



HAL
open science

A density-fitting implementation of the density-based basis-set correction method

Andreas Hesselmann, Emmanuel Giner, Peter Reinhardt, Peter Knowles, Hans-Joachim Werner, Julien Toulouse

► To cite this version:

Andreas Hesselmann, Emmanuel Giner, Peter Reinhardt, Peter Knowles, Hans-Joachim Werner, et al.. A density-fitting implementation of the density-based basis-set correction method. 2023. hal-04210752v1

HAL Id: hal-04210752

<https://hal.sorbonne-universite.fr/hal-04210752v1>

Preprint submitted on 19 Sep 2023 (v1), last revised 9 Feb 2024 (v2)

HAL is a multi-disciplinary open access archive for the deposit and dissemination of scientific research documents, whether they are published or not. The documents may come from teaching and research institutions in France or abroad, or from public or private research centers.

L'archive ouverte pluridisciplinaire **HAL**, est destinée au dépôt et à la diffusion de documents scientifiques de niveau recherche, publiés ou non, émanant des établissements d'enseignement et de recherche français ou étrangers, des laboratoires publics ou privés.

A density-fitting implementation of the density-based basis-set correction method

Andreas Heßelmann,^{1,*} Emmanuel Giner,^{2,†} Peter Reinhardt,^{2,‡} Peter J. Knowles,^{3,§} Hans-Joachim Werner,^{1,¶} and Julien Toulouse^{2,4,**}

¹*Institute for Theoretical Chemistry, University of Stuttgart, 70569, Stuttgart, Germany*

²*Laboratoire de Chimie Théorique, Sorbonne Université and CNRS, F-75005 Paris, France*

³*School of Chemistry, Cardiff University, Cardiff CF10 3AT, United Kingdom*

⁴*Institut Universitaire de France, F-75005 Paris, France*

(Dated: September 18, 2023)

This work reports an efficient density-fitting implementation of the density-based basis-set correction (DBBSC) method in the MOLPRO software. This method consists in correcting the energy calculated by a wave-function method with a given basis set by an adapted basis-set correction density functional incorporating the short-range electron correlation effects missing in the basis set, resulting in an accelerated convergence to the complete-basis-set limit. Different basis-set correction density-functional approximations are explored and the complementary-auxiliary-basis-set single-excitation correction is added. The method is tested on a benchmark set of reaction energies at the second-order Moller-Plesset (MP2) level and a comparison with the explicitly correlated MP2-F12 method is provided. The results show that the DBBSC method greatly accelerates the basis convergence of MP2 reaction energies, without reaching the accuracy of the MP2-F12 method but with a lower computational cost.

Keywords: basis-set convergence; Moller-Plesset perturbation theory; density-functional theory; reaction energies

I. INTRODUCTION

One of the main goals of quantum chemistry is the accurate prediction of molecular properties, which requires to tackle the electron correlation problem. For this, there are two main families of computational electronic-structure methods: wave-function theory (WFT) [1] which targets the complicated N -electron wave function, and density-functional theory (DFT) [2] which uses the simpler one-electron density. While DFT has become the workhorse of quantum chemistry thanks to its appealing balance between computational cost and accuracy, the lack of a systematic scheme to improve the quality of the density-functional approximations has renewed the interest in the development of WFT methods in the last few decades.

A serious limitation of WFT methods is their slow convergence of the correlation energy with the size of the one-electron basis set. This slow convergence originates from the short-range singularity of the Coulomb electron-electron repulsion which induces a derivative discontinuity in the exact eigenstate wave functions, known as the electron-electron cusp condition [3]. There are two main approaches for dealing with this problem. The first approach consists in extrapolating the results to the complete-basis-set (CBS) limit by using increasingly large basis sets [4, 5]. The second approach consists in using explicitly correlated R12 or F12 methods which incorporate in the wave function a correlation factor reproducing the electron-electron cusp (see, e.g., Refs. 6–16).

An alternative approach to accelerate basis-set convergence was recently proposed, which we will refer as the density-based basis-set correction (DBBSC) method [17]. It consists in correcting the energy calculated by a WFT method with a given basis set by an adapted basis-set correction density functional incorporating the short-range electron correlation effects missing in the basis set, resulting in an accelerated convergence to the CBS limit. In practice, this basis-set correction density functional is constructed from range-separated DFT [18] by defining a basis-dependent local range-separation parameter which provides a local measure of the incompleteness of the basis set. This DBBSC method was validated for configuration-interaction and coupled-cluster calculations of atomization energies [19–21], excitation energies [22], dissociation energy curves [23], and dipole moments [24, 25]. It was also extended to GW calculations [26] and to linear-response theory [27], and some mathematical aspects of the method were studied in detail on a one-dimensional model system [28].

In this work, we report an efficient implementation of the DBBSC method in the MOLPRO software [29–31] in which density fitting [32] is used to alleviate the computational bottleneck of the method, namely the calculation of the local range-separation parameter. This allows us to use the DBBSC method on larger molecular systems than what was previously possible. We thus apply the DBBSC method for correcting the basis-set errors in the molecular reaction energies of the FH51 benchmark set [33, 34] at the second-order Moller-Plesset (MP2) level. We also test different basis-set correction density-functional approximations, as well as the addition of a single-excitation correction for one-electron basis-set errors. Finally, we compare the performance of the DBBSC method with the explicitly correlated MP2-F12 method [11].

The paper is organized as follows. In Section II, we explain the theory of the present implementation of the DBBSC method. Section III provides computational details for the cal-

*Electronic address: andreas.hesselmann@theochem.uni-stuttgart.de

†Electronic address: emmanuel.giner@lct.jussieu.fr

‡Electronic address: peter.reinhardt@lct.jussieu.fr

§Electronic address: KnowlesPJ@Cardiff.ac.uk

¶Electronic address: werner@theochem.uni-stuttgart.de

**Electronic address: toulouse@lct.jussieu.fr

culations on the FH51 benchmark set. In Section IV, we give and discuss our results on the reaction energies. Finally, Section V contains our conclusions.

II. THEORY

For simplicity, we give the equations for closed-shell states and we assume real-valued HF spatial orbitals $\{\varphi_p\}$.

A. The DBBSC method at the MP2 level

Given the MP2 total energy $E_{\text{MP2}}^{\mathcal{B}}$ in a basis set \mathcal{B} , we apply the non-self-consistent basis-set correction [17, 19] as

$$E_{\text{MP2+DFT}}^{\mathcal{B}} = E_{\text{MP2}}^{\mathcal{B}} + \bar{E}^{\mathcal{B}}[n_{\text{HF}}^{\mathcal{B}}], \quad (1)$$

where $\bar{E}^{\mathcal{B}}[n_{\text{HF}}^{\mathcal{B}}]$ is the basis-correction density functional evaluated the active HF density $n_{\text{HF}}^{\mathcal{B}}$ (i.e., excluding core orbitals in case of frozen-core calculations). In order not to affect the CBS limit, this functional $\bar{E}^{\mathcal{B}}[n]$ must be such that it vanishes when the basis set \mathcal{B} is complete. Moreover, provided a good enough approximation is used for $\bar{E}^{\mathcal{B}}[n]$, the basis-set corrected MP2 energy, referred to as ‘‘MP2+DFT’’, is expected to converge faster to the MP2 CBS limit.

B. Local range-separation parameter

The dependence on the basis set of the basis-correction density functional $\bar{E}^{\mathcal{B}}[n]$ comes from the local range-separation parameter $\mu^{\mathcal{B}}(\mathbf{r})$. It is defined as [17, 19]

$$\mu^{\mathcal{B}}(\mathbf{r}) = \frac{\sqrt{\pi}}{2} W^{\mathcal{B}}(\mathbf{r}), \quad (2)$$

where $W^{\mathcal{B}}(\mathbf{r})$ is the on-top value of the effective interaction localized with the HF wave function

$$W^{\mathcal{B}}(\mathbf{r}) = \begin{cases} \frac{f_{\text{HF}}^{\mathcal{B}}(\mathbf{r})}{n_{2,\text{HF}}^{\mathcal{B}}(\mathbf{r})}, & \text{if } n_{2,\text{HF}}^{\mathcal{B}}(\mathbf{r}) \neq 0, \\ \infty, & \text{otherwise.} \end{cases} \quad (3)$$

In Eq. (3), $n_{2,\text{HF}}^{\mathcal{B}}(\mathbf{r})$ is the HF on-top pair density

$$n_{2,\text{HF}}^{\mathcal{B}}(\mathbf{r}) = \frac{n_{\text{HF}}^{\mathcal{B}}(\mathbf{r})^2}{2}, \quad (4)$$

with the active HF density $n_{\text{HF}}^{\mathcal{B}}(\mathbf{r}) = 2 \sum_i^{\text{act}} \varphi_i(\mathbf{r})^2$, and $f_{\text{HF}}^{\mathcal{B}}(\mathbf{r})$ has the expression

$$f_{\text{HF}}^{\mathcal{B}}(\mathbf{r}) = 2 \sum_{p,q}^{\text{all}} \sum_{i,j}^{\text{act}} \varphi_p(\mathbf{r}) \varphi_i(\mathbf{r}) (\varphi_p \varphi_i | \varphi_q \varphi_j) \varphi_q(\mathbf{r}) \varphi_j(\mathbf{r}), \quad (5)$$

where p and q run over all (occupied + virtual) HF spatial orbitals, i and j run over active HF spatial orbitals, and $(\varphi_p \varphi_i | \varphi_q \varphi_j)$ are the two-electron Coulomb integrals in chemists’ notation. We recall that by active orbitals we mean

occupied orbitals without the frozen-core orbitals, in case of frozen-core calculations.

The local range-separation parameter $\mu^{\mathcal{B}}(\mathbf{r})$ provides a local measure of the incompleteness of the basis set. A straightforward calculation of $f_{\text{HF}}^{\mathcal{B}}(\mathbf{r})$ in Eq. (5) requires to first calculating the molecular-orbital two-electron integrals $(\varphi_p \varphi_i | \varphi_q \varphi_j)$ with a dominant scaling of $O(N_{\text{act}} N_{\text{all}}^4)$, and then performing the sums at each grid point which scales as $O(N_{\text{act}}^2 N_{\text{all}}^2 N_{\text{grid}})$, where N_{act} is the number of active orbitals, N_{all} is the total number of orbitals in the basis, and N_{grid} is the number of spatial grid points. This is the computational bottleneck of the basis-set correction calculation.

This scaling can be reduced by density fitting [32]. Introducing an auxiliary fitting basis set $\{\chi_A\}$, the orbital product is approximated as

$$\varphi_p(\mathbf{r}) \varphi_i(\mathbf{r}) \approx \sum_A^{\text{fit}} d_A^{pi} \chi_A(\mathbf{r}), \quad (6)$$

where d_A^{pi} are the Coulomb-fitting coefficients

$$d_B^{pi} = \sum_A^{\text{fit}} (\varphi_p \varphi_i | \chi_A) [\mathbf{J}^{-1}]_{AB}, \quad (7)$$

with

$$J_{AB} = \iint \frac{\chi_A(\mathbf{r}_1) \chi_B(\mathbf{r}_2)}{\|\mathbf{r}_2 - \mathbf{r}_1\|} d\mathbf{r}_1 d\mathbf{r}_2, \quad (8)$$

and

$$(\varphi_p \varphi_i | \chi_A) = \iint \frac{\varphi_p(\mathbf{r}_1) \varphi_i(\mathbf{r}_1) \chi_B(\mathbf{r}_2)}{\|\mathbf{r}_2 - \mathbf{r}_1\|} d\mathbf{r}_1 d\mathbf{r}_2. \quad (9)$$

Orthonormalizing the auxiliary fitting basis functions with respect to the metric \mathbf{J} ,

$$\tilde{\chi}_A = \sum_B^{\text{fit}} [\mathbf{J}^{-1/2}]_{AB} \chi_B, \quad (10)$$

we can approximate the two-electron integrals as

$$(\varphi_p \varphi_i | \varphi_q \varphi_j) \approx \sum_A^{\text{fit}} (\varphi_p \varphi_i | \tilde{\chi}_A) (\tilde{\chi}_A | \varphi_q \varphi_j), \quad (11)$$

and the quantity $f_{\text{HF}}^{\mathcal{B}}(\mathbf{r})$ in Eq. (5) as

$$f_{\text{HF}}^{\mathcal{B}}(\mathbf{r}) \approx 2 \sum_A^{\text{fit}} \left[\sum_p^{\text{all}} \sum_i^{\text{act}} \varphi_p(\mathbf{r}) \varphi_i(\mathbf{r}) (\varphi_p \varphi_i | \tilde{\chi}_A) \right]^2. \quad (12)$$

Thus, with density fitting, there is no need to build explicitly the two-electron integrals anymore and the calculation of $f_{\text{HF}}^{\mathcal{B}}(\mathbf{r})$ in Eq. (12) now scales as $O(N_{\text{act}} N_{\text{all}} N_{\text{fit}} N_{\text{grid}})$ where N_{fit} is the number of auxiliary fitting basis functions. In practice, the same auxiliary fitting basis sets optimized for density fitting in MP2 can be used here.

C. Approximate basis-correction density functional

We approximate the basis-correction density functional with the local form [19]

$$\bar{E}^{\mathcal{B}}[n] \approx \int \bar{e}_{\text{c,md}}^{\text{sr}}(n(\mathbf{r}), \nabla n(\mathbf{r}), \mu^{\mathcal{B}}(\mathbf{r})) d\mathbf{r}, \quad (13)$$

where $\bar{e}_{\text{c,md}}^{\text{sr}}(n, \nabla n, \mu)$ is the complementary multi-determinant short-range correlation functional energy density [19, 35]

$$\bar{e}_{\text{c,md}}^{\text{sr}}(n, \nabla n, \mu^{\mathcal{B}}) = \frac{e_c(n, \nabla n)}{1 + \frac{e_c(n, \nabla n)}{c n_2(n)} \mu^3}, \quad (14)$$

where $c = (2\sqrt{\pi}(1 - \sqrt{2}))/3$ and $n_2(n)$ is a model of the on-top pair density. In Eq. (14), $e_c(n, \nabla n)$ is a standard Kohn-Sham correlation functional energy density. As in previous works, the default choice is the PBE correlation functional [36]. In this work, we also test using the LDA [37], LYP [38], TPSS [39], and SCAN [40] correlation functionals. Note that the TPSS and SCAN functionals are meta-GGA functionals, i.e. they depend also on the non-interacting positive kinetic energy density $\tau(\mathbf{r}) = (1/2) \sum_i^{\text{act}} |\nabla \varphi(\mathbf{r})|^2$, and thus constitute a slight extension of Eqs. (13) and (14).

The default choice [19] for $n_2(n)$ is to use the on-top pair density of the uniform-electron gas (UEG)

$$n_2^{\text{UEG}}(n) = n^2 g_0(n), \quad (15)$$

where the on-top pair-distribution function $g_0(n)$ is parametrized in Eq. (46) of Ref. 41. In this work, we also explore two other on-top pair-density models. The first one is the CS model [42–44]

$$n_2^{\text{CS}}(n) = \frac{n^2}{2} \Phi_{\text{CS}}(n)^2, \quad (16)$$

where

$$\Phi_{\text{CS}}(n) = \frac{\sqrt{\pi} \beta(n)}{1 + \sqrt{\pi} \beta(n)}, \quad (17)$$

and

$$\beta(n) = q n^{1/3}, \quad (18)$$

where q is an empirical parameter. The second one is the Hollett-Pegoretti (HP) model [45]

$$n_2^{\text{HP}}(n) = \frac{n^2}{2} \Phi_{\text{HP}}(n), \quad (19)$$

where

$$\Phi_{\text{HP}}(n) = \frac{2\sqrt{\pi} \beta(n)^2}{2\beta(n) e^{-\frac{1}{4\beta(n)^2}} + \sqrt{\pi} (1 + 2\beta(n)^2) \left[1 + \text{erf}\left(\frac{1}{2\beta(n)}\right)\right]}. \quad (20)$$

We may choose the value of the parameter q , e.g., by imposing that the integral of the model on-top pair density equals the integral of the exact on-top pair density, $\int n_2^{\text{model}}(n(\mathbf{r})) d\mathbf{r} =$

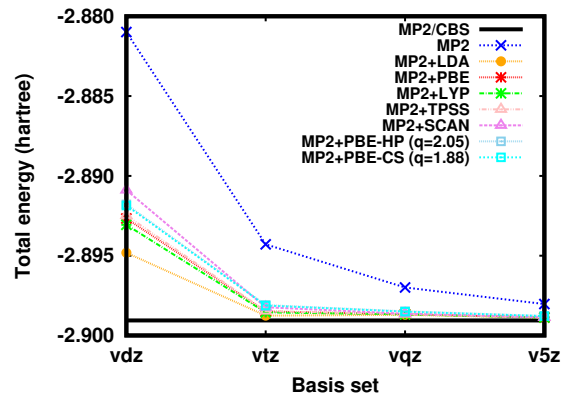


FIG. 1: Basis-set convergence of the total MP2 ground-state energy of the He atom with different basis-set correction density-functional approximations (evaluated at the HF density) using vnz basis sets.

$\int n_2^{\text{exact}}(\mathbf{r}) d\mathbf{r}$, in the helium atom. Estimating $n_2^{\text{exact}}(\mathbf{r})$ with a highly accurate 418-term Hylleraas-type wave function [46–48], we find $q = 1.88$ for the CS model and $q = 2.05$ for the HP model. When these on-top pair-density models are used with the PBE correlation functional in Eq. (14), we call the resulting basis-set correction functionals PBE-CS and PBE-HP, respectively.

As a first test, we compare in Fig. 1 these different basis-set correction density-functional approximations for the basis-set convergence of the total MP2 ground-state energy of the He atom with cc-pVnZ basis sets [49] (abbreviated as vnz). We see that all these density-functional approximations lead to a big acceleration of the MP2 total energy toward its CBS limit. We thus conclude at this point that all the proposed density-functional approximations provides a reasonable basis-set correction, at least for the total energy.

D. CABS single-excitation correction

For small basis sets \mathcal{B} , the HF energy can have a substantial basis-set error. This HF basis-set error is not corrected by the approximate basis-set correction functionals in Section II C since they only correct for missing short-range correlation. The HF basis-set error can however be easily corrected by using the complementary auxiliary basis set (CABS) [10] used in explicitly correlated R12/F12 methods. In this approach, a large orthonormal basis set is formed by the occupied+virtual HF orbitals obtained in the normal basis set \mathcal{B} and an additional set of virtual orbitals obtained from the CABS. The HF energy correction due to the addition of the CABS is estimated by second-order perturbation theory, leading to the expression, in a closed-shell formalism, [12, 50, 51]

$$\Delta E_{\text{HF}}^{\mathcal{B}, \text{CABS}} = 2 \sum_i^{\text{act}} \sum_{\alpha}^{\text{vir}} t_{\alpha}^i f_i^{\alpha}, \quad (21)$$

where i runs over active HF orbitals and α runs over all virtual orbitals (obtained in the normal basis set \mathcal{B} and from the

CABS). In Eq. (21), f_i^α are Fock matrix elements and t_α^i are single-excitation coefficients found by solving the first-order perturbation equations

$$f_\alpha^i = \sum_j^{\text{act}} t_\alpha^j f_j^i - \sum_\beta^{\text{vir}} f_\alpha^\beta t_\beta^i. \quad (22)$$

The correction is often referred to as the CABS single-excitation correction. The total basis-set corrected MP2 energy is thus

$$E_{\text{MP2+CABS+DFT}}^{\mathcal{B}} = E_{\text{MP2}}^{\mathcal{B}} + \Delta E_{\text{HF}}^{\mathcal{B},\text{CABS}} + \bar{E}^{\mathcal{B}}[n_{\text{HF}}^{\mathcal{B}}], \quad (23)$$

and will be referred to as ‘‘MP2+CABS+DFT’’. For comparison, we will also present MP2 results only corrected by the CABS single-excitation correction, which will be referred to as ‘‘MP2+CABS’’.

III. COMPUTATIONAL DETAILS

The DBBSC method with density fitting has been implemented in the MOLPRO software [29–31]. We have performed tests on the FH51 benchmark set. The FH51 set [33, 34] is a set of 51 reaction energies for various organic molecules. It is included in the GMTKN55 database [52]. As regards the basis set \mathcal{B} , we use the aug-cc-pVnZ basis sets [53] for first-row atoms and the aug-cc-pV(n+d)Z basis sets [54] for second-row atoms, which we jointly abbreviate as *avnz*, for $n = 2$ (d), 3 (t), 4 (q), and 5.

We perform canonical-orbital density-fitting HF [55] and density-fitting MP2 [32] calculations with the frozen-core approximation. The MP2/CBS reference values are estimated from the two largest basis sets ($n = 4$ and $n = 5$) by using the two-point extrapolation formula of Ref. 56 for the HF energy and the standard two-point extrapolation formula of Refs. 4, 5 for the MP2 correlation energy. We calculate the basis-set correction with different functionals evaluated at the active HF density, and including the CABS single-excitation correction [12, 50, 51]. The basis-set correction is consistently calculated in the frozen-core approximation, corresponding to using only active orbitals in Eq. (5) and in the HF density used in Eq. (1). For the $n = 2, 3$, for comparison, we also perform canonical-orbital density-fitting MP2-F12 (in the default 3C(F) variant) [11] calculations, implicitly including the CABS single-excitation correction.

For a given basis set \mathcal{B} , the density-fitting basis sets used are the corresponding \mathcal{B}/JKFIT and $\mathcal{B}/\text{MP2FIT}$ basis sets of Weigend *et al.* [57, 58] (and their extensions [51]) for the HF and MP2 calculations, respectively. The \mathcal{B}/JKFIT basis set is also used as CABS for the CABS single-excitation correction. We have checked the density-fitting errors and found them to be insignificant. For large systems, density-fitting calculations of the basis-set correction can be orders of magnitude faster than non-density-fitting calculations.

IV. RESULTS

The errors on the reaction energies of the FH51 set with respect to MP2/CBS calculated with MP2, MP2+CABS, MP2+CABS+PBE, and MP2-F12 are reported in Fig. 2. With the avdz basis set, MP2 can have quite large basis errors for some reaction energies, up to about 12 kcal/mol. Obtaining MP2 reaction energies with all basis errors below 1 kcal/mol requires the use of the av5z basis set. The CABS single-excitation correction is crucial to reduce the largest basis errors on MP2 reaction energies obtained with the avdz basis set. Even with larger basis sets, the CABS single-excitation correction still helps reducing the basis errors for some reaction energies. Adding the PBE-based basis-set correction further reduces the basis errors, albeit not always in a systematic way since there are a few cases where the basis error increases. It is noteworthy that the basis errors of the MP2+CABS+PBE reaction energies are all smaller than 1 kcal/mol with the avtz basis set and larger basis sets. MP2-F12 globally outperforms MP2+CABS+PBE, giving reaction energies with basis errors below about 1 kcal/mol already with the avdz basis set.

In Table I, we report the mean absolute errors (MAEs) on the reaction energies of the FH51 set with respect to MP2/CBS obtained with the methods already discussed, as well as with additional basis-set correction functionals, namely LDA, LYP, TPSS, SCAN, PBE-CS ($q = 1.88$), and PBE-HP ($q = 2.05$). For the methods already discussed, the mean errors are consistent with the observations made previously. For the avdz basis set, we go from a MAE of 2.07 kcal/mol for uncorrected MP2 to a MAE of 0.60 kcal/mol for MP2+CABS+PBE and a MAE of 0.33 kcal/mol for MP2-F12. For the avtz basis set, we go from a MAE of 0.73 kcal/mol for uncorrected MP2 to a MAE of 0.21 kcal/mol for MP2+CABS+PBE and a MAE of 0.15 kcal/mol for MP2-F12. For the avqz and av5z basis sets, the PBE-based basis-set correction is still effective in reducing the basis errors, as we go from MAEs of 0.25 and 0.12 kcal/mol, respectively, for uncorrected MP2 to MAEs of 0.12 and 0.07 kcal/mol, respectively, for MP2+CABS+PBE. Thus, MP2+CABS+PBE with an *avnz* basis set globally gives uncorrected MP2 reaction energies with slightly higher *av(n+1)z* quality, whereas MP2-F12 with an *avnz* basis set roughly gives uncorrected MP2 reaction energies with slightly lower *av(n+2)z* quality.

With the other basis-set correction functionals tested, the MAEs are very similar, except for the LYP correlation functional which gives much larger basis errors. We have also tested optimizing the parameter q in the CS and HP on-top pair density-density models in Eq. (18) and the parameter c in front of the on-top pair density in Eq. (14), but we did not obtain significant improvements. Thus, if we set aside LYP, we find a rather small sensitivity of the method to the underlying correlation functional for calculating reaction energies.

Finally, as regards computational costs, we consistently observe, for all basis sets, that MP2+CABS+PBE is approximately 10 times faster than MP2-F12 in the default 3C variant. However, we note that MP2-F12 can be made faster using the 3*A approximation [11] without losing much accuracy in most cases, and MP2+CABS+PBE is only approximately 3 to

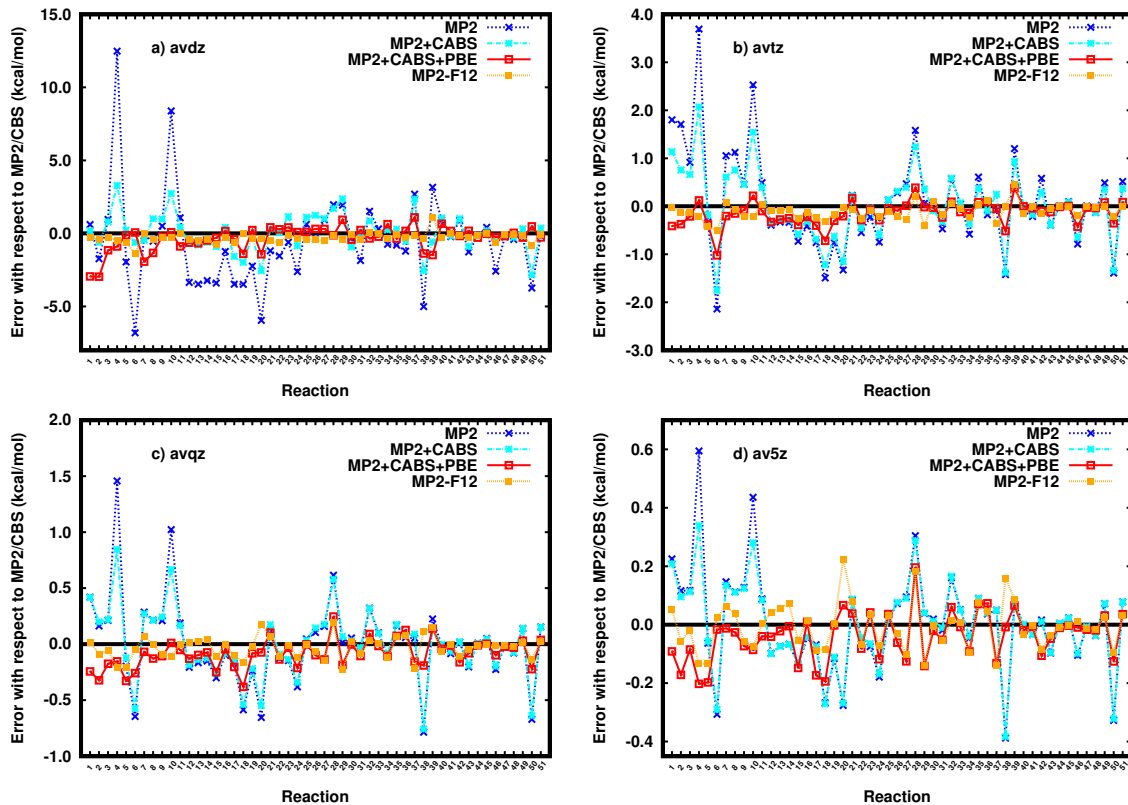


FIG. 2: Errors on reaction energies of the FH51 set with respect to MP2/CBS calculated with MP2, MP2+CABS, MP2+CABS+PBE, and MP2-F12 with *avnz* basis sets. The order of reactions is the one from Refs. 33, 34.

TABLE I: Mean absolute errors (in kcal/mol) on reaction energies of the FH51 set with respect to MP2/CBS with *avnz* basis sets.

	avdz	avtz	avqz	av5z
MP2	2.07	0.73	0.25	0.12
MP2+CABS	0.92	0.55	0.22	0.11
MP2+PBE	1.72	0.35	0.15	0.07
MP2+CABS+PBE	0.60	0.21	0.12	0.07
MP2+CABS+LDA	0.64	0.19	0.11	0.07
MP2+CABS+LYP	1.10	0.46	0.23	0.23
MP2+CABS+TPSS	0.61	0.21	0.12	0.07
MP2+CABS+SCAN	0.64	0.24	0.12	0.07
MP2+CABS+PBE-CS ($q = 1.88$)	0.60	0.22	0.11	0.06
MP2+CABS+PBE-HP ($q = 2.05$)	0.61	0.22	0.12	0.07
MP2-F12	0.33	0.15	0.08	0.06

4 times faster than this cheaper MP2-F12 variant. Of course, the relative gains in computational cost would be much less for more expensive wave-function methods such as CCSD(T).

V. CONCLUSION

We have reported an efficient density-fitting implementation of the DBBSC method in the MOLPRO software us-

ing different basis-set correction density-functional approximations and including the CABS single-excitation correction. We have tested the method on the FH51 benchmark set of reaction energies at the MP2 level and provided a comparison with the explicitly correlated MP2-F12 method.

For the smallest basis sets, the CABS single-excitation correction provides an important correction on reaction energies which is not included in the basis-set correction density-functional approximations. The basis-set corrected reaction energies are quite insensitive to the choice of the basis-set correction density-functional approximation, with the notable exception of the LYP functional which gives much worse results. This point should be further analyzed in the future. Overall, the basis-set corrected MP2 reaction energies calculated with a n -zeta basis set are of slightly higher quality than uncorrected MP2 reaction energies calculated with $(n+1)$ -zeta quality. However, the explicitly correlated MP2-F12 method is consistently more accurate, with reaction energies calculated with a n -zeta basis set being of slightly lower quality than uncorrected MP2 reaction energies calculated with $(n+2)$ -zeta quality. We believe that the DBBSC method is still valuable for accelerating the basis convergence of MP2 due to the fact that it has a lower computational cost than MP2-F12.

Finally, let us mention that the present implementation of the DBBSC method can be applied to any other wave-function methods, such as CCSD(T), with expected similar gains in accuracy.

Acknowledgments

It is a pleasure to dedicate the present paper to Carlo Adamo on the occasion of his 60th birthday.

Conflict of interest

None of the authors have a conflict of interest to disclose.

-
- [1] J. A. Pople, *Rev. Mod. Phys.* **71**, 1267 (1999).
- [2] W. Kohn, *Rev. Mod. Phys.* **71**, 1253 (1999).
- [3] T. Kato, *Comm. Pure Appl. Math.* **10**, 151 (1957).
- [4] T. Helgaker, W. Klopper, H. Koch and J. Noga, *J. Chem. Phys.* **106**, 9639 (1997).
- [5] A. Halkier, T. Helgaker, P. Jorgensen, W. Klopper, H. Koch, J. Olsen and A. K. Wilson, *Chem. Phys. Lett.* **286**, 243 (1998).
- [6] W. Klopper and C. M. Samson, *J. Chem. Phys.* **116**, 6397 (2002).
- [7] F. R. Manby, *J. Chem. Phys.* **119**, 4607 (2003).
- [8] S. Ten-no, *J. Chem. Phys.* **121**, 117 (2004).
- [9] S. Ten-no, *Chem. Phys. Lett.* **398**, 56 (2004).
- [10] E. F. Valeev, *Chem. Phys. Lett.* **395**, 190 (2004).
- [11] H.-J. Werner, T. B. Adler and F. R. Manby, *J. Chem. Phys.* **126**, 164102 (2007).
- [12] G. Knizia, T. B. Adler and H.-J. Werner, *J. Chem. Phys.* **130**, 054104 (2009).
- [13] S. Ten-no and J. Noga, *WIREs Comput. Mol. Sci.* **2**, 114 (2012).
- [14] C. Hättig, W. Klopper, A. Köhn and D. P. Tew, *Chem. Rev.* **112**, 4 (2012).
- [15] L. Kong, F. A. Bischoff and E. F. Valeev, *Chem. Rev.* **112**, 75 (2012).
- [16] T. Shiozaki and H.-J. Werner, *Mol. Phys.* **111**, 607 (2013).
- [17] E. Giner, B. Pradines, A. Ferté, R. Assaraf, A. Savin and J. Toulouse, *J. Chem. Phys.* **149**, 194301 (2018).
- [18] J. Toulouse, F. Colonna and A. Savin, *Phys. Rev. A* **70**, 062505 (2004).
- [19] P.-F. Loos, B. Pradines, A. Scemama, J. Toulouse and E. Giner, *J. Phys. Chem. Lett.* **10**, 2931 (2019).
- [20] Y. Yao, E. Giner, J. Li, J. Toulouse and C. J. Umrigar, *J. Chem. Phys.* **153**, 124117 (2020).
- [21] Y. Yao, E. Giner, T. A. Anderson, J. Toulouse and C. J. Umrigar, *J. Chem. Phys.* **155**, 204104 (2021).
- [22] E. Giner, A. Scemama, J. Toulouse and P.-F. Loos, *J. Chem. Phys.* **151**, 144118 (2019).
- [23] E. Giner, A. Scemama, P.-F. Loos and J. Toulouse, *J. Chem. Phys.* **152**, 174104 (2020).
- [24] E. Giner, D. Traore, B. Pradines and J. Toulouse, *J. Chem. Phys.* **155**, 044109 (2021).
- [25] D. Traore, J. Toulouse and E. Giner, *J. Chem. Phys.* **156**, 174101 (2022).
- [26] P.-F. Loos, B. Pradines, A. Scemama, E. Giner and J. Toulouse, *J. Chem. Theory Comput.* **16**, 1018 (2020).
- [27] D. Traore, E. Giner and J. Toulouse, *J. Chem. Phys.* **158**, 234107 (2023).
- [28] D. Traore, E. Giner and J. Toulouse, *J. Chem. Phys.* **156**, 044113 (2022).
- [29] H.-J. Werner, P. J. Knowles, G. Knizia, F. R. Manby and M. Schütz, *WIREs Comput. Mol. Sci.* **2**, 242 (2012).
- [30] H.-J. Werner, P. J. Knowles, F. R. Manby, J. A. Black, K. Doll, A. Hesselmann, D. Kats, A. Köhn, T. Korona, D. A. Kreplin, Q. Ma, I. Miller, Thomas F., A. Mitrushchenkov, K. A. Peterson, I. Polyak, G. Rauhut and M. Sibaev, *J. Chem. Phys.* **152** (2020).
- [31] H.-J. Werner, P. J. Knowles, and others, MOLPRO, version 2023, a package of ab initio programs, see <http://www.molpro.net>.
- [32] H.-J. Werner, F. R. Manby and P. J. Knowles, *J. Chem. Phys.* **118**, 8149 (2003).
- [33] J. Friedrich and J. Hänchen, *J. Chem. Theory Comput.* **9**, 5381 (2013).
- [34] J. Friedrich, *J. Chem. Theory Comput.* **11**, 3596 (2015).
- [35] A. Ferté, E. Giner and J. Toulouse, *J. Chem. Phys.* **150**, 084103 (2019).
- [36] J. P. Perdew, K. Burke and M. Ernzerhof, *Phys. Rev. Lett.* **77**, 3865 (1996).
- [37] J. P. Perdew and Y. Wang, *Phys. Rev. B* **45**, 13244 (1992).
- [38] C. Lee, W. Yang and R. G. Parr, *Phys. Rev. B* **37**, 785 (1988).
- [39] J. Tao, J. P. Perdew, V. N. Staroverov and G. E. Scuseria, *Phys. Rev. Lett.* **91**, 146401 (2003).
- [40] J. Sun, A. Ruzsinszky and J. P. Perdew, *Phys. Rev. Lett.* **115**, 036402 (2015).
- [41] P. Gori-Giorgi and A. Savin, *Phys. Rev. A* **73**, 032506 (2006).
- [42] R. Colle and O. Salvetti, *Theor. Chim. Acta* **37**, 329 (1975).
- [43] R. Colle and O. Salvetti, *Theor. Chim. Acta* **53**, 55 (1979).
- [44] F. Moscardó, E. San-Fabián and L. Pastor-Abia, *Theor. Chem. Acc.* **115**, 334 (2006).
- [45] J. W. Hollett and N. Pegoretti, *J. Chem. Phys.* **148**, 164111 (2018).
- [46] D. E. Freund, B. D. Huxtable, and J. D. Morgan III, *Phys. Rev. A* **29**, 980 (1984).
- [47] J. D. Baker, D. E. Freund, R. N. Hill, and J. D. Morgan III, *Phys. Rev. A* **41**, 1247 (1990).
- [48] C. J. Umrigar and X. Gonze, *Phys. Rev. A* **50**, 3827 (1994).
- [49] D. Woon and T. Dunning, *J. Chem. Phys.* **100**, 2975 (1994).
- [50] T. B. Adler, G. Knizia and H.-J. Werner, *J. Chem. Phys.* **127**, 221106 (2007).
- [51] G. Knizia and H.-J. Werner, *J. Chem. Phys.* **128**, 154103 (2008).
- [52] L. Goerigk, A. Hansen, C. Bauer, S. Ehrlich, A. Najibi and S. Grimme, *Phys. Chem. Chem. Phys.* **19**, 32184 (2017), Online database: <https://www.chemiebn.uni-bonn.de/pctc/mulliken-center/s>
- [53] R. A. Kendall, T. H. Dunning and R. J. Harrison, *J. Chem. Phys.* **96**, 6796 (1992).
- [54] T. Dunning, K. Peterson and A. Wilson, *J. Chem. Phys.* **114**, 9244 (2001).
- [55] R. Polly, H.-J. Werner, F. R. Manby and P. J. Knowles, *Mol. Phys.* **102**, 2311 (2004).
- [56] A. Karton and J. M. L. Martin, *Theor. Chem. Acc.* **115**, 330 (2006).
- [57] F. Weigend, *Phys. Chem. Chem. Phys.* **4**, 4285 (2002).
- [58] F. Weigend, A. Köhn and C. Hättig, *J. Chem. Phys.* **116**, 3175 (2002).