Palladium-Catalyzed [3 + 2]-C–C/N–C Bond-Forming Annulation

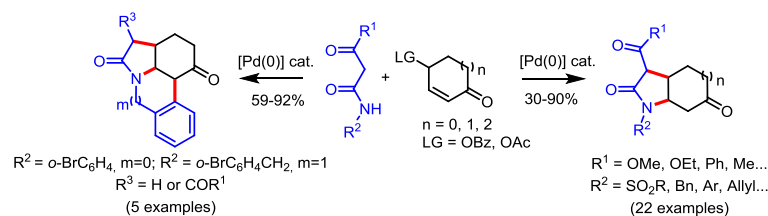Yang Liu, Zhongyi Mao, Alexandre Pradal, Pei-Qiang Huang, Julie Oble,
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Palladium-Catalyzed [3+2]-C-C/N-C Bond Forming Annulation

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Supporting Information Placeholder

R[2] = *o*-BrC₆H₄, m=0; R[2] = *o*-BrC₆H₄CH₂, m=1
R[3] = H or COR[1]
(5 examples)

R[1] = OMe, OEt, Ph, Me...
R[2] = SO₂R, Bn, Ar, Allyl...
(22 examples)

ABSTRACT: The synthesis of bi- and tri-cyclic structures incorporating pyrrolidone rings are disclosed, starting from resonance-stabilized acetamides and cyclic α,β-unsaturated-γ-oxy-carbonyl derivatives. This process involves an intermolecular Tsuji-Trost allylation / intramolecular nitrogen 1,4-addition sequence. Crucial for the success of this bis-nucleophile / bis-electrophile [3+2] annulation is its well-defined step chronology in combination with the total chemoselectivity of the former step. When the newly formed annulation product carries a properly located *ortho*-haloaryl moiety at the nitrogen substituent, a further intramolecular keto α-arylation can join the cascade, thereby forming two new cycles and three new bonds in the same synthetic operation.

The efficient assembly of complex polycyclic structures in a single synthetic operation is a major endeavor for chemists, and domino reactions[1] represent one of the most interesting ways to attain this objective with high step economy.[2] In particular, annulation reactions, the most efficient methods for the generation of cyclic molecules, pioneered by O. Diels and K. Alder,[3] and Sir R. Robinson,[4] allow building up a cyclic structure through the concerted or stepwise creation of two new bonds from two separated components.[5] In the frame of our current studies on η³-allylpalladium chemistry and domino sequences,[6] we recently reported the synthesis of hexahydroindole derivatives[7] through Pd-catalyzed pseudo-domino sequences[8] using β-amido-ester as bis-nucleophiles and cyclic bis-allylic substrates as bis-electrophiles (Scheme 1, eq 1).[9] Cyclic α,β-unsaturated-γ-oxy-carbonyls represent another interesting family of bis-electrophiles, which can be engaged in synthetically interesting Pd-catalyzed cascades, as reported by A. Fürstner, J. E. Harvey, S. S. Ramasastry and R. Tong (Scheme 1, eq 2).[10,11] Despite these works, the knowledge of the behavior of these bis-electrophiles is still far from mature. Herein, we disclose a Pd-catalyzed cascade between resonance-stabilized acetamides and various cyclic α,β-unsaturated-γ-oxy-carbonyls (Scheme 1, eq 3), which allows the achievement of a [3+2]-C-C/N-C bond forming annulation.[12] Importantly, the desired transformation is possible only if the intermolecular C-C bond formation at the electrophilic γ position (umpoled position with respect to the carbonyl function) precedes the intramolecular N-C bond formation at the electrophilic β position. Furthermore, the use

of acetamides bearing a properly located *ortho*-haloaryl moiety at the nitrogen substituent sets the stage for an additional intramolecular keto α-arylation at the end of the cascade, thereby allowing the selective formation of two new cycles and three new bonds in the same synthetic operation (Scheme 1, eq 4).

Scheme 1. Domino reactions with bis-electrophiles and bis-nucleophiles
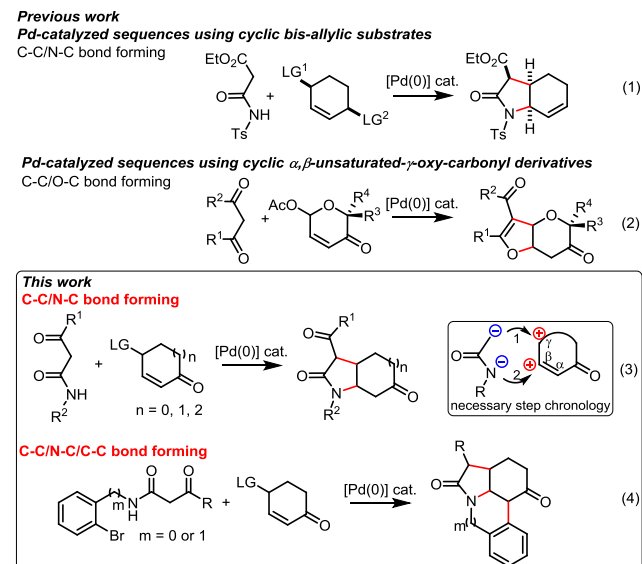
We started the investigation of this [3+2]-C-C/N-C bond forming annulation using the *N*-tosyl amido-ester **1a**[13] and 2-cyclohexenone 4-benzoate **2a**[14] as bis-nucleophile and bis-electrophile model substrates, respectively (Table 1). The first test was performed using the conditions previously developed with cyclic bis-allylic substrates (Scheme 1, eq 1)[7,9] [Pd(OAc)$_2$ (5 mol %), dppb (10 mol %), NaH (3 equiv) in CH$_3$CN at 50 °C]. However, we obtained only a very low amount of the expected fused pyrrolidone **3a** (18% yield), along with a significant amount of phenol, arising from **2a** through the elimination of the benzoate anion under basic medium. To suppress the formation of this by-product, the optimization of the reaction was performed in the absence of base, at room temperature, in THF (0.1 M) and with a slight excess of bis-electrophile **2a** (1.3 equivalents of **2a** relative to **1a**; see Supporting Information for detailed optimization). Subsequent screening of the palladium sources, such as: Pd(PPh$_3$)$_4$ (entry 1), Pd(OAc)$_2$ (Table 1, entries 2-3), Pd$_2$(dba)$_3$ (entries 4-5) or [Pd(η^3-C$_3$H$_5$)Cl]$_2$ (entry 6) and the phosphine ligand (PPh$_3$ or dppf), allowed us to reach the optimal conditions [Pd(η^3-C$_3$H$_5$)Cl]$_2$ (5 mol %), dppf (15 mol %) in THF at rt: *protocol A*], which led to the desired bicyclic pyrrolidone **3a** in 90% yield (Table 1, entry 6).[15]

Table 1. Optimizations on the model bis-nucleophile/bis-electrophile pair 1a / 2a

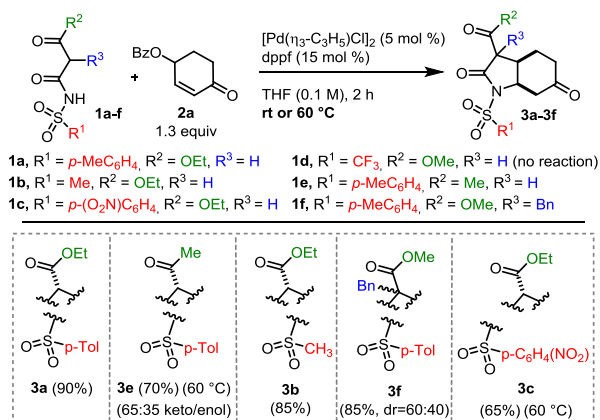| entry	[Pd] (x mol%)	ligand (y mol%)[a]	yield (%)[b]
1	Pd(PPh$_3$)$_4$ (10)	-	67
2	Pd(OAc)$_2$ (10)	PPh$_3$ (30)	73
3	Pd(OAc)$_2$ (10)	dppf (15)	60
4	Pd$_2$(dba)$_3$ (5)	PPh$_3$ (30)	81
5	Pd$_2$(dba)$_3$ (5)	dppf (15)	82
6	**[Pd(η^3-C$_3$H$_5$)Cl]$_2$ (5)**	**dppf (15)**	**90**

[a] dppb: 1,4-bis(diphenylphosphino)butane; dppf: bis(diphenylphosphino)ferrocene. [b] Isolated yield.

With the optimized conditions in hand, the scope and the limitations of this [3+2]-C-C/N-C bond forming annulation between the bis-electrophile **2a** and a range of resonance-stabilized *N*-sulfonyl acetamides **1a**-**f**, modified at the sulfonyl or at the carbonyl substituents, were next investigated (Scheme 2). All the *N*-sulfonyl acetamides tested behaved satisfactorily (65-90% yield), including acetoacetamide **1e**, which afforded the expected annulated product as a 65:35 keto/enol mixture, whereas the *N*-triflyl derivative **1d** did not show reactivity. In particular, in the case of **1c** and **1e**, the reaction temperature had to be raised to 60 °C, due to their manifest lower reactivity with respect to the other bis-nucleophiles. It is worth noting that the annulation works satisfactorily when starting from the benzylated bis-nucleophile **1f**, too.
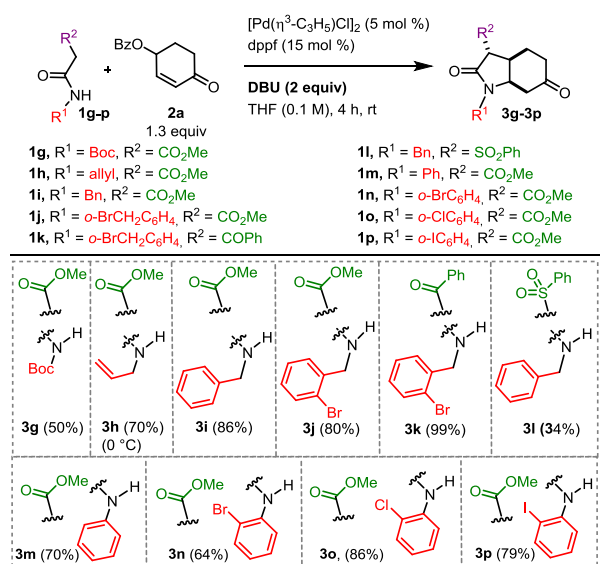
Different *N*-protections on the resonance-stabilized acetamides were next evaluated. Surprisingly, application of the above-optimized protocol to the *N*-carbamate bis-nucleophile **1g** did not afford the expected annulation product. After several trials and in deep contrast with our previous

outcomes, we found that this type of substrate requires the presence of a base. A re-optimization effort established the following new conditions as optimal [Pd(η^3-C$_3$H$_5$)Cl]$_2$ (5 mol %), dppf (15 mol %), DBU (2.0 equiv) in THF (0.1 M) at rt: *protocol B*] (Scheme 3).

Scheme 2. Variations of the resonance-stabilized *N*-sulfonyl acetamide in the reaction with 2a

1a, R^1 = *p*-MeC$_6$H$_4$, R^2 = OEt, R^3 = H **1d**, R^1 = CF$_3$, R^2 = OMe, R^3 = H (no reaction)
1b, R^1 = Me, R^2 = OEt, R^3 = H **1e**, R^1 = *p*-MeC$_6$H$_4$, R^2 = Me, R^3 = H
1c, R^1 = *p*-(O$_2$N)C$_6$H$_4$, R^2 = OEt, R^3 = H **1f**, R^1 = *p*-MeC$_6$H$_4$, R^2 = OMe, R^3 = Bn

3a (90%)	**3e** (70%) (60 °C) (65:35 keto/enol)	**3b** (85%)	**3f** (85%, dr=60:40)	**3c** (65%) (60 °C)

Scheme 3. Variations of the *N*-protection on the resonance-stabilized acetamide in the reaction with 2a

1g, R^1 = Boc, R^2 = CO$_2$Me **1l**, R^1 = Bn, R^2 = SO$_2$Ph
1h, R^1 = allyl, R^2 = CO$_2$Me **1m**, R^1 = Ph, R^2 = CO$_2$Me
1i, R^1 = Bn, R^2 = CO$_2$Me **1n**, R^1 = *o*-BrC$_6$H$_4$, R^2 = CO$_2$Me
1j, R^1 = *o*-BrCH$_2$C$_6$H$_4$, R^2 = CO$_2$Me **1o**, R^1 = *o*-ClC$_6$H$_4$, R^2 = CO$_2$Me
1k, R^1 = *o*-BrCH$_2$C$_6$H$_4$, R^2 = COPh **1p**, R^1 = *o*-IC$_6$H$_4$, R^2 = CO$_2$Me

3g (50%)	**3h** (70%) (0 °C)	**3i** (86%)	**3j** (80%)	**3k** (99%)	**3l** (34%)

| **3m** (70%)	**3n** (64%)	**3o** (86%)	**3p** (79%)

Under these conditions, the Boc derivative **1g** afforded the expected product **3g** in 50% yield. Good yields of annulated products were also obtained with *N*-allyl (**1h**), -benzyl (**1i**, **1l**), -*o*-bromobenzyl (**1j**, **1k**), -phenyl (**1m**), -*o*-bromophenyl (**1n**), -*o*-chlorophenyl (**1o**), -*o*-iodophenyl (**1p**) derivatives, carrying a methoxycarbonyl, a benzoyl or a phenylsulfonyl moiety as the resonance stabilizing group for the acetamide. In the case of *N*-allyl amido-ester **1h**, the reaction must be performed at 0 °C for maximum yield. Interestingly, the *o*-bromo and *o*-chloro (but not the *o*-iodo) -phenyl substituted amides **1n** and **1o** can react smoothly also in the absence of DBU, albeit in lower yields. This behavior is very likely due to the N-H acidity enhancement brought about by the highly electronegative bromine and chlorine atoms, which activate the nitrogen atom.[16] The above results suggest that the acidity

of the two mobile hydrogen atoms of the bis-nucleophile have to fit in an appropriate pK_a window to allow the annulations.

Encouraged by these results, we next turned our attention to explore the [3+2]-annulation involving five- and seven-membered cyclic α,β-unsaturated-γ-oxy-carbonyls (Scheme 4). Treatment of cyclopentenone 4-benzoate **2b**[17] with **1a** under the conditions of *protocol A* gave the expected annulated product **4a** in a very low yield. However, after substantial experimentation, we found that performing the reaction in toluene at 110 °C (*protocol C*) led to the annulation product **4a** as a mixture of diasteoisomers (62% yield). Under these new conditions and in the presence of DBU (*protocol D*), the *N*-benzyl bis-nucleophile **1i** furnished the corresponding annulated product **4i**, though in 49% yield. The seven-membered bis-electrophile **2c**[18] reacted smoothly under the original optimized conditions in THF. However, while reaction with **1a** required heating at 60 °C, reaction with **1g** and **1i** called for the presence of DBU at rt. As the Tsuji-Trost step is expected to be equally favored in all the cases studied, these experiments suggest that the intramolecular 1,4-addition step is easier for the 6-*exo-trig* and 7-*exo-trig* than the 5-*exo-trig* ring closures.

Scheme 4. Variations of the cyclic α,β-unsaturated-γ-oxy-carbonyl
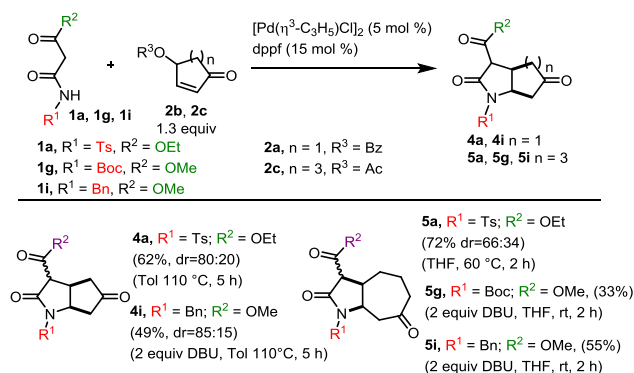
The use of cyclic α,β-unsaturated-γ-oxy-carbonyls bearing a heteroatom was then investigated. We initially considered the formal incorporation of an oxygen or a nitrogen atom at position 6 of the parent cyclohexenone 4-benzoate (cyclohexenone numbering). However, preliminary experiments showed that η³-allyl complexes deriving from substrates of this type have a strong tendency to β-eliminate leading to 2-pyrone or -pyridone, the driving force being very likely the aromatic character of these heterocycles (see SI). We therefore decided to consider the heteroatom at position 5 through the use of the 5-oxo-5,6-dihydro-2H-pyran-2-yl benzoate **2e**[19] in the presence of **1a**. After several relatively fruitless trials, we found that the use of the above-optimized system in the presence of 2.2 equivalents of **2e**, provided the adduct **7a** (30% yield), which resulted from a second allylation taking place besides the "normal" allylation / conjugate addition sequence (Scheme 5, top). This result suggests that in this particular case, the initially formed adduct is more activated toward over-allylation than the starting bis-nucleophile **1a**. We speculate that an intramolecular H-bond in the first generated intermediate **A** may be responsible of such original behavior. As expected, reaction of the benzylated β-

amido-ester **1f** gave, under identical condition, the regular adduct **7f** in satisfactory yield (Scheme 5, bottom).

Scheme 5. Reactivity of cyclic α,β-unsaturated-γ-oxy-carbonyls bearing an oxygen at position 5
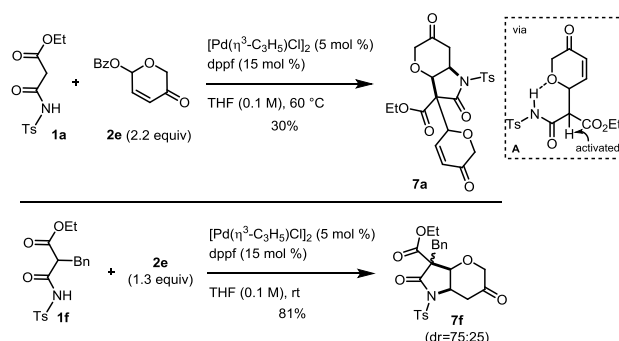
Finally, we reasoned that by using a bis-nucleophile bearing a juxtaposed halo-aryl moiety, the allylation / conjugate addition sequence could be coupled to an additional Pd-catalyzed intramolecular arylation.[20] The success of such an ambitious plan depends, *inter alia*, on the possibility of intramolecularly trapping the enolate issued from the nitrogen 1,4-addition with the catalytically generated arylpalladium(II) moiety, without premature proton transfer from the acid carbon atom of the bis-nucleophile component. With the above *caveat* in mind,[21] we initially investigated the reaction between the bis-nucleophile **1s**, bearing a methyl group on the α-position, and bis-electrophile **2a** (Scheme 6). After considerable experimentations (see SI), we found that treatment of the brominated bis-nucleophile **1s** with cyclic α,β-unsaturated-γ-oxy-carbonyl **2a** in the presence of the system [Pd(OAc)₂ (10 mol %), dppb (15 mol %), DBU (2 equiv)] in toluene at rt for 10 min, followed by addition of K₃PO₄ and further 2 h heating (145 °C oil bath in sealed tube), gave the desired *cis/cis* fused tricyclic compound **10s** in 85% yield as major diastereoisomer, along with 7% yield of the minor diastereoisomer **10s'** (Scheme 6).[15]
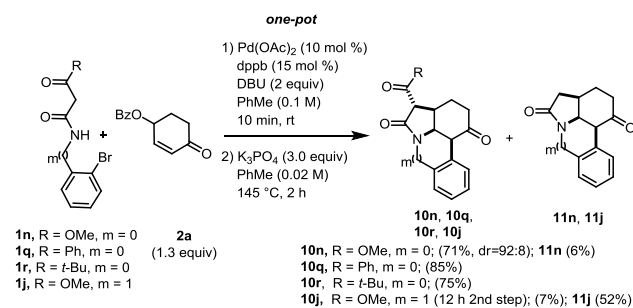
Scheme 6. Triple domino allylation / *N*-conjugate addition / enolate α-arylation
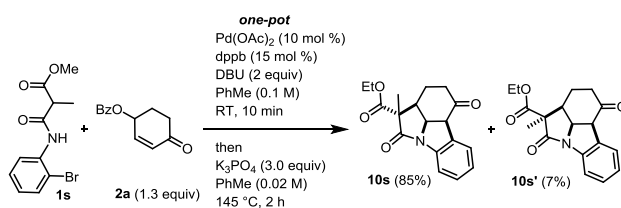
Following this satisfactory result, the triple domino process was tested using the unsubstituted amides bis-nucleophiles **1n**, **1q**, **1r**, and **1j** (Scheme 7). To our delight, the targeted tricycle compounds **10n, 10q** and **10r**[22] were successfully isolated in good yields. In the case of **1n**, the tricycle was accompanied by a minor amount (6% yield) of the demethoxycarbonylated product **11n**. The homologated bromo-benzyl derivative **1j** led to the desired triple domino product **10j**, too. However, in this case, a prolonged reaction time was necessary and the major product consisted in the demethoxycarboylated product **11j**. It

thus appears that the intramolecular arylation is sensibly easier for the formation of 6/5/5 fused tricycles than 6/5/6 ones.

Scheme 7. Triple domino allylation/*N*-conjugate addition/enolate α-arylation

1n, R = OMe, m = 0
1q, R = Ph, m = 0
1r, R = *t*-Bu, m = 0
1j, R = OMe, m = 1

2a
(1.3 equiv)

10n, 10q,
10r, 10j

11n, 11j

10n, R = OMe, m = 0; (71%, dr=92:8); **11n** (6%)
10q, R = Ph, m = 0; (85%)
10r, R = *t*-Bu, m = 0; (75%)
10j, R = OMe, m = 1 (12 h 2nd step); (7%); **11j** (52%)

A plausible mechanism for this annulation is presented in Scheme 8 for the reaction between *N*-*o*-bromophenyl amido-ester **1n** and 2-cyclohexenone 4-benzoate **2a**. The sequence starts with an oxidative addition of the bis-electrophile **2a** onto the Pd(0) complex to generate the corresponding transient η^3-allyl-complex **B** through the η^2-alkene complex **A** (*steps a and b*). The amidinium-enolate of the bis-nucleophile **1n**,[23,24] formed in the meantime (*step c*), enters the catalytic cycle through C-allylation. Then, a new deprotonation by DBU and an oxidative addition of its aryl halide to Pd(0) generates amidate **C** (*step d*). Pd(0) decoordination (*step e*), to form **D**, opens the way to the intramolecular aza-conjugate addition to afford bicyclic enolate **E** (*step f*). Subsequent enolate-palladation (*step g*) followed by reductive elimination generates the final tricyclic product **10n** and the Pd(0) complex. A notable feature of this process is the fact that the *catalytic* generation of the η^3-allyl-complex **B** completely wins against the alternative *stoichiometric* conjugate addition reactivity, a *conditio sine qua non* step chronology for the annulation. Indeed, a premature (N or C) conjugate addition reactivity would impede generation of the η^3-allyl-complex.

In conclusion, through the reaction between resonance-stabilized acetamides as bis-nucleophiles and cyclic α,β-unsaturated-γ-oxy-carbonyl derivatives as bis-electrophiles, we have successfully developed a new domino transformation consisting of an intermolecular Tsuji-Trost allylation / intramolecular nitrogen 1,4 addition sequence. The success of this [3+2] C-C/N-C bond forming annulation is due to the total chemoselectivity of the former step (C-allylation) as well as to the well-defined chronology of the following steps. When the newly formed annulation product contains an appropriately located *ortho*-haloaryl moiety at the nitrogen substituent, a further intramolecular keto α-arylation can follow the cascade, thereby forming two new cycles and three new bonds in the same synthetic action. In view of the several synthetically interesting structures incorporating the bicyclic 4,5-fused pyrrolidine motif, such as lycorine-type alkaloids,[25] daphniphyllum alkaloids[26] and aeruginosins[27] just to mention a few examples, and the many methods developed for the chemoselective and direct transformation of amides,[28] the current method is expected to find wide application in organic synthesis. Further work is ongoing to develop enantioselective versions of this new transformation.

ASSOCIATED CONTENT

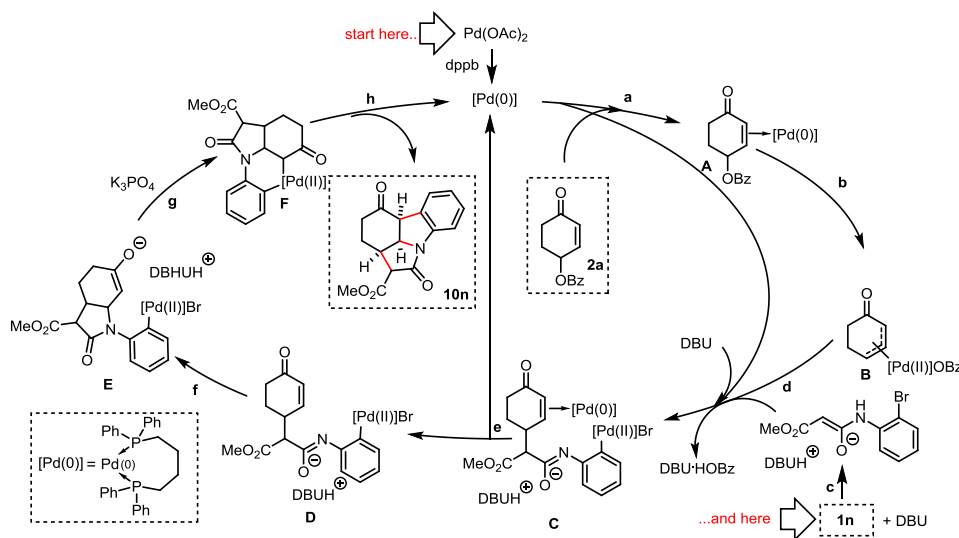### Supporting Information

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Scheme 8. Plausible mechanism of domino Tsuji-Trost allylation / aza-Michael / keto α-arylation

4

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(21)   Treatment of annulated products **3j** and **3n** with excess LDA and TBSOTf gave tricycle aldol adducts **9** in high yields (see SI). This result shows the difficulty of accomplishing a double depronation on this type of substrates, as the first generated resonance-stabilized enolate undergoes a fast intramolecular carbonyl addition, thereby impeding a second deprotonation in α position to the keto function.
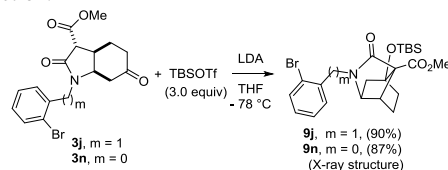
(22) The identity and the stereochemistry of compound **10r** could be unambiguously proven by X-ray diffraction of a single crystal of it (see SI).

(23) We found the annulation very sensitive to proton transfer factors. Thus, while the *N*-sulfonyl bis-nucleophile based annulations could be achieved only in the absence of base, most of the *N*-alkyl based ones were successful only in the presence of base. In line with this pronounced proton transfer sensitivity, it is also remarkable that the formal introduction of an *o*-bromo or *o*-chloro (but not an *o*-iodo) substituent to the *N*-phenyl amido-ester permits base-free conditions.

(24) In the cases in which the base is not required, it is likely that the benzoate anion, counter-ion of the η^3-allylPd complex, acts as the base to deprotonate the pro-nucleophile. See: Giambastiani; G.; Poli, G. *J. Org. Chem.* **1998**, *63*, 9608.

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