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Inverse Conformational Selection in Lipid–Protein Binding

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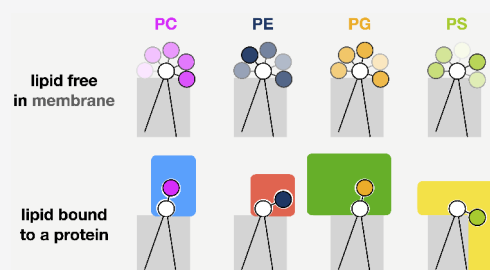


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Supporting Information

ABSTRACT: Interest in lipid interactions with proteins and other biomolecules is emerging not only in fundamental biochemistry but also in the field of nanobiotechnology where lipids are commonly used, for example, in carriers of mRNA vaccines. The outward-facing components of cellular membranes and lipid nanoparticles, the lipid headgroups, regulate membrane interactions with approaching substances, such as proteins, drugs, RNA, or viruses. Because lipid headgroup conformational ensembles have not been experimentally determined in physiologically relevant conditions, an essential question about their interactions with other biomolecules remains unanswered: Do headgroups exchange between a few rigid structures, or fluctuate freely across a practically continuous spectrum of conformations? Here, we combine solid-state NMR experiments and molecular dynamics simulations from the NMRlipids Project to resolve the conformational ensembles of headgroups of four key lipid types in various biologically relevant conditions. We find that lipid headgroups sample a wide range of overlapping conformations in both neutral and charged cellular membranes, and that differences in the headgroup chemistry manifest only in probability distributions of conformations. Furthermore, the analysis of 894 protein-bound lipid structures from the Protein Data Bank suggests that lipids can bind to proteins in a wide range of conformations, which are not limited by the headgroup chemistry. We propose that lipids can select a suitable headgroup conformation from the wide range available to them to fit the various binding sites in proteins. The proposed *inverse conformational selection model* will extend also to lipid binding to targets other than proteins, such as drugs, RNA, and viruses.



INTRODUCTION

Lipid interactions with other biomolecules are gaining interest not only in molecular cell biology¹ but also in nanobiotechnology applications, such as the design of antibodies² and carriers for mRNA vaccines.^{3,4} Interactions of cellular membranes or lipid nanoparticles with other biomolecules, such as proteins, drugs, or RNA, are regulated by lipid headgroups, the outward-facing components of lipid bilayers.^{1,5} Chemical compositions of lipid headgroups vary between different organelles and organisms, and specific interactions with certain lipid headgroups are known to be essential for the function of several proteins.^{1,5} However, the driving forces for specific interactions between lipids and membrane-binding substances are not fully understood because conformational ensembles of lipids in the physiologically relevant liquid state have not been experimentally determined. Therefore, it is not clear if the specificity arises from the differences in accessible conformations between lipid types or from specific intermolecular lipid–protein interactions.

Structures of protein-bound lipids are available in the Protein Data Bank (PDB, <http://www.rcsb.org/>),⁶ and crystal structures of lipids have been determined,^{7,8} but their relation to lipids in bulk membranes in the liquid lamellar phase

remains unclear.⁹ Similarly to disordered proteins,¹⁰ lipid molecules sample conformational ensembles that are composed of sets of individual structures whose occurrence probabilities are determined by the Boltzmann distribution.¹¹ In the liquid bilayer state, individual phosphatidylcholine lipids sample the conformational ensemble on the nanosecond time scale.^{12,13} The most accurate experimental information on conformational ensembles of lipids in this biologically relevant phase is typically derived from NMR experiments, particularly from the C–H bond order parameters, $S_{\text{CH}} = \frac{1}{2} (3 \cos^2 \theta - 1)$, where θ is the angle between the C–H bond and membrane normal, and the average is taken over the conformational ensemble of lipids.^{14–16} Because these order parameters are similar to those in living cells,^{17–19} the model membranes can be used to resolve the lipid headgroup conformations in biological conditions. According to these

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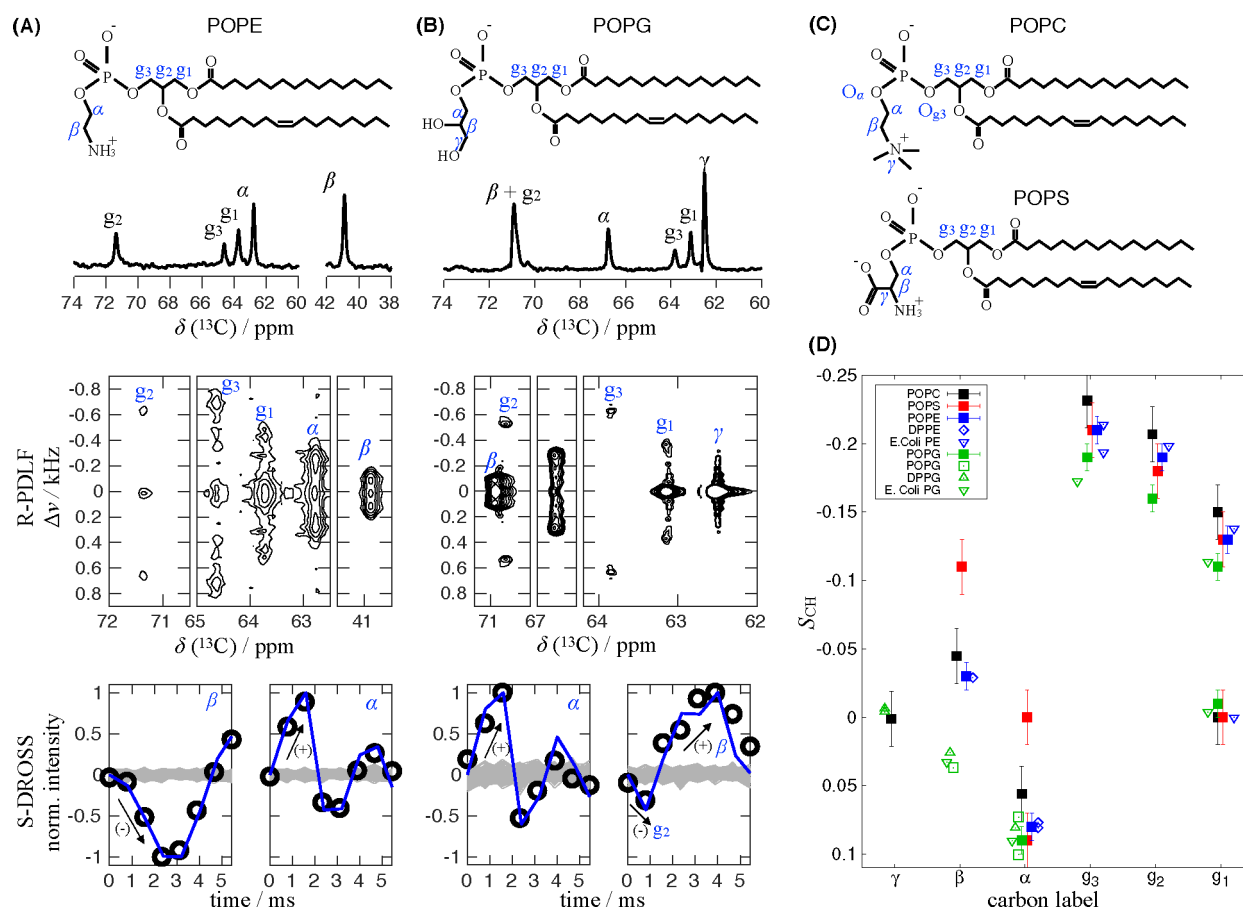


Figure 1. Chemical structures, refocused-INEPT spectra, 2D R-PDLF spectra, and S-DROSS dipolar slices (from top to bottom) for (A) POPE and (B) POPG multilamellar vesicles measured by ^1H - ^{13}C solid-state NMR. The S-DROSS dipolar slices show both experimental data (open circles) and the result of NMR numerical simulations (blue solid lines). Full NMR spectra are shown in Figures S1 and S2. (C) Chemical structures of POPC and POPS. (D) Headgroup and glycerol backbone C-H bond order parameters S_{CH} for different phospholipids measured in the lamellar liquid disordered phase. Filled squares show S_{CH} magnitudes and signs determined by ^1H - ^{13}C NMR spectroscopy for POPE (310 K) and POPG (298 K), as measured in this work, and from data previously reported for POPS (298 K)³² and POPC (300 K).^{33,34} Empty symbols show headgroup and glycerol backbone S_{CH} magnitudes measured previously with ^2H NMR and are also plotted as real values using the signs determined in this work for POPG with 10 mM PIPES (298 K),³⁵ DPPG with 10 mM PIPES and 100 mM NaCl (314 K),²⁰ DPPE (341 K),³⁶ and *E. coli* PE and *E. coli* PG with 10 mM PIPES and 100 mM NaCl (310 K).¹⁷

experiments, the glycerol backbone conformations are largely similar, irrespective of the headgroup,¹⁷ and the headgroup conformations are similar in the phosphatidylcholine (PC), phosphatidylethanolamine (PE), and phosphatidylglycerol (PG) lipids, whereas the phosphatidylserine (PS) headgroup is more rigid.^{7,20} Notably, however, the signs of S_{CH} are not accessible from ^2H NMR experiments,²¹ and universal models to map order parameters to structural ensembles are not available.^{22,23} Therefore, it is not clear if the lipid headgroups in a liquid lamellar bilayer exchange between a few restricted conformations or can fluctuate freely across a wide conformational range.

Here, we use natural abundance ^1H - ^{13}C solid-state NMR experiments and MD simulations from the NMRlipids Project to resolve differences in the conformational ensembles of PC, PE, PG, and PS lipid headgroups. Zwitterionic PCs are the most common lipids in eukaryotes and PEs in bacteria;^{5,24} PE is also the second most abundant glycerophospholipid in eukaryotic cells and has been related to various diseases.²⁵⁻²⁷ PC lipids are also used in mRNA COVID-19 vaccines.⁴ PSs are the most common negatively charged lipids in eukaryotes and PGs in bacteria, and both are known to affect membrane

protein functionality and signaling.^{24,28-30} All the four studied lipid types specifically bind to various proteins,³¹ we therefore elucidate also the effect of protein binding on headgroup conformations. The resulting *inverse conformational selection model* for lipid-protein binding can be also applied to understand interactions between membranes and biomolecules other than proteins, such as drugs, RNA, or viruses. On the other hand, the detailed understanding of lipid bilayer interfaces can facilitate the design of lipid nanoparticle carriers for mRNA vaccines with fewer side effects.

RESULTS AND DISCUSSION

Differences between Lipid Headgroups from ^{13}C NMR Experiments. To experimentally characterize the differences in headgroup conformational ensembles of lipids that are not bound to proteins, we measured the C-H bond order parameters S_{CH} and their signs for 1-palmitoyl-2-oleoyl-*sn*-glycero-3-phospho-(1'-*rac*-glycerol) (POPG) and 1-palmitoyl-2-oleoyl-*sn*-glycero-3-phosphoethanolamine (POPE) in the liquid lamellar phase, as we did previously for POPC and POPS.³²⁻³⁴ Determination of headgroup and glycerol backbone S_{CH} and their signs was straightforward from the data in

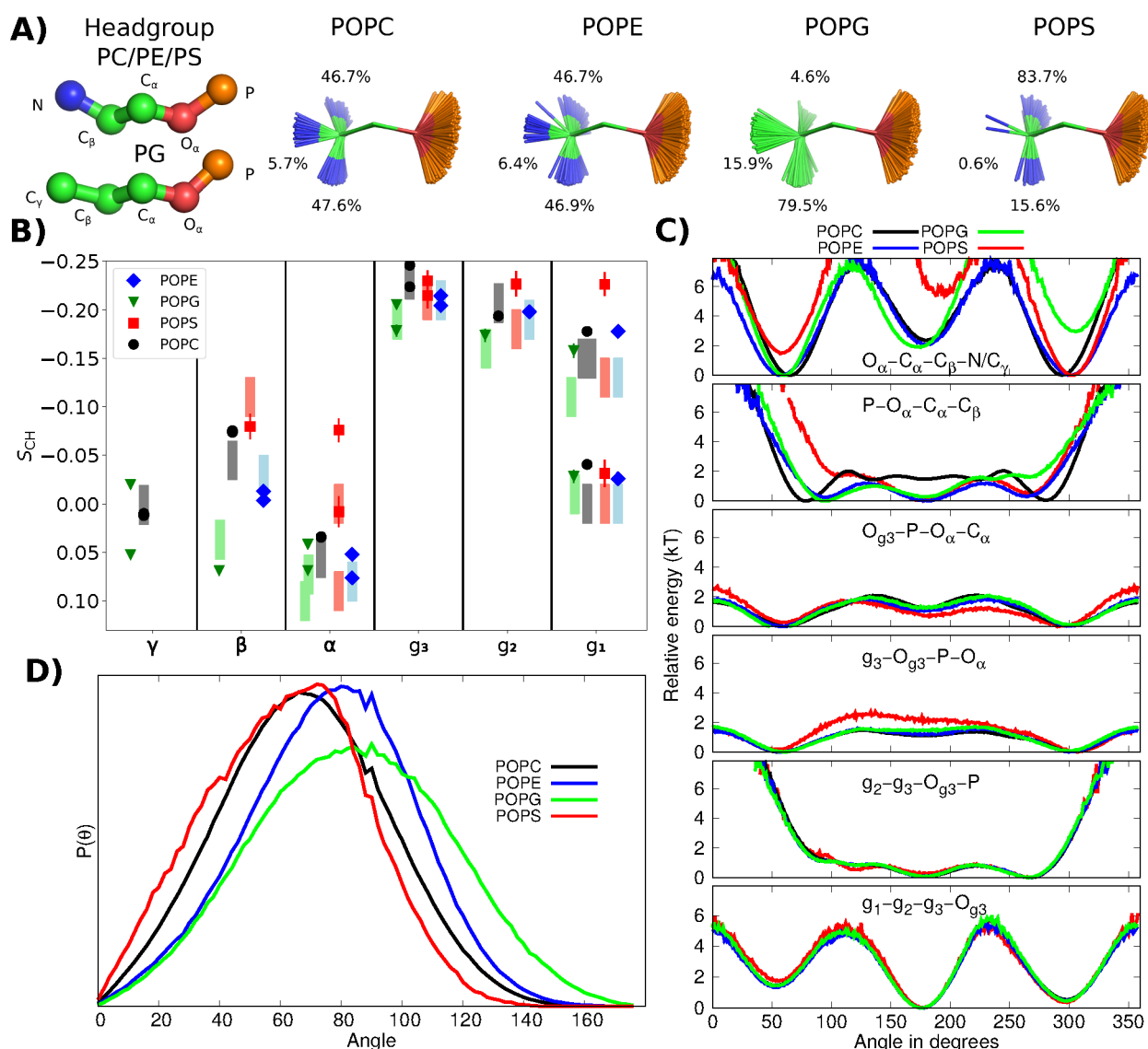


Figure 2. Results from the best simulation model demonstrating the differences in conformational ensembles between different lipid types. The best simulation model (CHARMM36) was selected from the comparison of 10 different force fields based on quality evaluation against experimental S_{CH} values in Figures S3 and S4. (A) Snapshots with overlaid C_β , C_ω and O_α atoms and occurrence of different conformations. (B) Headgroup and glycerol backbone region order parameters of the different lipid types. Colored shaded regions show the range of experimental values from Figure 1. (C) Relative free energies for individual heavy-atom dihedral angles estimated from the inverse Boltzmann formula. Angles corresponding to free energies above $8 k_B T$ are not shown because they are not observed in simulations. (D) Distributions of the P-N/ C_γ vector angle with respect to the membrane normal.

Figures 1, S1, and S2 for all the C-H bonds, except for the β and g_2 carbons in POPG. These carbons have overlapping peaks in the INEPT spectra due to their similar chemical environments, and only the magnitude of the larger order parameter could be determined from the R-PDLF spectra shown in Figure 1B. Based on previous 2H NMR measurements,^{17,20,35} we assigned the larger S_{CH} to the g_2 carbon and used the literature value for the β -carbon in SIMPSON simulations to determine the signs. The decrease in the beginning of the S-DROSS curve suggests that the sign of the larger S_{CH} from g_2 is negative, and the later increase suggests that the sign of the smaller S_{CH} from β is positive, as confirmed by the SIMPSON simulations (Figure 1B).

Experimental order parameters of POPC, POPE, POPG, and POPS glycerol backbones and headgroups from this and previous studies are displayed in Figure 1D, where the signs

from ^{13}C NMR experiments are used also for the 2H NMR data from the literature. The overall agreement of S_{CH} determined by different research teams and different techniques for the same lipid headgroup was excellent (within ± 0.02) here and in previous studies.^{21,32,37} Therefore, the differences between lipid types are dictated by the headgroup chemistry rather than inaccuracies in experiments or differences in the acyl chains or in the experimental conditions.

The most distinct order parameters are observed for PS headgroups, for which the α -carbon S_{CH} exhibits significant forking (different values for different hydrogens bound to the same carbon) and the β -carbon has a more negative value than in other studied lipid types. On the other hand, the β -carbon S_{CH} of PG headgroups has a positive sign, in contrast to all the other lipid types. The sign has not been previously known from 2H NMR experiments, which enable detecting only absolute

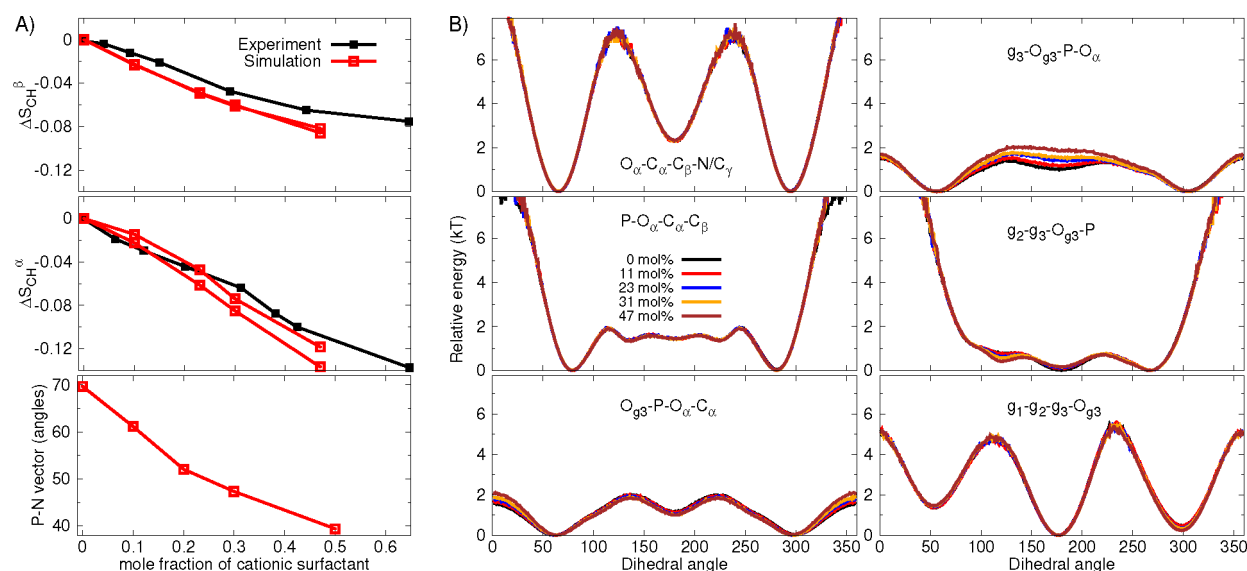


Figure 3. (A) Modulation of PC headgroup order parameters and P–N vector tilt (average angle between the P–N vector and the bilayer normal) upon addition of cationic surfactant (dihexadecyldimethylammonium) from CHARMM36 simulations compared with the experimental data.³⁹ (B) Relative free energies for individual dihedral angles estimated from inverse Boltzmann distributions of heavy-atom dihedral angles with different amounts of cationic surfactant from CHARMM36 simulations.

values for order parameters.^{17,20,35} The glycerol backbone order parameters are similar for all the lipid types, although they move slightly toward positive values (closer to zero) in the order PC < PE < PS < PG. Only minor differences between PC and PE headgroups are observed.

Conformational Ensembles of Different Lipid Headgroups from MD Simulations. Ideal MD simulations would correctly predict the conformational ensembles of lipids, thereby giving an interpretation for the experimental C–H bond order parameters, S_{CH} , due to their direct connection.²¹ To find the best MD simulation models (force fields) for this purpose, we took advantage of the power of the NMRlipids Project, an open science collaboration that has gathered a massive number of simulation trajectories, allowing the evaluation of all the available force fields of PC,³⁷ PS,³² PE (Figure S3), and PG (Figure S4) lipids against NMR data. The comparison of hundreds of simulation trajectories, simulated with approximately 20 different force fields, revealed that none of them can correctly capture the lipid conformational ensembles and reproduce the experimental S_{CH} of all segments within experimental error (for a detailed discussion on the quality of the force fields, see section S3 in the Supporting Information and refs 13, 21, 32, and 37). However, CHARMM36, the force field showing the best agreement with experiments, correctly captures the essential differences between the PC, PE, PG, and PS headgroup order parameters, particularly the positive order parameter of the β -carbon in PG and the forking of the α -carbon in PS (Figure 2B).

To understand the structural origin of distinct order parameters for the distinct lipids, we first calculated and visualized the heavy-atom dihedral angle distributions from the best-performing simulations that reproduced the differences between headgroups (Figures 2A and S11). Then, we used the inverse Boltzmann formula to estimate the free energy costs for different dihedral angle orientations. The results in Figure 2C suggest that all lipid headgroups are very flexible, and free energy costs for rotating individual dihedrals to almost any angle is low (below $8 k_B T$). Only *cis* states of P– O_α – C_α – C_β

and g_2 – g_3 – O_{g_3} –P dihedrals have larger relative free energies and are not observed for any lipids during the simulations.

Major differences between headgroups are observed for the last two dihedrals in the headgroup end, O_α – C_α – C_β – N/C_γ and P– O_α – C_α – C_β , which prefer *gauche*[−] conformations for PG and *gauche*⁺ for PS, whereas PC and PE exhibit symmetric distributions. Also, the free energy barriers for O_α – C_α – C_β – N/C_γ dihedral rotations between *gauche* and *trans* states are larger for PS and PG lipids than for PC and PE lipids. The rest of the dihedrals are similar between different lipids, with the exception of PS lipids, for which a slightly larger free energy for *eclipsed anti* conformation in the g_3 – O_{g_3} –P– O_α dihedral was observed. Therefore, we suggest that the main differences between lipid headgroups leading to distinct order parameters occur in the headgroup end, while in the phosphate region they remain similar in different lipids with the exception of PS. The increased barriers for dihedral rotations may explain the more rigid headgroup structures in PS.^{7,38} Furthermore, the angle between headgroup dipole and membrane normal decreases in the order of PG > PE > PC > PS (Figure 2D). However, the differences between PC and PE in P– O_α – C_α – C_β dihedral and P–N vector dipole may be artificial, as the β -carbon order parameter in PC is too negative, even in the best available force field,³⁷ thereby not being equal to the order parameter in PE as observed in experiments.

In conclusion, the rotation of dihedral angles to almost any position bears a relatively low free energy cost, and the sampled dihedral angles are within similar ranges for all studied headgroup types. In other words, all studied lipid headgroups sample very broad conformational ensembles in the liquid lamellar phase. Therefore, the lipid headgroups are able to adopt a wide range of conformations when interacting with proteins, ions, or other biomolecules. Furthermore, the structures in lipid crystals^{7,8} are not representative of the liquid state, and models aiming to explain NMR data using only a few conformations^{14–16,23} are not sufficient to capture the large conformational space of lipids in the liquid lamellar state. However, conformational ensembles of lipids in MD

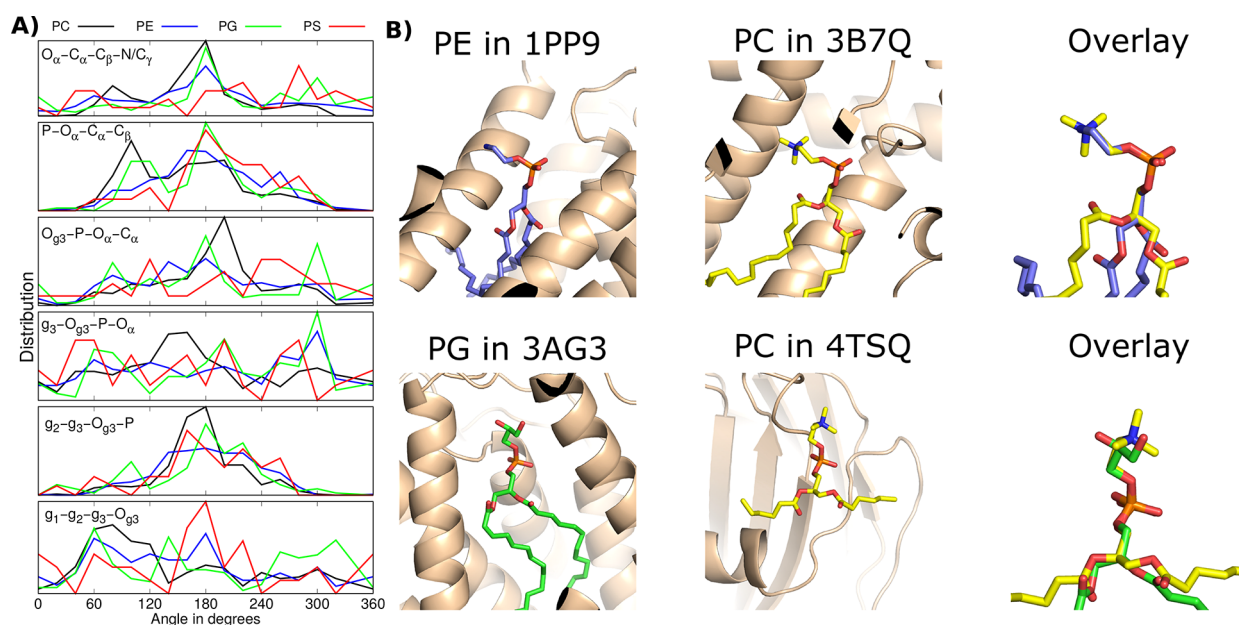


Figure 4. (A) Heavy-atom dihedral angle distributions calculated from lipid structures in the PDB. (B) The structure of PE headgroup bound to cytochrome *bc*₁ complex (PDB ID: 1PP9⁴²) with identical conformation as the PC headgroup bound to yeast Sec14 (PDB ID: 3B7Q⁴³), and the structure of PG headgroup bound to bovine cytochrome *c* oxidase (PDB ID: 3AG3⁴⁴) with identical conformation as the PC headgroup bound to FraC, a pore-forming toxin (PDB ID: 4TSQ⁴⁵).

simulation trajectories with accurate force fields that capture the experimental NMR order parameters can be considered realistic, and thereby used to interpret the experimental data.

Lipid Conformational Ensembles in Charged Membranes. Charged entities, such as lipids, proteins, surfactants, drugs, and ions, incorporated in membranes can change the orientation of the headgroup dipole in PC lipids and affect the order parameters of the lipid headgroups.⁴⁰ However, a detailed understanding of the structural and energetic responses of lipids to membrane-bound electric charge is still lacking.¹⁶ For example, it is not clear if the membrane-bound charges restrain the lipids into specific conformations or just redistribute the probabilities of states by keeping the accessible conformations similar.

To resolve lipid headgroup conformational ensembles in cellular membranes bearing positive charge, we calculated the heavy-atom dihedral angle distributions from the subset of simulations that correctly captured the experimentally measured decrease in PC headgroup order parameters upon addition of cationic surfactants into a bilayer (Figure 3A). The dihedral angle distributions and relative free energies in Figures S12 and 3B reveal that the addition of positive charge into a membrane decreases the abundance of *trans* states in $g_2-g_3-O_{g_3}-P$ and $g_3-O_{g_3}-P-O_{\omega}$ and *cis* states in $O_{g_3}-P-O_{\alpha}-C_{\alpha}$ dihedrals. The choline region remains essentially unchanged, and only minor changes are observed in the other dihedrals, even at a surfactant molar fraction of 0.47.

In addition, binding of ions to the membrane may affect the lipid headgroup conformational ensembles under physiological conditions. The Ca^{2+} ion bound to the PC headgroup leads to similar decreases in *trans* state probability for the $g_3-O_{g_3}-P-O_{\alpha}$ dihedral in the most realistic MD simulation models as observed for cationic surfactants (lipid17ecc and CHARMM36 in Figures S7, S13, and S14). The dihedral distributions of the PG headgroup are more sensitive to the bound ions in the most realistic simulations, but upward tilting of the headgroup

dipole upon the addition of $CaCl_2$ is weaker than in PC (Lipid17 and Slipids in Figures S7, S15, and S16). However, the changes in PG lipid dihedrals upon the addition of $CaCl_2$ differ between the best models (Figures S15 and S16), and none of the simulations captures the Ca^{2+} ion binding affinity and conformational ensemble of PG lipids simultaneously. Moreover, experimental data to evaluate the response of α -carbon order parameters to the added $CaCl_2$ in PG is not available. Additionally, the headgroup conformational ensembles in mixtures of PC and charged (PG or PS) or zwitterionic (PE) lipids could not be resolved with the currently available force fields and experimental data (Figures S5 and S6, and refs 32 and 41).

In conclusion, the MD simulation results suggest that the structural response of lipid headgroups to membrane-bound charges arises from mild changes in dihedral angle distributions, rather than restriction of lipids to fixed conformations. Therefore, lipid headgroups sample a large space of different conformations also in charged membranes, thereby retaining the capability to adopt multiple conformations when interacting with proteins or other molecules.

Protein-Bound Lipid Conformations. Interpretation of experimental order parameters using MD simulations in previous sections suggests that PC, PE, PG, and PS lipid headgroups are very flexible, allowing them to bind proteins in various different conformations. To test this prediction, we analyzed the conformations of all protein-bound lipids from structures deposited in the PDB.⁶ We found 311 PC, 394 PE, 154 PG, and 35 PS lipid conformations that were bound to different types of proteins (including both integral and peripheral membrane proteins) and were determined as a part of protein structure using crystallography or cryo-EM. The full list of structures with bound lipids is available from *Data/PDBdistributions/pdbtable.pdf* in ref 46.

The heavy-atom dihedral angle distributions calculated from these conformations (Figure 4 A) reveal that the protein-

bound lipids indeed exhibit a wide range of conformations, independent of the headgroup type. As for bulk lipid bilayers (Figures 2C and S11), only *cis* conformations of $P-O_{\alpha}-C_{\alpha}-C_{\beta}$ and $g_2-g_3-O_{g_3}-P$ dihedrals are almost completely absent in all lipids, and significant differences between different headgroups are not observed. Deviation of protein-bound lipid structures from protein-free lipid structures in crystals has been previously proposed to indicate inaccuracies in lipid structures in PDB.^{9,22} However, we observe large deviations from the protein-free lipid crystal structures also in conformational ensembles that reproduce the NMR data in the liquid lamellar phase, indicating that such deviations are realistic also in the protein-bound states.

Our results suggest that flexible lipid headgroups can optimize the intermolecular interactions with proteins by binding in a wide range of conformations. Therefore, the specific binding of lipids to proteins is not driven by differing structural preferences between headgroups. This is demonstrated in Figure 4B with two examples where different lipid types bound to different proteins have almost identical headgroup conformations: PE in cytochrome *bc*₁ complex is similar to PC bound to yeast Sec14, and PG bound to bovine cytochrome *c* oxidase is similar to PC bound to a pore-forming toxin (FraC). On the other hand, a single lipid headgroup type is capable of accommodating various binding positions and as such would be able to specifically bind to many different kinds of binding sites.

CONCLUSIONS

C–H bond order parameters S_{CH} from NMR experiments suggest that lipid headgroup conformational ensembles depend on lipid type (PC, PE, PG, or PS) and membrane charge (cationic lipids, surfactants, accumulated ions, or drugs). Our analysis of these data, using MD simulations collected within the NMRlipids Project, revealed that the differences in S_{CH} can be explained by small changes in the dihedral angle probability distributions. All four studied headgroup types (PC, PE, PG, and PS) are flexible and access a similarly wide range of conformations, with very few restrictions in dihedral orientations—also when membranes are charged. In conclusion, the observed differences in S_{CH} originate from slightly reweighed conformational probabilities rather than changes in accessible structures.

The flexibility and wide conformational space of headgroups suggest that protein-bound lipids can adapt to various binding sites to optimize the lipid–protein interactions. We tested this prediction by analyzing from PDB the conformations of lipids that are tightly bound to proteins. We found that also the protein-bound lipids exhibit a wide range of conformations, without significant differences between different lipid types: The specificity of lipid binding to proteins is not regulated by accessible lipid structures, but a single lipid type can adapt to various binding pockets.

Our results suggest that lipid headgroup binding to proteins can be described by the *inverse conformational selection model*, where unbound lipids sample a wide conformational ensemble offering the large selection of conformations that can bind to various binding sites in substrates, such as protein, drug, RNA, or virus, that remain conformationally fixed. In this model, the binding energy is dominated by intermolecular interactions and changes in lipid conformational entropy, while conformational restrictions inherent to the lipids play a minor role. Here

we have focused only on lipid headgroups, while potential applications of the proposed model on interactions between hydrophobic acyl chains and proteins are left for future studies. Applications of the proposed model in molecular cell biology include the understanding of lipid-mediated cell signaling and functions of lipid-regulated membrane proteins in general. In bionanotechnology, our model facilitates, for example, the design of phospholipid-specific antibodies and lipid nanoparticle carriers with minimal side effects for mRNA vaccines. Furthermore, our work has demonstrated the power of open-access MD simulation data to complement the PDB data in elucidating the behavior of complex systems of disordered biomolecules.

METHODS

Experimental C–H Bond Order Parameters. The headgroup and glycerol backbone C–H bond order parameters of 1-palmitoyl-2-oleoyl-*sn*-glycero-3-phosphoethanolamine (POPE) and 1-palmitoyl-2-oleoyl-*sn*-glycero-3-phospho-(1'-*rac*-glycerol) (POPG), purchased from Avanti Polar Lipids, were measured using natural abundance ¹H–¹³C solid-state NMR spectroscopy as described previously.^{33,34} The samples were prepared by simply mixing the lipids with water until a homogeneous preparation of multi-lamellar vesicles (MLVs) was attained and left to equilibrate for approximately 24 h before measurements. The magnitudes of the order parameters were determined using an R-type Proton Detected Local Field (R-PDLF) experiment⁴⁷ and the order parameter signs from S-DROSS experiments⁴⁸ combined with SIMPSON simulations.⁴⁹ The setup of the solid-state NMR experiments was identical to that in our previous work.³² The POPE experiments were recorded at 310 K and POPG experiments at 298 K, where the bilayers are in the liquid disordered phase.⁵⁰

Glycerol backbone peaks from both lipids, and the α -carbon peak from POPE in the INEPT spectra, were assigned based on previously measured POPC spectra.³³ The β -carbon peak from POPE was assigned based on ¹³C chemical shift table for amines available at <https://www.chem.wisc.edu/areas/reich/nmr/c13-data/cdata.htm>. The assignment of the α , β , and γ peaks of POPG was based on the measured R-PDLF dipolar couplings in comparison to the known order parameters from previous ²H NMR experiments.^{20,35} The β -carbon peak from POPG overlapped with the g_2 peak from the glycerol backbone because their chemical environments are similar.

Molecular Dynamics Simulations. Molecular dynamics simulation data were collected and analyzed using the methods from the NMRlipids Open Collaboration project (nmrlipids.blogspot.fi).^{21,32,37,51} Simulation details, accessibility information, and quality evaluation of approximately 70 combinations of lipid headgroup and force field simulated for this work are in the Supporting Information.

Best models for the interpretation of lipid headgroup conformational ensembles from the experimental data were selected using quality evaluation measures defined in the NMRlipids project. Conformational ensembles of headgroup and glycerol backbone in PE and PG simulations were evaluated using the C–H bond order parameters.³⁷ Interactions between different headgroups were evaluated by monitoring the changes in headgroup order parameters upon mixing the lipids.³² The ion binding affinities and response of lipids to bound charge were evaluated by monitoring the changes in lipid headgroup order parameters.^{32,51}

The differences in headgroup conformation ensembles were analyzed by calculating the distributions of six dihedral angles, $O_{\alpha}-C_{\alpha}-C_{\beta}-N/C_{\gamma}$, $P-O_{\alpha}-C_{\alpha}-C_{\beta}$, $O_{g_3}-P-O_{\alpha}-C_{\omega}$, $g_3-O_{g_3}-P-O_{\omega}$, $g_2-g_3-O_{g_3}-P$, and $g_1-g_2-g_3-O_{g_3}$, labeled in Figure 1C. Relative free energy costs for turning dihedral angles with respect to the most probable value (lowest free energy) were estimated from the inverse Boltzmann formula $\Delta E(\theta) = -kT[\ln[p(\theta)] - \ln[p(\theta_0)]]$, where $p(\theta)$ is the dihedral angle distribution and θ_0 is the most probable angle from MD simulation.

Analysis of Protein-Bound Lipid Conformations. Lipid structures from the Protein Data Bank (PDB⁶) were searched using PDBe REST API (www.ebi.ac.uk/pdbe/pdbe-rest-api) and the ligand names listed in the Supporting Information. Dihedral distributions from the lipid structures were calculated using the MDAnalysis Python library^{52,53} and Jupyter notebook available from *scripts/PDBanalysis.ipynb* folder in ref 46. Only the structures determined using X-ray crystallography or cryo-EM with a resolution higher than 3.2 Å were used in the analysis. Some structures of multimeric proteins contained multiple lipids in the same conformation. The conformations were considered identical if all dihedral angles were the same within 3°. In these cases, this conformation was counted only once for the subsequent analyses to avoid its overweighting.

To demonstrate the existence of similar structures of different lipids bound to different proteins, we searched for pairs having the last five heavy-atom dihedrals angles in the headgroup end (excluding $g_1-g_2-g_3-O_g$, from the list in the previous section) within 30° of each other among the structures with resolution higher than 2.5 Å. The two most representative examples were handpicked from the results.

■ ASSOCIATED CONTENT

SI Supporting Information

The Supporting Information is available free of charge at <https://pubs.acs.org/doi/10.1021/jacs.1c05549>.

Additional NMR data, ligand names in PDB used to analyze the protein-bound lipid conformations, evaluation of MD simulations against NMR experiments, dihedral angle distributions from simulations, and description of simulated systems with simulation details (PDF)

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Notes

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