

# Magnetic properties, magnetic structure, and possible magnetoelectric effect of orthorhombic corundum-like



**Authors:** Simon Redor<sup>1,2</sup>, Maxim Avdeev<sup>3,4</sup>, David Hrabovsky<sup>2,5</sup>, Jean-Marie Tarascon<sup>1</sup> and Gwenaëlle Rousse<sup>1,2,\*</sup>

**Institution adress:** Chimie du Solide et de l'Energie, UMR 8260, Collège de France, 75231, Cedex 05 Paris, France.

## Affiliations:

1 – Chimie du Solide et de l'Energie, UMR 8260, Collège de France, 75231, Cedex 05 Paris, France.

2 – Sorbonne Université, 4 Place Jussieu, F-75005, Paris, France

3 – Australian Nuclear Science and Technology Organisation, Lucas Heights, New South Wales 2234, Australia

4 – School of Chemistry, The University of Sydney, Sydney, New South Wales 2006, Australia

5 – Plateforme Mesures Physiques à Basses Températures (MPBT), Sorbonne Université, 75005 Paris, France

## Abstract

The magnetoelectric effect is an appealing property displayed by most compounds in the corundum-derived “429” series ( $\text{M}_4\text{A}_2\text{O}_9$ ,  $\text{A} = \text{Nb}$  or  $\text{Ta}$ ). The exception is  $\text{Ni}_4\text{Nb}_2\text{O}_9$ , crystallizing in the orthorhombic  $Pbcn$  space group: because of symmetry considerations, its magnetic space group forbids the occurrence of the magnetoelectric effect.  $\text{Li}_2\text{Ni}_2\text{W}_2\text{O}_9$  crystallizes in the same space group as  $\text{Ni}_4\text{Nb}_2\text{O}_9$ ; however, as half the  $\text{Ni}^{2+}$  cations are replaced by  $\text{Li}^+$  cations, the interactions between the remaining magnetic elements are drastically modified. In this report, we present the results of magnetic bulk measurements, showing an antiferromagnetic ordering coupled with strong ferromagnetic interactions below  $T_N = 8$  K in  $\text{Li}_2\text{Ni}_2\text{W}_2\text{O}_9$ . Through neutron diffraction, the magnetic structure of  $\text{Li}_2\text{Ni}_2\text{W}_2\text{O}_9$

is solved, and it is confidently established that it presents the  $Pb'cn'$  Shubnikov space group (#60.425), whose point symmetry  $m'm'm'$  allows the magnetoelectric effect. This could possibly open the door to a new family of magnetoelectric, orthorhombic compounds.

## Introduction

The “429” family is a series of compounds with general formula  $M_4A_2O_9$  ( $M = Mg, Mn, Fe, Co, Ni; A = Nb, Ta$ ). Described for the first time by Bertaut et al. in 1961 [1], these materials sparked interest in recent years because of their magnetic properties; especially their magnetoelectric properties [2–6]. The linear magnetoelectric effect is a coupling between the electric and magnetic responses of a material: an electric polarization of the material can be induced by a magnetic field, and its magnetization can be induced by an electric field. It has been demonstrated that the existence of this effect in a material relies on its magnetic structure, and particularly its magnetic point symmetry [7–11]. Most 429-compounds crystallize in the trigonal space group  $P\bar{3}c1$ , in a structure derived from that of corundum ( $\alpha\text{-Al}_2\text{O}_3$ ) [1, 12]. This often leads to magnetic structures with point symmetries compatible with the magnetoelectric effect, such as  $2/m$  in  $\text{Co}_4\text{Nb}_2\text{O}_9$  [13–16],  $2/m'$  and  $2'/m$  in  $\text{Fe}_4\text{Nb}_2\text{O}_9$  [17], or  $\bar{3}'m'$  in  $\text{Mn}_4\text{Nb}_2\text{O}_9$  [1], to give a few examples.

$\text{Ni}_4\text{Nb}_2\text{O}_9$  is an outcast in the 429 family. Although its crystal structure is very similar to corundum, slight differences in the Ni positions make it crystallize in the orthorhombic  $Pbcn$  space group [18], instead of the usual trigonal  $P\bar{3}c1$  seen in other 429-compounds.  $\text{Ni}_4\text{Nb}_2\text{O}_9$  displays interesting magnetic properties originating from the  $3d^8$  electronic configuration of  $\text{Ni}^{2+}$ ; specifically, it has been studied for its ferrimagnetism [19, 20] and magnetization reversal effects [21–23]. However, the magnetic space group of this material is  $Pb'cn'$  with magnetic point group  $m'm'm'$  (#8.4.27), and thus, it does not allow a magnetoelectric coupling [24].

A few studies have been conducted on exotic magnetic materials based on  $\text{Ni}_4\text{Nb}_2\text{O}_9$ . These materials are usually prepared through the partial substitution of the  $\text{Ni}^{2+}$  cations by another divalent cation. Tarakina et al. reported on compounds with the general formula  $\text{Mg}_{4-x}\text{Ni}_x\text{Nb}_2\text{O}_9$  ( $0 \leq x \leq 4$ ), showing that for  $x > 2.75$ , the  $Pbcn$  structure and the magnetic properties of  $\text{Ni}_4\text{Nb}_2\text{O}_9$  were maintained, although planar defects were introduced in the structure [25]. Later, Bolleta et al. worked on tuning the magnetization reversal effect in  $\text{Ni}_4\text{Nb}_2\text{O}_9$  through partial substitution of nickel with zinc, showing significant changes of magnetic properties in  $\text{Ni}_{4-x}\text{Zn}_x\text{Nb}_2\text{O}_9$  compounds, from  $x = 0$  to  $x = 0.75$  [21]. More

recently, Jiongo-Dongmo et al. studied the  $\text{Co}_4\text{Nb}_2\text{O}_9 - \text{Ni}_4\text{Nb}_2\text{O}_9$  phase diagram ( $\text{Ni}_x\text{Co}_{4-x}\text{Nb}_2\text{O}_9$ ), aiming to combine the linear magnetoelectric effect of the former with the magnetization reversal of the latter. They showed a nice transition from the  $Pbcn$  structure to the  $P\bar{3}c1$  structure for  $x$  between 2.2 and 2.5. However, the linear magnetoelectric effect could not be observed in the  $Pbcn$  structure for this series of compounds [22].

The crystal structure of  $\text{Li}_2\text{Ni}_2\text{W}_2\text{O}_9$  was recently reported [26]. Similarly to  $\text{Ni}_4\text{Nb}_2\text{O}_9$ ,  $\text{Li}_2\text{Ni}_2\text{W}_2\text{O}_9$  crystallizes in a corundum-like  $Pbcn$  structure. However, the  $\text{Nb}_2\text{O}_9$  dimers found in  $\text{Ni}_4\text{Nb}_2\text{O}_9$  are replaced by  $\text{W}_2\text{O}_9$  dimers, allowing half the  $\text{Ni}^{2+}$  cations to be substituted by  $\text{Li}^+$  cations to ensure charge neutrality. Specifically, the  $\text{Li}^+$  cations are arranged in a honeycomb configuration inbetween the (Ni, W) layers (Figure 1). This structural peculiarity might change radically the magnetic interactions between  $\text{Ni}^{2+}$  cations in  $\text{Li}_2\text{Ni}_2\text{W}_2\text{O}_9$  compared to  $\text{Ni}_4\text{Nb}_2\text{O}_9$ .

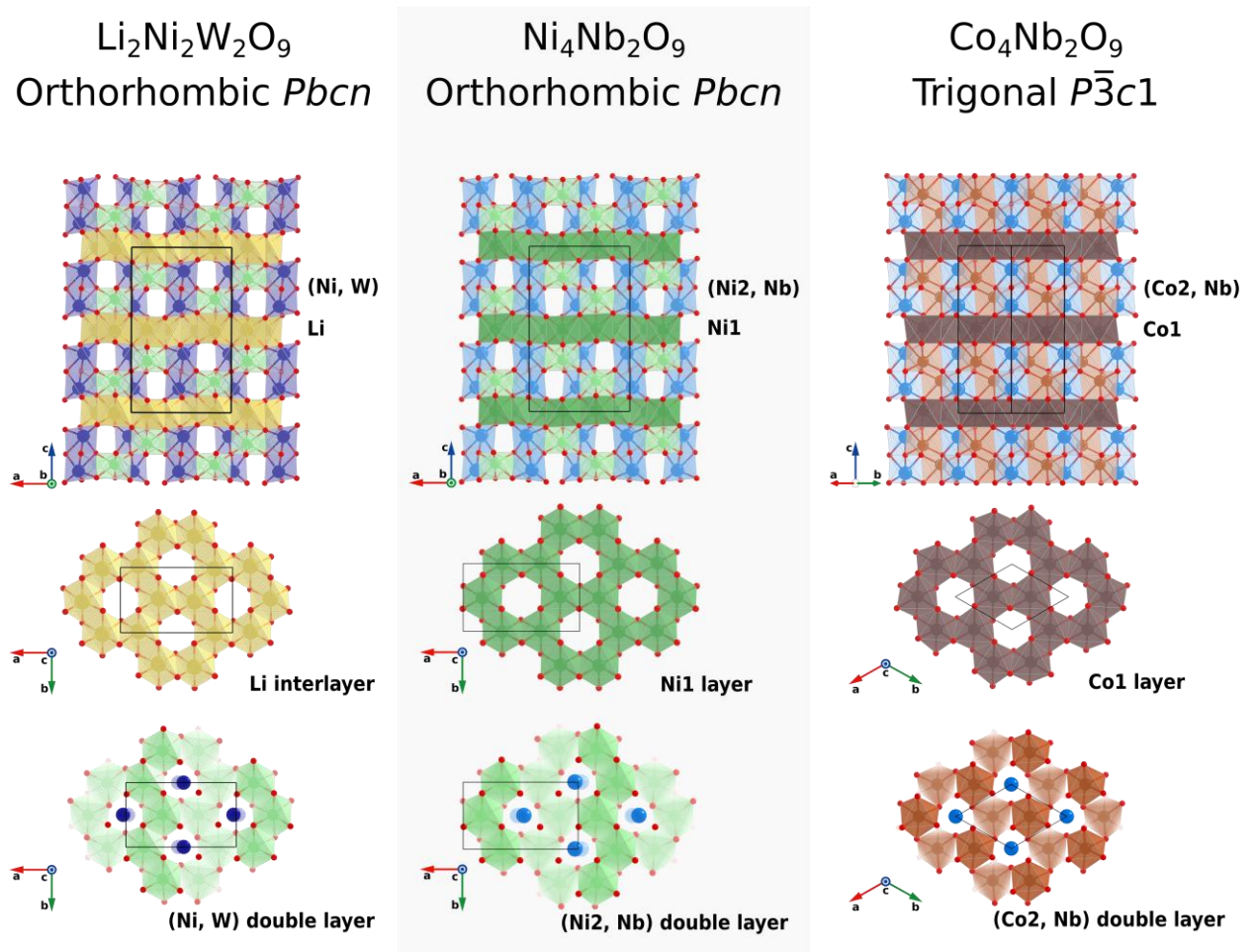


Figure 1 – Crystal structures of  $\text{Li}_2\text{Ni}_2\text{W}_2\text{O}_9$  (left),  $\text{Ni}_4\text{Nb}_2\text{O}_9$  (middle), and  $\text{Co}_4\text{Nb}_2\text{O}_9$  (right). Oxygen atoms are red, tungsten atoms are dark blue, and niobium atoms are sky blue; lithium atoms are yellow, nickel atoms in various shades of green, cobalt atoms in various shades of brown. For the double layer view, atoms in a paler color are at lower z-coordinates.

Herein, we present a preliminary study of the magnetic properties of  $\text{Li}_2\text{Ni}_2\text{W}_2\text{O}_9$ . An antiferromagnetic ordering is observed below  $T_N = 8$  K, while modified Curie-Weiss fits suggest strong ferromagnetic interactions. Neutron powder diffraction allowed us to confidently solve the magnetic structure of the material; the magnetic properties are well explained by a magnetic lattice in the  $Pb'c'n'$  Shubnikov space group (#60.425) with magnetic point group  $m'm'm'$  (#8.5.28), which is neither polar nor ferromagnetic but allows magnetoelectric coupling in the material [8–11].

### Experimental methods

**Reagents** –  $\text{Li}_2\text{Ni}_2\text{W}_2\text{O}_9$  powder was synthesized from lithium carbonate powder ( $\text{Li}_2\text{CO}_3$ , Sigma-Aldrich,  $\geq 99.0\%$ ), nickel oxide powder ( $\text{NiO}$ , Sigma-Aldrich, 99%), and tungsten trioxide powder ( $\text{WO}_3$ , Alfa Aesar, 99.8%).

**Synthesis** –  $\text{Li}_2\text{Ni}_2\text{W}_2\text{O}_9$  powder was obtained by solid-state synthesis. The full process was detailed elsewhere [26]. In summary,  $\text{Li}_2\text{CO}_3$ ,  $\text{NiO}$ , and  $\text{WO}_3$  powders were mixed in stoichiometric amounts, first with an agate mortar, then for 30 minutes in a steel jar with one steel ball using a mechanical SPEX SamplePrep 8000M Mixer/Mill. The resulting mix was then heated in an alumina crucible in air, using a Carbolite CWF 1200 muffle furnace. The mix was first heated at 650 °C for 12 h (heating rate: +5 °C/min), then to 700 °C for 24 h (heating rate: +2 °C/min). Then, the annealed powder was ball milled for 30 minutes and annealed again at 700 °C for 24 h in air (heating rate: +5 °C/min). This milling-annealing step was repeated one more time in order to remove most of the impurities in the powder.

The purity of the final product was evaluated through powder X-ray diffraction measurements (Figure 2), using a Bruker D8 Advance lab diffractometer equipped with a Cu  $K\alpha$  source ( $\lambda_1 = 1.54056$  Å and  $\lambda_2 = 1.54439$  Å), and a LynxEye-2 detector. The measurements were performed in the Bragg-Brentano configuration.

**Magnetic properties** – The magnetic properties of  $\text{Li}_2\text{Ni}_2\text{W}_2\text{O}_9$  were assessed using a Quantum Design MPMS3 Superconducting QUantum Interference Device (SQUID). A few milligrams of  $\text{Li}_2\text{Ni}_2\text{W}_2\text{O}_9$  powder were compressed in a plastic capsule and placed in a brass sample holder for the measurements.

Longitudinal magnetic moment (M) measurements were performed in a DC magnetic field ( $H = 200$  Oe, 1 kOe or 10 kOe), with temperatures ranging from 1.8 K to 300 K; the measurements were performed in field-cooled (FC) and zero-field-cooled (ZFC) modes. Magnetization curves were obtained for  $T = 2$  K and  $T = 100$  K, with  $H$  ranging from -70 to 70 kOe.

**Neutron diffraction** – Neutron powder diffraction measurements were performed using the Echidna diffractometer [27] ( $\lambda_{LT} = 2.4395$  Å or  $\lambda_{RT} = 1.6220$  Å) at the Open Pool Australian Lightwater (OPAL) reactor of Australia’s Nuclear Science and Technology Organisation (ANSTO). The measurements were performed at room temperature ( $\lambda = \lambda_{RT}$ ), then at  $T = 15$  K, 7 K, 5 K, and 3.5 K ( $\lambda = \lambda_{LT}$ ), using roughly 2 g of  $\text{Li}_2\text{Ni}_2\text{W}_2\text{O}_9$  powder placed in a vanadium can.

The powder diffraction patterns were analysed according to the Rietveld method [28] with the FullProf suite [29]. The magnetic form factor of  $\text{Ni}^{2+}$  was used to solve and refine the magnetic structure.

## Results

### 1 – Structural characterization

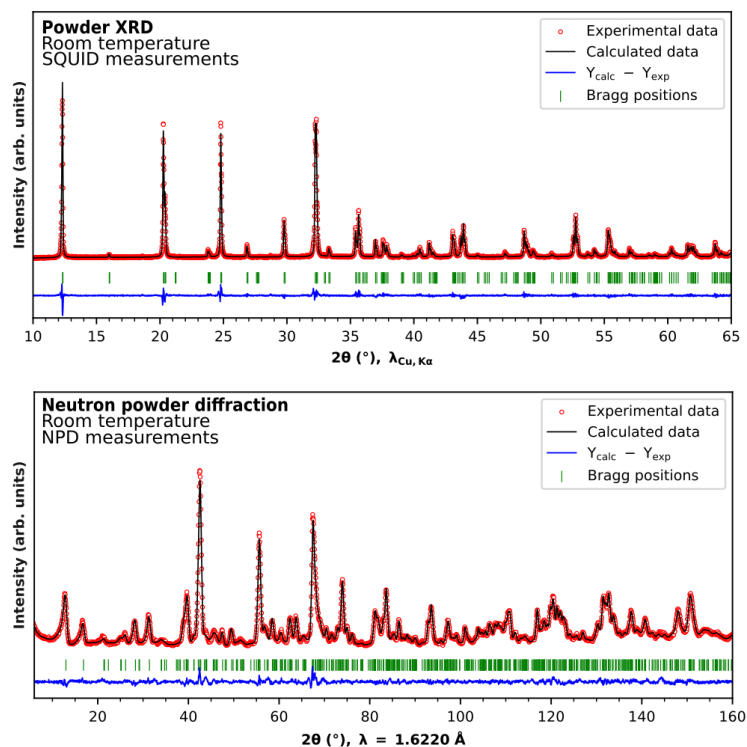


Figure 2 – Top: Rietveld refinement of the powder XRD pattern for a typical  $\text{Li}_2\text{Ni}_2\text{W}_2\text{O}_9$  sample used for the magnetic bulk measurements. Bottom: Rietveld refinement of the neutron powder diffraction pattern at room temperature for the  $\text{Li}_2\text{Ni}_2\text{W}_2\text{O}_9$  sample used for all neutron diffraction measurements.

In this study, different  $\text{Li}_2\text{Ni}_2\text{W}_2\text{O}_9$  samples were used for magnetic bulk measurements and for low-temperature neutron powder diffraction measurements. The Rietveld refinements for their room temperature XRD and neutron powder diffraction data, respectively, are shown in Figure 2.

For both samples, the experimental patterns can be nicely fitted with the crystal structure reported for  $\text{Li}_2\text{Ni}_2\text{W}_2\text{O}_9$ , in the  $Pbcn$  space group and with lattice parameters  $a \approx 8.70 \text{ \AA}$ ,  $b \approx 5.06 \text{ \AA}$ , and  $c \approx 14.35 \text{ \AA}$  (Tables I, S1, S2). Both samples are of high purity, and were deemed of good enough quality for the subsequent measurements.

Sample	a (Å)	b (Å)	c (Å)	R <sub>Bragg</sub> (%)
LNWO SQUID	8.69663(7)	5.06468(4)	14.3514(11)	6.60
LNWO Neutron	8.69523(8)	5.06407(5)	14.3451(18)	3.32

Table I – Lattice parameters deduced from the Rietveld refinements of the room temperature powder XRD and neutron powder diffraction data for  $\text{Li}_2\text{Ni}_2\text{W}_2\text{O}_9$  samples.

## 2 – Magnetic properties

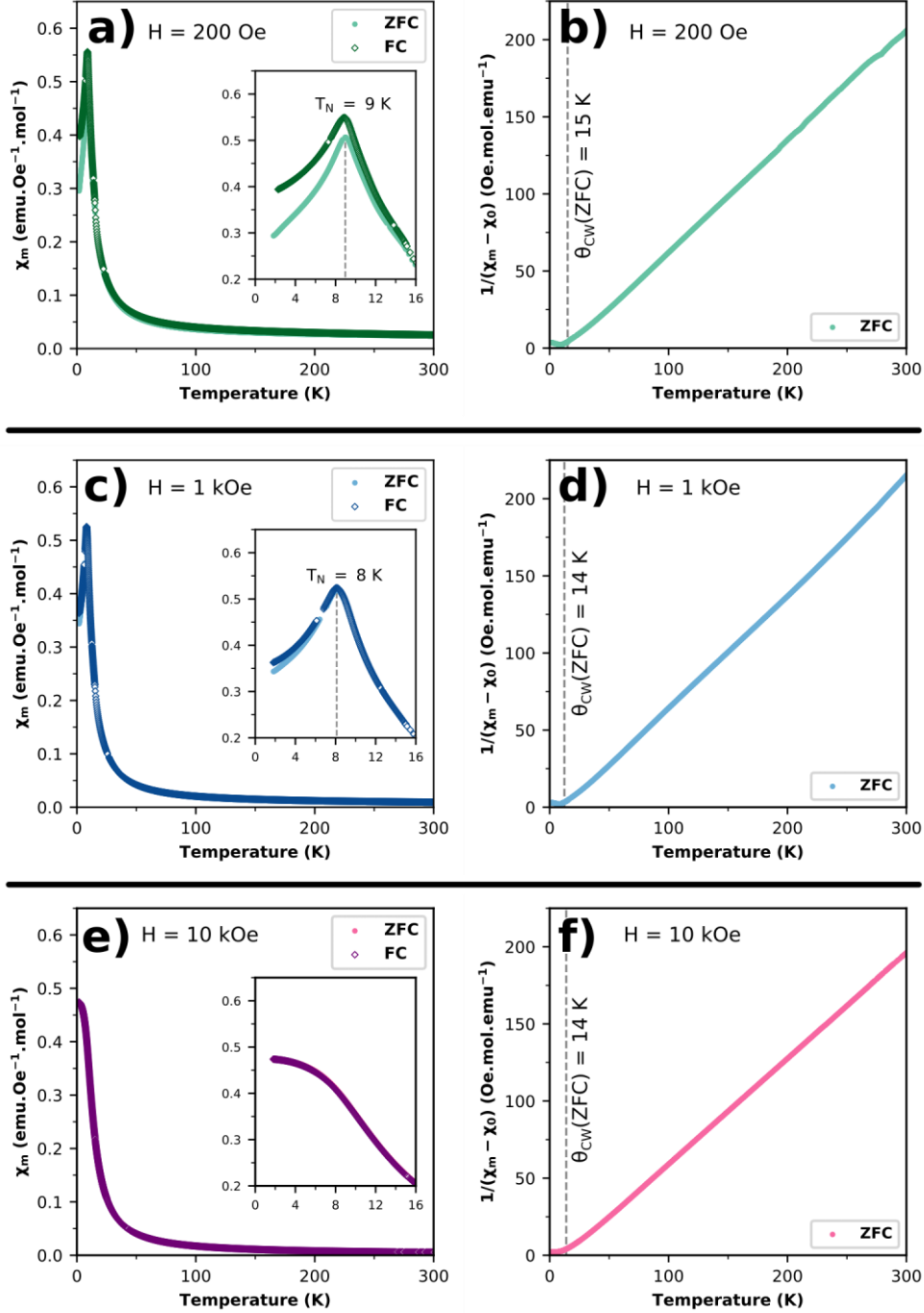


Figure 3 – Molar magnetic susceptibility ( $\chi_m$ ) vs T curves and  $1/(\chi_m - \chi_0)$  vs T curves for (a, b) H = 200 Oe, (c, d) H = 1 kOe, and (e, f) H = 10 kOe.

Molar magnetic susceptibility ( $\chi_m$ ) measurements performed on  $\text{Li}_2\text{Ni}_2\text{W}_2\text{O}_9$  powder, in zero-field cooled (ZFC) and field cooled (FC) modes, are shown in Figure 3. For an applied magnetic field H of 200 Oe or 1 kOe, the material displays a paramagnetic behavior down to a temperature  $T_N$  of roughly 8 to 9 K. At this Néel point, a cusp can be observed, characteristic of an antiferromagnetic transition. However, such a transition is no longer apparent when the applied field is increased to 10 kOe, suggesting that the magnetic

properties of the material are field-dependent. Magnetization measurements were performed to confirm this observation, and are displayed in Figure 4.

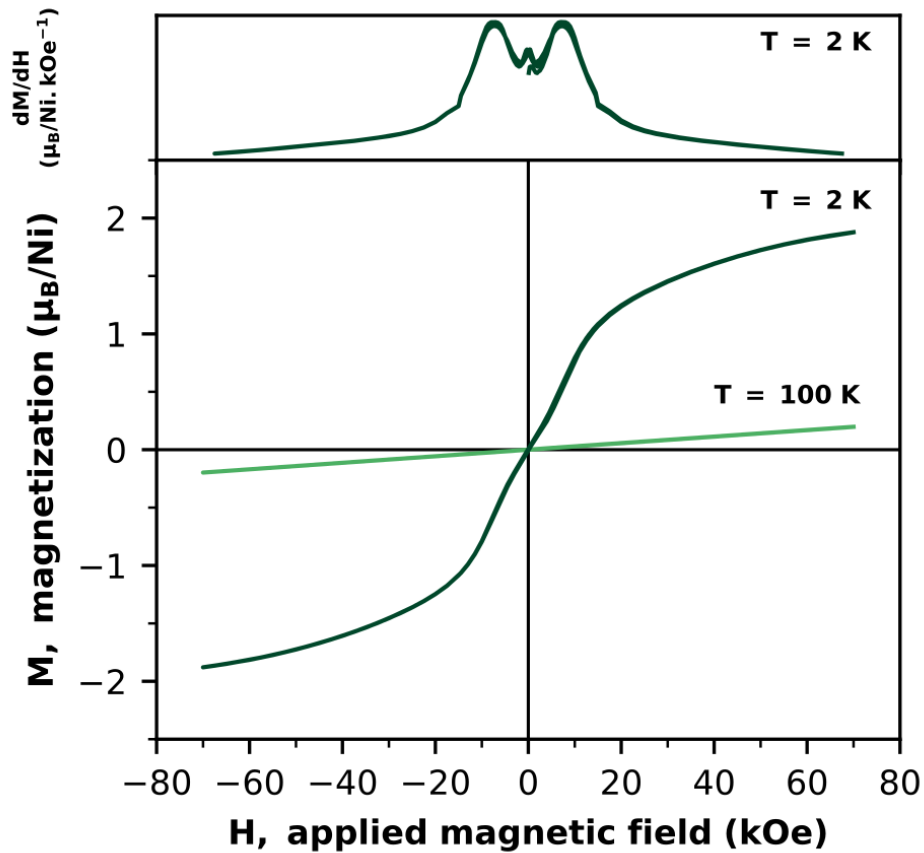


Figure 4 – Magnetization curves for  $\text{Li}_2\text{Ni}_2\text{W}_2\text{O}_9$  powder at  $T = 2$  K (dark green) and  $T = 100$  K (light green). The derivative curve for  $T = 2$  K is displayed on the upper part of the figure.

At 100 K, the magnetization curve is linear, indicating a fully paramagnetic behavior, which does not seem impacted by the applied magnetic field. In contrast, at 2 K, below  $T_N$ , inflexion points can be observed in the magnetization curve for  $|H| \approx 8$  kOe, indicating a change in magnetic behavior. This behavior could be associated to a metamagnetic transition, [30], but this needs to be confirmed through neutron diffraction measurements in an applied magnetic field, which were not conducted in this study.



		$\theta_{CW}$ (K)	C ( $10^{-5} \text{ m}^3 \cdot \text{K} \cdot \text{mol}^{-1}$ )	$\mu_{eff}$ ( $\mu_B$ )	Fitting range (K)
H = 200 Oe	ZFC	15.09(2)	1.7228(9)	3.32(8)	[50; 300]
	FC	8.8(1)	2.208(6)	3.8(2)	[50; 200]
H = 1 kOe	ZFC	13.68(3)	1.680(1)	3.28(8)	[50; 250]
	FC	12.85(4)	1.731(1)	3.33(8)	[50; 250]
H = 10 kOe	ZFC	13.99(3)	1.8303(4)	3.42(5)	[50; 300]
	FC	13.98(1)	1.8196(4)	3.41(5)	[50; 300]

Table II – Results of the modified Curie-Weiss fits for the  $1/\chi_m$  vs T curves.

The curves for  $1/\chi_m$  vs T were fitted with a modified Curie-Weiss law (Equation 1):

$$\chi_m = \chi_0 + \frac{C}{T - \theta_{CW}} \quad (1)$$

where  $\chi_0$  is a temperature-independent contribution to the magnetic susceptibility in  $\text{mol}^{-1}$ , C is the Curie constant in  $\text{m}^3 \cdot \text{K} \cdot \text{mol}^{-1}$  and  $\theta_{CW}$  is the Curie-Weiss temperature in K. Results of the fits are shown in Table II. The fitting range had to be adapted from one measurement to the next to account for deviations from the model due to magnetic impurities. Interestingly, even though the shape of the susceptibility curves clearly proves that  $\text{Li}_2\text{Ni}_2\text{W}_2\text{O}_9$  goes through an antiferromagnetic ordering at low temperatures (for  $H < 8$  kOe), the values of  $\theta_{CW}$  are all positive, which indicates that there are strong ferromagnetic interactions in the material.

One can calculate the effective moment per magnetic ion,  $\mu_{eff}$ , from the values of C, using equation 2 [31]:

$$\mu_{eff} = \sqrt{\frac{3k_b}{N_A \mu_0 \mu_B^2}} \sqrt{C} \approx 800 \sqrt{C} \quad (2)$$

where  $k_B$  is the Boltzmann constant ( $1.38 \times 10^{-23} \text{ J} \cdot \text{K}^{-1}$ ),  $N_A$  is the Avogadro constant ( $6.022 \times 10^{23} \text{ mol}^{-1}$ ),  $\mu_0$  is the magnetic constant ( $4\pi \times 10^{-7} \text{ H} \cdot \text{m}^{-1}$ ), and  $\mu_B$  is the Bohr magneton ( $9.274 \times 10^{-24} \text{ J} \cdot \text{T}^{-1}$ ). Aside from the aberrant  $3.8 \mu_B$  value calculated for the FC curve at  $H = 200$  Oe (which is probably related to impurities in the sample), the calculated values of  $\mu_{eff}$  for the  $\text{Ni}^{2+}$  cations in  $\text{Li}_2\text{Ni}_2\text{W}_2\text{O}_9$  range from 3.3 to  $3.4 \mu_B$ . These experimental values land between the theoretical values for  $\text{Ni}^{2+}$  in an octahedral coordination, with and without orbital angular momentum ( $4.47$  and  $2.83 \mu_B$ , respectively); thus, there must be some weak contribution from the orbital angular momentum to the magnetic moment. This matches quite well with empirical values previously reported in the literature [24, 32].

The frustration index for  $\text{Li}_2\text{Ni}_2\text{W}_2\text{O}_9$ ,  $f$ , can also be calculated from  $\theta_{CW}$  and  $T_N$ , using equation 3:

$$f = \left| \frac{\theta_{CW}}{T_N} \right| \quad (3)$$

Estimating that  $T_N$  is roughly equal to 8 K leads to values of  $f$  ranging from 1.1 to 1.9: the magnetic structure is not heavily frustrated.

### 3 – Magnetic structure

. In order to solve the magnetic structure of  $\text{Li}_2\text{Ni}_2\text{W}_2\text{O}_9$ , and obtain a better understanding of the results obtained from the bulk magnetic measurements, low temperature powder neutron diffraction experiments were conducted. Neutron diffraction scans were performed between 15 K and 3.5 K, above and below the Neel point (Figure 5).

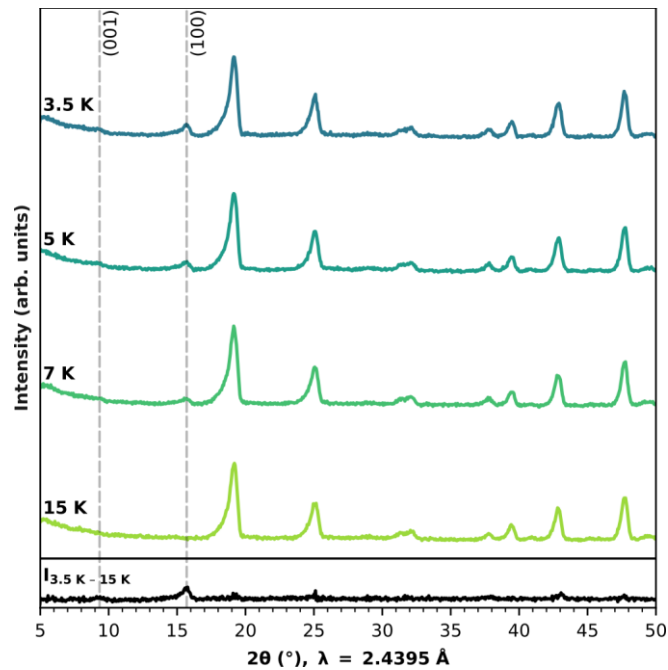


Figure 5 – Powder neutron diffraction scans for  $\text{Li}_2\text{Ni}_2\text{W}_2\text{O}_9$  powder, from  $T = 15$  K (light green) to  $T = 3.5$  K (blue). The difference between the diffractograms for  $T = 3.5$  K and  $T = 15$  K is plotted in black at the bottom of the figure.

Starting at 7 K, a peak appears at  $2\theta = 16.06^\circ$ ; this peak is not present in the diffractogram for  $T = 15$  K, and it keeps increasing as the temperature gets lower. This confirms that there is indeed a magnetic ordering below 8 K. Similarly, another peak seems to increase in intensity as the temperature is decreased, at  $2\theta = 9.55^\circ$ . Aside from these changes, there is no significant shift in positions or intensities for the nuclear Bragg peaks

upon the sample's cooling down, indicating that the nuclear crystal structure remains the same below the magnetic transition temperature.

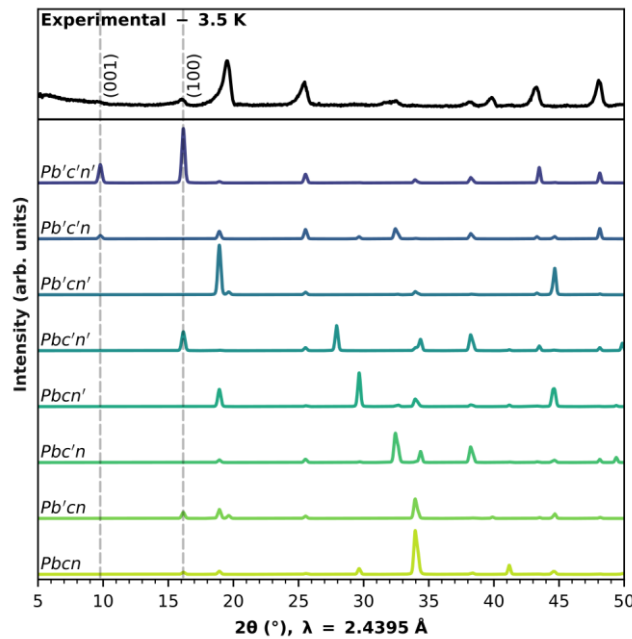


Figure 6 – Experimental (black curve, top) and simulated (colored curves) diffraction patterns of  $\text{Li}_2\text{Ni}_2\text{W}_2\text{O}_9$ . The simulated curves were plotted with  $m_b = 0.5 \mu_B$  and  $m_c = 1.5 \mu_B$ . Theoretical data for the 8 magnetic space groups were taken from the MAXMAGN software [7] of the Bilbao Crystallographic Server [8–10].

The nuclear structure of  $\text{Li}_2\text{Ni}_2\text{W}_2\text{O}_9$  was solved in a previous work [26] and was used as a basis to solve the magnetic structure.  $\text{Li}_2\text{Ni}_2\text{W}_2\text{O}_9$  crystallizes in the orthorhombic  $Pbcn$  space group. The positions of the noticeable magnetic reflections at  $2\theta = 9.55^\circ$  and  $2\theta = 16.06^\circ$  in the neutron diffraction patterns match with the (001) and (100) Bragg positions for the parent space group, in the same unit cell as the nuclear structure. Therefore, the propagation vector for the magnetic structure is  $\mathbf{k} = (0, 0, 0)$ , as the magnetic and nuclear unit cells are identical.

Based on the Miller indexes for the magnetic peaks, (001) and (100), it can be ascertained that components of the nickel cations' magnetic moments are perpendicular to the reciprocal space basis vectors  $\mathbf{c}^*$  and  $\mathbf{a}^*$ ; for an orthorhombic space group,  $\mathbf{c}^*$  and  $\mathbf{a}^*$  are collinear to the  $c$ -axis and  $a$ -axis, respectively. Simulations, based on the nuclear structure of  $\text{Li}_2\text{Ni}_2\text{W}_2\text{O}_9$ , were performed using the FullProf software [29] in the eight Shubnikov magnetic space groups ( $Pbcn$ ,  $Pb'cn$ ,  $Pbc'n$ ,  $Pbcn'$ ,  $Pbc'n'$ ,  $Pb'cn'$ ,  $Pb'c'n$ , and  $Pb'c'n'$ ), which are compatible with  $\mathbf{k} = (0, 0, 0)$  and a magnetic atom in the  $8d$  (x,y,z) site. Only two magnetic space groups lead to the appearance of the (001) and (100) diffraction peaks

(Figures S1 to S8):  $Pbcn$ , with two magnetic components along the  $a$ -axis and  $b$ -axis (Figure S1), and  $Pb'c'n'$ , with two magnetic components along the  $b$ -axis and  $c$ -axis (Figures 6, S2). However, the best Rietveld [28] refinement was obtained when using the latter as the model for the calculations (Figures 7, S9). The symmetry of  $Pb'c'n'$  allows all three magnetic moment components for the Ni site, but the data showed no evidence of  $M_x$  and it was fixed to zero in the final refinement. The results of this refinement can be found in Table III, and the resulting magnetic structure in Figure 8.

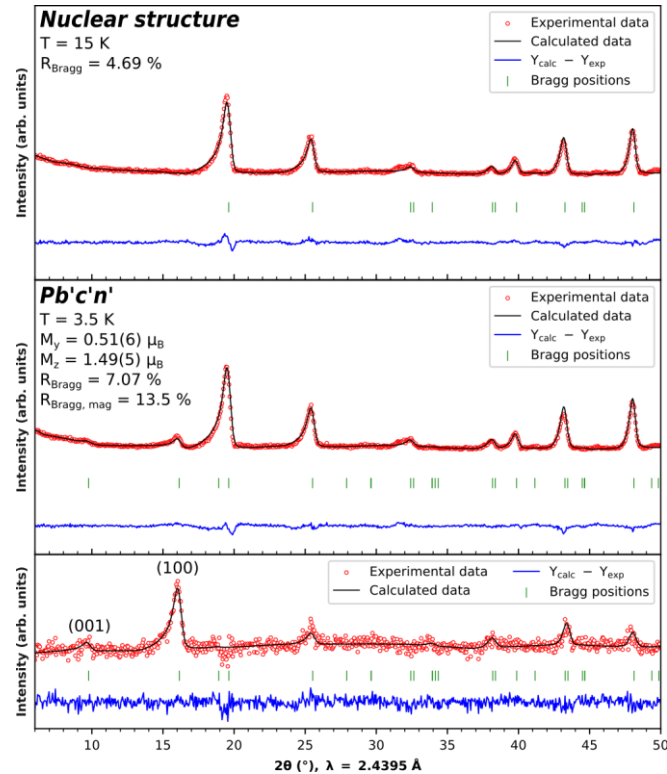


Figure 7 – Rietveld refinement for the neutron powder diffraction data of  $\text{Li}_2\text{Ni}_2\text{W}_2\text{O}_9$  at  $T = 15$  K in the  $Pbcn$  nuclear space group (top) and at  $T = 3.5$  K in the  $Pb'c'n'$  Shubnikov space group (#60.425) (middle). Bottom: Rietveld refinement against the difference curve between the neutron powder diffraction data of  $\text{Li}_2\text{Ni}_2\text{W}_2\text{O}_9$  at  $T = 3.5$  K and  $T = 15$  K.

Cation	Coordinates	$x/a$	$y/b$	$z/c$	$M_y$ ( $\mu_B$ )	$M_z$ ( $\mu_B$ )
Ni(1)	$x, y, z$	0.6648(5)	0.512(11)	0.8136(18)	+0.51(6)	+1.49(5)
Ni(2)	$\bar{x}+1/2, \bar{y}+1/2, z+1/2$	0.8352(5)	0.988(11)	0.3136(18)	-0.51(6)	+1.49(5)
Ni(3)	$\bar{x}, y, \bar{z}+1/2$	0.3352(5)	0.512(11)	0.6864(18)	+0.51(6)	-1.49(5)
Ni(4)	$x+1/2, \bar{y}+1/2, \bar{z}$	0.1648(5)	0.988(11)	0.1864(18)	-0.51(6)	-1.49(5)
Ni(5)	$\bar{x}, \bar{y}, \bar{z}$	0.3352(5)	0.488(11)	0.1864(18)	-0.51(6)	-1.49(5)
Ni(6)	$x+1/2, y+1/2, \bar{z}+1/2$	0.1648(5)	0.012(11)	0.6864(18)	+0.51(6)	-1.49(5)
Ni(7)	$x, \bar{y}, z+1/2$	0.6648(5)	0.488(11)	0.3136(18)	-0.51(6)	+1.49(5)
Ni(8)	$\bar{x}+1/2, y+1/2, z$	0.8352(5)	0.012(11)	0.8136(18)	+0.51(6)	+1.49(5)

All nickel cations have a small component along the  $b$ -axis ( $|M_y| = 0.51(6) \mu_B$ ) and a larger component along the  $c$ -axis ( $|M_z| = 1.49(5) \mu_B$ ), for a total magnetic moment of  $1.58(5) \mu_B$  per nickel cation. This is in reasonable agreement with the expected  $2 \mu_B$  for  $Ni^{2+}$ , with  $S = 1$  in a  $d^8$  electronic configuration. The spin-sequence along the  $b$ -axis is (+ - + - - + - +), and it is (+ + - - - + +) along the  $c$ -axis. This leads to a canted, antiferromagnetic structure, in accordance with the antiferromagnetic transition observed in the magnetization measurements. Moreover, the nickel cations can be paired in such a way that the magnetic components of each cation in a pair are reversed, and their coordinates systems are opposite to one another. Such pairs are Ni(1)/Ni(5), Ni(2)/Ni(6), Ni(3)/Ni(7), and Ni(4)/Ni(8). This indicates that the magnetic structure contains a time reversal center, as well as an inversion center; the combination of these two symmetry elements is required in linear magnetoelectric materials.

Table III – Magnetic structure of  $Li_2Ni_2W_2O_9$  as deduced from the Rietveld refinement of the neutron diffraction pattern measured at 3.5 K.

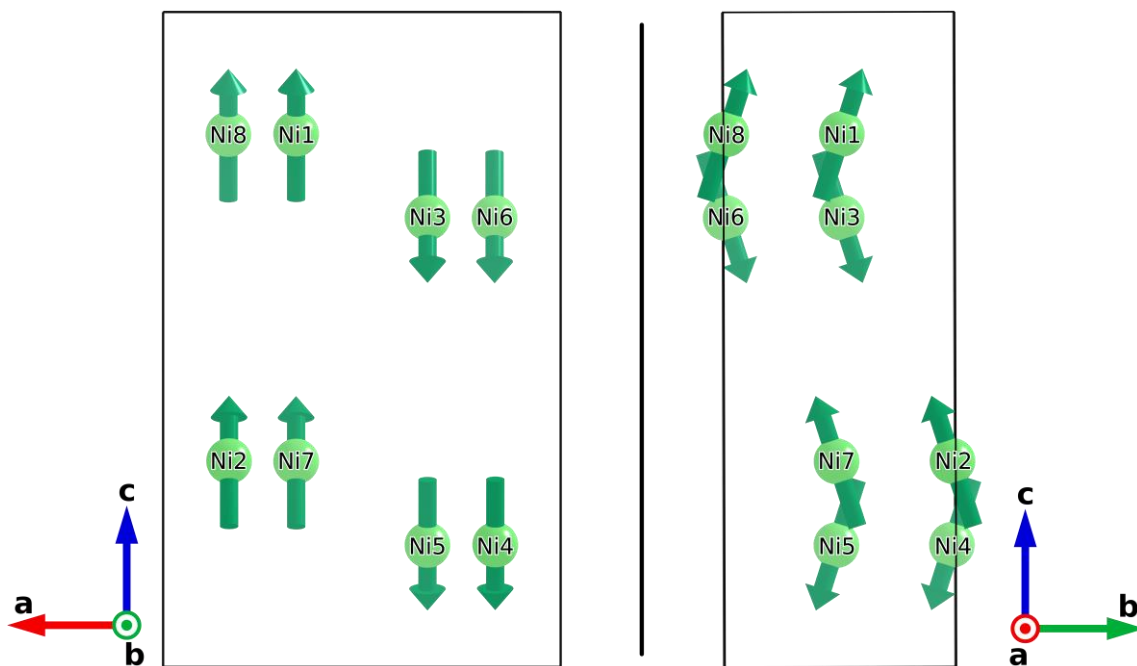


Figure 8 – Magnetic structure of  $\text{Li}_2\text{Ni}_2\text{W}_2\text{O}_9$ . Magnetic moments are represented by green arrows. For the sake of clarity, only nickel atoms are shown in the figure.

To get into more details, the magnetic structure of  $\text{Li}_2\text{Ni}_2\text{W}_2\text{O}_9$  can be broken down into three different scales: chain, layer, and stack of layers. Inside a Ni-Ni chain, a parallel alignment of the magnetic moments is observed. Meanwhile, inside a Ni-W layer, the magnetic moments of two neighboring chains display weak parallel alignment along the  $b$ -axis, and a comparatively stronger antiparallel alignment along the  $c$ -axis. Finally, these layers are piled into stacks along the  $c$ -axis; the resulting magnetic moments of two neighboring layers in a stack are antiparallel to one another along the  $b$ -axis (Figure 9).

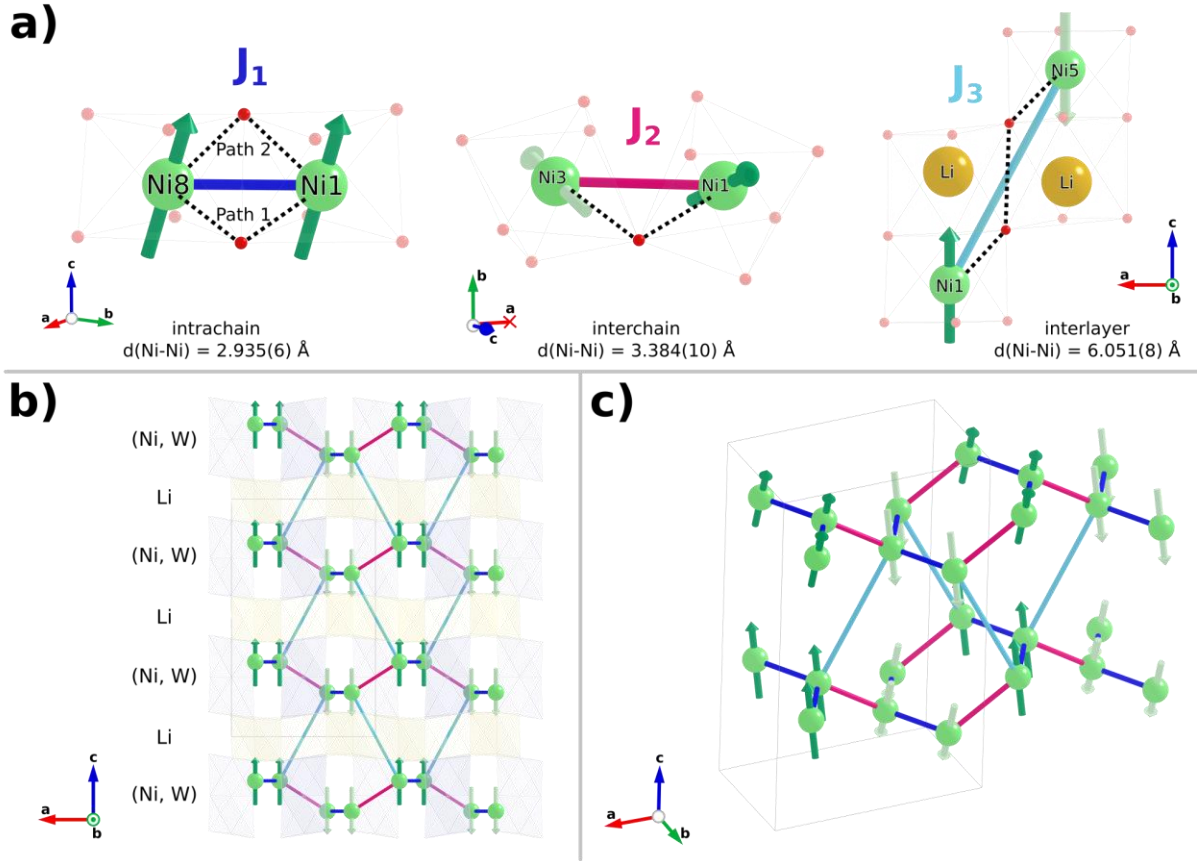


Figure 9 – Magnetic interaction paths in  $\text{Li}_2\text{Ni}_2\text{W}_2\text{O}_9$ . In a), super-exchange and super-super-exchange paths are shown as black dotted lines. Nickel atoms are shown in green, lithium atoms in yellow and oxygen atoms in red. In b),  $\text{WO}_6$  and  $\text{LiO}_6$  octahedra are respectively shown in pale blue and in pale yellow. Magnetic moments with a positive  $M_z$  component are shown in dark green, while those with a negative  $M_z$  component are shown in light green.

## Discussion

$\text{Li}_2\text{Ni}_2\text{W}_2\text{O}_9$ 's crystal structure leads to both super-exchange (Ni-O-Ni) and super-super-exchange (Ni-O-O-Ni) interactions in the material. As the magnetic structure is long-range ordered, three distinct isotropic exchange interactions need to be considered: an intrachain double interaction  $J_1$  (2.93 Å), an interchain interaction  $J_2$  (3.40 Å), and an interlayer interaction  $J_3$  (6.05 Å) (Figure 9).

Path	Distances (Å)	Angles (deg.)
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	d(Ni-Ni)	d(Ni-O)	d(O-O)	d(O-Ni)	$\overline{\text{Ni-O-Ni}}$	$\overline{\text{Ni-O-O}}$	$\overline{\text{O-O-Ni}}$	Ni-O-O-Ni
J1 - Path 1 - super	2.935(6)	1.982(7)	-	2.047(6)	93.51(17)	-	-	-
J1 - Path 2 - super	2.935(6)	2.104(5)	-	2.118(7)	88.07(16)	-	-	-
J2 - super	3.384(10)	2.033(6)	-	2.033(6)	112.7(4)	-	-	-
J3 - super-super	6.051(8)	2.108(5)	3.19738(5)	2.108(5)	-	108.64(16)	108.64(16)	180.00

Between two closest neighbors in a NiO<sub>6</sub> chain (J<sub>1</sub>), two super-exchange paths exist. The super-exchange paths are characterized by a ~ 90° Ni-O-Ni angle (Table IV), leading to ferromagnetic interactions in accordance to the Goodenough-Kanamori-Anderson rules [33–35]. Similarly, for interchain interaction J<sub>2</sub>, the orientation of the magnetic moments of the nickel cations (**m<sub>b</sub>** components parallel and **m<sub>c</sub>** components antiparallel) suggest some competition between ferromagnetic and antiferromagnetic interactions. The former could be coming from J<sub>2</sub>'s super-exchange path, as the 112.01° Ni-O-Ni angle is closer to 90° than 180°, probably below the ferromagnetic-to-antiferromagnetic critical angle described by Goodenough [33]; but again, this is mostly speculative, and an analysis based solely on geometry might miss some other important factors influencing the magnetic structure. J<sub>3</sub>'s super-super-exchange path leads to antiferromagnetic interactions between the (Ni, W) layers without too much ambiguity; this can be explained quite well, again, by the Goodenough-Kanamori-Anderson rules, and by the 180° Ni-O-O-Ni dihedral angle.

Table IV – Geometric characteristics of the super-exchange (Ni-O-Ni) and super-super-exchange (Ni-O-O-Ni) paths for exchange interactions J <sub>1</sub> , J <sub>2</sub> , and J <sub>3</sub> . Ni-O-O-Ni is the dihedral angle in degrees for the super-super-exchange path.
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In any case, the nature of these magnetic interactions explain the unusual behavior observed in the magnetization measurements: although the structure is antiferromagnetic overall, the Curie-Weiss temperature for Li<sub>2</sub>Ni<sub>2</sub>W<sub>2</sub>O<sub>9</sub> is positive because of the intrachain and interchain ferromagnetic interactions. Moreover, looking at the topology of J<sub>1</sub>, J<sub>2</sub>, and J<sub>3</sub> (Figure 9b, 9c), it is interesting to observe that the magnetic connectivity of the nickel cations leads to hexagonal magnetic loops; as there are no odd-sided loops, the magnetic structure is not geometrically frustrated. This observation matches well with the low values of the frustration index calculated earlier.

These considerations not only fit with the *Pb'c'n'* space group, they also make it rather unlikely that Li<sub>2</sub>Ni<sub>2</sub>W<sub>2</sub>O<sub>9</sub>'s magnetic space group is *Pbcn*. Looking at the theoretical magnetic structure associated with the *Pbcn* Shubnikov space group (Figure S10), one can notice that the intrachain magnetic interactions would be mostly antiferromagnetic. Not only



would this go against the Goodenough-Kanamori-Anderson rules, it would also go against the experimental results, namely the high ferromagnetic interactions observed in the magnetic measurements. Considering said measurements, and taking into account that  $J_1$ , as the shortest magnetic interaction path is also likely to be the strongest, it can be confidently ascertained that the intrachain interactions in  $\text{Li}_2\text{Ni}_2\text{W}_2\text{O}_9$  have to be ferromagnetic. Thus, the  $Pbcn$  magnetic space group is ruled out, and only  $Pb'c'n'$  remains.

In Figure 10 is displayed a phase diagram showing the possible ground-state magnetic structures of  $\text{Li}_2\text{Ni}_2\text{W}_2\text{O}_9$ , depending on the values of the propagation vector  $\mathbf{k}$  and of the exchange integrals  $J_1$ ,  $J_2$ , and  $J_3$ .

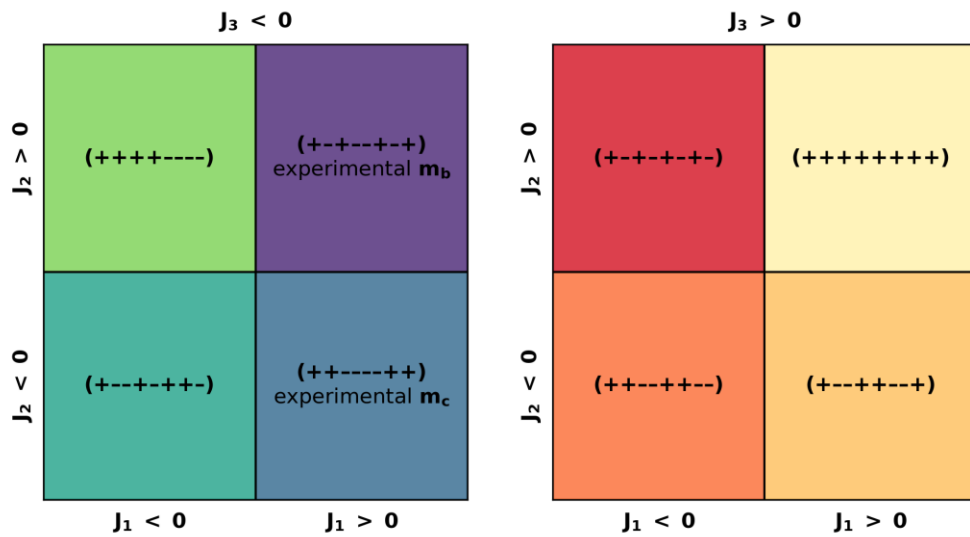


Figure 10 – Phase diagram showing the possible magnetic ground states of  $\text{Li}_2\text{Ni}_2\text{W}_2\text{O}_9$  below  $T_N$  as a function of the values of the exchange integrals  $J_1$ ,  $J_2$  and  $J_3$ .  $\mathbf{k} = (0, 0, 0)$  in all ground-states. The ground-state magnetic structures observed experimentally from the neutron diffraction data shown in Figure 3 are indicated in the purple and blue parts of the diagram.

This phase diagram was calculated by using the programs SIMBO and ENERMAG; El Khayati et al. gave a full description of these programs and the theory behind them in other works [36–40]. The magnetic ground state (or first-ordered state) is obtained, as a function of  $\mathbf{k}$  and the exchange integrals  $J_{ij}$ , as the eigenvector corresponding to the lowest eigenvalue of the negative Fourier transform of the exchange integral matrix:

$$\xi_{ij}(\mathbf{k}) = - \sum_m J_{ij}(\mathbf{R}_m) e^{-2\pi i \mathbf{k} \cdot \mathbf{R}_m} \quad (4)$$

where  $J_{ij}(\mathbf{R}_m)$  is the isotropic exchange interaction between the spins of atoms  $i$  and  $j$  in unit cells separated by the lattice vector  $\mathbf{R}_m$ . As a convention, a negative value of the exchange

interaction  $J_{ij}$  means that there is an antiparallel coupling between the spins  $\mathbf{S}_i$  and  $\mathbf{S}_j$ , leading to the expression of the pair interaction energy shown in equation (5):

$$E_{ij} = -J_{ij}\mathbf{S}_i \cdot \mathbf{S}_j \quad (5)$$

To build the phase diagram, we used a procedure which was applied to several magnetic structures in the past [32, 41–46]. In a few words, the exchange integral matrixes  $\xi(\mathbf{k}, J_1, J_2, J_3)$  were calculated for values of  $J_1, J_2, J_3$  ranging from -100 to 100, and for each set of values ( $J_1, J_2, J_3$ ), the magnetic structure of lowest energy was selected. For all regions of the phase diagram, the lowest energy was obtained for  $\mathbf{k} = (0, 0, 0)$ ; thus, only the spin sequence vary from one magnetic structure to the other.

The magnetic ground state of  $\text{Li}_2\text{Ni}_2\text{W}_2\text{O}_9$  is split between two parts of the phase diagram. Although it is clear that  $J_1 > 0$  and  $J_3 < 0$ , the value of  $J_2$  is not the same depending on the component of the magnetic moments which is considered: indeed,  $J_2 > 0$  for  $\mathbf{m}_b$ , and  $J_2 < 0$  for  $\mathbf{m}_c$ . This, again, highlights the dual nature of the  $J_2$  interaction, both ferromagnetic and antiferromagnetic.

Another point to comment on relates to the magnetic space group of  $\text{Li}_2\text{Ni}_2\text{W}_2\text{O}_9$ , its nuclear structure, and potential magnetoelectric properties. It has been shown that the existence of this property in a material is dependent on the point symmetry of its magnetic structure [7–11]. In particular, the  $m'm'm'$  point group allows for the magnetoelectric effect to manifest, as the diagonal components of the associated magnetoelectric tensor are non-zero [9, 10]. This means that this effect could be present in  $\text{Li}_2\text{Ni}_2\text{W}_2\text{O}_9$  because of its  $Pb'c'n'$  magnetic space group. This is interesting because, as said earlier, the corundum-like structure of  $\text{Li}_2\text{Ni}_2\text{W}_2\text{O}_9$  is very similar to that of "429" magnetic compounds [1], especially  $\text{Ni}_4\text{Nb}_2\text{O}_9$ , which crystallizes in the same orthorhombic  $Pbcn$  space group [18]. However, ferrimagnetic ordering in  $\text{Ni}_4\text{Nb}_2\text{O}_9$  occurs in the  $Pb'cn'$  magnetic space group [19, 24], corresponding to the magnetic point group  $m'm'm$  (#8.4.27) which is not compatible with the linear magnetoelectric effect. Meanwhile, 429 compounds crystallizing in a trigonal space group  $\bar{P}3c1$  ( $\text{Co}_4\text{Nb}_2\text{O}_9$ ,  $\text{Co}_4\text{Ta}_2\text{O}_9$ ,  $\text{Mn}_4\text{Nb}_2\text{O}_9$ ,  $\text{Mn}_4\text{Ta}_2\text{O}_9$ , etc.) usually display linear magnetoelectric coupling [2, 13, 47, 48]. Thus,  $\text{Li}_2\text{Ni}_2\text{W}_2\text{O}_9$  might be the gateway to the exploration of a magnetoelectric effect based on  $\text{Ni}^{2+}$  in a corundum-like orthorhombic structure. Moreover, as far as we can tell, substitution of the  $\text{Ni}^{2+}$  cations with other magnetic 3d transition metal ( $\text{Co}^{2+}$ ,  $\text{Fe}^{2+}$ , etc.) has not yet been achieved in  $\text{Li}_2\text{Ni}_2\text{W}_2\text{O}_9$ ; such an endeavor could potentially lead to other interesting magnetic compounds.

Compared to members of the 429 family, the magnetic transition temperature for  $\text{Li}_2\text{Ni}_2\text{W}_2\text{O}_9$  ( $T_N = 8$  K) is much lower.  $\text{Fe}_4\text{Nb}_2\text{O}_9$  displays two temperatures of transition,  $T_{N1} \approx 90$  K and  $T_{N2} \approx 80$  K [5, 17]. In  $\text{Co}_4\text{Nb}_2\text{O}_9$ , the Neel point has been reported at  $T_N = 27$  K [16, 49]; meanwhile, in  $\text{Co}_4\text{Ta}_2\text{O}_9$ , various studies find  $T_N \approx 20$  K [47, 50, 51]. Even  $\text{Ni}_4\text{Nb}_2\text{O}_9$ , a material with  $\text{Ni}^{2+}$  as the paramagnetic cation, displays a Curie temperature  $T_C$  equal to 76 K [19, 20, 24]. These major differences suggest that the magnetic interactions are weaker in  $\text{Li}_2\text{Ni}_2\text{W}_2\text{O}_9$  than in 429 compounds. This could be due to a number of factors. First of all, the lower amount of magnetic cations per formula unit could introduce some sort of "diamagnetic dilution" effect. Such a phenomenon was reported by Tarakina et al. in  $\text{MgNi}_3\text{Nb}_2\text{O}_9$  [25]. Another factor, related to the one described above, is the difference in crystal structure between  $\text{Li}_2\text{Ni}_2\text{W}_2\text{O}_9$  and the  $\text{M}_4\text{A}_2\text{O}_9$  compounds. Whereas  $\text{Ni}_4\text{Nb}_2\text{O}_9$ , for instance, displays two types of magnetic layers (the (Ni2, Nb) double layer and the Ni1 single layer), paramagnetic  $\text{Ni}^{2+}$  cations in  $\text{Li}_2\text{Ni}_2\text{W}_2\text{O}_9$  are only present in (Ni, W) double layers, separated by **diamagnetic**  $\text{Li}^+$  interlayers (Figure 1). As such, the distance along the  $c$ -axis between two **paramagnetic**  $\text{Ni}^{2+}$  cations is much longer in  $\text{Li}_2\text{Ni}_2\text{W}_2\text{O}_9$  ( $d_{J3}(\text{Ni-Ni}) = 6.051(8)$  Å) than in  $\text{Ni}_4\text{Nb}_2\text{O}_9$  ( $d(\text{Ni1-Ni2}) = 2.659(3)$  Å) [24]. This means that the magnetic interactions are likely much weaker between two successive paramagnetic layers in  $\text{Li}_2\text{Ni}_2\text{W}_2\text{O}_9$  than in other 429 compounds. Finally, one may wonder whether or not replacing the  $\text{Nb}^{5+}/\text{Ta}^{5+}$  cations by  $\text{W}^{6+}$  cations has an impact on the magnetic properties, even though  $\text{NbO}_6$ ,  $\text{TaO}_6$ , and  $\text{WO}_6$  are all distorted by the same second-order Jahn-Teller effect [52, 53].

It is also interesting to observe that, although it displays the same antiferromagnetic ordering as many of the 429 compounds [1, 5, 6, 50],  $\text{Li}_2\text{Ni}_2\text{W}_2\text{O}_9$  distinguishes itself by displaying a positive Curie-Weiss temperature  $\theta_{\text{CW}} \approx 14$  K, whereas all 429 compounds show  $\theta_{\text{CW}} < 0$ . Even  $\text{Ni}_4\text{Nb}_2\text{O}_9$  displays a strongly negative Curie-Weiss temperature ( $\theta_{\text{CW}} = -233$  K), even though it is distinct from other 429 compounds for having a parallel alignment of the  $\text{Ni}^{2+}$  magnetic moments in its Ni1 single layer, and of those in its (Ni2, Nb) double layer [19, 24]. This again shows how the introduction of the  $\text{Li}^+$  interlayer changes the magnetic interactions in  $\text{Li}_2\text{Ni}_2\text{W}_2\text{O}_9$ . Compared to  $\text{Ni}_4\text{Nb}_2\text{O}_9$ , for which the Ni1-Ni2 antiferromagnetic interactions are the strongest in the material, the intrachain ferromagnetic Ni-Ni interactions dominate in  $\text{Li}_2\text{Ni}_2\text{W}_2\text{O}_9$ . Meanwhile, the  $P\bar{3}c1$  structure of other 429 compounds usually leads to strong antiferromagnetic intralayer interactions and weak ferromagnetic interlayer interactions [15, 17, 50, 54]. As such, the crystal structure of  $\text{Li}_2\text{Ni}_2\text{W}_2\text{O}_9$  introduces favored

ferromagnetic interactions in the general antiferromagnetic order usually found in 429 compounds.

## Conclusion

Herein, we gave a first overview of the magnetic properties of layered  $\text{Li}_2\text{Ni}_2\text{W}_2\text{O}_9$ . The material becomes antiferromagnetically ordered below  $T_N = 8$  K, but only for  $|\mathbf{H}|$  below 8 kOe. Moreover, although magnetic measurements clearly indicate an antiferromagnetic ordering, subsequent modified Curie-Weiss fits point to the presence of strong ferromagnetic interactions, with low frustration. This apparent contradiction was explained by solving the magnetic structure of  $\text{Li}_2\text{Ni}_2\text{W}_2\text{O}_9$ , through low-temperature neutron powder diffraction ( $T < T_N$ ). Although the magnetic moments are arranged in an antiferromagnetic configuration overall, the magnetic lattice is organized in three different subscales: in a (Ni,W) layer, ferromagnetic  $\text{NiO}_6$  chains are coupled antiferromagnetically along  $\mathbf{c}$  and ferromagnetically along  $\mathbf{b}$ . Meanwhile, the coupling between two subsequent (Ni,W) layer is antiferromagnetic. This peculiar magnetic structure fits in the  $Pb'c'n'$  magnetic space group, which by symmetry allows the linear magnetoelectric effect. Further experiments are required to fully explore all the magnetic properties of  $\text{Li}_2\text{Ni}_2\text{W}_2\text{O}_9$ , especially regarding possible metamagnetism and magnetoelectric effects.

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See Supplemental Material at [URL will be inserted by publisher] for the detailed results of the Rietveld refinements performed on the laboratory XRD and neutron powder diffraction data for  $\text{Li}_2\text{Ni}_2\text{W}_2\text{O}_9$  samples. Simulations of neutron powder diffraction patterns for different models of the magnetic structure, and the .mcif file for the final magnetic structure of  $\text{Li}_2\text{Ni}_2\text{W}_2\text{O}_9$  are also included.

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