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Influence of an Oriented External Electric Field on Double Proton Transfer Reactions

Geoffroy-Neveux Antoine*, Labet Vanessa, Alikhani Esmail Laboratoire MONARIS, Sorbonne Université, Paris, France

antoine.geoffroy_neveux@sorbonne-universite.fr, vanessa.labet@sorbonne-universite.fr, esmail.alikhani@sorbonne-universite.fr



1. Introduction

Work dealing with the influence of an oriented external electric field (OEEF) on the reactivity of organic compounds has increased in recent years. This type of external stimulus, called a "smart reagent", makes it possible to control the kinetics and selectivity of a reaction. [1] In this study, [2] we investigated the influence of an OEEF on the mechanism of double proton transfer in the pyrazole-guanidine complex. This reaction is concerted in the absence of an OEEF, but its potential energy profile exhibits a broad plateau in the region of the transition state, suggesting a strongly asynchronous mechanism.[3]

2. Conventions and methods			
The reaction studied is the double proton transfer in the	$ \xrightarrow{\mathbf{F}_{z} > 0} Axe z $ $ \xrightarrow{\mathbf{F}_{z} > 0} \bigoplus $ $ \xrightarrow{\mathbf{F}_{z} < 0} \bigoplus $	$\frac{\overline{F_z \text{ conversion}}}{\overline{Gaussian au MV.cm^{-1}}} \bullet DFT \ \omega B97X-D/6-311++G(2d,2p)$	



3. Without electric field

• Potential energy profile : Concerted mecanism



 \star Principal component analysis of IRC struc-

4. Effect of the Oriented External Electric Field

• Potential energy profile with electric field



$$F_z < -20 \implies Concerted \rightarrow Stepwise$$







Both protons are transferred in the transition state region

 \star ELF Analysis along the reaction path



Proton transfer at the edge of the plateau \implies Concerted asynchronous mecanism

5. Conclusions

We have provided a detailed analysis of the effect of an OEEF on an atypical mechanism: the plateau reaction in the Pyrazole-Guanidine complex (asynchronous concerted mechanism). The electric field is a "smart reagent" that allows us to modulate the nature of the reaction mechanism to transfer protons stepwise, or in a concerted synchronous manner. It is a powerful tool for analyzing the arrangement of primitive processes in concerted mechanisms.

6. References

[1] S. Shaik *et al.*, J. Am. Chem. Soc. **2021**, 142(29):12551 - 12562 [2] A. Geoffroy-Neveux et al., J. Phys. Chem. A 2022, 126(20):3057-3071 [3] G. Rauhut *et al.*, Phys. Chem. Chem. Phys. **2003**, 5(5):791 - 800