e, where R-TiO₂ is exposed to H₂O, under a 38 eV electron beam (healing V-BGS) and then exposed to O₂ (healing I-BGS). Notably, the plateau reached by the intensity of the ESD-induced BGS supports their V-BGS nature because, if I-BGS were created, the inability of H₂O to heal them would result in a continuous increase in their intensity.

Finally, the concentration profiles of the excess electrons and the dispersion of the associated BGS were explored by off-specular EELS (Fig. 4) where the cross section gets more surface sensitive by switching from dipolar to impact regime⁷⁹. The inverse of the modulus of the wave vector transfer parallel to the surface gives an estimate of the probing depth⁷⁹:

$$k_{\parallel} = \frac{\sqrt{2mE_I}}{\hbar} \sin \Theta_I - \frac{\sqrt{2m(E_I - \hbar\omega)}}{\hbar} \sin \Theta_S \quad (1)$$

where E_I is the incident electron energy, $\hbar\omega$ the energy loss, $2\pi\hbar$ the Planck constant, Θ_I and Θ_S the incident and scattering angle, respectively (Sect. S3 in Supplemental Material⁵⁹). BGS were recorded off-specular at 300 K, before and after 20 L O₂ exposure, on R-TiO₂ (Fig. 4a) and SA-TiO₂ (Fig. S5 in Supplemental Material⁵⁹). Data were normalized to interband transitions whose probing depth does not change significantly with Θ_S (Fig. 4a-inset and Fig. S2c in Supplemental Material⁵⁹). The dramatic decrease in BGS intensity with Θ_S shows that excess charges are not located at

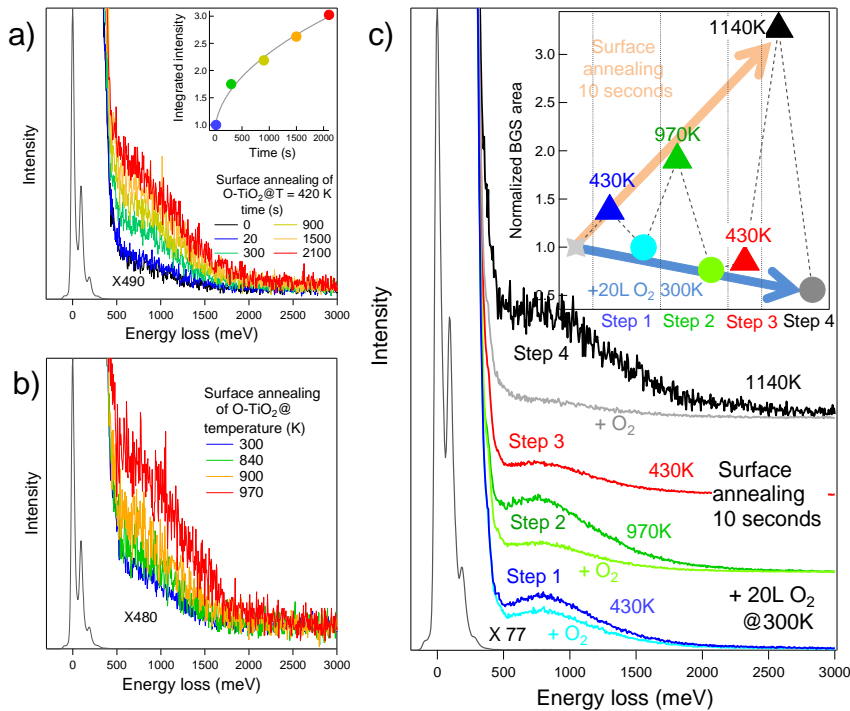


FIG. 2. Singling out BGS from Ti_{int} by surface annealing of O-TiO₂. EELS measurements at 300 K: a) BGS recorded on SA-420K-TiO₂ versus time; inset: time evolution of the BGS area (500-2500 meV) fitted by a square-root function; b) BGS from SA-TiO₂ prepared at different temperatures during 10 s; c) BGS recorded upon successive treatments (step 1 to 4): surface annealing (10 s) at given temperatures (strong color) and then 20 L of O₂ (light color) at 300 K (from bottom to top). Inset: BGS intensities step by step: black dotted lines show the order of treatments, the orange and blue eye-guide arrows evidence an increase in BGS intensity upon increase in annealing temperature and a corresponding decrease in residual intensity after 20 L O₂.

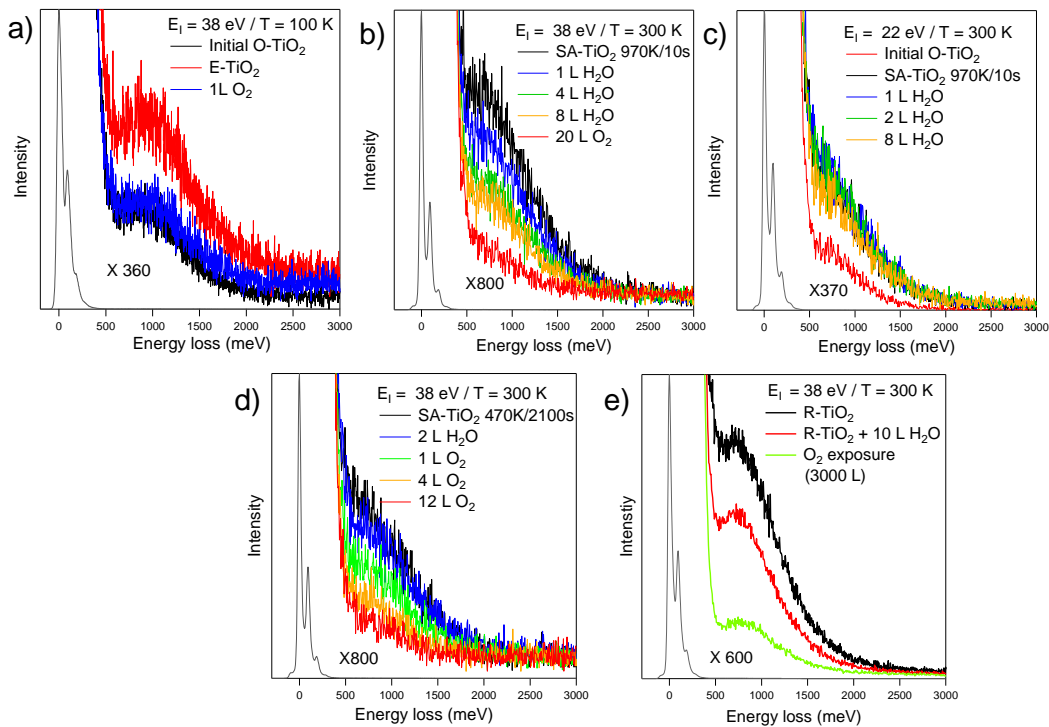


FIG. 3. Distinguishing V-BGS from I-BGS by exposure to H₂O and O₂: a) BGS of a E-TiO₂ surface are totally healed by 1 L of O₂ at 100 K; b) BGS of SA-970K-TiO₂ surface (10 s annealing) are healed at 300 K by H₂O under an electron beam at 38 eV; c) Same experiment as in b), but at 22 eV; d) BGS of SA-470K-TiO₂ (2100 s annealing) at 300 K are not affected by H₂O under electrons at 38 eV but, in contrast, are healed by O₂; e) BGS from R-TiO₂ are partially healed at 300 K by H₂O under electrons at 38 eV an even more cancelled by further O₂ exposure.

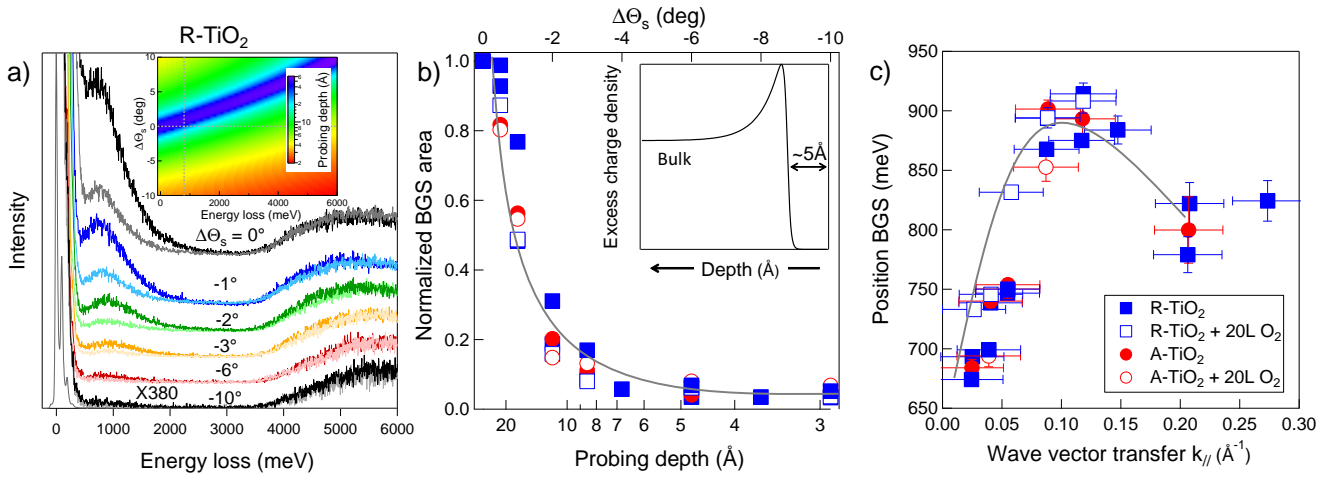


FIG. 4. Concentration profile and dispersion of BGS at 300 K: a) BGS of R-TiO₂ at different collection angles relative to specular direction $\Delta\Theta_S = \Theta_S - \Theta_I$ at fixed incidence $\Theta_I = 60^\circ$, as prepared (strong color) and after exposure to 20 L of O₂ (light color); inset: calculated off-specular probing depth (see text) averaged over a detector aperture of $\sim 1^\circ$ versus energy loss and angular position $\Delta\Theta_S$ at fixed incident angle at an energy of 38 eV; the grey dotted lines correspond to specular geometry and to BGS energy loss; b) BGS area normalized to that recorded in specular geometry (maximum probing depth) versus probing depth (bottom scale) and $\Delta\Theta_S$ (top): R-TiO₂ and SA-970K-TiO₂, as prepared and after 20 L of O₂ at 300 K, are compared. Inset: schematic concentration profile of excess electrons in the subsurface region; c) Dispersion of the position of the BGS with the wave vector transfer $k_{||}$. $k_{||}$ -error stems from a detector aperture of $\sim 1^\circ$. Grey lines in (b,c) are guide for eyes. Color codes for (b,c) are given in (c).

the extreme surface, as confirmed by the poor effect of O₂ at the highest Θ_S . Normalized BGS area versus probing depth from R-TiO₂ and SA-TiO₂ (Fig. 4b) and BGS positions versus $k_{||}$ (Fig. 4c) show that the four series of data are strikingly similar (Fig. 4b,c). This shows that the electron localization and state dispersion are independent of whether they are related to O_b-vac (R-TiO₂ and SA-TiO₂) or to Ti_{int} (after O₂ exposure). Generalizing previous theory⁸⁰ and experiment⁶¹, the observation directly proves that the surface/subsurface electrostatic potentials dictate the location of the excess charges. A unique representation of the charge distribution can be drawn for all surfaces, either reduced or oxidized (Fig. 4b-inset). Going inward, after an almost charge-free atomic layer, excess charges pass through a maximum, consistently with photoelectron diffraction^{60,61} and calculations^{62–64,78,81}. Going deeper, the defect density lowers to reach the bulk value. The dispersive trend up to a cut-off of $k_{||} \sim 0.1 \text{ \AA}^{-1}$ (Fig. 4c), demonstrates the transport behavior of excess electrons down to a distance ($\sim 10 \text{ \AA}$) that can be interpreted as the polaron radius at 300 K^{63,64,76,78,81,82}.

To summarize, vacancy- and interstitial-related BGS have been singled out by EELS experiments based on annealing only the very surface. The associated excess electrons contribute to a unique subsurface distribution whose profile and dispersive behavior are robust with respect to the nature of the defects. In contrast, their role in surface chemistry relies on the specificity of the defects. The dual origin of BGS allows the understanding of seemingly contradictory observations and opens up new ways of interpreting experiments carried out on TiO₂ surfaces.

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* jingfeng.li1987@gmail.com

† Corresponding author :remi.lazzari@insp.jussieu.fr

‡ stephane.chenot@insp.jussieu.fr

§ jacques.jupille@insp.jussieu.fr

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SUPPLEMENTARY INFORMATION

Contributions of oxygen vacancies and titanium interstitials to the band gap states of reduced titania

Jingfeng Li,^{*} Rémi Lazzari,[†] Stéphane Chenot,[‡] and Jacques Jupille[§]
CNRS, UMR 7588, Institut des NanoSciences de Paris, 75005 Paris, France and
Sorbonne Universités, UPMC Univ Paris 06, UMR 7588,
Institut des NanoSciences de Paris, 75005 Paris, France

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The supplementary information describes the experimental set-up (Sect. I), the annealing of the surface by a hot filament placed in front of the sample (Sect. II), the determination of the probing depth in electron energy loss spectroscopy (Sect. III). It also involves complementary figures (Sect. IV).

I. EXPERIMENTAL

Experiments were performed in an ultra-high vacuum vessel¹ including a preparation chamber equipped with a LEED-Auger device (Low Energy Electron Diffraction) (base pressure 3.10^{-10} mbar) and a main chamber (base pressure 5.10^{-11} mbar) housing a High Resolution Electron Energy Loss spectrometer (HREELS). The thermal contact between the TiO₂ sample (from Mateck GmbH²) and the metallic backplate was insured by firmly clamping the double-side polished substrate onto the polished backplate with an intermediate gold foil in between. TiO₂(110) surfaces were prepared through several cycles of sputtering/annealing in vacuum (1 keV/1100 K) until reaching a sharp (1×1) LEED pattern and a surface free of contaminants as judged from Auger spectroscopy and EELS signal. The stoichiometry defects in these reduced sample (R-TiO₂) yielded a sufficiently high conductivity to prevent any charge effect during measurements.

EELS experiments have been performed with a LK-2000 spectrometer^{3,4} which consists of a dual cylindrical monochromator at 127° and a rotating analyzer⁵. While the incident angle is fixed at $\Theta_I = 60^\circ$, the collection angle can be changed from $\Theta_S = 20^\circ$ to $\Theta_S = 70^\circ$. Most EELS spectra have been acquired along the specular direction at a beam energy of $E_I = 38$ eV with a resolution around 50 meV as given by the full-width at half-maximum of the elastic peak (counting rate close to 10^6 s⁻¹). The beam energy was slightly higher than the minimum energy of the Ti 3p core level excitation of the Auger transition (32-34 eV) for vacancy creation by the Knotek-Feibelman process^{6,7}. However, as estimated from the cross section of the literature at this energy, any sizable beam damage is excluded since it would require several days of exposure at the current density of a few tens of nA which is used herein. Test experiments after irradiation over a time scale much larger than regular experiments showed no Band-Gap States (BGS) inten-

sity change. Complementary phonon spectra recorded in HREELS at higher resolution (around 8 meV, $E_I = 8$ eV, counting rate $> 10^5$ s⁻¹) were used to determine surface temperature during annealing with a filament (see Sect. II). Surface exposure was performed by backfilling the EELS chamber with research-grade pure O₂ and distilled H₂O that was outgassed through several pump-freeze cycles. Exposure are given in Langmuir, 1 L corresponding to an exposure of $1.33 \cdot 10^{-6}$ mbar.s⁻¹. Cooling at 100 K was achieved through a copper braid connecting the end of the 5-axis manipulator of the EELS chamber and a tank cooled through circulating liquid nitrogen. All spectra have been acquired with the scattering plane perpendicular to the oxygen bridging rows *i.e.* along the $[1\bar{1}0]$ bulk direction. BGS area and position have been determined through a Lorentzian fit of peaks between 500 and 2500 meV. As described in Refs. 8–10, three Fuchs-Kliwer phonons at 45, 55 and 95 meV and their multiple/comboination excitations dominate the loss spectrum of TiO₂(110) at low energy losses before the BGS and the onset of interband transitions above 3.2 eV (see for instance Fig.4-a of the article).

Besides the above mentioned reduced R-TiO₂ surface, several types of TiO₂(110) surface preparations were used. R-TiO₂ could be reoxidised (O-TiO₂) by annealing 30 mins at 1100 K and cooling down to 300 K, both in O₂ ($p = 5 \times 10^{-6}$ mbar), a treatment known to induce a regrowth of surface TiO₂ islands by diffusion of Ti_{int} and to heal O_b-vac^{11,12} (rate ~ 30 K.min⁻¹). A E-TiO₂ surface was prepared by electron bombardment with a defocused electron beam (current of $\sim 1 \mu\text{A.cm}^{-2}$, beam energy of 75 eV to increase the oxygen electron stimulated desorption cross section) produced by an ancillary electron gun that was well-degassed prior to be used by irradiating the sample manipulator head; most importantly, the distance to the sample and the screening of the gun filament exclude any sample heating (See Section II). The last preparation (labeled SA-TiO₂) consisted in annealing only the sample surface by means of a hot filament placed in front of the crystal. It is described in what follows.

II. SURFACE ANNEALING WITH A FILAMENT

In the EELS measurement position, the surface could be radiatively heated in a controlled way by means of a tungsten filament set in front of the grounded sample. Those samples of which Surfaces have been Annealed at temperature T (K) are labelled SA- T -TiO₂. Prior to thermal treatment, the filament as well as the sample manipulator were carefully outgased, and during surface heating, the chamber pressure kept in the low 10^{-10} mbar; no burst of pressure upon switching on the filament was observed. After treatment, no trace of contamination was found by Auger spectroscopy (by photoemission performed in an other vacuum system) and the EELS spectrum could be cycled reversibly back and forth with oxygen exposure. Surface temperature could be changed by modifying either the filament-surface distance (a few mms to 20 mm) or the filament power (up to 25 W). Such a surface annealing avoids the surface/bulk thermal equilibrium which is established during annealing by the back of the sample. An additional advantage of the HREELS technique is the ability to determine the temperature of the surface during annealing through the Bose-Einstein distribution of losses and gains in the phonon region¹³⁻²¹. Their ratio is given by the Boltzmann factor $\exp(-\hbar\omega/k_B T)$ where $\hbar\omega$ is the energy loss/gain and k_B the Boltzmann constant (Fig. S1-a). The obtained stationary temperature corresponds to that of the near surface region, and more precisely to the EELS probing depth of phonon losses which typically amounts to about ten nanometers (see Sect. III and Fig. S2-b). A reproducible calibration specific to the present work could be performed as a function of the filament power P at fixed filament-sample distance d (Fig. S1-b) or as a function of the distance at fixed power (Fig. S1-c). Temperatures could not be measured beyond 650 K (Fig. S1-c) because the trajectories of the HREELS electron beam were too strongly perturbed by the filament to maintain a reasonable counting rate. Therefore, extrapolated values were used from the fitted $T \sim P/d^2$ behavior (Figs. S1-b,c). Creation of defects through spurious electrons emitted by the filament^{6,7,22-24} is excluded. Indeed, a surface was annealed at high temperature by the hot filament which was either grounded or polarized at +100 V to insure that absolutely no electrons were emitted. The overlapping EELS spectra confirmed that only heating affects the BGS intensity. In addition, no change could be observed when the filament was negatively polarized to illuminate the surface by electrons which means that, within the specific geometry of the filament-substrate used herein, annealing effects fully dominate. Generally speaking, these observations suggest that care must be taken to avoid surface annealing when creating point defects by electron bombardment. BGS increase through atomic hydrogen adsorption due to thermal cracking of residual background molecules

(H₂, H₂O) on the hot filament can be ruled out from a straightforward evaluation of the order of magnitude of exposure. Assuming a perfect dissociation coefficient (H₂ → 2H) over the whole filament surface of 1 mm² (length 60 mm, diameter=0.15 mm), a partial pressure of H₂ of $\sim 10^{-10}$ mbar would give a yield of $2.2 \cdot 10^9$ H atom.s⁻¹ (the impinging rate is given by $p/\sqrt{2\pi mk_B T}$; T=300 K). Since the solid angle of the sample (1 cm²) seen from the filament (in the punctual filament approximation!) is around 0.02 rad (distance of 2 cm during the low temperature annealing), only $4.4 \cdot 10^7$ H atom.cm⁻².s⁻¹ can reach the surface. Since the monolayer (ML(O_b)) of bridging oxygen row is $5 \cdot 10^{14}$ cm⁻², the creation of 0.1 ML(O_b) would require an exposure of $1.1 \cdot 10^6$ s which is already ~ 500 times the longest exposure performed in our experiments. Indeed, literature shows that to obtain significant effect of atomic hydrogen on TiO₂ by dissociation on heated tungsten, exposures that are orders of magnitude above ours are required^{25,26}.

In the present work, two specific annealing conditions were chosen: (i) 420 K where creation of O_b-vac cannot happen, but diffusion of Ti_{int} is possible^{12,27-30} and (ii) around 1000 K where the creation of O_b-vac is systematically observed by bulk annealing³¹⁻³⁷. Two durations were compared: (i) a short one ($t = 10$ s) mainly at the highest temperature corresponding to a flash annealing and (ii) longer ones (up to $t > 1000$ s). When flash annealing at high temperature, any increase of the temperature has never been observed as measured by the thermocouple on our manipulator. After annealing the sample for 2100 s at 420 K, this measured "bulk" temperature was increased by only few degrees over a sample thickness of 0.5 mm. This gradient of temperature is due to the thermal mass of the Cu manipulator that acts as a thermal sink. Upon a short annealing of 10 s, a simple square root of time estimate indicates that a similar temperature increase would be observed at a distance of a few tens of microns of the surface and that, consequently the depth at which the temperature is homogeneous is much below the micron. All surface annealing treatments were performed on samples that had been oxidized by saturating amounts of oxygen at room temperature. The term "saturation" has a relative sense. It corresponds to exposures high enough to heal the largest part of the BGS intensity (*i.e.* at least 20 L at 300 K), as shown in Fig. S3.

III. PROBING DEPTH AND SPECTRA NORMALIZATION

In an EELS experiments, an incoming electron of wave vector \mathbf{k}_I and incident energy $E_I = \hbar^2 k_I^2 / 2m$ (m mass of the electron, $2\pi\hbar$ Planck constant) is scattered along a wave vector \mathbf{k}_S and loses/gains an energy $\hbar\omega$ ($E_S = \hbar^2 k_S^2 / 2m = E_I \pm \hbar\omega$) before being detected by

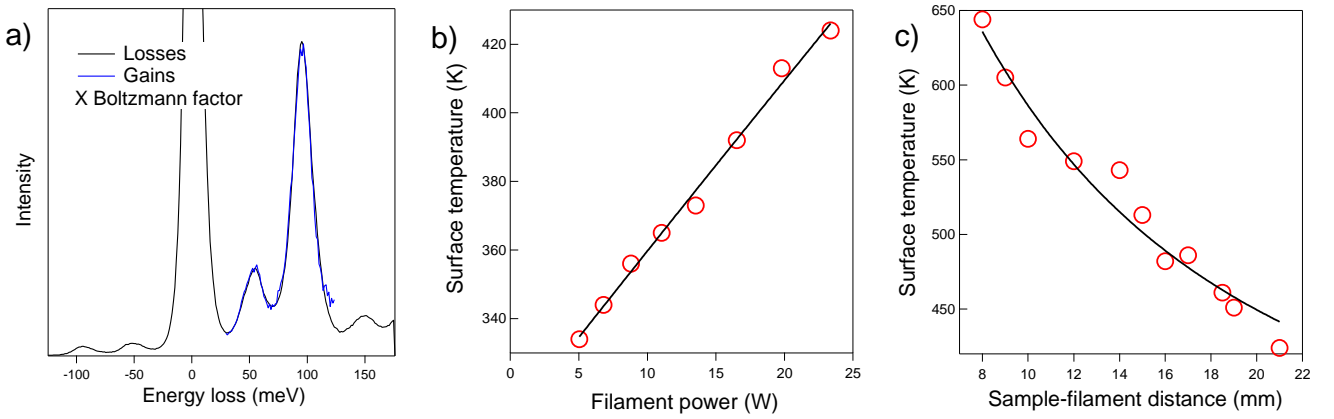


FIG. S1. a) HREELS spectrum of the phonon losses (black line) of $\text{TiO}_2(110)$ overlapped with the folded gains (blue line) multiplied by the Boltzmann factor. The accuracy of the temperature determination is of the order of 1 K. Spectra have been recorded at a beam energy of $E_I = 8$ eV and a resolution ~ 8 meV in the specular geometry. Measured surface temperature (red circles) b) versus filament power P at fixed sample-filament distance $d = 20$ mm or c) versus d at fixed $P = 23$ W. Best fits with a $T \sim P/d^2$ behavior are shown as black lines.

a spectrometer of acceptance θ_c . By moving the detector away from the specular direction, the interaction between the electron beam and the surface changes progressively from a long-range dipolar mode to a short-range impact scattering regime. The depth sensitivity of the measurement depends on the impact energy E_I , the energy loss $\hbar\omega$ and scattering geometry defined by the incident angle Θ_I , the average scattering angle Θ_S and the polar coordinates (θ_S, ϕ_S) that span the collection aperture θ_c of the detector (see Fig. S2-a). There is no precise definition of the probing depth in electron energy loss spectroscopy^{13,38}. Indeed, the successful dielectric theory¹³⁻²¹ that accounts for the dipolar scattering for $\Theta_I = \Theta_S$ shows a complex interplay between the profile of dielectric function and the cross section of energy loss through an angular integration over the so-called sensitivity function. But a reasonable estimate³⁸ is given by the evanescent decay length d_p of the electric field produced by the moving electron and through which it interacts with the substrate¹³⁻²¹. It is worth noticing that this probing depth is quite different from photoemission spectroscopy for which it is based on an escape depth of inelastically scattered electrons. According to the dielectric theory¹³⁻²¹, d_p is given by the inverse of the parallel wave vector transfer of the electron $k_{\parallel} = (k_S - k_I) \sin \Theta_I$ along the specular trajectory. To first order in energy loss ($\theta_E = \hbar\omega/2E_I \ll 1$), $d_p(\theta_E, \Theta_I) = 1/(k_I \theta_E \sin \Theta_I)$ is inversely proportional to the energy loss $\hbar\omega$ and the higher the beam energy the larger the probing depth. d_p has been plotted in Fig. S2-b for two beam energies $E_I = 8$ and 38 eV corresponding to high and low resolution measurements as performed in this work. It is also compared to its average over a typical circular detector

aperture^{5,13} of $\theta_c = 1^\circ$ as defined by:

$$\langle d_p \rangle (\theta_E, \Theta_I, \theta_c) = \frac{1}{\pi \theta_c^2} \int_0^{\theta_c} \theta_S d\theta_S \int_0^{2\pi} d\phi_S d_p(\theta_E, \Theta_I, \theta_S, \phi_S)$$

$$\frac{1}{d_p(\theta_E, \Theta_I, \theta_S, \phi_S)} = k_I [(\theta_E \sin \Theta_I - \theta_S \cos \phi_S \cos \Theta_I)^2 + (\theta_S \sin \phi_S)^2]^{1/2}. \quad (1)$$

The main effect of this detector integration is to smear out the nonphysical divergence at zero loss (Fig. S2-b,c; circles vs lines). Because of the bulk sensitivity of the phonon losses ($\langle d_p \rangle \simeq 100$ Å at $E_I = 8$ eV and $\langle d_p \rangle \simeq 50$ Å at $E_I = 38$ eV), specular spectra have been normalized throughout this work to the most intense phonon peak around 100 meV. Indeed, due to gas interaction with the filament of the EELS apparatus, absolute counting rate evolves during gas exposure. At the BGS energy, EELS at $E_I = 38$ eV probes the subsurface ($\langle d_p \rangle \simeq 30$ Å) and it becomes very surface sensitive in the case of interband transitions ($\langle d_p \rangle < 10$ Å). Regarding out-of-specular measurements, the definition of the probing depth is even less straightforward since there is a continuous transition from a strongly peaked long-range dipolar interaction close to the specular direction to a short-range impact interaction. Formally, the calculation of the loss cross section changes from an efficient description by the dielectric theory to more complex quantum mechanical treatment. But, owing to the same underlying electrostatic interaction between the electron and the sample, this transition of probing depth is still well accounted by the inverse of the wave vector transfer

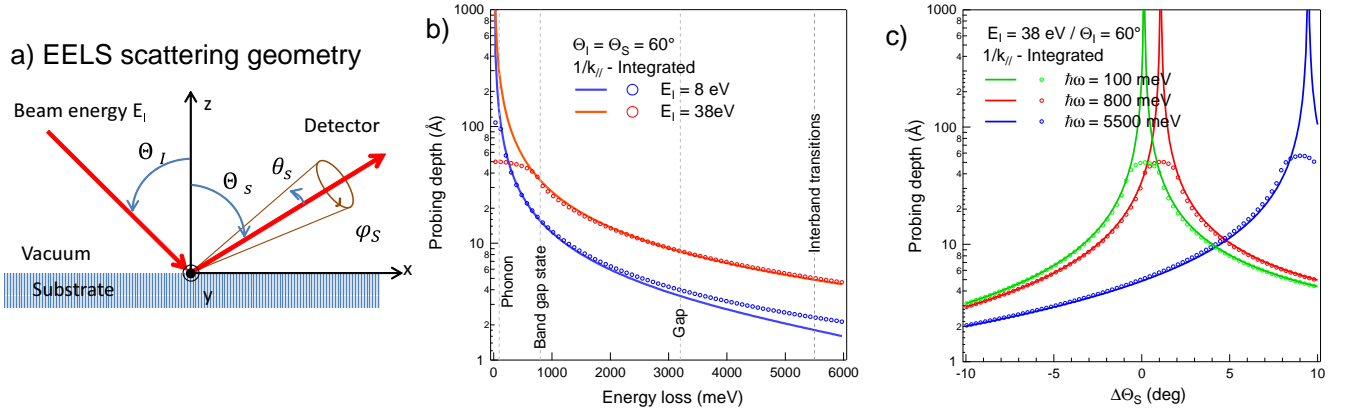


FIG. S2. a) Scattering geometry in an (HR)EELS experiment. An incoming electron of impact energy E_I impinges on the surface of the substrate at an incident angle Θ_I ; it is scattered off along a direction coplanar to the z -axis by its angles $(\Theta_S, \theta_s, \phi_s)$. $(\Theta_I = \Theta_S, \theta_s = 0)$ corresponds to the specular scattering geometry. Along its path, the electron exchanges quanta of energy $\hbar\omega \ll E_I$ with the substrate and is subjected to a total in-plane wave vector transfer k_{\parallel} . b) Probing depth d_p in the specular scattering geometry ($\Theta_I = \Theta_S = 60^\circ$) as function of energy loss $\hbar\omega$ at the two impact energies $E_I = 8$ and 38 eV (blue/red) used in this work. d_p and $\langle d_p \rangle$ (see text for definition) are compared. Typical loss energies for phonons, BGS, band gap or interband transitions are highlighted as grey dotted lines. c) Same as Fig. b but in out-of-specular geometry as a function of $\Delta\Theta_S = \Theta_S - \Theta_I$ for three energy losses representative of the phonon ($\hbar\omega = 100$ meV), BGS ($\hbar\omega = 800$ meV) and interband transitions ($\hbar\omega = 5500$ meV). In Figs. b,c, the maxima corresponds to conditions for which k_{\parallel} is close to zero.

$d_p^{os} = 1/k_{\parallel}^{os}$ (Fig. S2-a) given by:

$$k_{\parallel}^{os}(\theta_E, \Theta_I, \Theta_S, \theta_s, \phi_s) = k_I \left[(\theta_E \sin \Theta_S + \sin \Theta_I - \sin \Theta_S - \theta_s \cos \phi_s \cos \Theta_S)^2 + (\theta_s \sin \phi_s)^2 \right]^{1/2} \quad (2)$$

as well as the corresponding integrated value $\langle d_p^{os} \rangle$ over the detector aperture. As shown in the inset of Fig. 4-a of the paper and reminded in Fig. S2-c, $\langle d_p^{os} \rangle$ at an energy loss typical for BGS ($\hbar\omega = 800$ meV) quickly drops from 30 Å down to 2 Å as soon as the detection angle Θ_S is in off-specular direction. To avoid the resonant condition seen for $\Delta\Theta_S > 0$ corresponding to

$k_{\parallel}^{os}(\theta_s = 0) = 0$ *i.e.* $\sin \Theta_S = \sin \Theta_I / (1 + \theta_E)$ and to be more surface sensitive at all energies, data have been recorded only for negative offset $\Delta\Theta_S < 0$ (see Fig. 4-a of the paper and Fig. S5). At the opposite to phonon losses (Fig. S2-c, green curves), the probing depth for interband transitions does not change significantly for $\Delta\Theta_S < 0$ (Fig. S2-c, blue curves); therefore, this energy range has been used to normalize spectra and to analyse the depth dependence of BGS intensity in out-of-specular geometry (Fig. 4-b of the paper). The overlapping of spectra above the band gap validates *a posteriori* this assumption.

IV. COMPLEMENTARY FIGURES

* jingfeng.li1987@gmail.com

† Corresponding author :remi.lazzari@insp.jussieu.fr

‡ stephane.chenot@insp.jussieu.fr

§ jacques.jupille@insp.jussieu.fr

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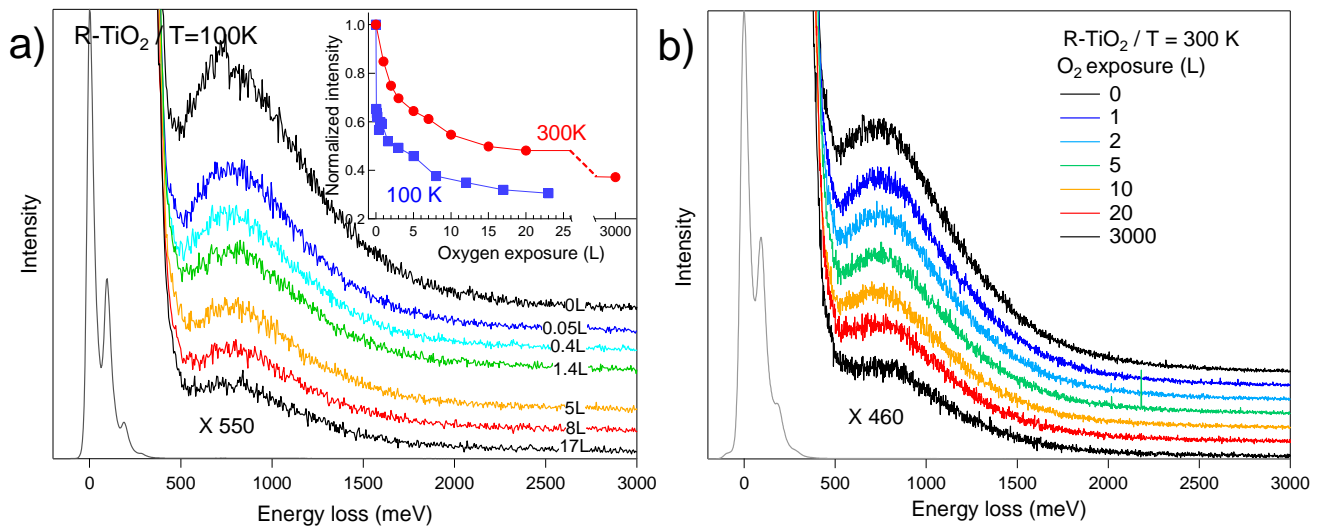


FIG. S3. Evolution of the intensity of BGS of R-TiO₂ upon O₂ exposure given in figure at a) 100 K and b) 300 K. Curves are offset for clarity. The healing of the BGS is much slower at 300 K. Inset of Fig. a shows the variation as a function of O₂ exposure of the BGS area (integrated intensity) between 500 and 2500 meV (normalized to the initial values) at 100 K (blue squares) and 300 K (red circles).

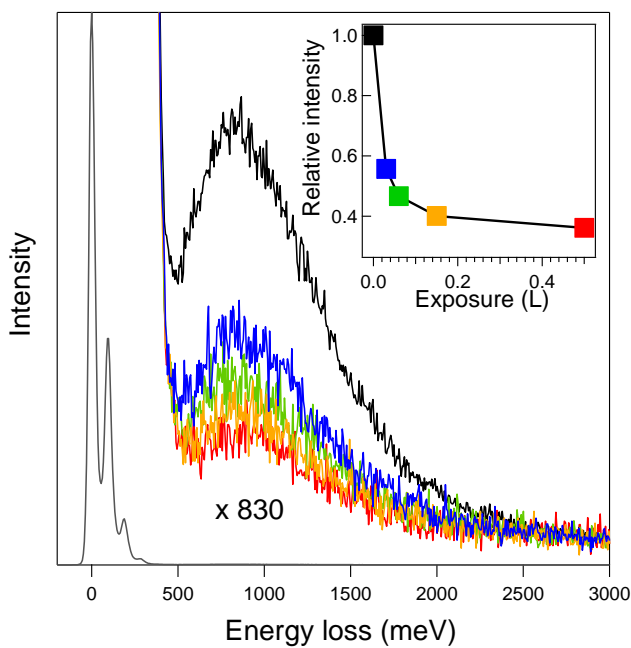


FIG. S4. Evolution of EELS spectra in the BGS region of a SA-TiO₂ surface annealed during 10 s at 970 K upon O₂ exposure at 100 K. The inset showing the evolution of the BGS integrated intensity demonstrates the very fast healing of BGS created by surface annealing.

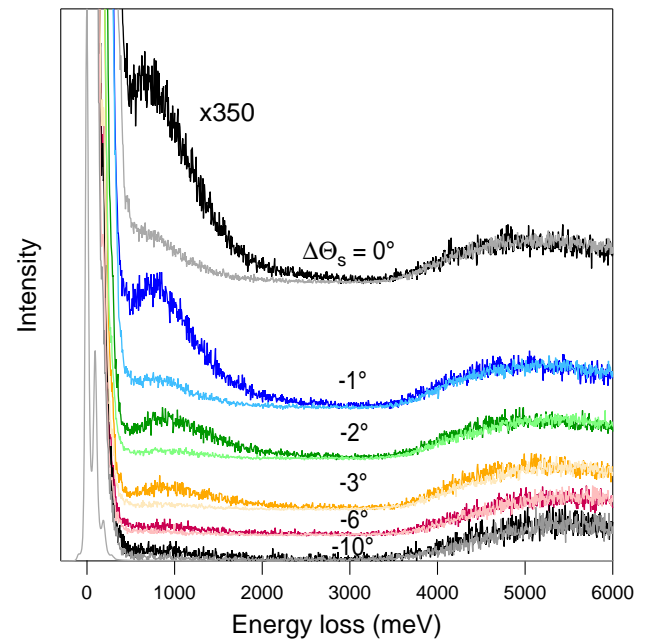


FIG. S5. Out-of-specular EELS measurements of BGS of a SA-TiO₂ surface annealed during 10 s at 970 K before (full line) and after exposure (light line) to 20 L of O₂ at 300 K. The offset angle $\Delta\Theta_s = \Theta_s - \Theta_I$ is indicated in figure. The corresponding evolutions of integrated intensities are displayed in Fig. 4-b of the article.

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